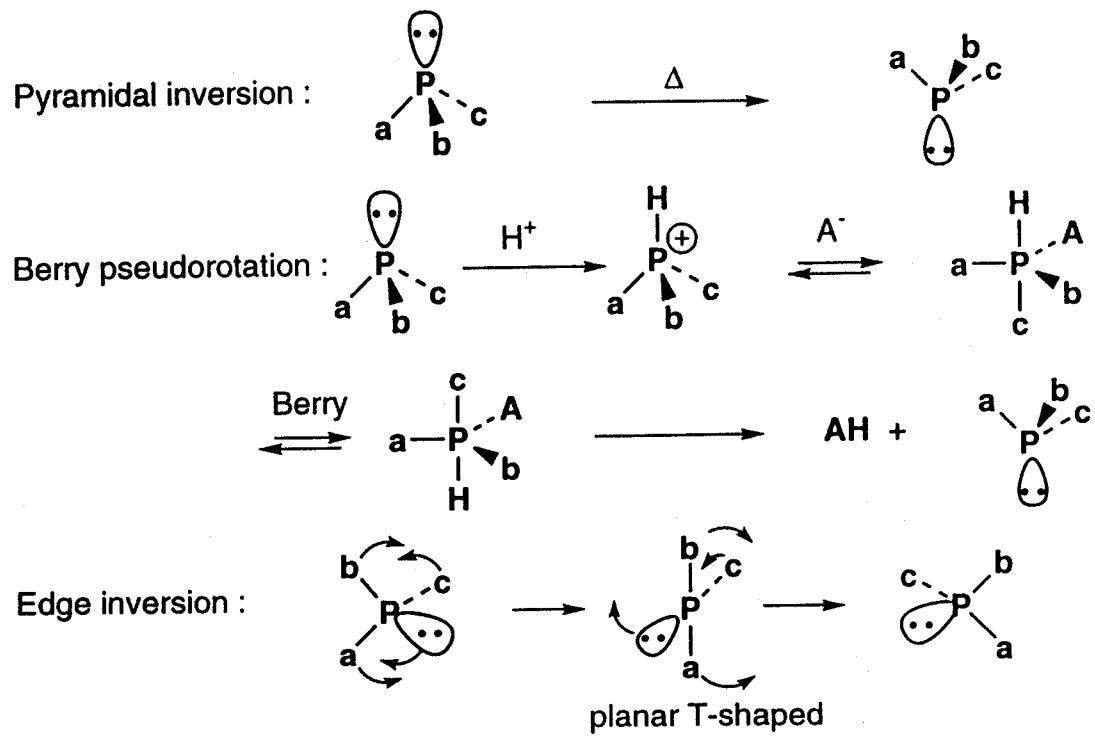


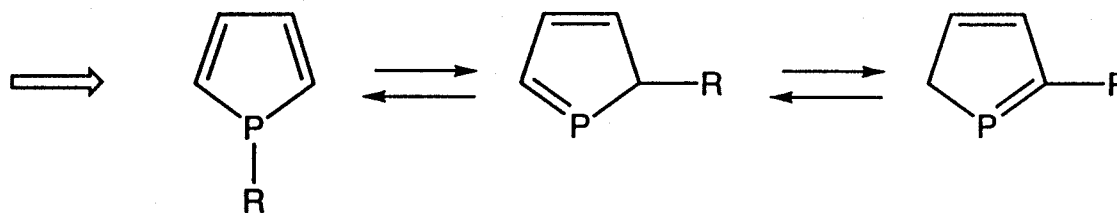
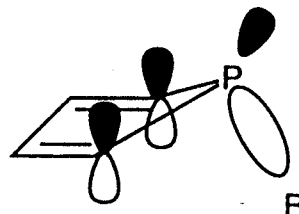
Some racemisation pathways for optically active phosphorus



Arduengo ..., JACS, 108, 2461 (1986)

The Phosphole - Cyclopentadiene Analogy

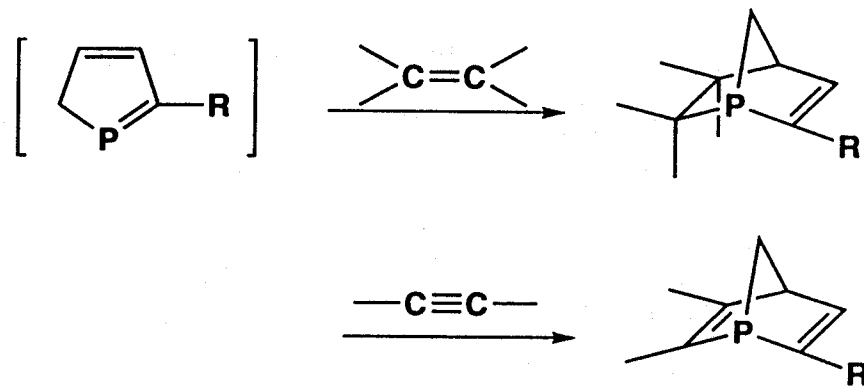
σ, π overlap in pyramidal phospholes:



R migrates rather easily: H \ll 0°C, Ar \approx 150°C, sp carbons (-C \equiv N, -C \equiv CPh) \approx 100°C, also -SR, -PR₂, -SiR₃. No competition with 3*H*-phospholes. They are much higher in energy [Bachrach, JOC, 58, 5414 (1993)].

