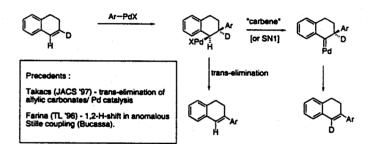
Summary Slides

TA500 2000

John M. Brown

Mechanisms for 2-arylation in Heck reactions of dihydronaphthalene



Hydroboration / amination : (Mark Hooper; Elena Fernandez)

(i) catecholborane, 0.2-1% catalyst, 20°C, thf, 1 h.; (ii) 2 MeMgCl (3M in thf), 20°C, 30 min; (iii) 3 H₂NOSO₃H, thf, 10 h; (iv) CODRhacac, Me₃SiOSO₂CF₃, thf then pentane

Route to secondary amines: Kenji Maeda

- (i) C₆H₁₁NHOSO₃H, diglyme, 24 h, 20°C;
- (ii) PhCH₂NHCl (ex. PhCH₂NH₂, aq. NaOCl in situ), 0°C, 5 min then 20°C, 1 h;
- (iii) Et₂NHCl (ex. Et₂NH, aq. NaOCl),*.

Amination of p-methoxystyrene

Reactant	Product	Yield and E.e.
	HN C ₆ H ₁₁	
H_2N	MeO	82% yield, 92% e.e.
	HN, CH ₂ C ₆ H ₅	
H ₂ N	MeO	75% yield, 91% e.e.
H ₂ N OMe N	Me O Me	77% yield, 90% e.e.
H ₂ N O	MeO HN O Me	76% yield, 91% e.e.

Amination with cyclohexylamine

Reactant	Product	Yield and E.e.
	н _ү - ^{С₆Н₁₁}	
	Me HŅ- ^C 6H ₁₁	71% yield, 87% e.e.
MeO	MeO HŅ C ₆ H ₁₁	82% yield, 92% e.e.
CI	CI Me	50% yield, 78% e.e.
		47% yield, 87% e.e.

"expected"

"anomalous"

$$Ar = Ph$$
 $X = I$ Condition A 3:2

$$Ar = 3,4-Cl_2Ph X = I$$
 Condition A 1:1

$$Ar = 4$$
-FPh $X = Br$ Condition B 1: 1

$$Ar = 4FPh$$
 $X = Br$ Condition C 1:20

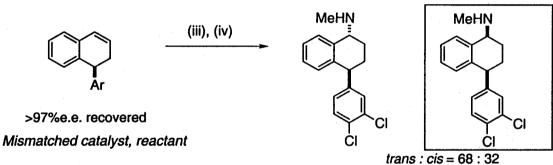
Pd PAr₂
AcO OAc
Ar₂P

1, Ar = o-tol

Conditions: **A,** 1 mol% Pd(OAc)₂, P(o-tol)₃, Et₃N, CH₃CN, reflux; **B**, 1 mol% complex **1**, NaOAc, DMA, 140°C; **C**, 5 mol% Pd(OAc)₂, Bu₄NOAc, DMF, 80°C

Step 1

Step 2



[low yielding]

Conditions : (i) Catecholborane (0.6 eq.), S-QuinapRh catalyst, C_7H_8 , 2 h., (ii) H_2O_2 , NaOH, H_2O ; (iii) Catecholborane (1.2 eq.), S-QuinapRh catalyst, C_7H_8 , 48 h.; (iv) Et_2Zn , C_7H_8 , then MeNHCl (in situ), Et_2O .

Ligands preparedby resolution procedures

$$OH$$
 \longrightarrow HO

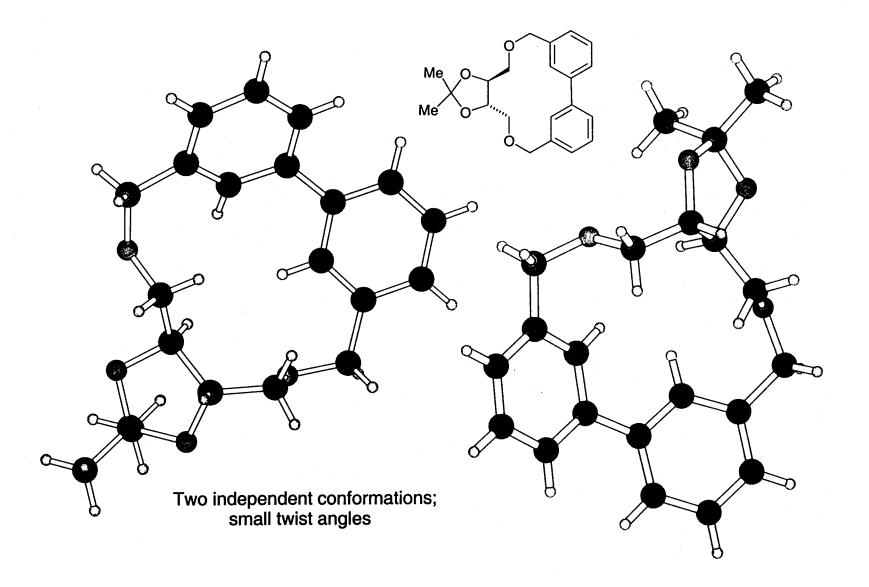
Ligand needs to be resolved; racemisation very slow at >100°C

