

Tutorial Session – Frontiers in Organic Chemistry

Enhancing the Potential of Organocatalysis with Light

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ICREA Research Professor

Institute of Chemical Research of Catalonia (ICIQ)
Barcelona Institute of Science & Technology

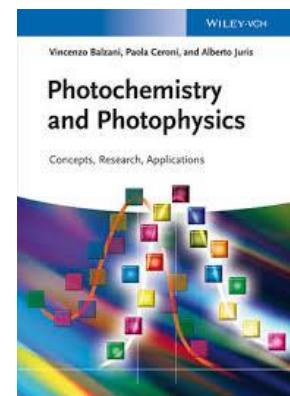
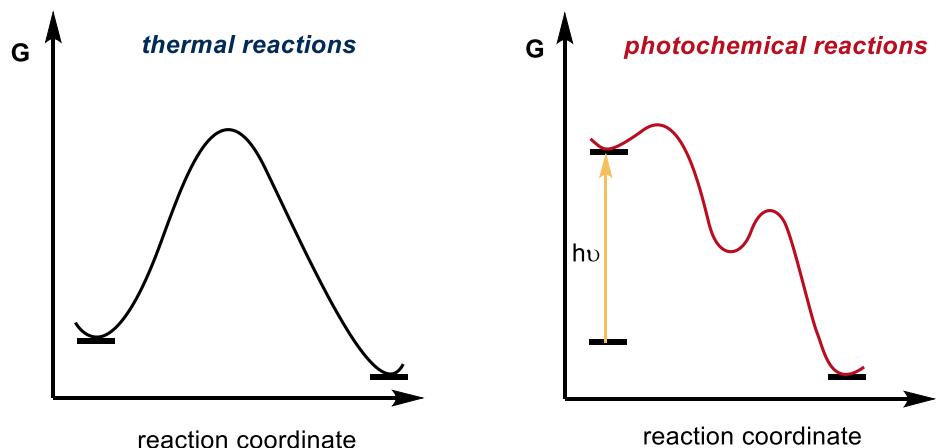
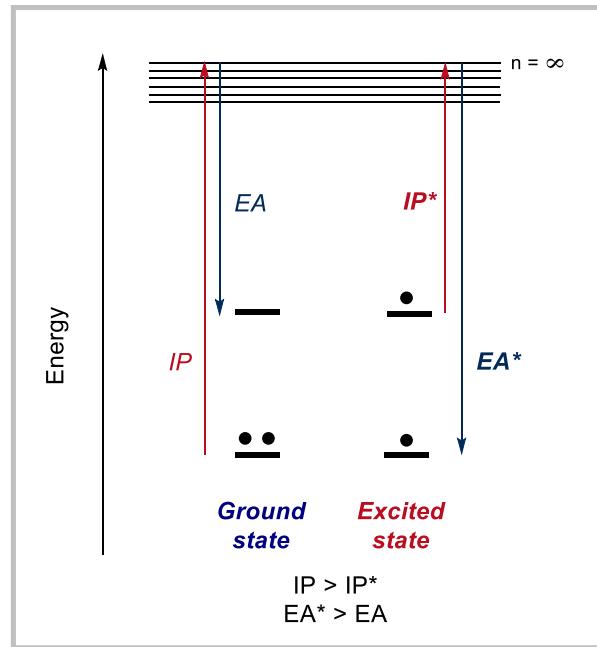
Tarragona – SPAIN
pmelchiorre@iciq.es

 Ischia Advanced School of Organic Chemistry | IASOC
21 Sept 2018

Photochemistry and Excited-State Reactivity

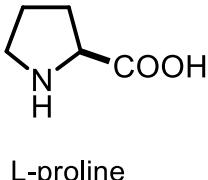
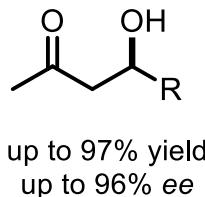
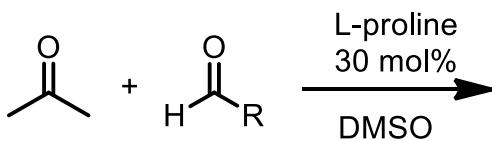
a molecule in the **excited state** is both a better **reductant** and a better **oxidant** than in the ground state

excited-state reactivity unlocks unconventional reaction pathways

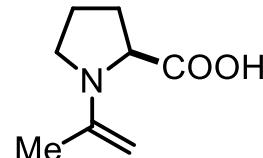


Activation Modes in Aminocatalysis

- Proline-catalyzed intermolecular aldol reaction



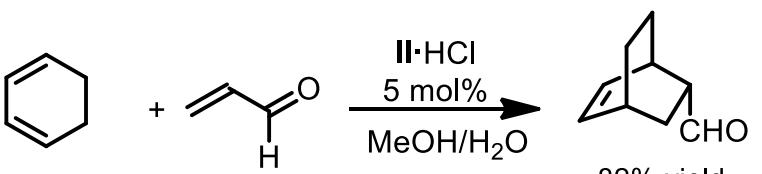
Enamine Catalysis



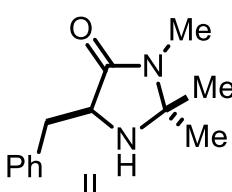
HOMO-Raising Activation

B. List, R. A. Lerner, C. F. Barbas III, *J. Am. Chem. Soc.* **2000**, 122, 2395-2396

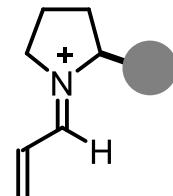
- Iminium ion catalyzed asymmetric Diels-Alder of enals



82% yield
94:6 endo:exo
94% ee



Iminium-Ion Catalysis

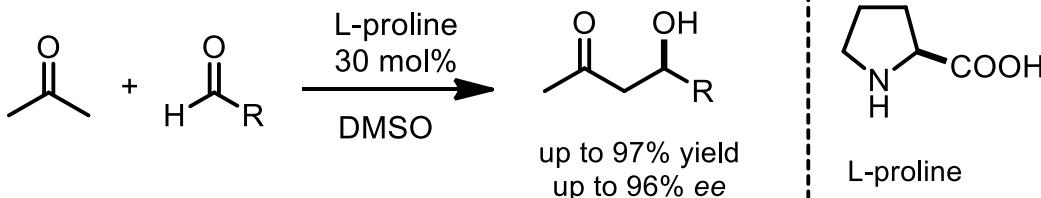


LUMO-Lowering Activation

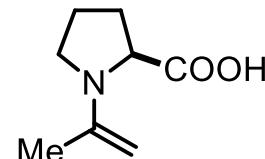
K. A. Ahrendt, C. J. Borths, D. W. C. MacMillan, *J. Am. Chem. Soc.* **2000**, 122, 4243-4244

Enamine in the Ground State

- Proline-catalyzed intermolecular aldol reaction



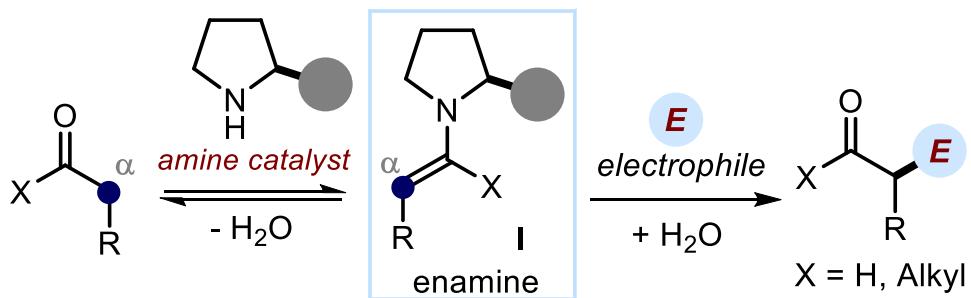
Enamine Catalysis



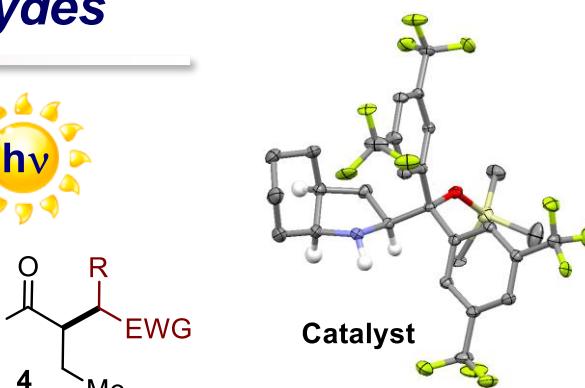
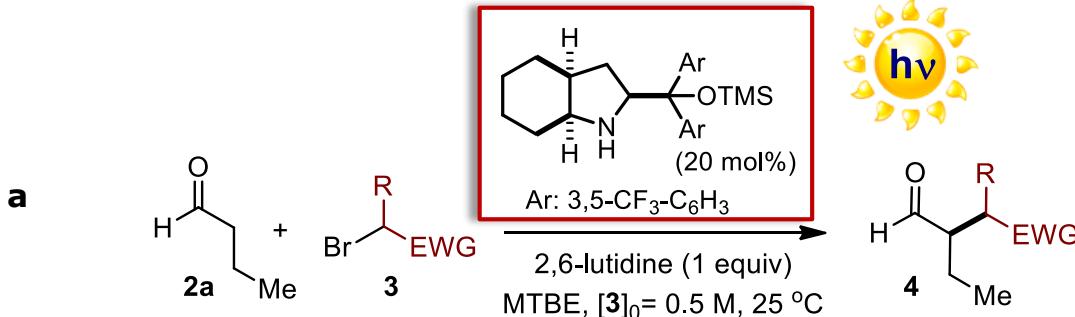
HOMO-Raising
Activation

B. List, R. A. Lerner, C. F. Barbas III, *J. Am. Chem. Soc.* **2000**, 122, 2395-2396

enamine-mediated catalysis



Photochemical Asymmetric Alkylation of Aldehydes

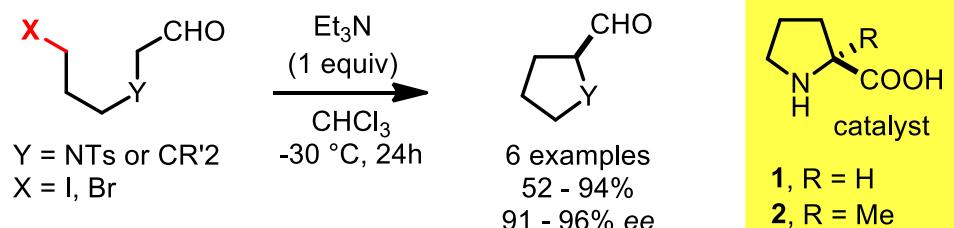


entry	alkyl-bromide	product	entry	alkyl-bromide	product	entry	alkyl-bromide	product
1		95% yield, 84% ee	5		86% yield, 86% ee	9		65% yield, 84% ee
2		95% yield, 93% ee	6		70% yield, 83% ee	10*†		98% yield, 91% ee
3		70% yield, 86% ee	7		96% yield, 87% ee	11*		75% yield, 85% ee
4		94% yield, 86% ee	8		92% yield, 87% ee	12*		77% yield, 90% ee

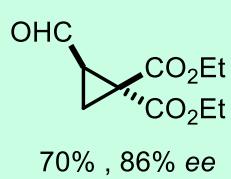
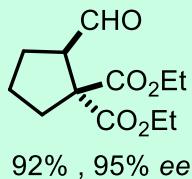
For a pertinent precedent, see: D. A. Nicewicz, D. W. C. MacMillan, *Science* **2008**, *322*, 77

with E. Arceo, I. Jurberg, A. Alvarez
Nature Chemistry, **2013**, *5*, 750-756

α -Alkylation of Aldehydes



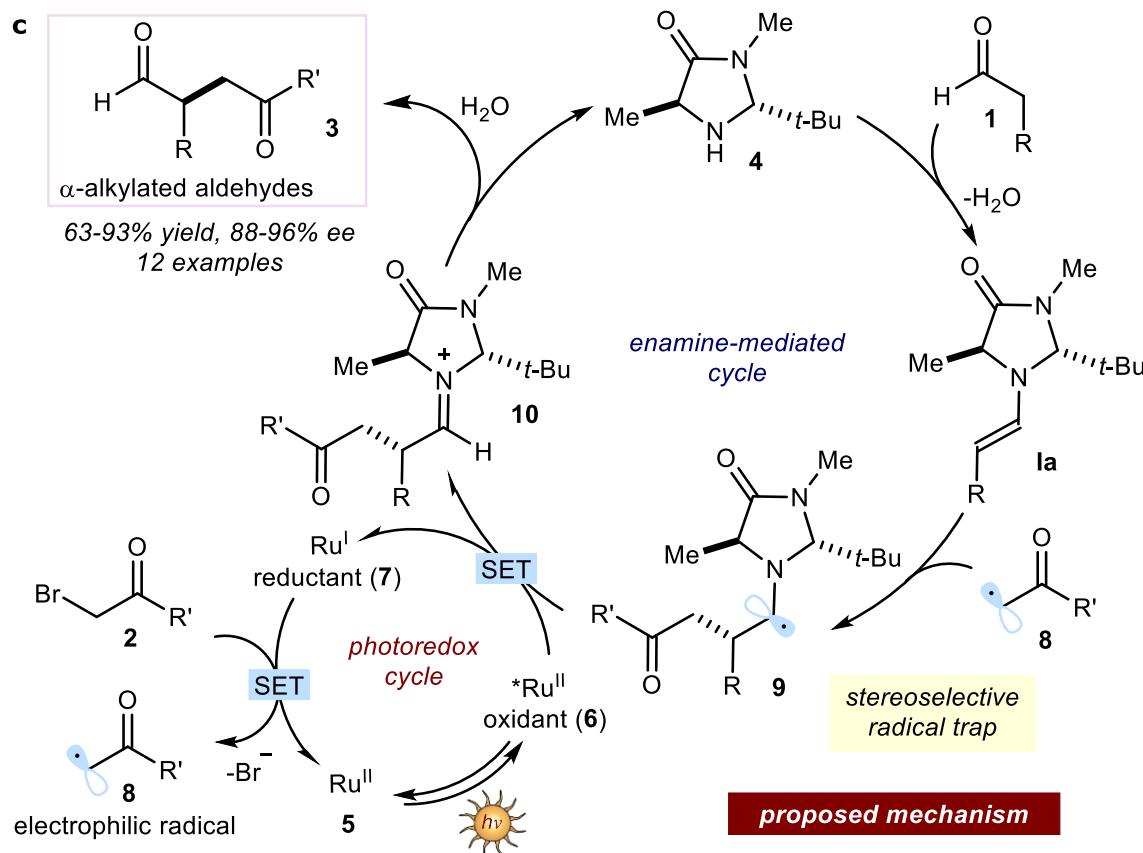
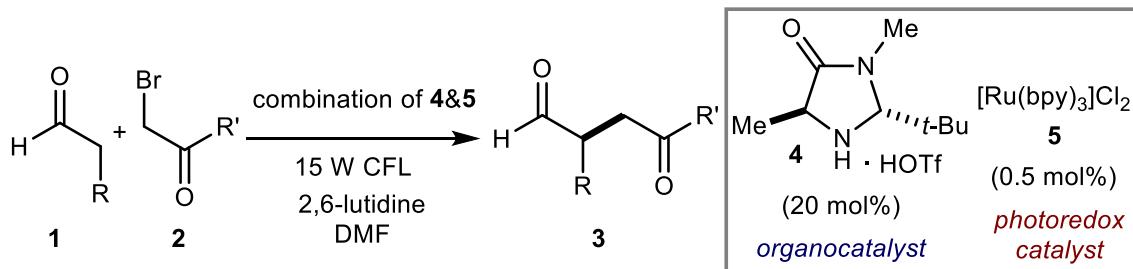
Selected Products:



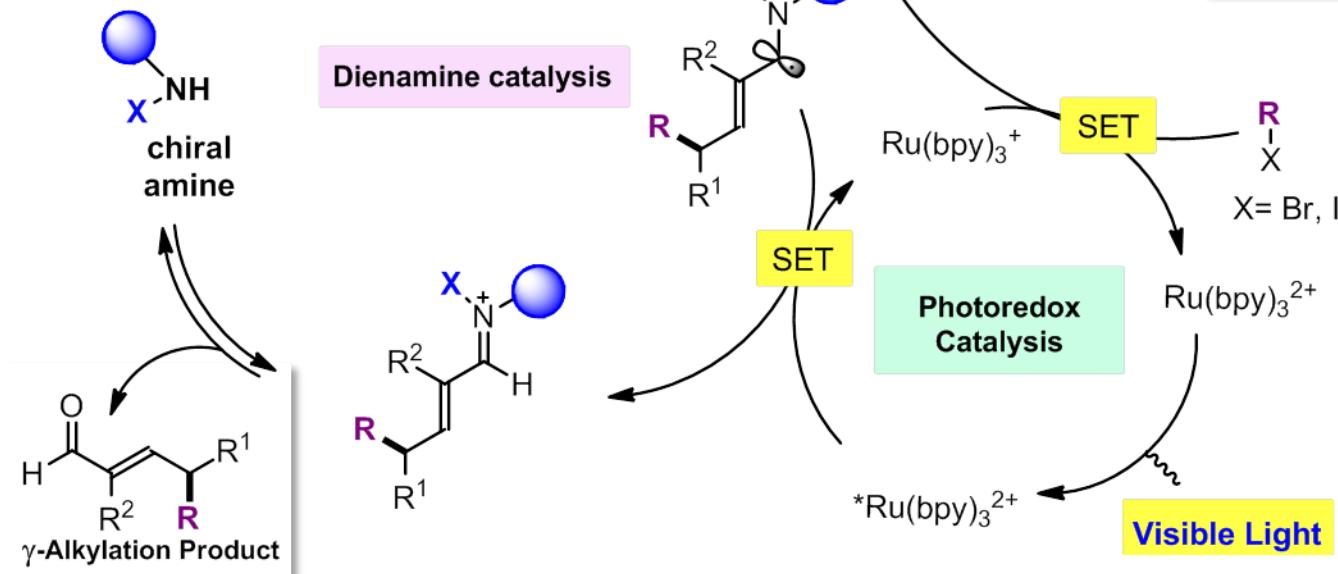
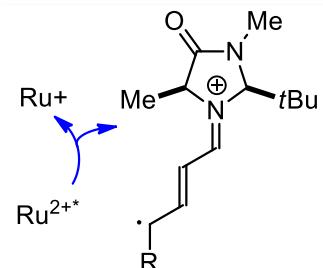
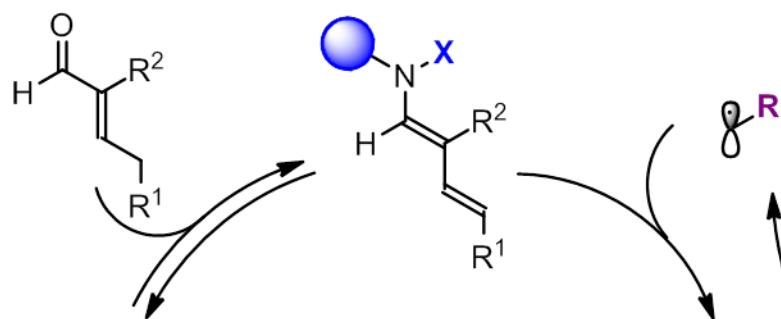
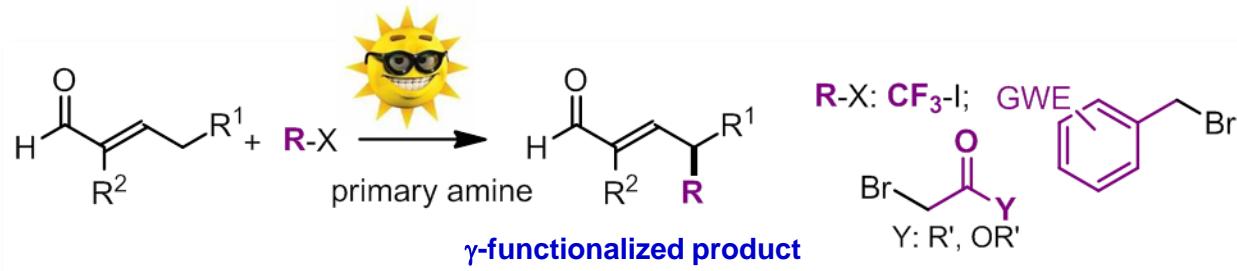
Intramolecular S_N2 type Reaction

N. Vignola, B. List, J. Am. Chem. Soc. 2004, 126, 450–451

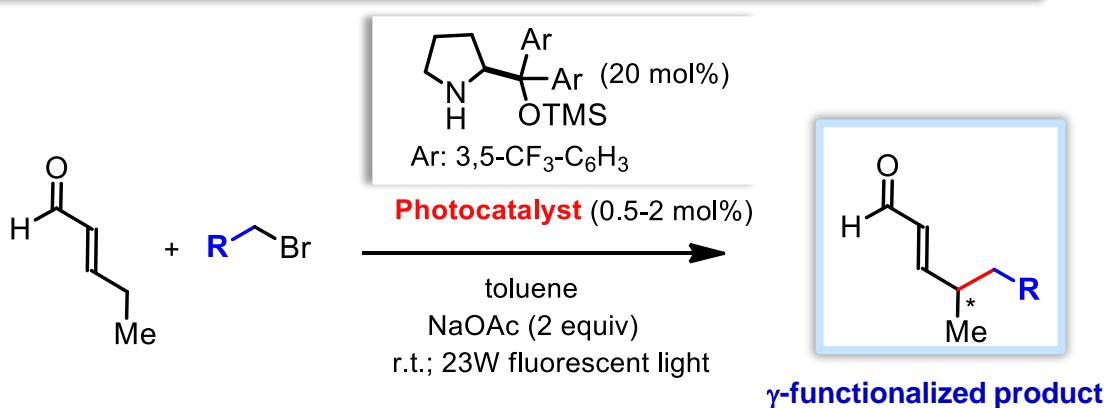
The Solution: Merging Photoredox- and Enamine-catalysis



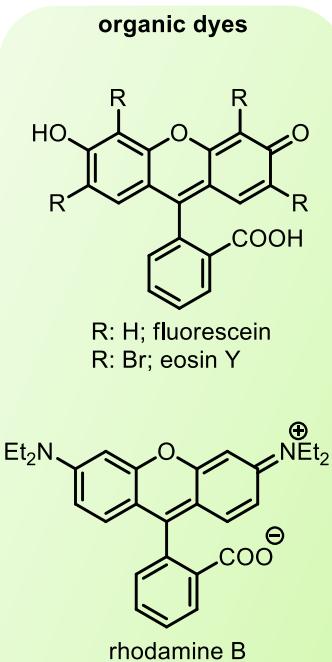
Merging Photoredox- and Amino-Catalysis



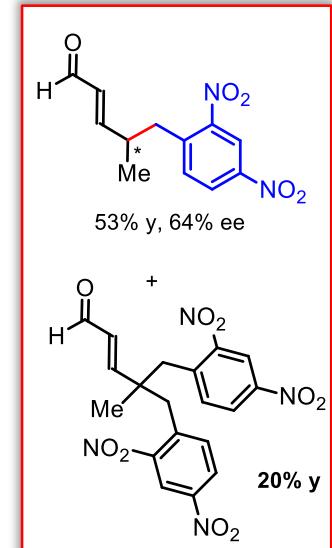
Vinylogous Reactivity in Radical Pathways



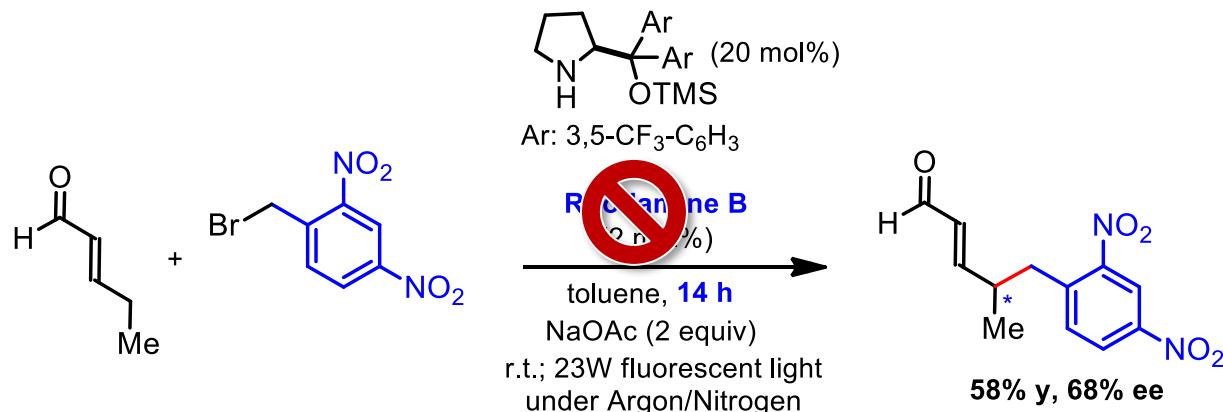
Photocatalyst	time	conv. R'Br	ee
	Fluorescein	48 h	100 50 %
	[Ru(bpy) ₃]Cl ₂	45 h	88 64 %
	[Ru(bpy) ₃]Cl ₂	40 h	75 66 %
	[Ru(bpy) ₃]Cl ₂	14 h	100 64 %



K. Zeitler *et al.*,
 ACIE 2011, 50, 951



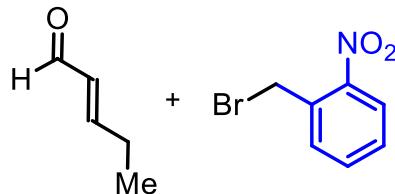
Vinylogous Reactivity in Radical Pathways

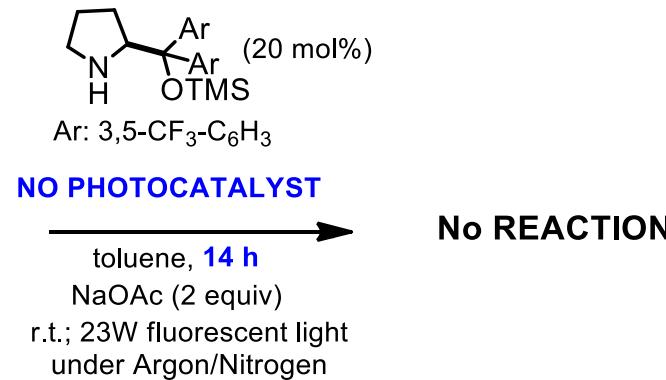


Elena Arceo

Control Experiment

The reaction works without any Photo-redox catalyst!

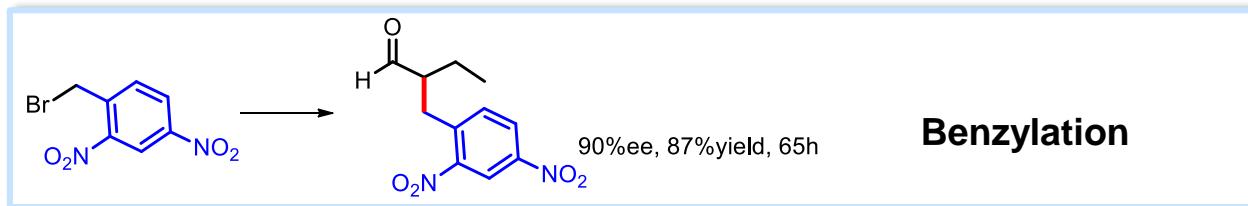
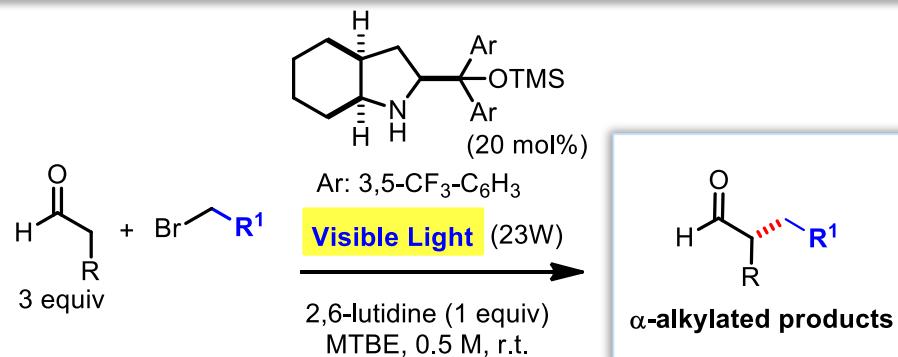




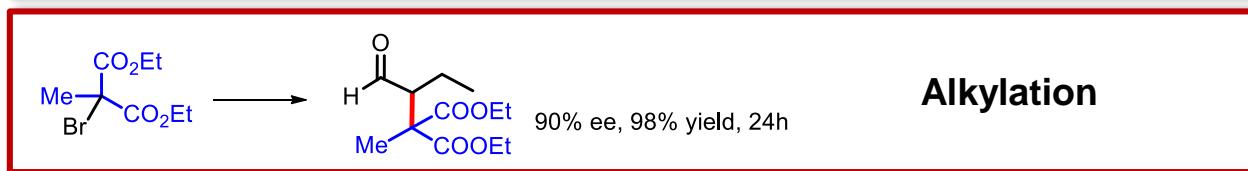
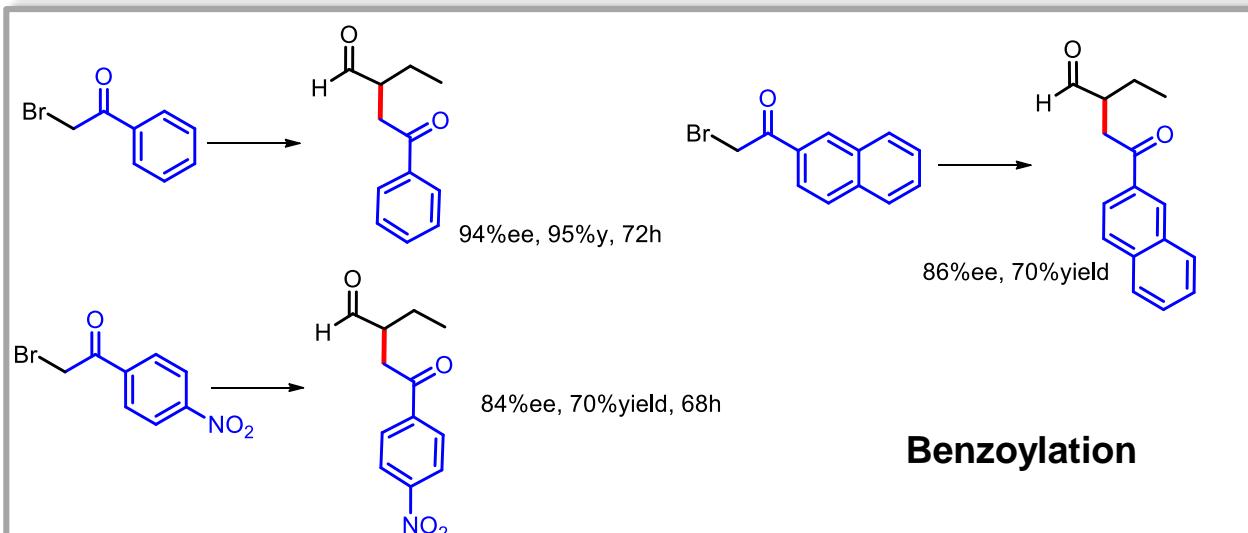
Expanding the Scope