C. Charrier, H. Bonnard, G. de Lauzon, F. Mathey, JACS, 1983;
F. Laporte, F. Mercier, L. Ricard, F. Mathey, BSCF, 1993

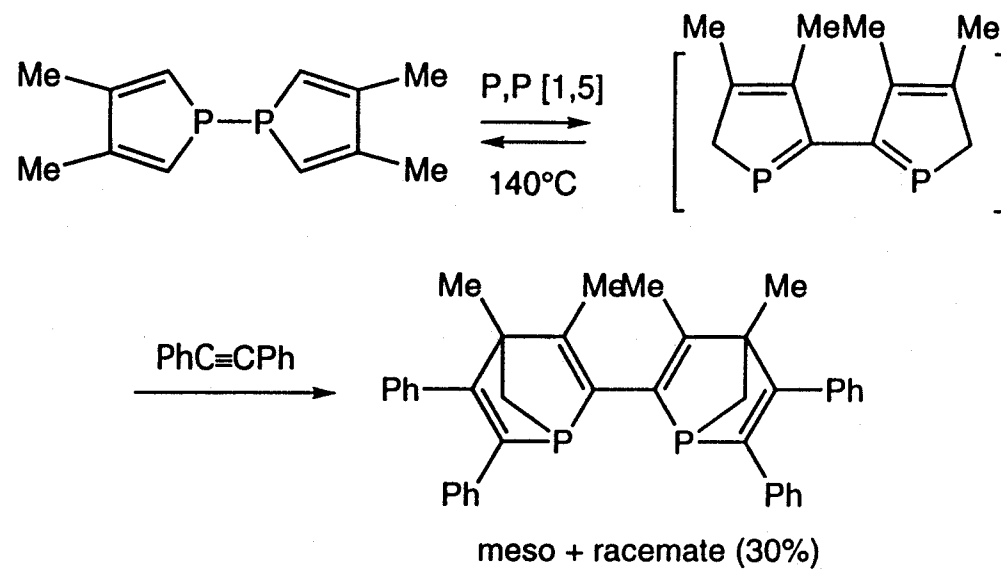
2-H-phospholes as 1-phosphadienes



LeGoff, Mathey, Ricard, *J. Org. Chem.* (1989)

Mathey, Mercier, Charrier, Fischer, Mitschler, *J. Am. Chem. Soc.* (1981)

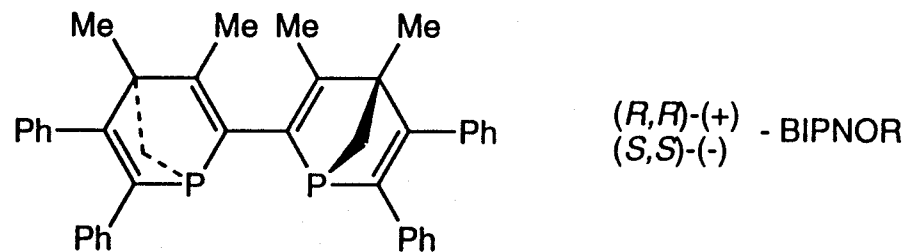
Synthesis of a bis-phosphanorbornadiene



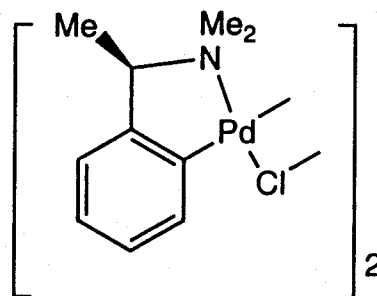
The two diastereomers can be separated as their PdCl_2 complexes

M.-O. Bévierre, F. Mercier, L. Ricard, F. Mathey, BSCF, 1992

The Performances of BIPNOR in Asymmetric Catalysis



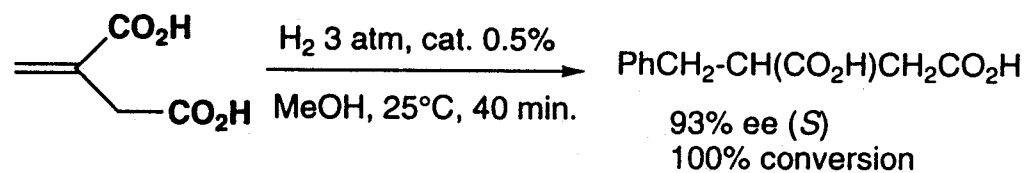
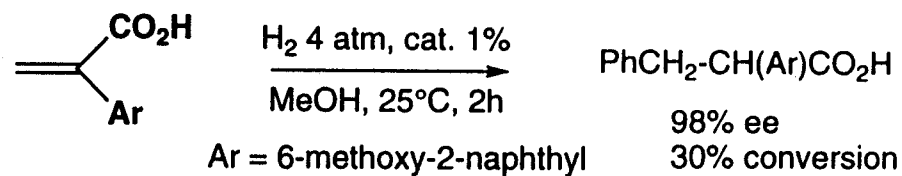
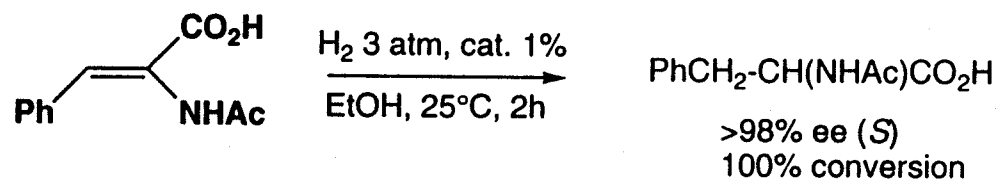
resolution by chromatography of the two diastereomeric Pd-BIPNOR-Pd dinuclear complexes obtained by reaction with the (*R*)-homochiral Pd₂ complex



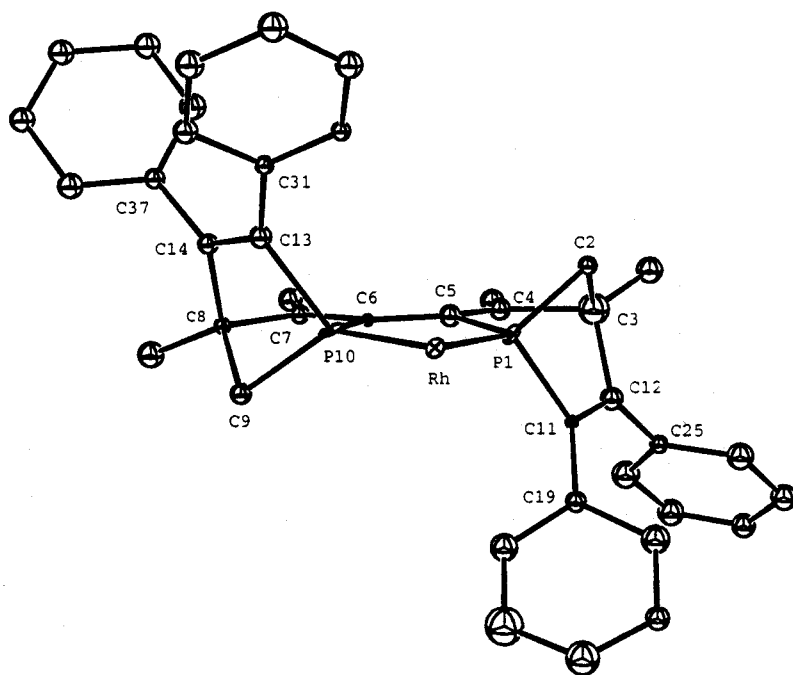
preliminary experiments indicate ee's in the same range as those obtained with BINAP in the Rh-catalyzed asymmetric hydrogenation of dehydroaminoacids