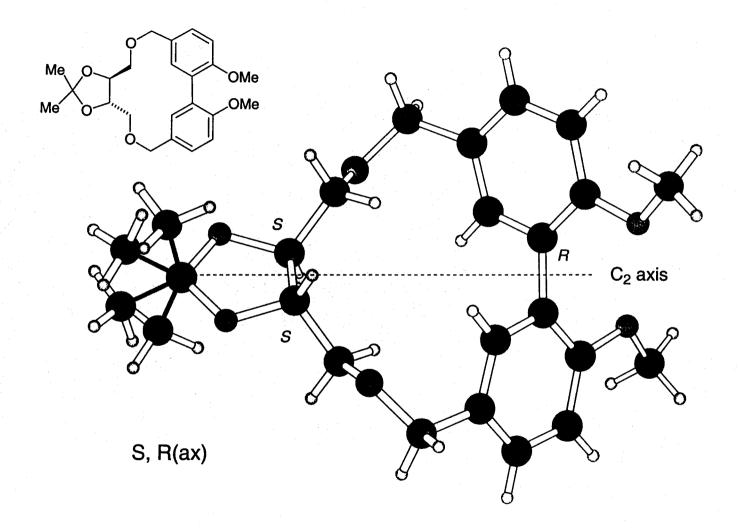
Diastereoselective biaryl synthesis [Lipshutz, TL. 1998]

Fast interconversion; X perturbs equilibrium

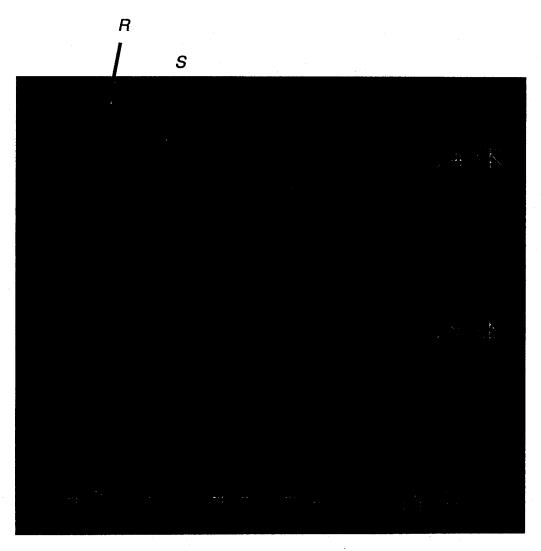
Successful cyclisation method

P. Caubere, 1989





Threitol cyclophane



S

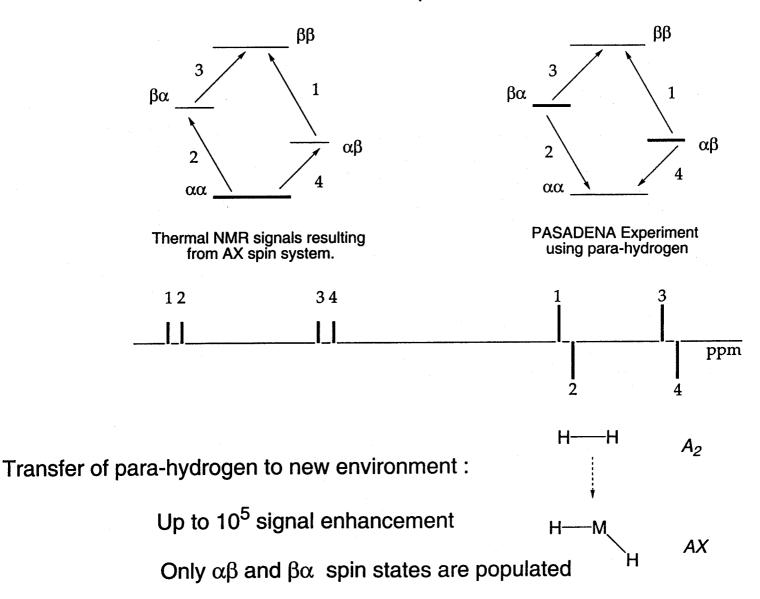
Biphenol Synthesis in the Mannitol Series

Biphenols show sharp CH₂ NMR at ambient temperature and below

Likewise Threitol, Iditol series

[Bargon, Giernoth U. Bonn]

PHIP Experiment



Summary of observations on the PHIP transient

The new species is associated with 1-H chemical shifts of ca. -2 and -19 ppm [-2 ppm is unusual - agostic?]

Formation is observed in the temperature range of -10°C to -25°C - occurs only when substrate remains

Concurrent with the observation of transient, polarised hydrogenation product is also seen.

Both rhodium norbornadiene and cyclooctadiene complexes can be employed.

The transient is associated with bound substrate, since different substrates give distinct spectra

Any exchange of substrate with unbound species is slow (half-life >300 ms) on the NMR time-scale

Conclusions on the PHIP-observed species

The complex is a rhodium enamide dihydride with one agostic hydrogen in flight to the beta-carbon; the second is trans- to a coordinated (enamide) oxygen

For dehydroalanine, the hydride exchanges rapidly with the beta-methylene group

The transient behaves as an intermediate in hydrogenation, being closely associated with the observation of spin-polarised product (INEPT)

Polarisation transfer to the reactant observed by INEPT requires some substrate dissociation from the transient - but does not reveal the mechanism by which this occurs.

Dynamic agostic methyl group for the dehydroalanine case

Rotation about the $C\alpha$ - $C\beta$ bond is rapid on the NMR timescale