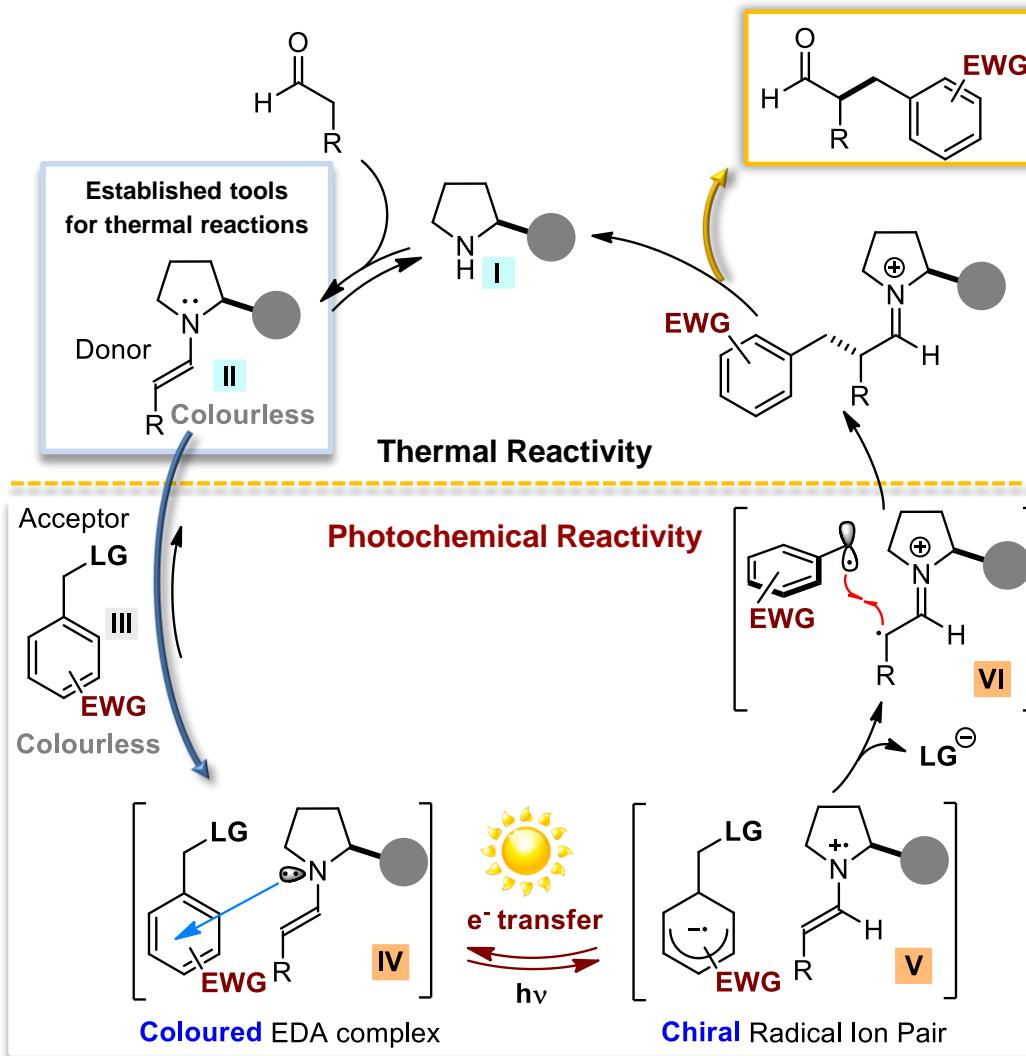
Igor Jurberg



Elena Arceo

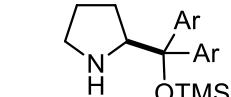
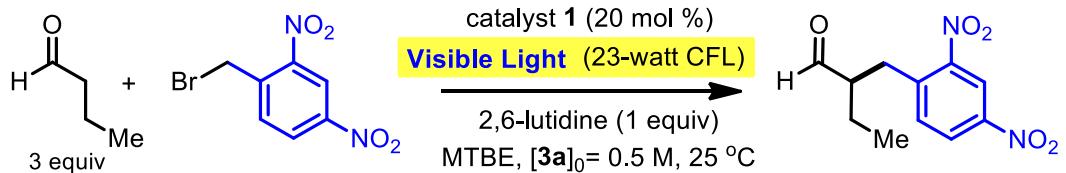


Photochemical Mechanism



EDA complexes and Charge Transfer theory
R. S. Mulliken, *J. Phys. Chem.* 1952, 56, 801

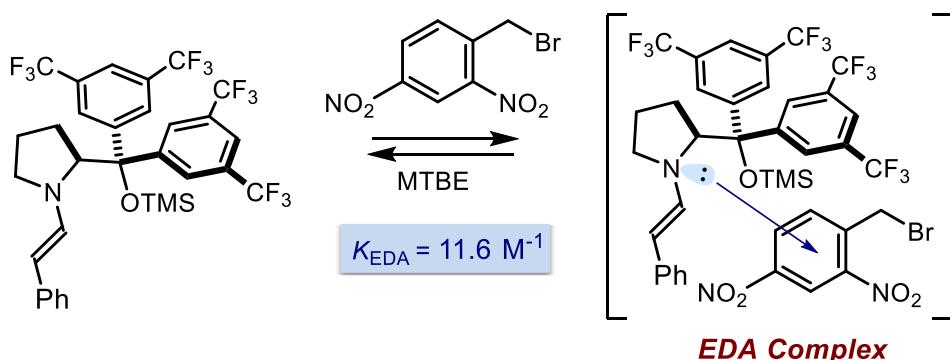
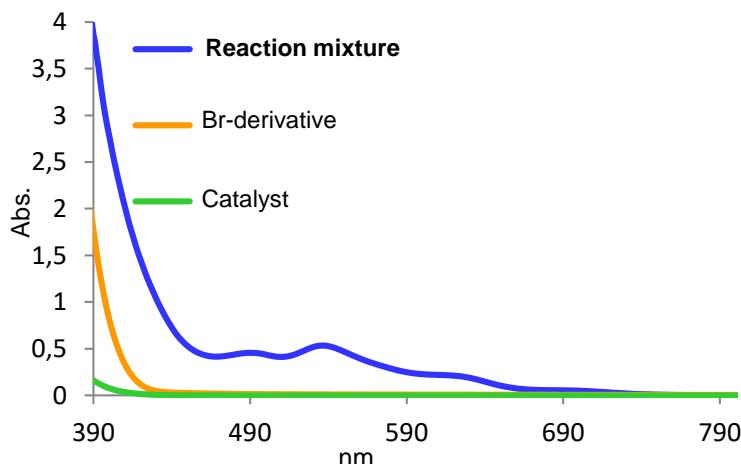
Electron Donor-Acceptor (EDA) Complex



1a: Ar = C₆H₅
1b: Ar = 3,5-(CF₃)₂-C₆H₃

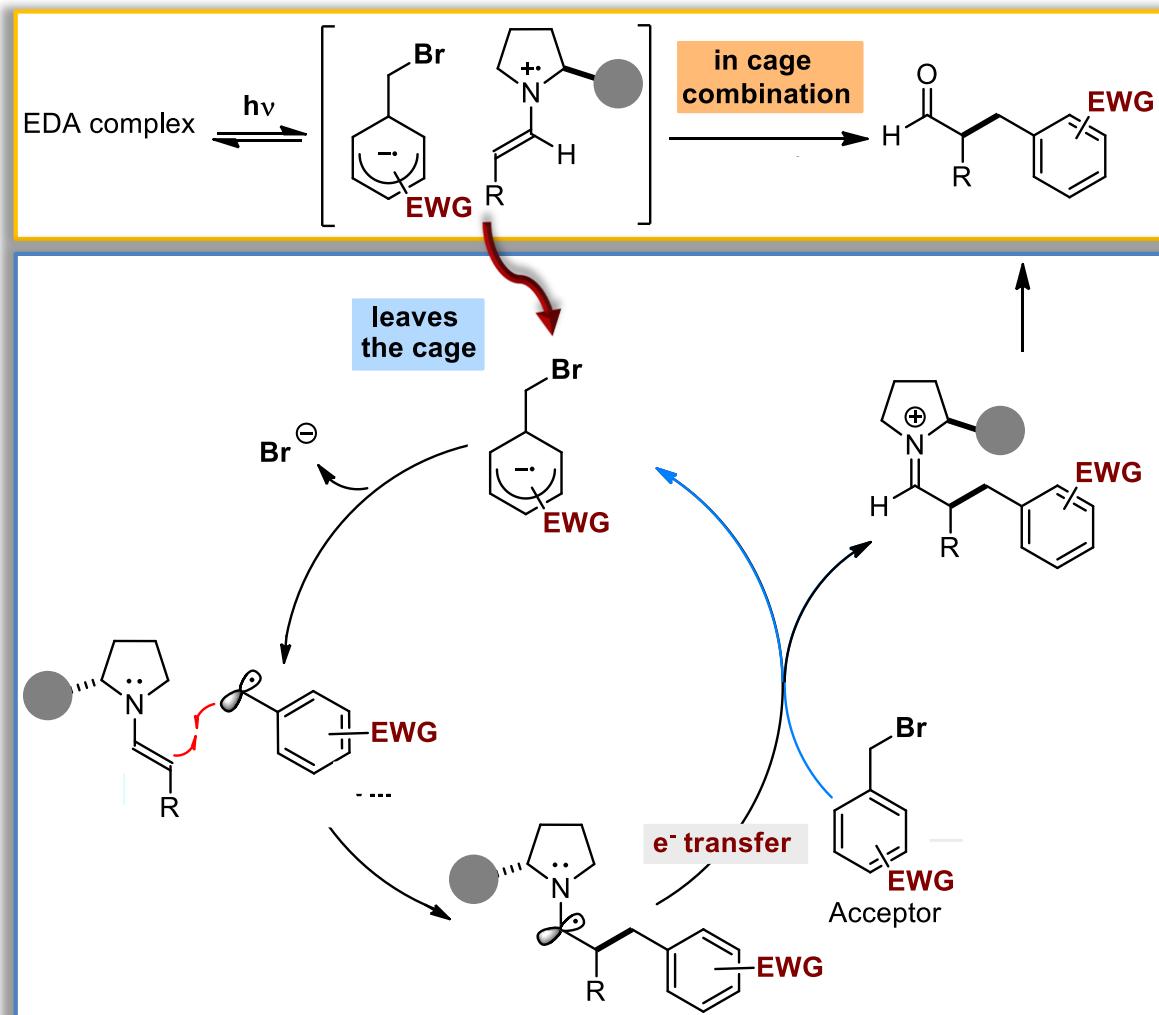
about Serendipity & Observations

visual observation



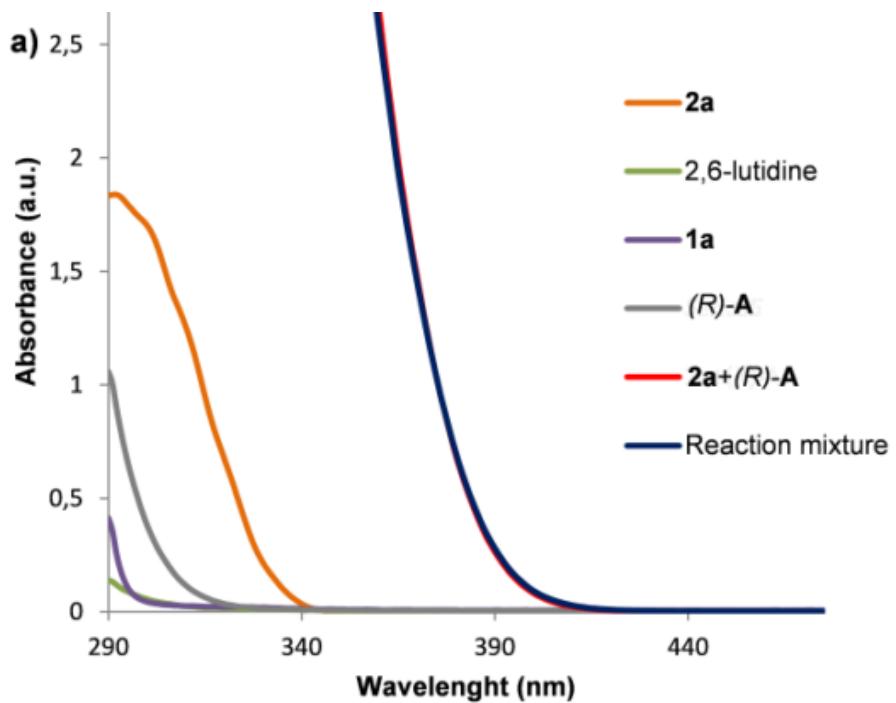
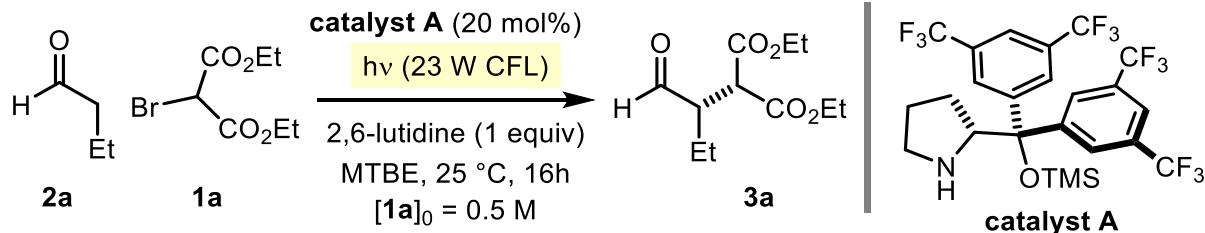
For mechanistic studies (with Ana Bahamonde)
J. Am. Chem. Soc. **2016**, *138*, 8019–8030