F. Robin, F. Mercier, L. Ricard, F. Mathey, M. Spagnol

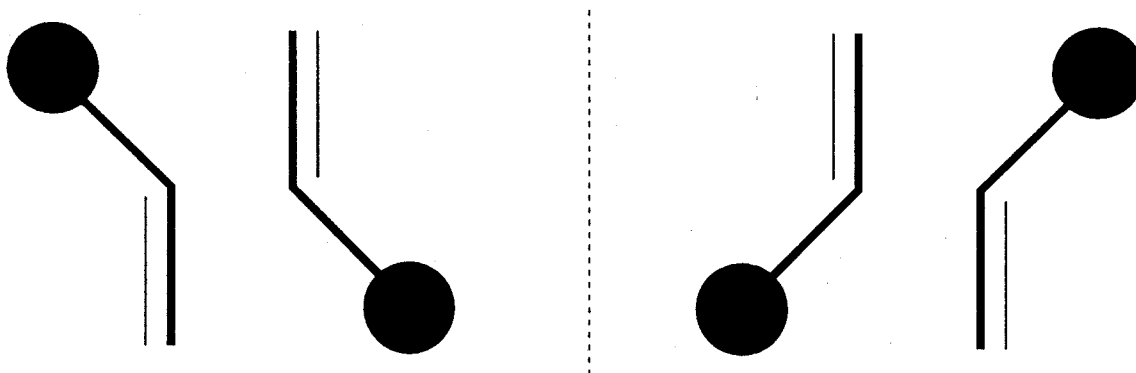
Asymmetric hydrogenation of functional alkenes



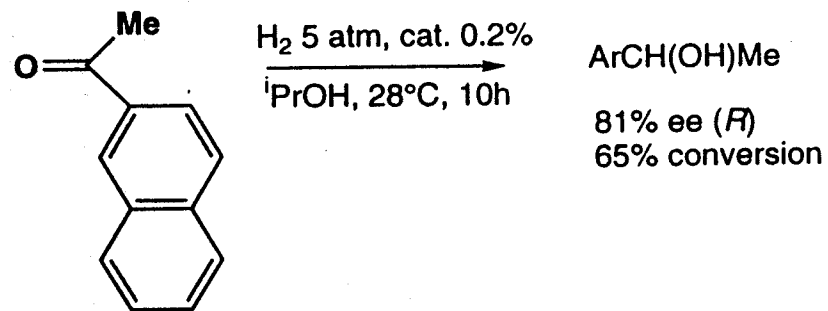
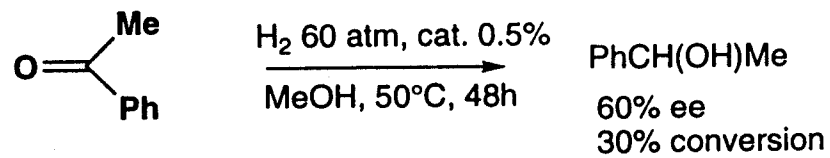
catalyst: [Rh(cod)(-)-BIPNOR]⁺PF₆⁻ for cases 1 and 2
 [Rh(cod)(+)-BIPNOR]⁺PF₆⁻ for case 3



View of the [Rh(-)-BIPNOR] moiety of [Rh(cod)(-)-BIPNOR]⁺PF₆⁻ from the PRhP plane

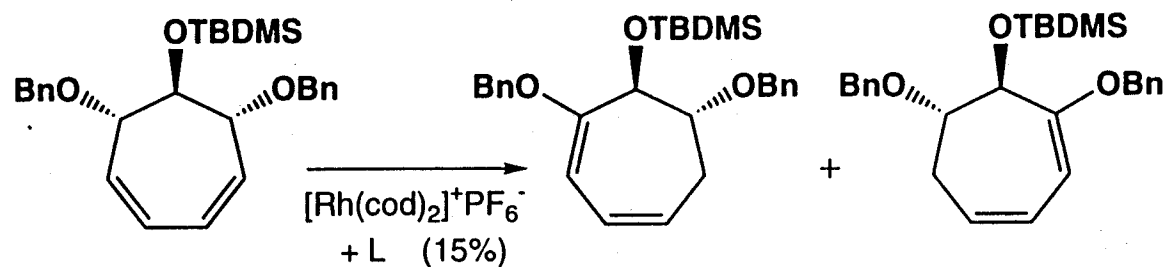


Asymmetric hydrogenation of ketones



catalyst: [RuBr₂(+)-BIPNOR] for case 1
[RuBr₂(+)-BIPNOR] + KOH + (*S,S*)-1,2-diphenyl-1,2-ethanediamine for case 2

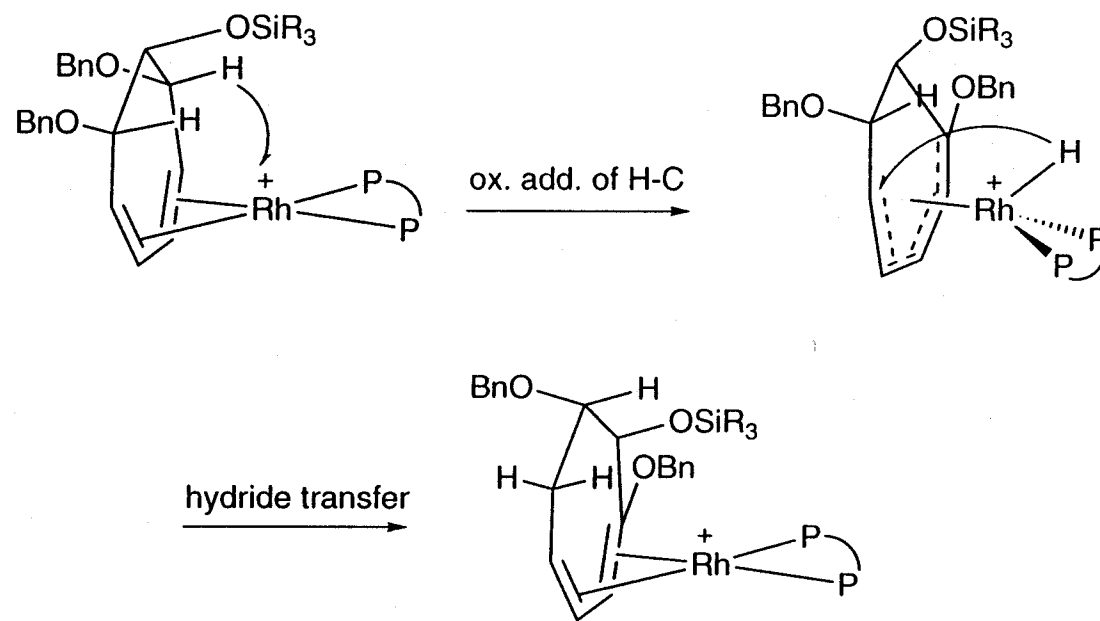
Asymmetric isomerisation of a cyclic diene