an Alternative Path: *Chain Mechanism*



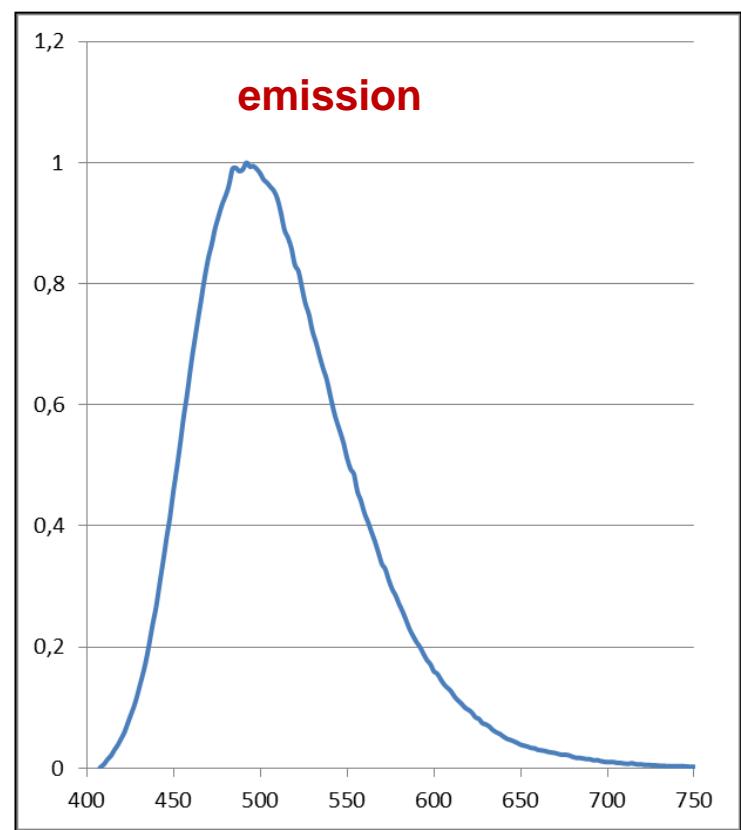
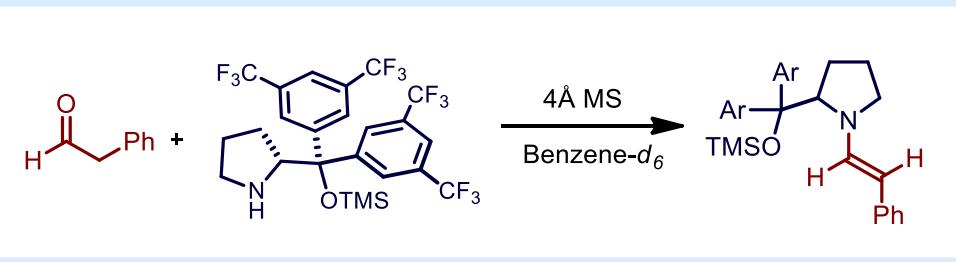
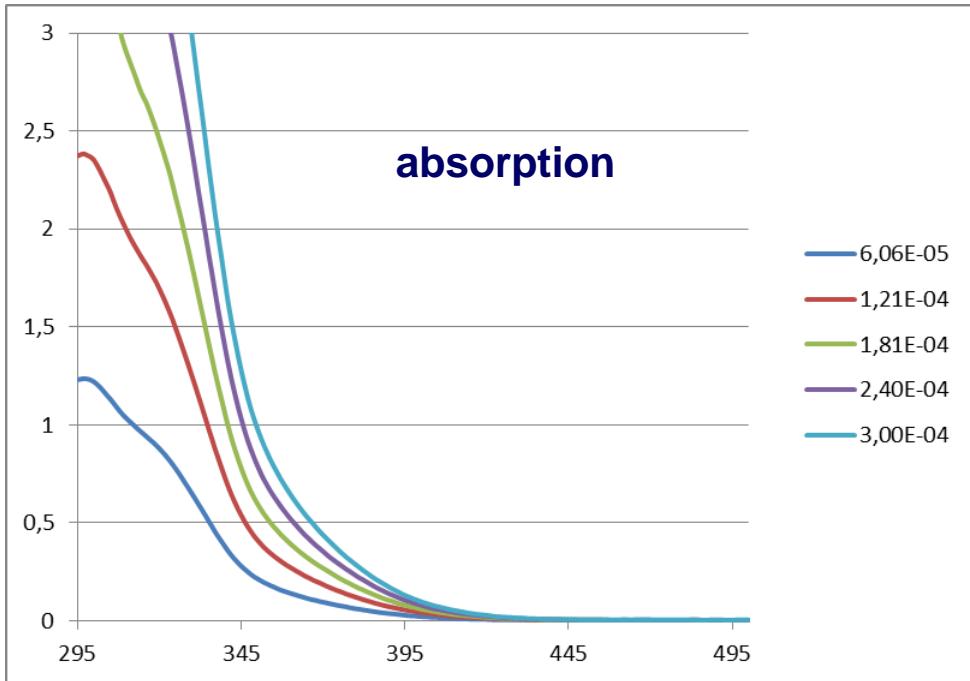
A quantum yield (Φ) of **25** was determined ($\lambda = 450 \text{ nm}$)

Asymmetric Photochemical Alkylation



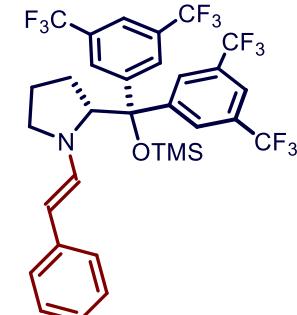
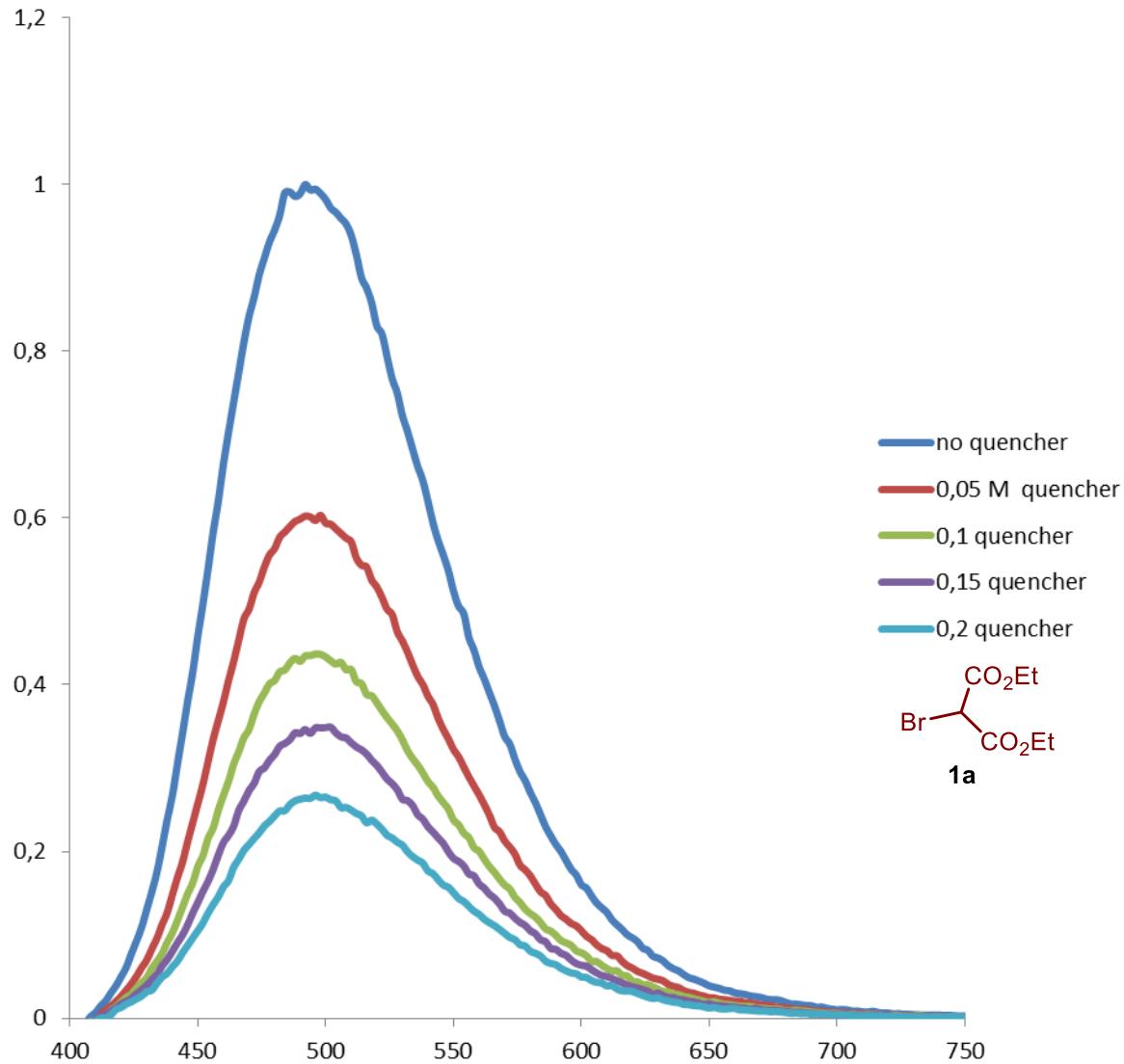
the enamine weakly absorbs visible light

Direct Excitation of Enamines

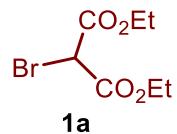


Mattia Silvi

Direct Excitation of Enamines



no quencher
0,05 M quencher
0,1 quencher
0,15 quencher
0,2 quencher

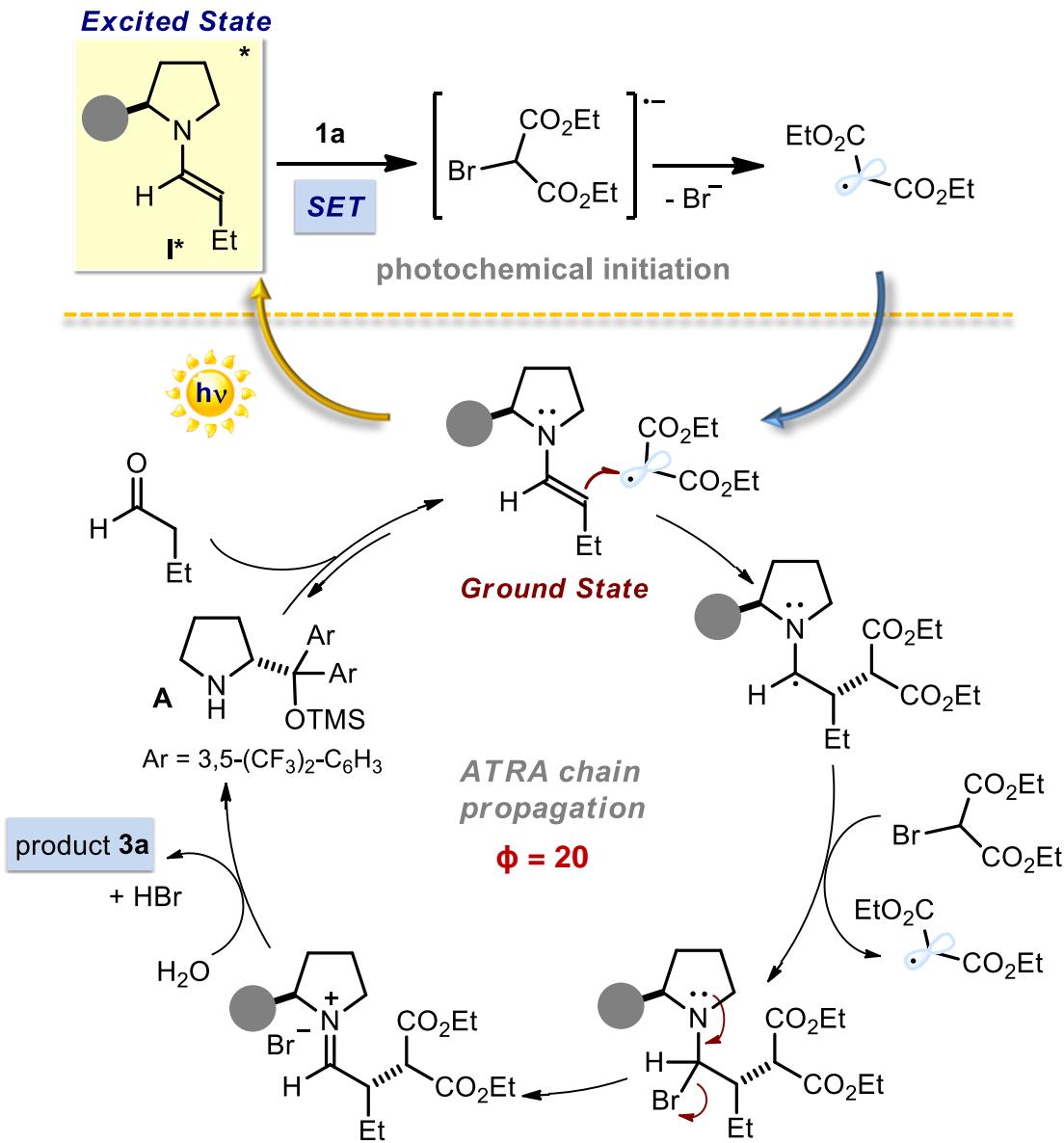


Stern–Volmer quenching studies



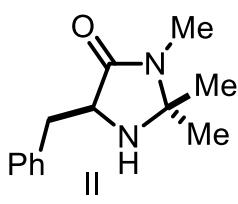
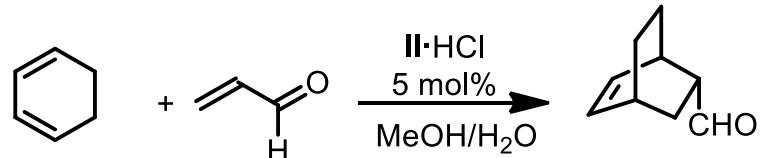
Mattia Silvi