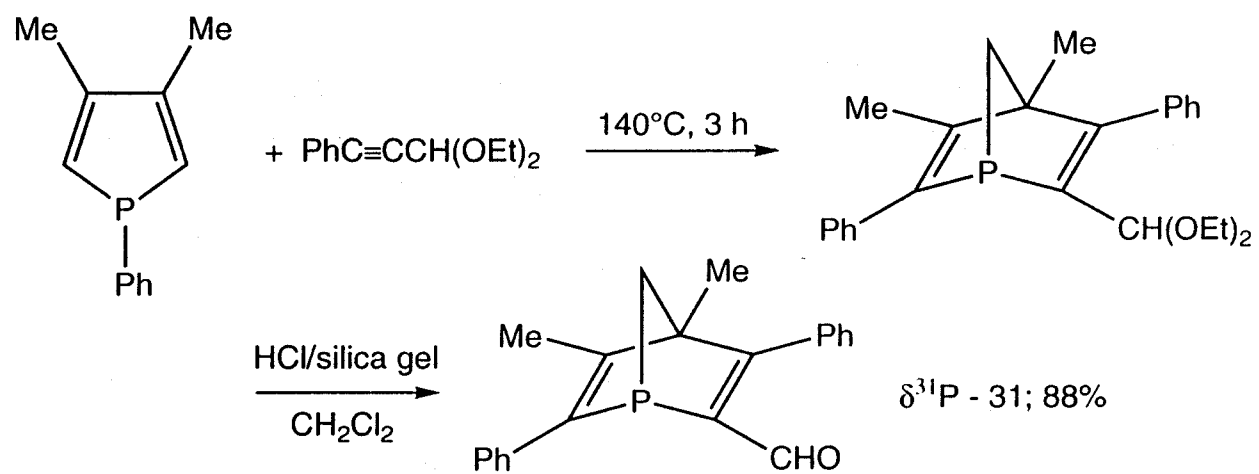
L	Solvents	T°C	t (100% conv.)	ee
(+)-BIPNOR	THF/tol (1:1)	50°C	32 h	55
(+)-BINAP	"	"	24h	31
(+)-BIPNOR	DME/tol (1:3)	90°C	5min.	91
(+)-BINAP	"	"	15min.	11

Th. Faitg, J. Soulié, J.-Y. Lallemand, F. Robin, F. Mercier, F. Mathey, to be published.