Photo-Excitation of Enamines

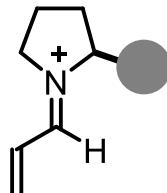


Iminium Ion in the Ground State

- Iminium ion catalyzed asymmetric Diels-Alder of enals



Iminium-Ion Catalysis



LUMO-Lowering
Activation

K. A. Ahrendt, C. J. Borths, D. W. C. MacMillan, *J. Am. Chem. Soc.* **2000**, 122, 4243-4244

iminium ion-mediated catalysis

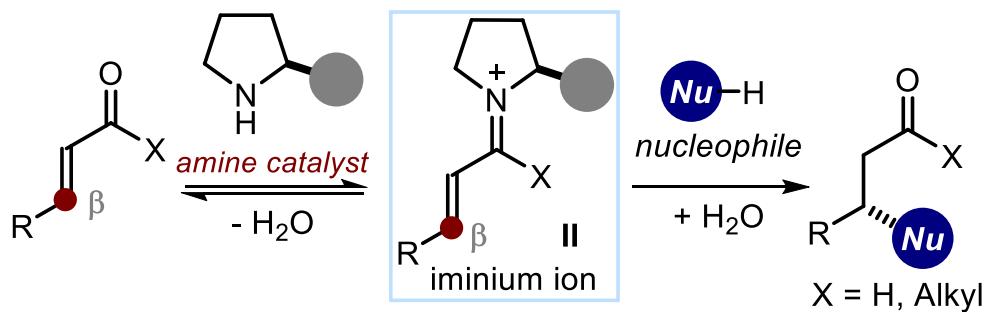
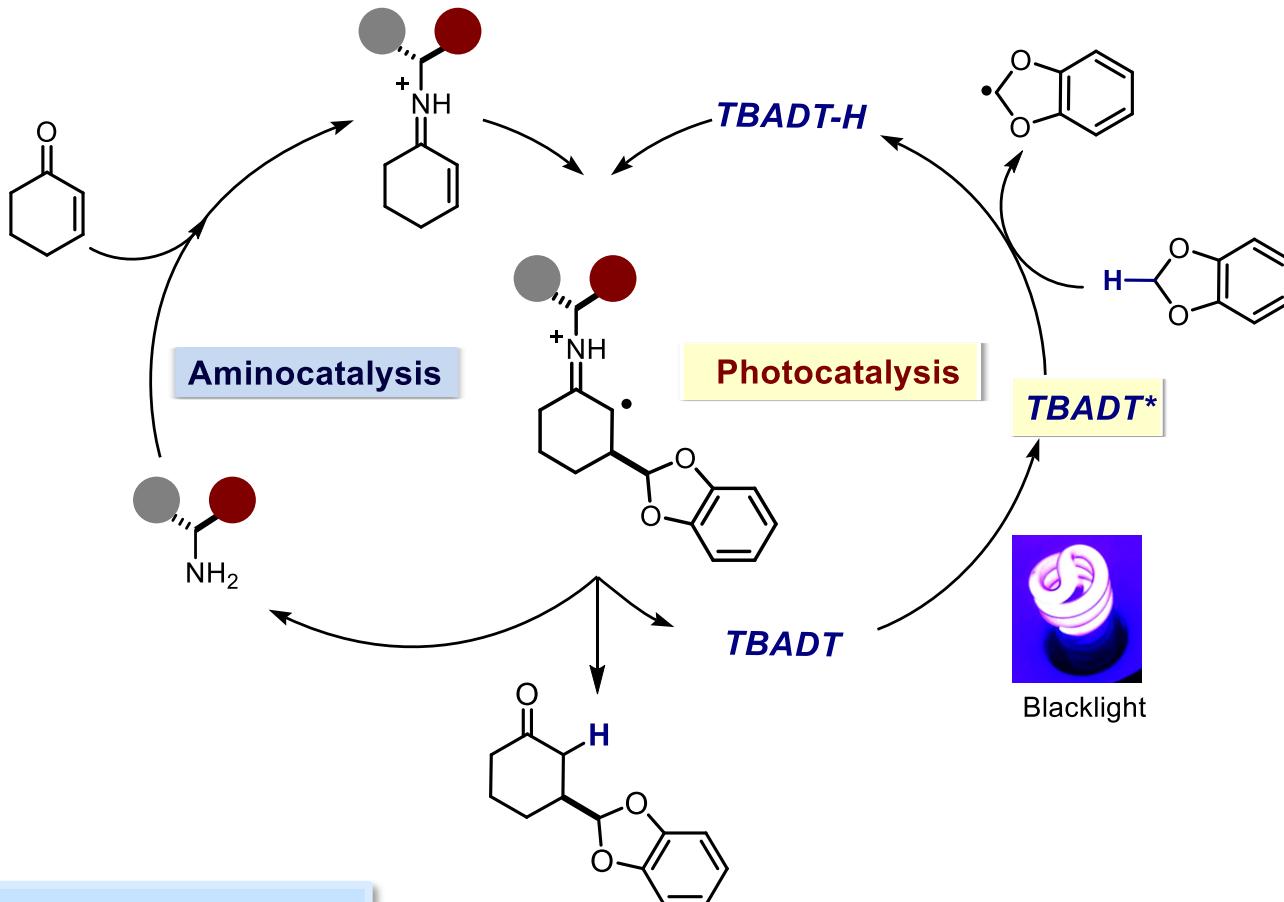


Photo-Organocatalytic Radical Conjugate Additions

HAT: Hydrogen Atom Transfer mechanism



collaboration with Maurizio Fagnoni
Università di Pavia (Italy)

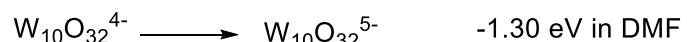
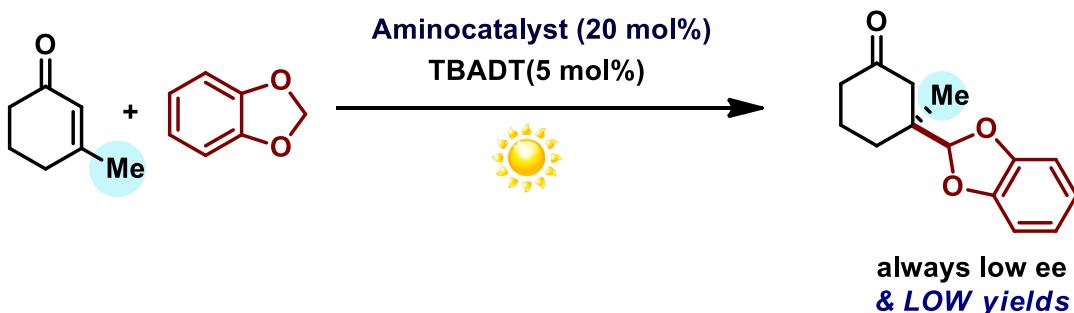


Photo-Organocatalytic Radical Conjugate Additions



Quaternary carbon stereocentre



David Bastida



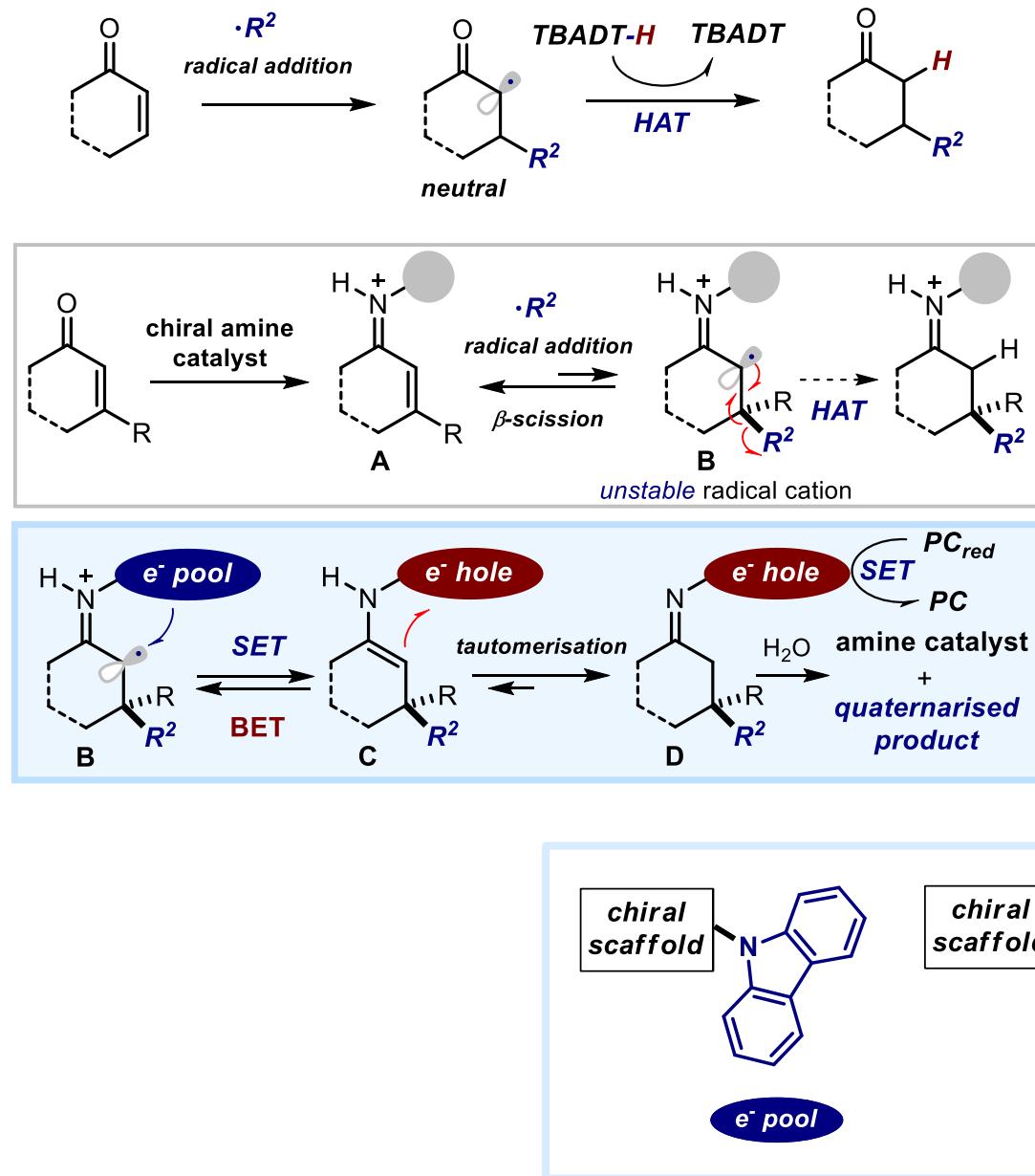
John Murphy

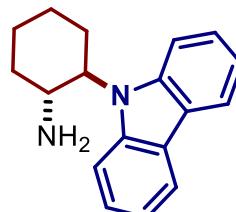
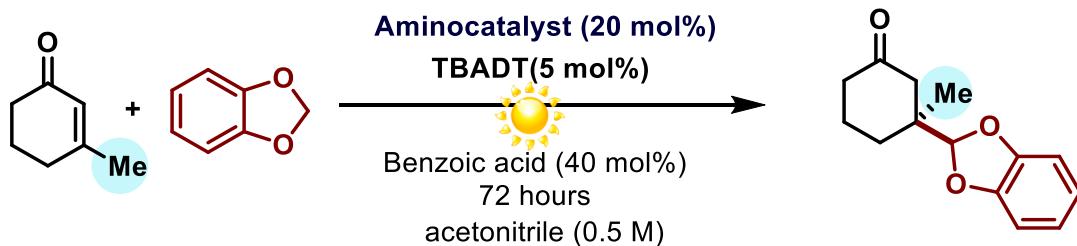
For pertinent precedents on enantioselective catalytic radical conjugate additions

T. P. Yoon et al. *J. Am. Chem. Soc.* **2015**, *137*, 2452

M. P. Sibi et al. *J. Am. Chem. Soc.* **2006**, *128*, 13346

Identifying the Problem





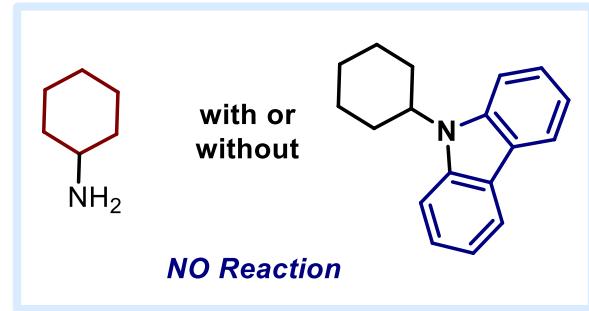
52% yield
80% ee

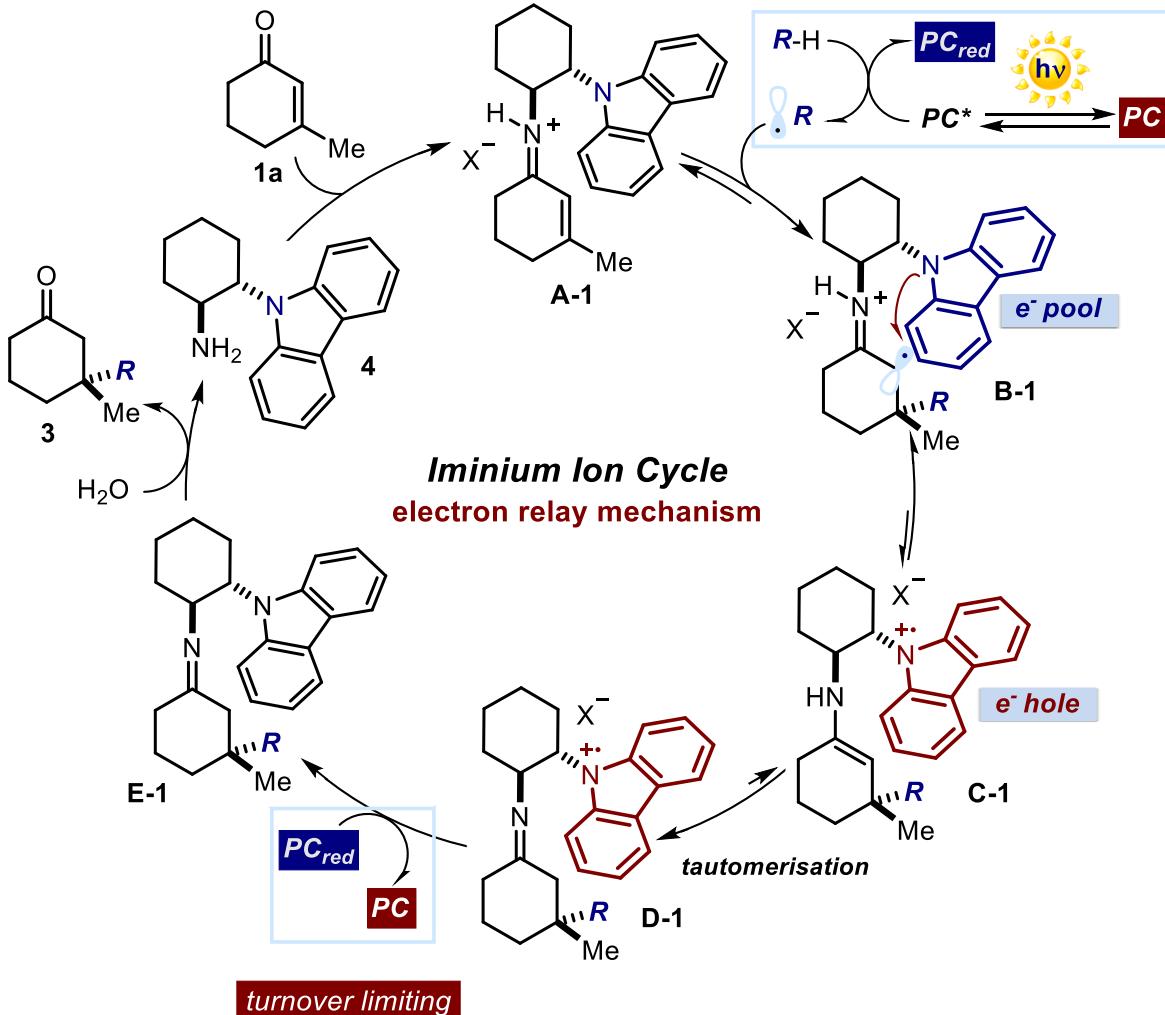


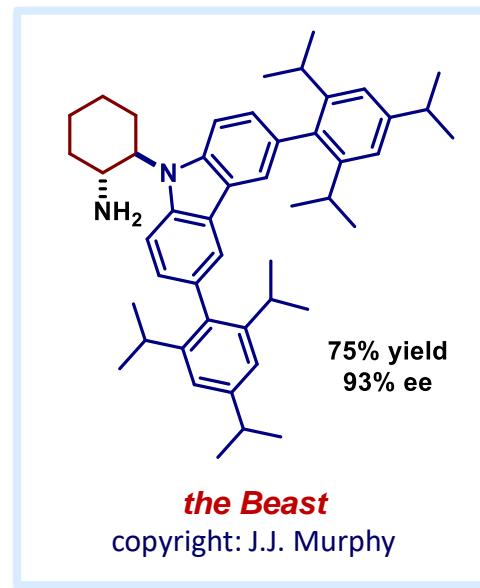
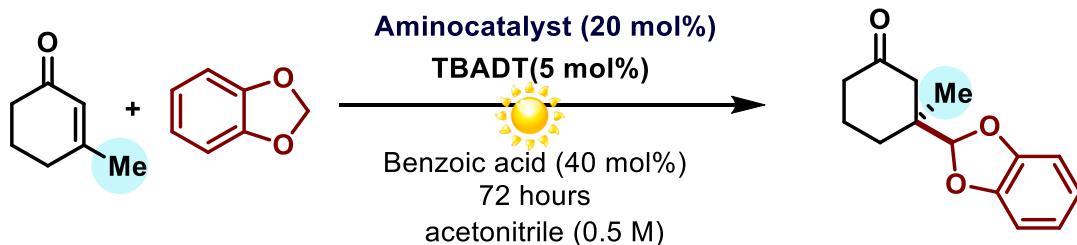
David Bastida