Proposed mechanism for the isomerisation of the cyclic diene

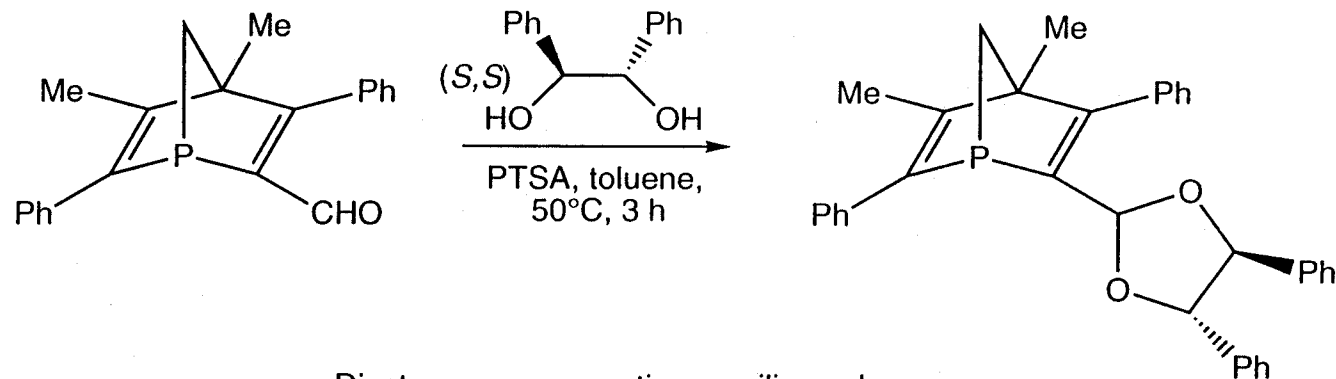


Synthesis of a 1-Phosphanorbornadiene-2-carboxaldehyde



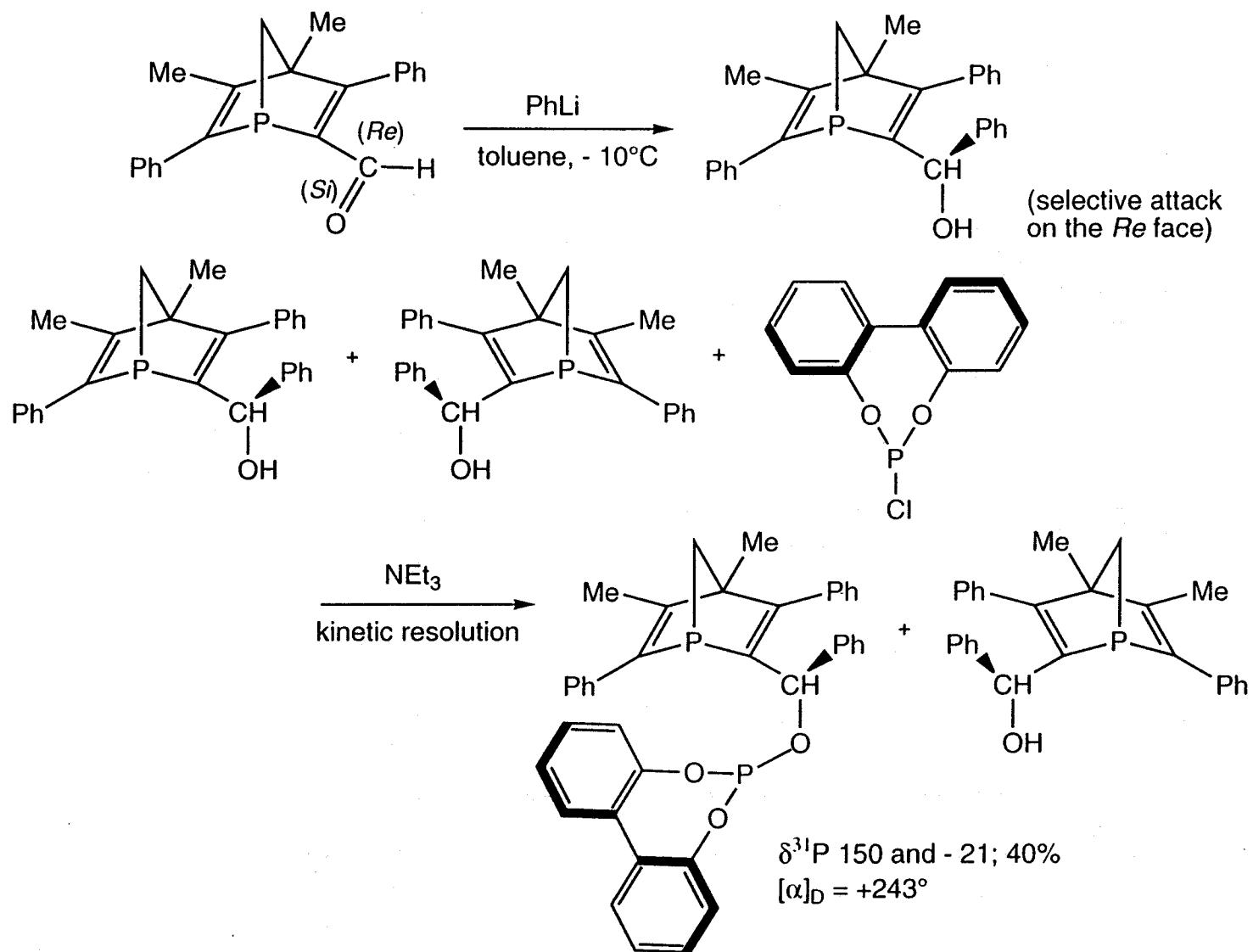
S. Lelièvre, F. Mercier, F. Mathey (1998)

Resolution of the Phosphanorbornadiene-1-carboxaldehyde

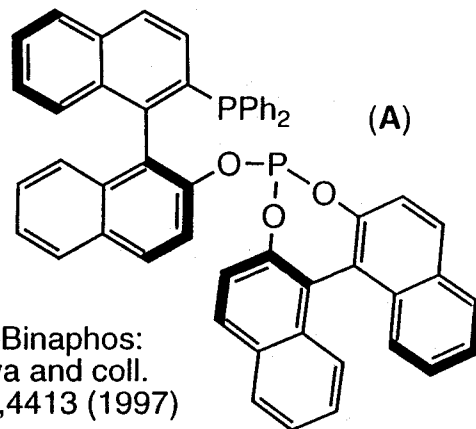


Diastereomer separation on silica gel
with hexane/dichloromethane 7/3
 $[\alpha]_D = +41.6$ and $+200^\circ$ ($c=1$, CH_2Cl_2)
Deprotection with HCl/silica gel/acetone
 $[\alpha]_D = +134.5$ and -133 ($c=0.55$, CDCl_3)

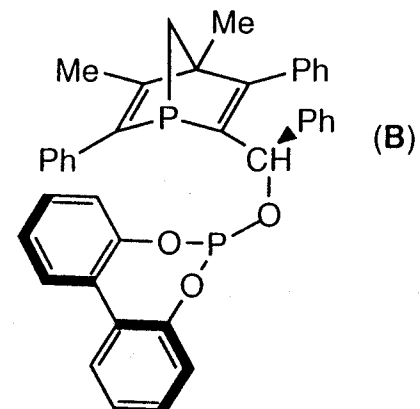
Synthesis of enantiopure Phosphanorbornadiene-phosphites