John Murphy





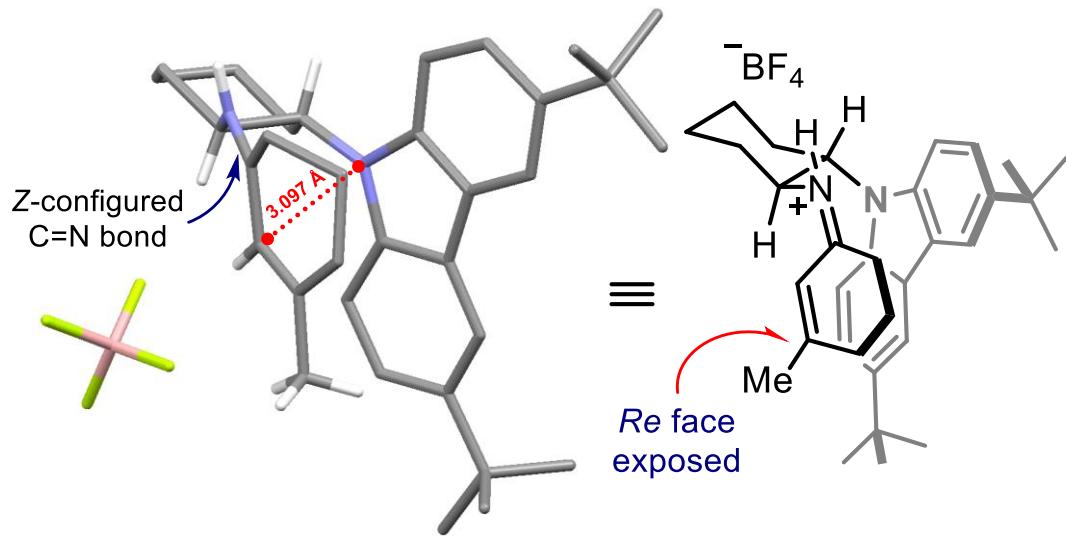


David Bastida



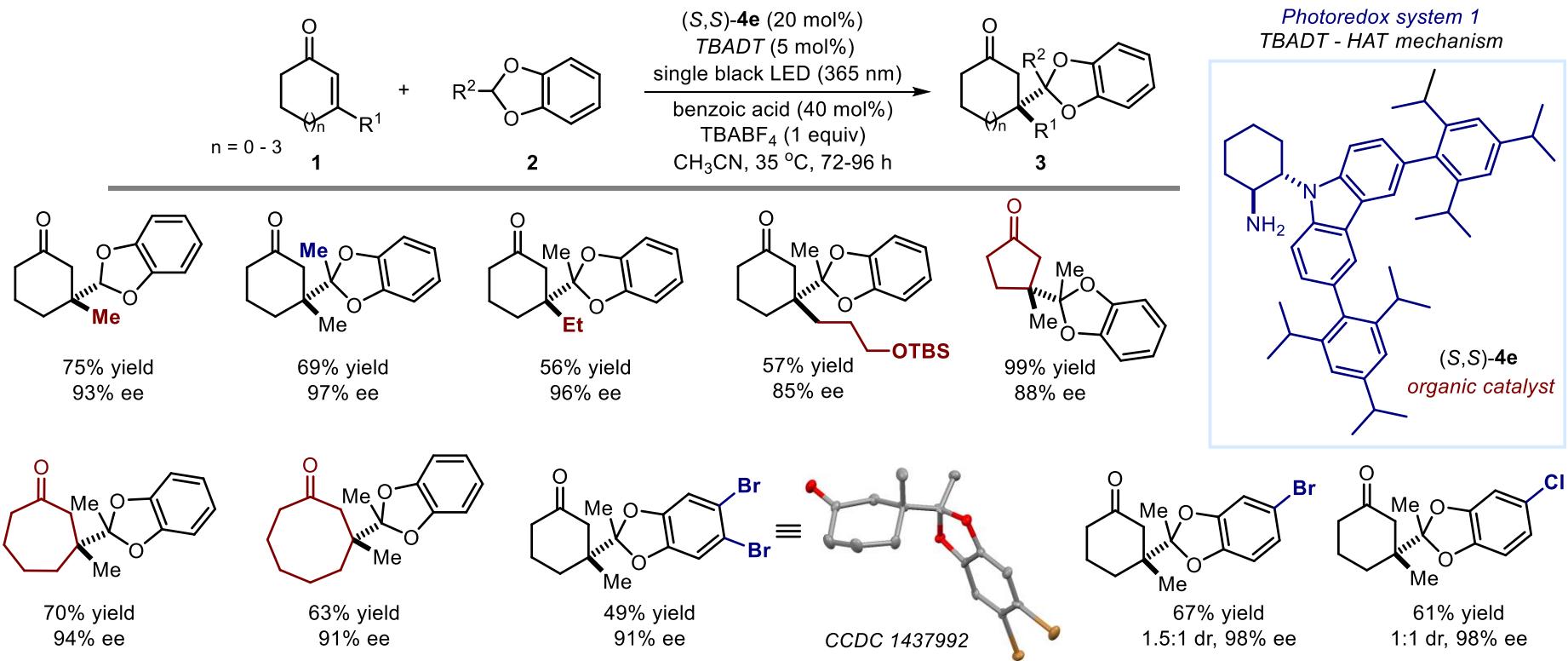
John Murphy

with J.J. Murphy, D. Bastida, S. Paria, M. Fagnoni
Nature **2016**, *532*, 218–222

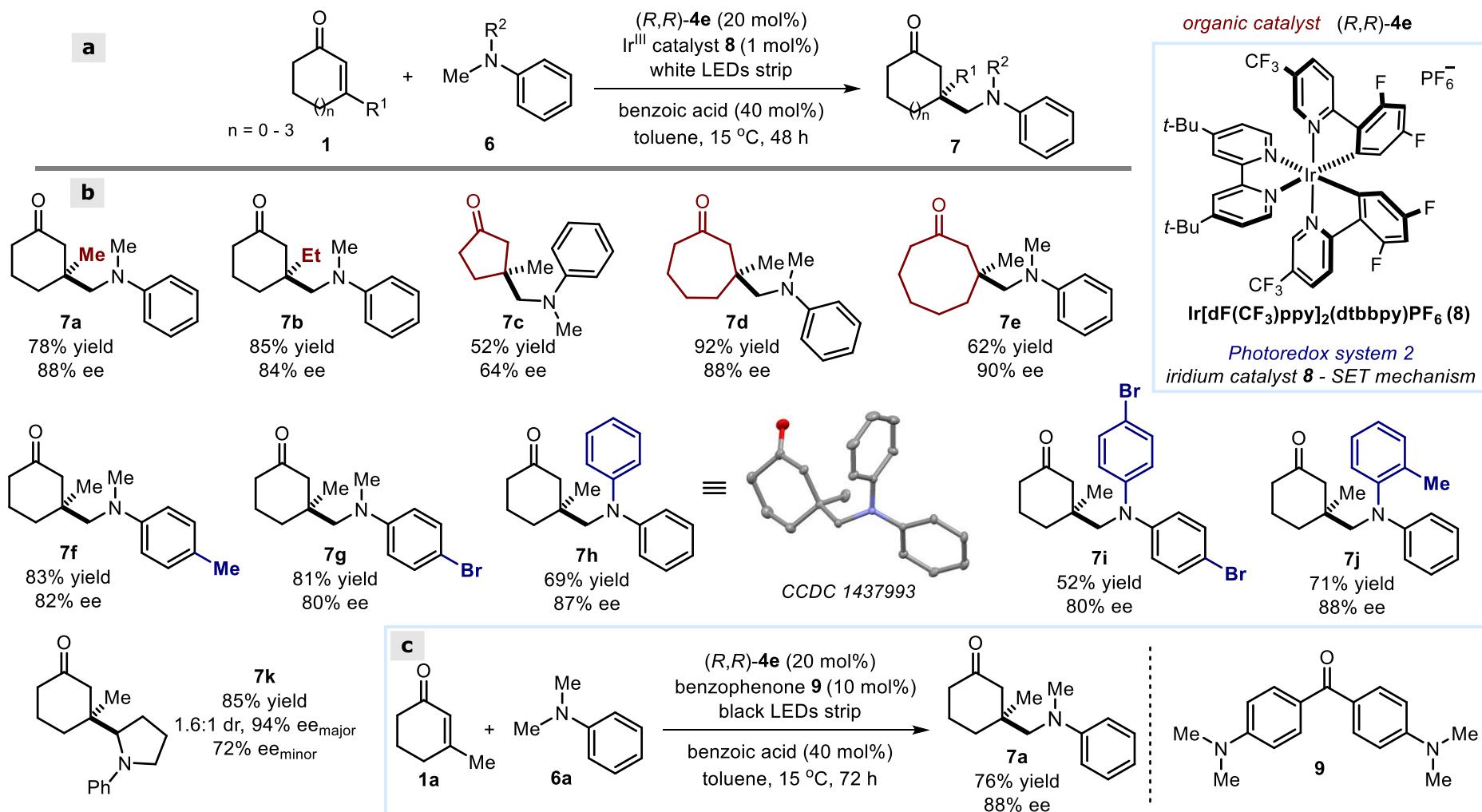


CCDC 1437991

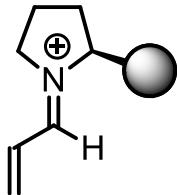
Results



Results

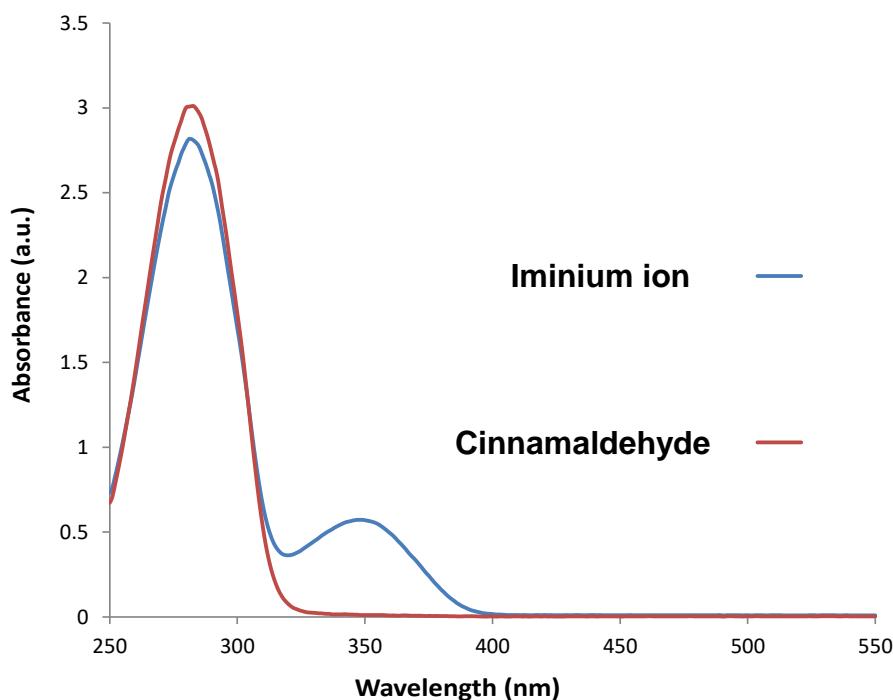
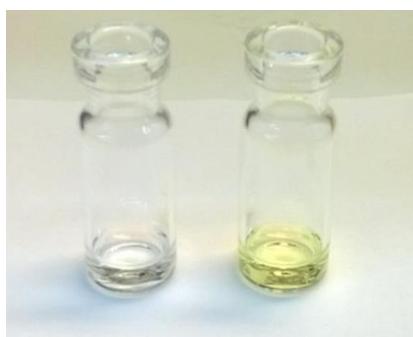
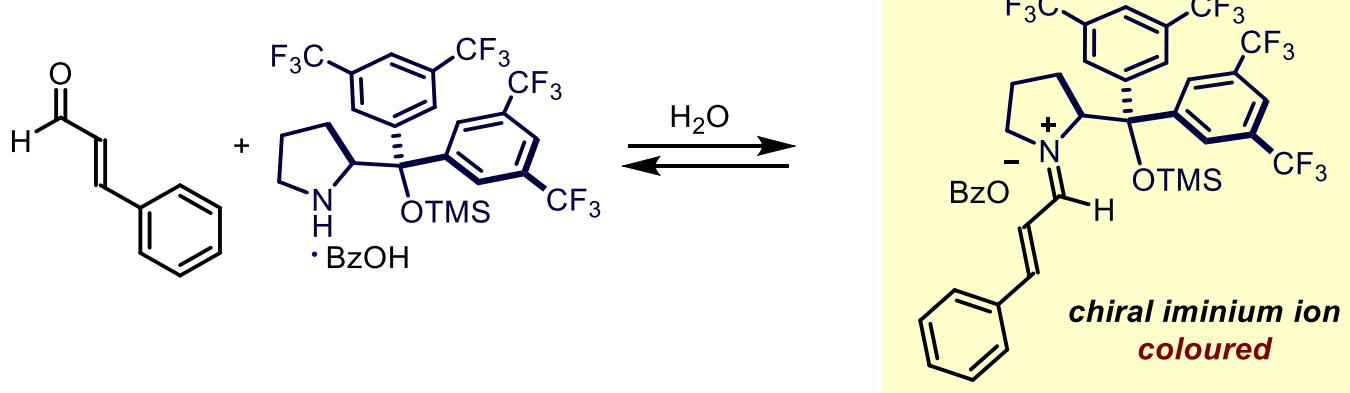


Iminium-Ion Catalysis



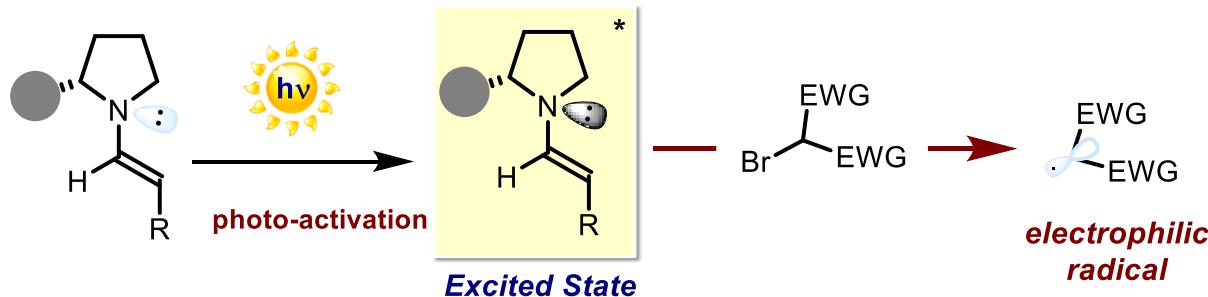
LUMO-Lowering
Activation

the Photochemistry of Iminium Ions



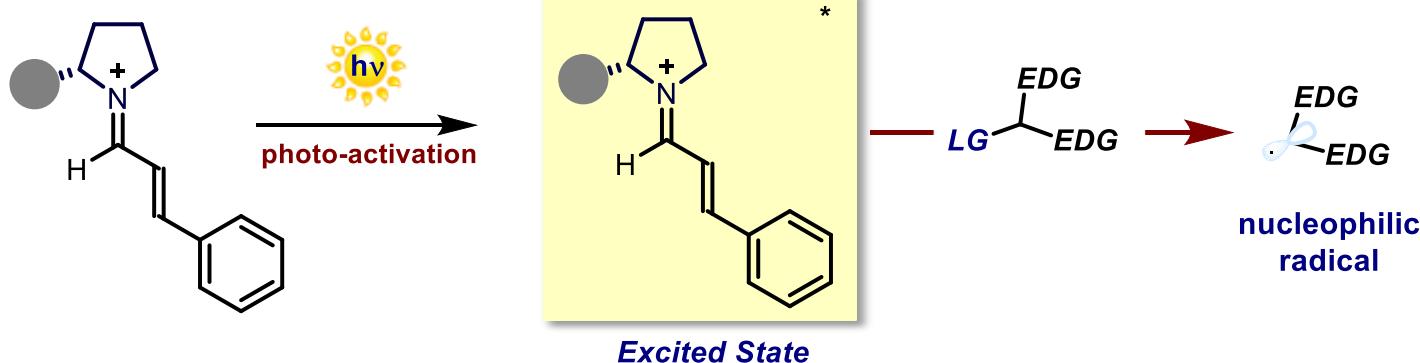
Photochemistry of Enamines in the Excited State

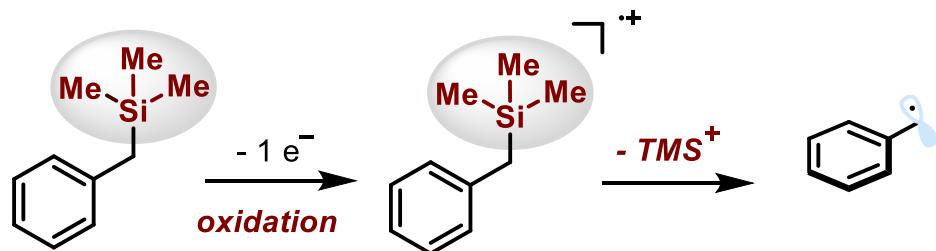
Strong reductant



Photochemistry of Iminium Ions in the Excited State

Strong oxidant?