Hydroformylation of Styrene

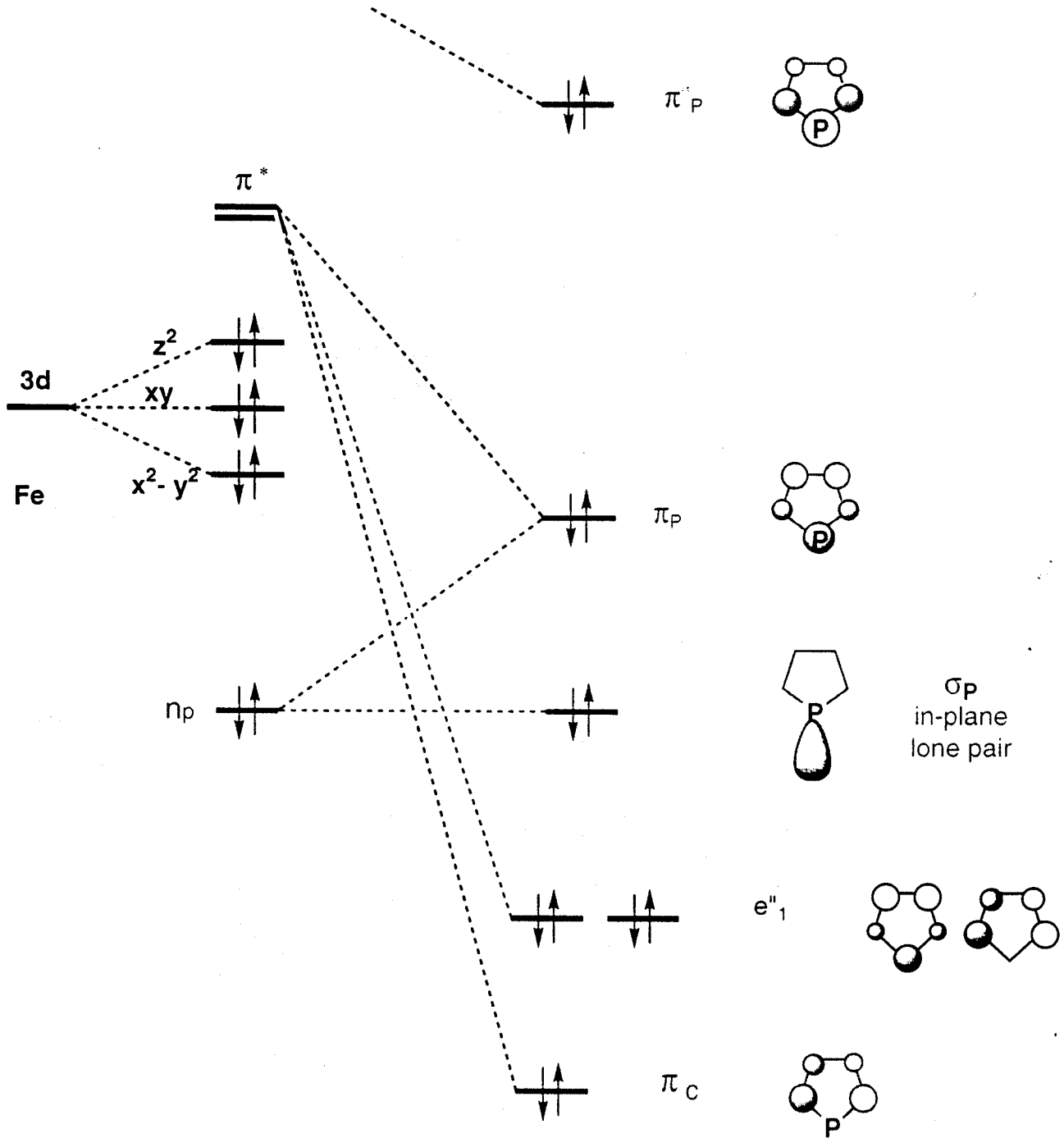


(*S,R*)-Binaphos:
Takaya and coll.
JACS, 4413 (1997)



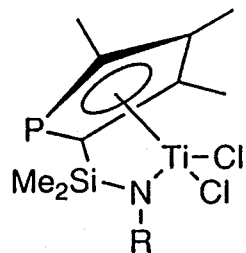
phosphine	solvent	P atm	T°C	Rh/P	b/l	ee	TOF h ⁻¹
(A)	(C ₆ H ₆)	100	50	1/8	88/12	94	46
(B)	toluene	100	50	1/2	92/8	0	91
(B)	MeOH	150	40	1/8	94/6	21	14

Electronic structure of phosphoferrocene

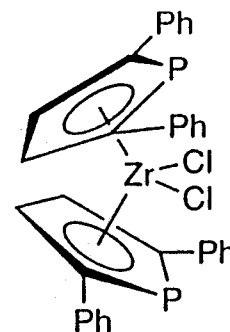


Phosphametalloenes in catalysis

Olefin polymerization

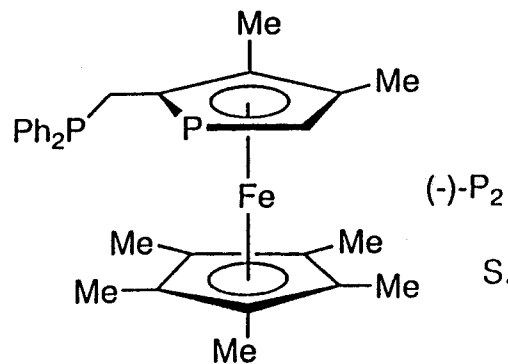


S.J Brown et al., *Organometallics*, 17, 5445 (1998)



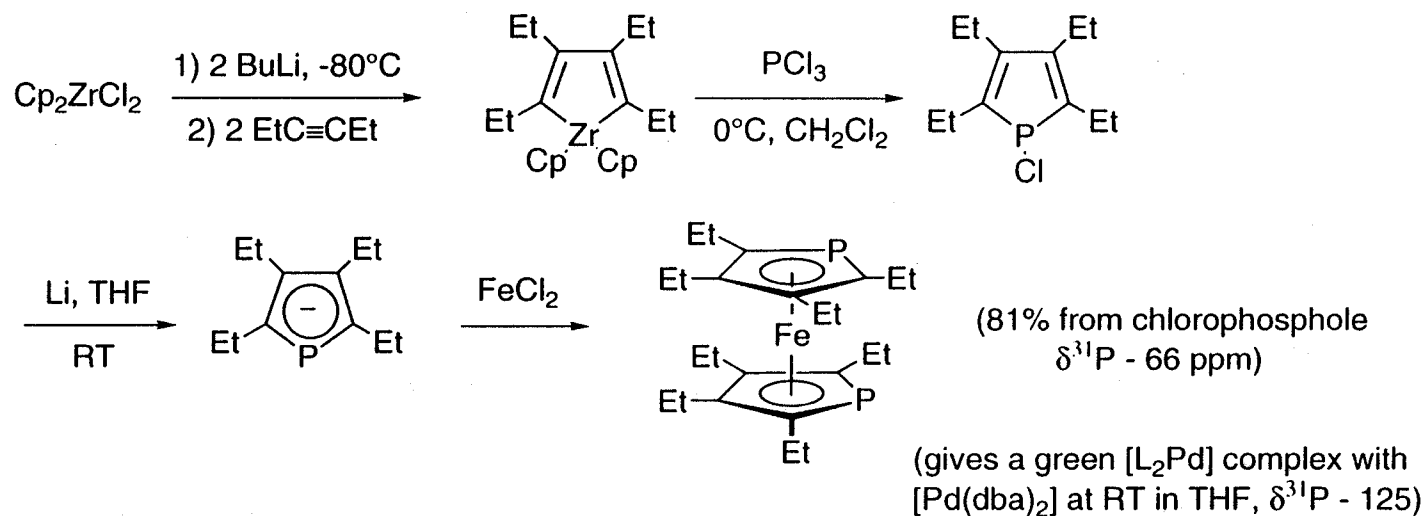
E.J.M. de Boer et al., *J. Mol. Cat. A*, 128, 155 (1998)

Asymmetric hydrogenation



S. Qiao, G.C. Fu, *J. Org. Chem.*, 63, 4168 (1998)

Synthesis of Octaethyl-1,1'-Diphosphaferrocene



X. Sava, L. Ricard, P. Le Floch, F. Mathey (1999)

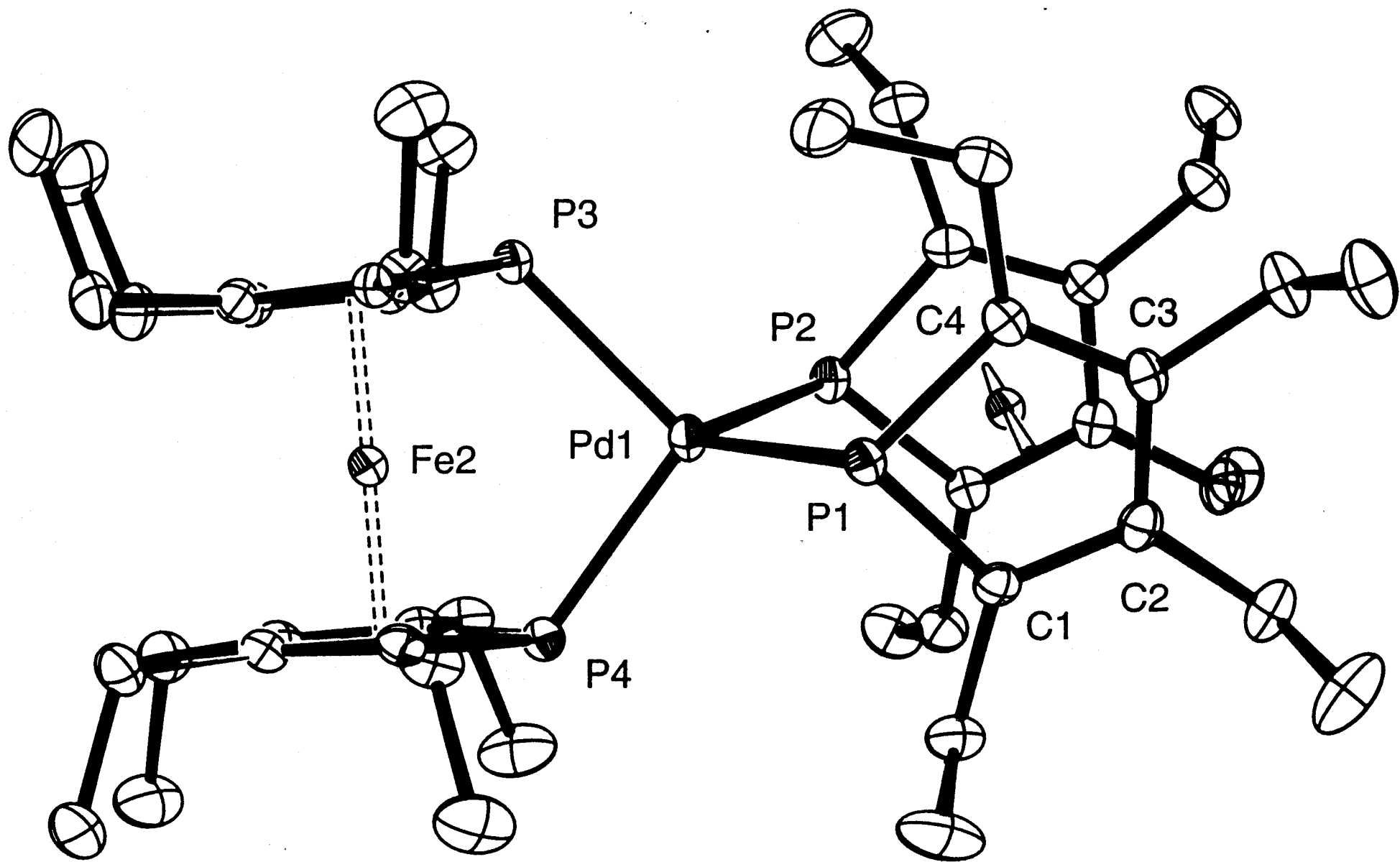
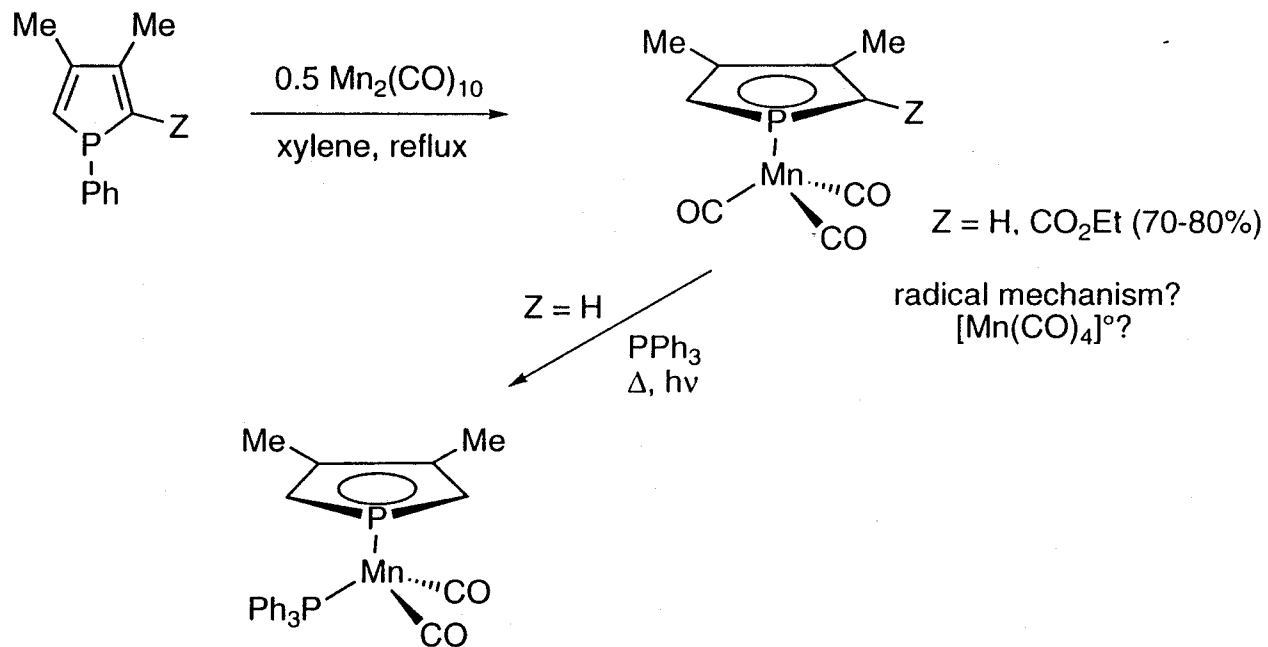