Mattia Silvi

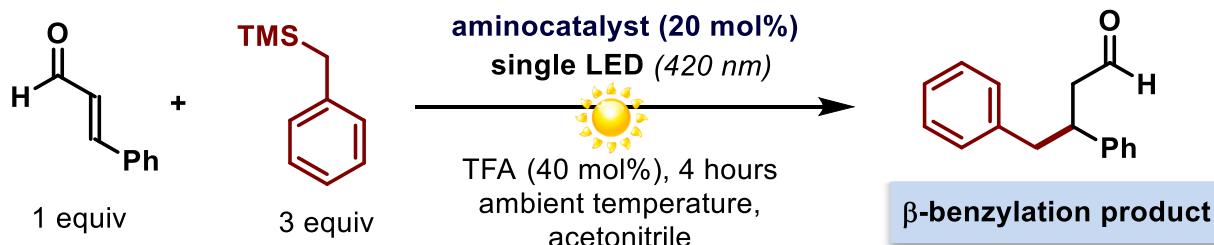
SILANE

- ✓ Low reduction potentials ($E_{ox} = +1.4 - 1.7$ V)
- ✓ Can easily fragment realeasing free radicals
- ✓ Cheap, easy to synthesize, low toxicity

For a pertinent precedent, see:

Ohga, K.; Mariano, P. S. *J. Am. Chem. Soc.* **1982**, 104, 617

Photo-excitation of Iminium Ions

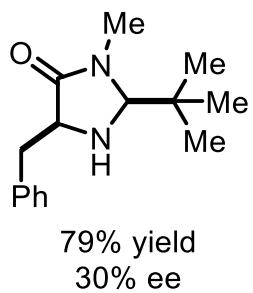
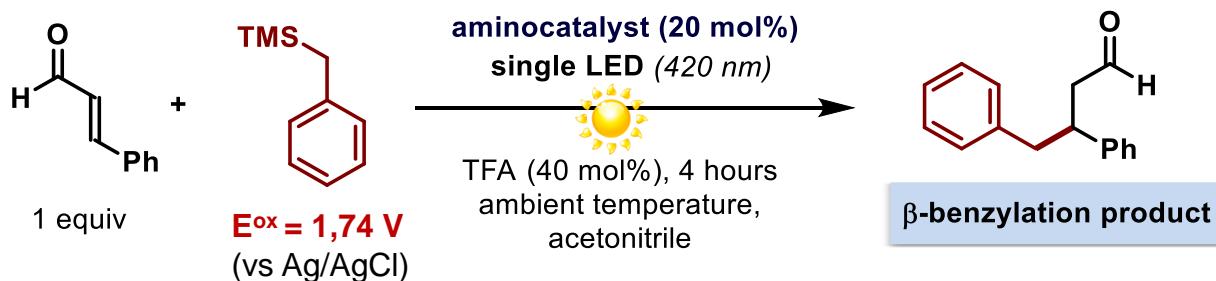


Mattia Silvi

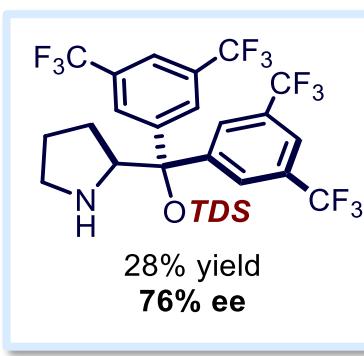


Charlie Verrier

Photo-excitation of Iminium Ions



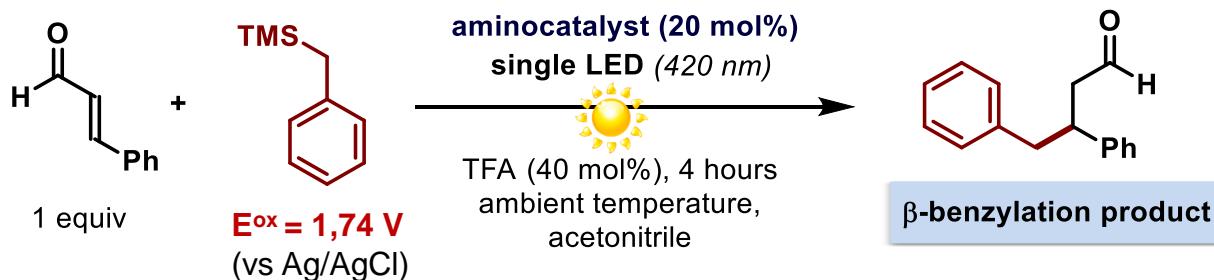
$E^{\text{ox}} = 1,80 \text{ V}$
(vs Ag/AgCl)



TDS: dimethylhexylsilyl

$E^{\text{ox}} = 1,57 \text{ V}$
(vs Ag/AgCl)

Photo-excitation of Iminium Ions



Plausible explanation

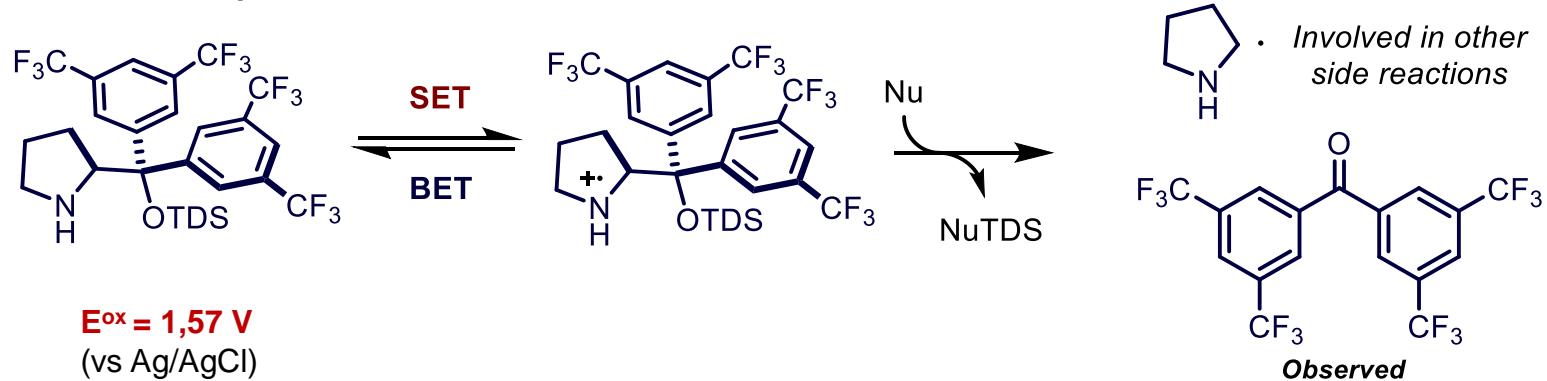
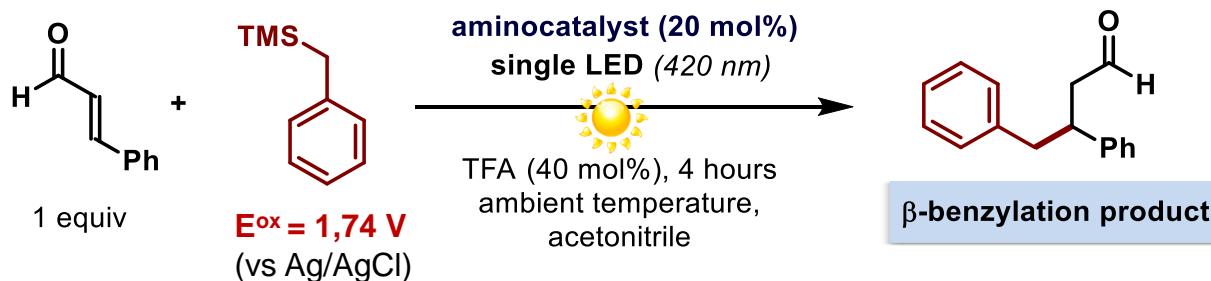
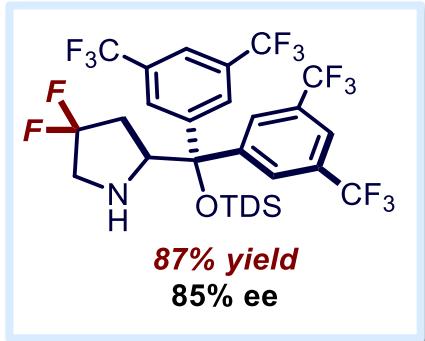


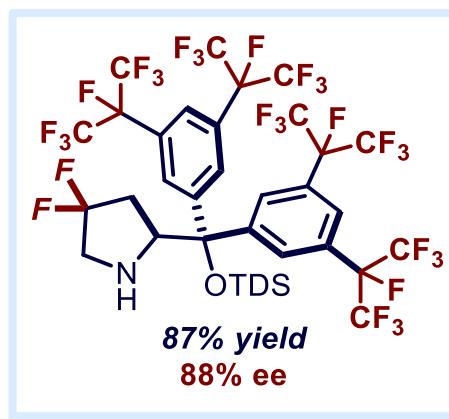
Photo-excitation of Iminium Ions



Charlie Verrier



$E^{\text{ox}} = 2,20 \text{ V}$
(vs Ag/AgCl)

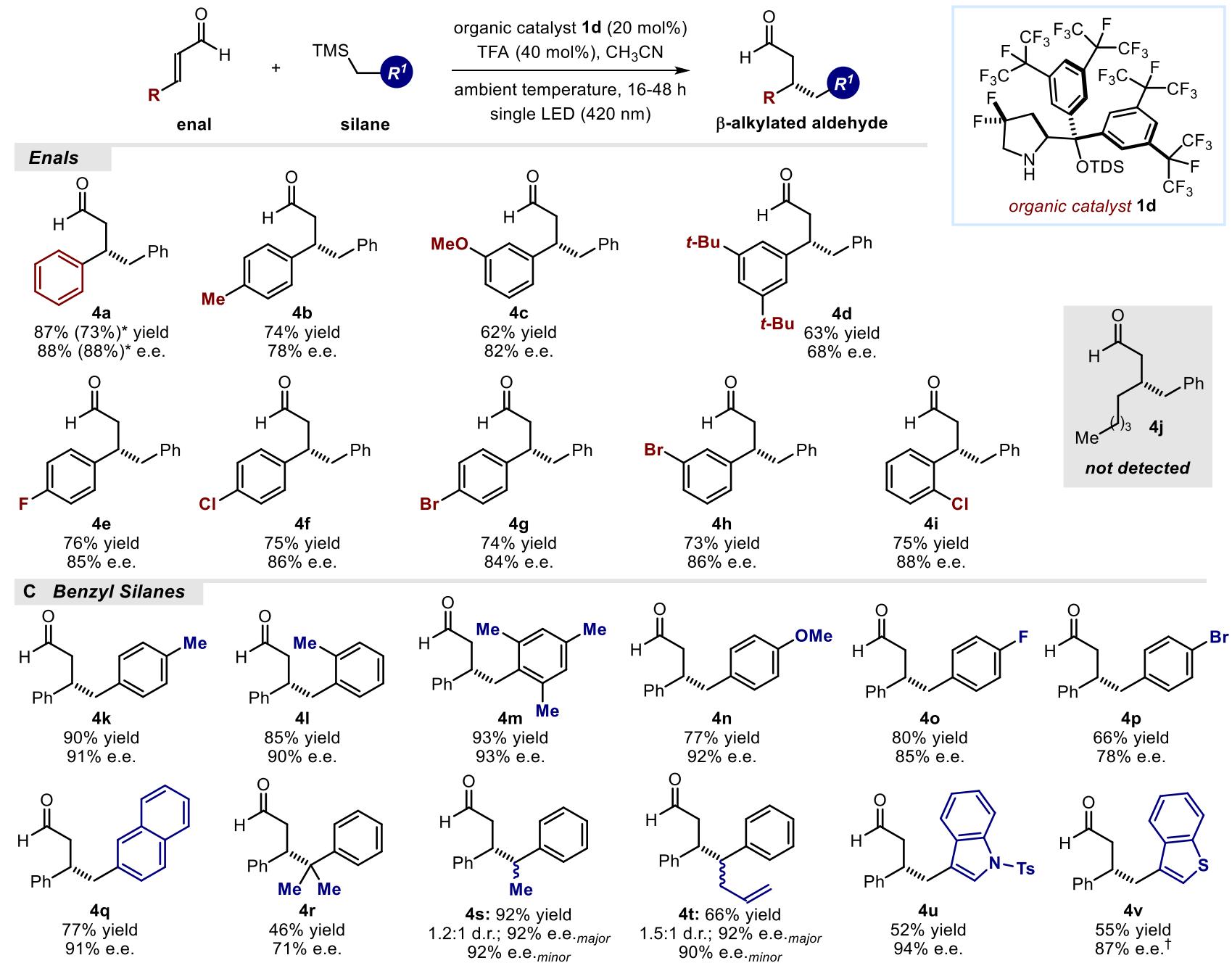


$E^{\text{ox}} = 2,40 \text{ V}$
(vs Ag/AgCl)