Table Suzuki Coupling of Aryl Bromides

Aryl bromide	mol%Pd	t/h	Conversion(%)	TON
4-bromoacetophenone	0.0001	1	77	770000
4-bromoacetophenone	0.0001	20	98	980000
3-bromothiophene	0.005	1	80	16000
3-bromothiophene	0.005	20	96	19200
2-bromoanisole	0.005	1	67	13400
2-bromoanisole	0.005	20	96	19200
bromobenzene	0.0001	1	46	460000

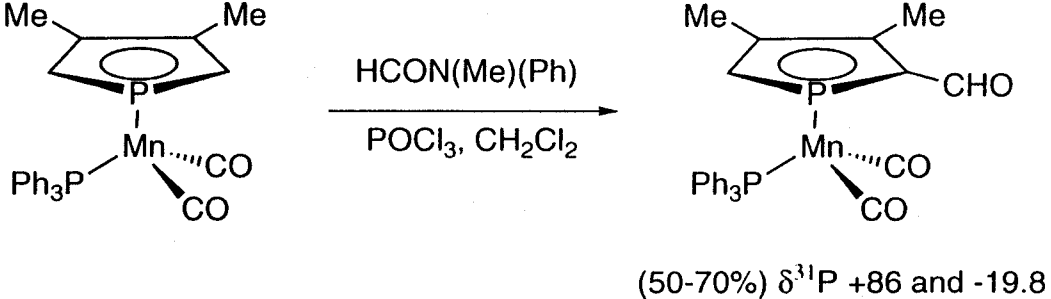
Reaction conditions : 1.0 equiv of aryl bromide, 1.5 equiv of phenylboronic acid, 2.0 equiv. K_2CO_3 . Temperature : 110°C, TON are expressed in mol product(mol Pd)⁻¹.

Synthesis of phosphacymantrenes



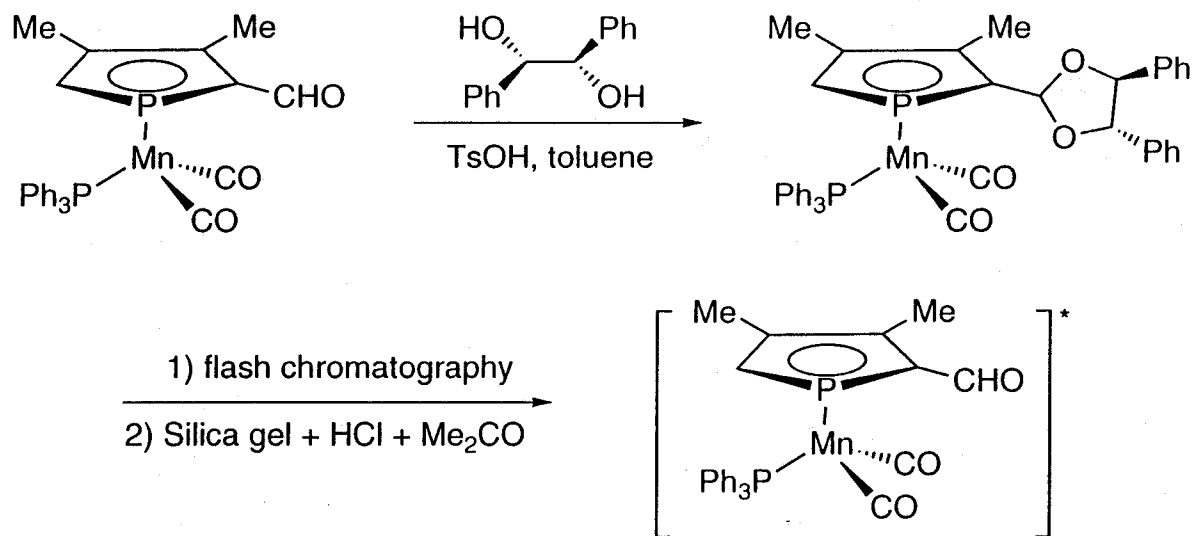
F. Mathey, *Tetrahedron Lett.* (1976)
B. Deschamps, L. Ricard, F. Mathey, *Organometallics* (1999)

Synthesis of a phosphacymantrene-2-carboxaldehyde

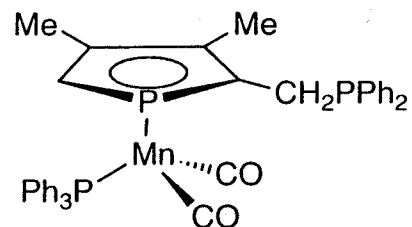
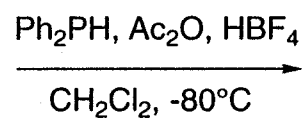
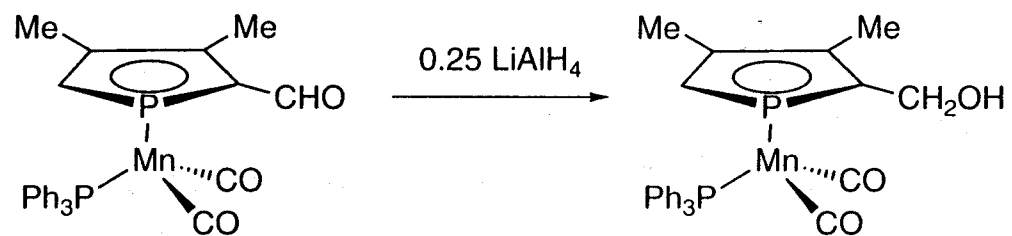


B. Deschamps, L. Ricard, F. Mathey (1999)

**Separation of the enantiomers of a phosphacymantrene
2-carboxaldehyde**



Synthesis of a 2-(phosphinomethyl)-phosphacymantrene



cyclic P: $\delta^{31}\text{P}$ -40 ppm (AB)
 PPh₂: $\delta^{31}\text{P}$ -9 ppm (AB, $^3J_{\text{AB}} = 25$ Hz)
 PPh₃: $\delta^{31}\text{P}$ +89 ppm