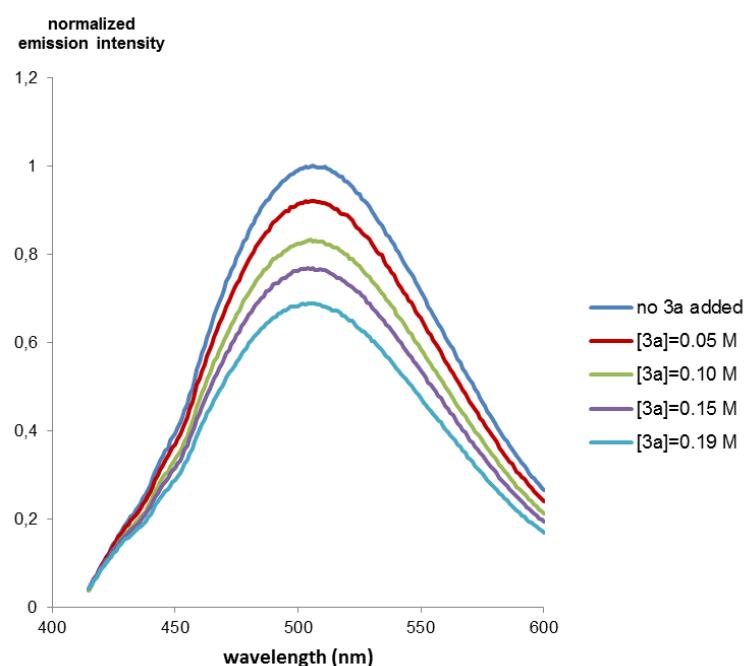
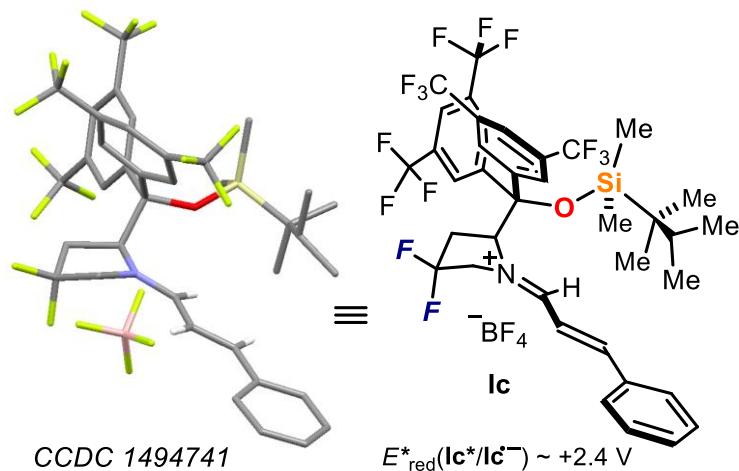
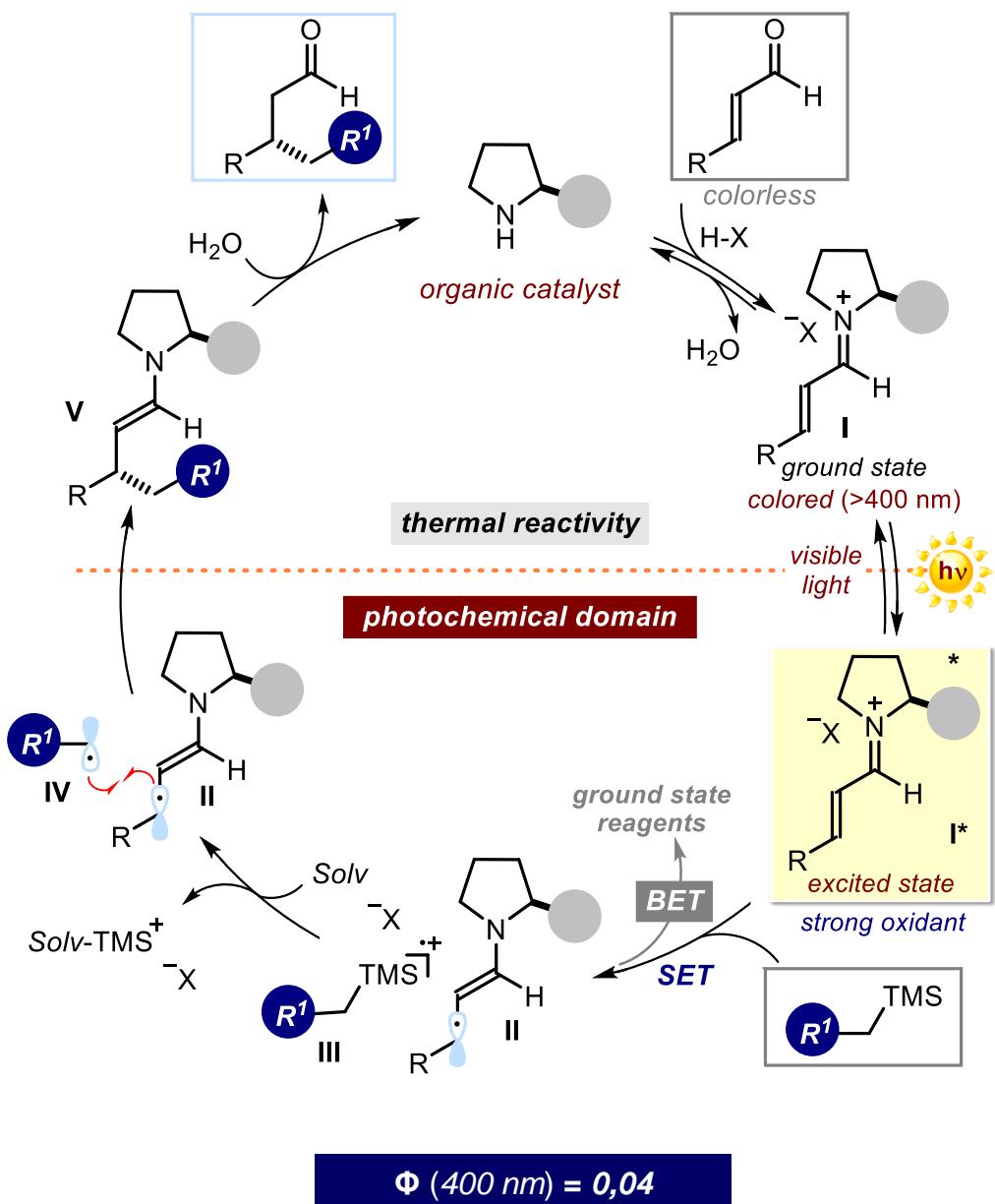
TDS: dimethylhexylsilyl

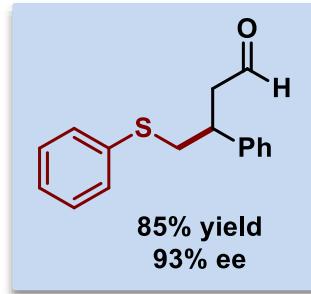
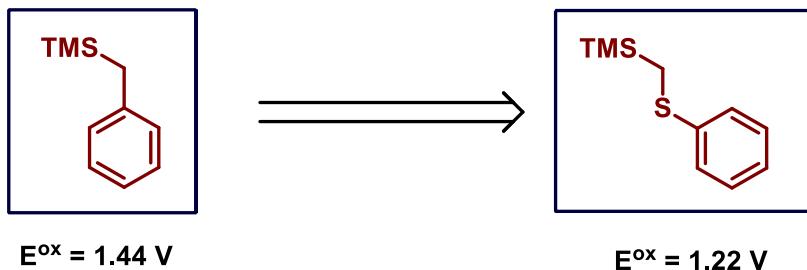


Yannick Rey

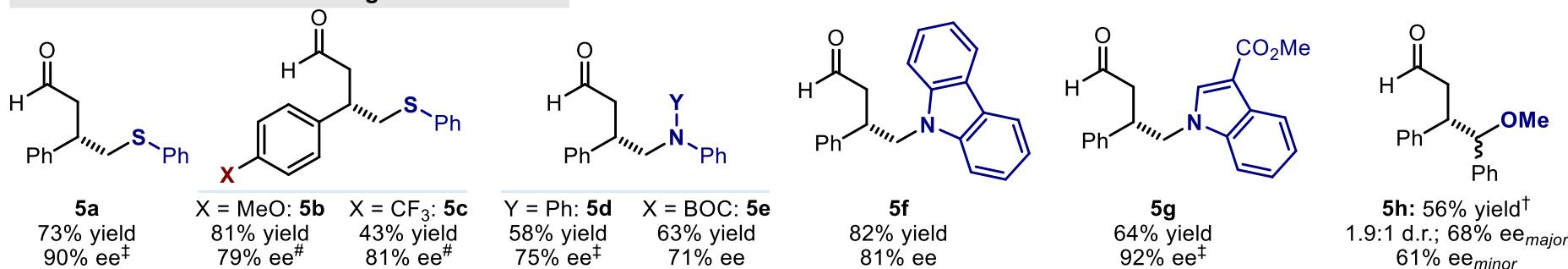


Proposed mechanism

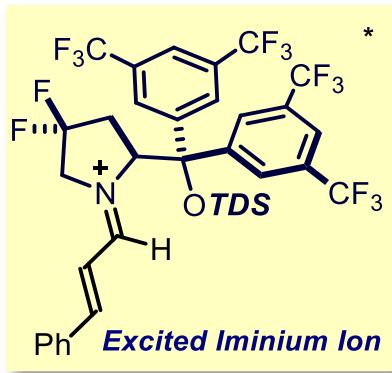




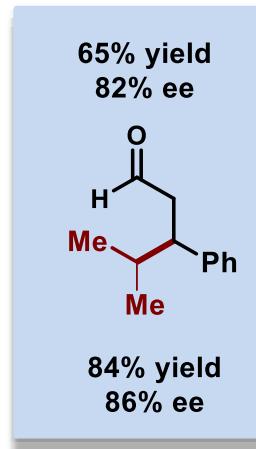
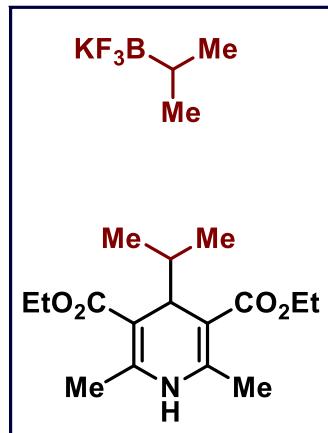
α -Heteroatom-substituted Organic Silanes



Photochemistry of Iminium Ions



Strong oxidant



with Charlie Verrier, Nurtalya Alandini, Luca Buzzetti
ACS Catal. **2018**, *8*, 1062–1066

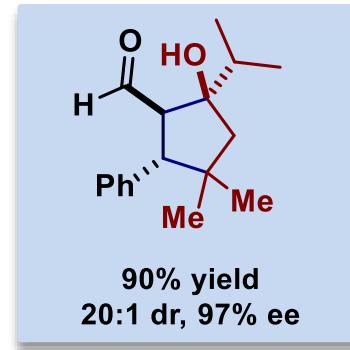
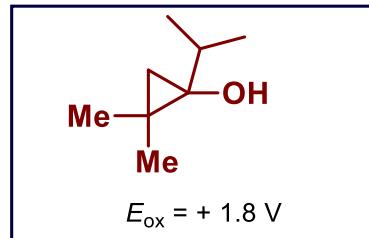
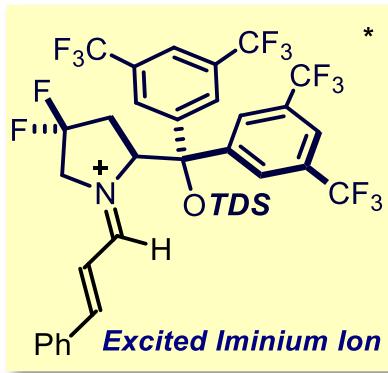
For pertinent precedents:

J. C. Tellis, D. N. Primer, G. A. Molander, *Science* **2014**, *345*, 433–436

K. Nakajima, S. Nojima, K. Sakata, Y. Nishibayashi, *ChemCatChem* **2016**, *8*, 1028–1032

G. A. Molander *et al.*, *ACS Catal.* **2016**, *6*, 8004–8008

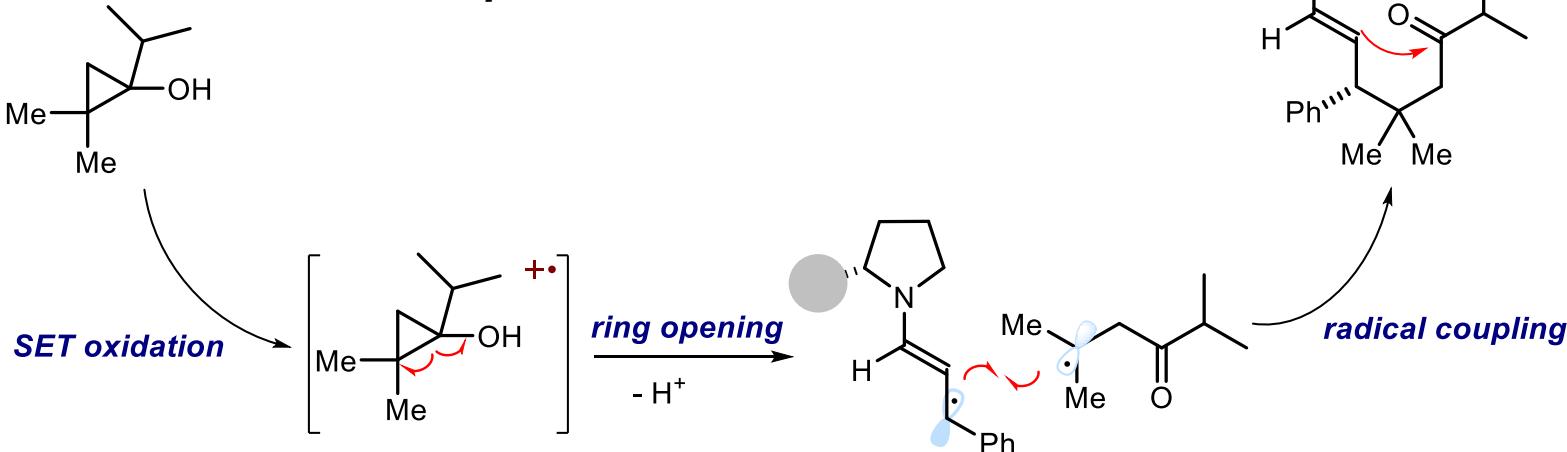
Asymmetric Organocatalytic Photo-Cascade



Strong oxidant

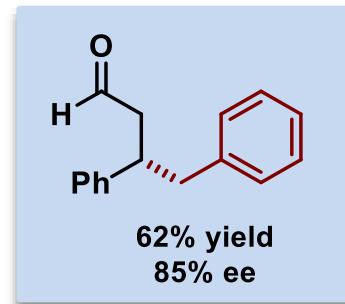
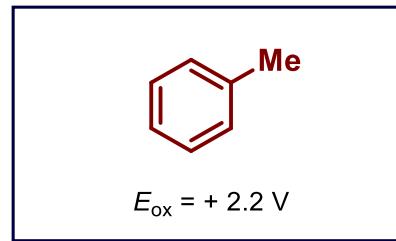
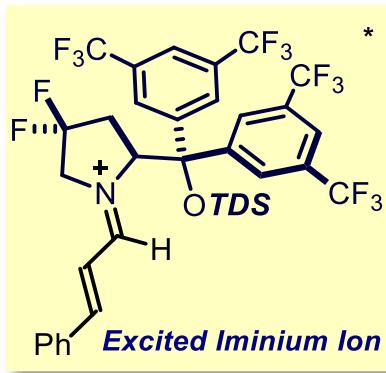
with Lukasz Wozniak & G. Magagnano,
Angew. Chem. Int. Ed. **2018**, 57, 1068

Mechanistic path



For a pertinent study, see:
P. Mariano, *Tetrahedron* **1981** 37, 3385-3395

Asymmetric Photocatalytic C-H Functionalization of Toluene



Strong oxidant

with D. Mazzarella, G. Crisenzia,
unpublished results

Mechanistic path

