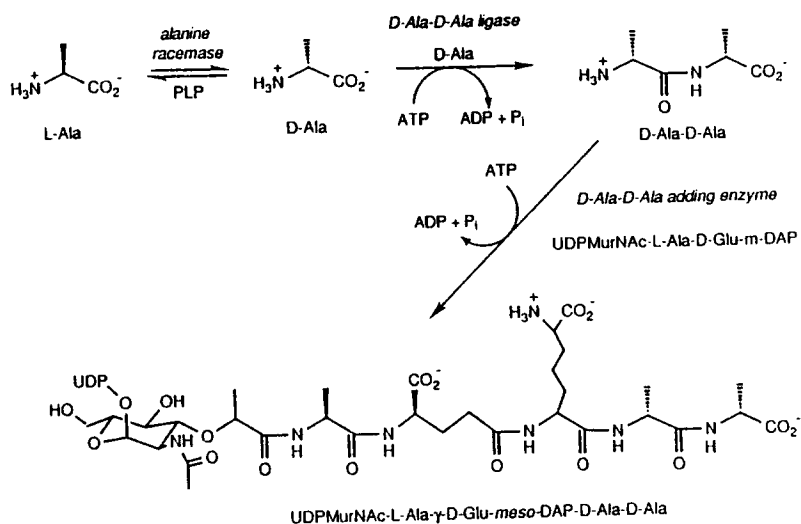


**Molecular Studies
on the
Genes and Enzymes
of the D-Alanine Branch of
Bacterial Peptidoglycan Biogenesis**



Peptidoglycan Biogenesis as a Target for Antibiotics

Enzyme	Antibiotic
Cytoplasmic Phase	
Alanine racemase	β-F-D-Alanine, Ala-P
D-Ala-D-Ala ligase	D-Cycloserine (+ Phosphinate analog of D-Ala-D-Ala)
D-Ala-D-Ala adding enzyme	None
Membrane Phase	
Undecaprenyl-PP transferase	Bacitracin
Cross-linking transpeptidase	β-Lactams (penicillin, cephalosporin)
Binding of peptidyl-D-Ala-D-Ala	Vancomycin, ristocetin

To study enzyme mechanism: first need pure enzyme in substantial (mg) quantity.

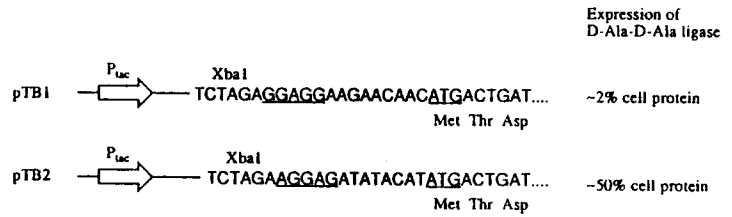
- Strategy:
1. Clone gene.
 2. Sequence gene.
 3. Construct an overproducing vector.
 4. Purify to homogeneity.
 5. Characterize for mechanistic/structural studies.

Two Approaches to Cloning:

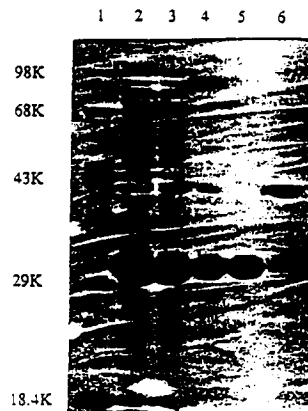
1. Hybridisation
 - purify a small amount of protein
 - obtain N-terminal sequence
 - make ³²P-labelled complementary oligonucleotide probe
 - hybridise probe to gene; clone.
2. Complementation
 - obtain a mutant strain conditionally defective for protein product
 - under non-permissive conditions complement phenotype with DNA library
 - select for complementing DNA; clone.

D-Ala-D-Ala ligase: - *E. coli* ST640, temperature-sensitive *ddl*- strain (chemical mutagenesis)
 - Complementation of ST640 with *E. coli* λ library gives two cloned genes
 ⇒ Two D-Ala-D-Ala ligases in *E. coli*?

Re-Engineering of Ribosome Binding Site



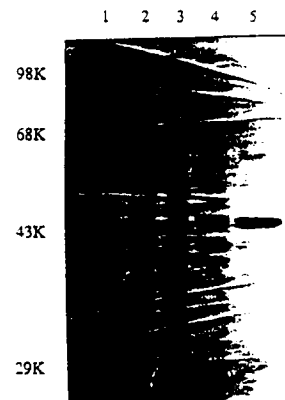
Purification of *E. coli* D-Ala-D-Ala Ligase from W3110/pTB2



Lane 1: Molecular weight markers
 Lane 2: Crude Extract
 Lane 3: Ammonium sulphate (25-50%) ppt.
 Lane 4: AcA54 Gel Filtration
 Lane 5: MonoQ FPLC
 Lane 6: *S. typhimurium* D-Ala-D-Ala Ligase

	Vol (ml)	Protein (mg/ml)	Total (mg)	units (u/ml)	Sp. Activity (u/mg)	Yield (%)	Purification (fold)
Crude Extract	20.9	42.6	890	579	13.6	100	1.0
(NH ₄) ₂ SO ₄ ppt.	11.0	64.0	704	1084	16.9	98.5	1.24
Gel Filtration	64.0	6.7	430	205	30.6	108.5	2.25
MonoQ FPLC	48.0	6.8	326	210	31.0	83.4	2.23

Purification of *E. coli* D-Ala-D-Ala Adding Enzyme from JM105/pTB4



Lane 1: Molecular weight markers
 Lane 2: Crude Extract
 Lane 3: Ammonium sulphate (25-50%) ppt.
 Lane 4: AcA54 Gel Filtration
 Lane 5: MonoQ FPLC

Strategy for Enzyme Mechanistic Study

Determination of **Kinetic** Mechanism - Order of addition of substrates, release of products.

Determination of **Chemical** Mechanism - Order of bond-breaking and bond-making steps, search for intermediates

Both aspects of enzyme mechanism can affect inhibitor design.

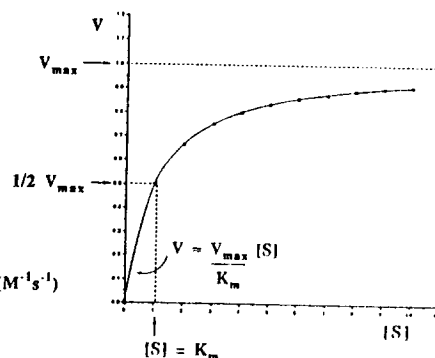


$$V = \frac{V_{max} [S]}{K_m + [S]}$$

$$V_{max} = k_{cat} [E]_{total}$$

$$k_{cat} = \text{turnover number (sec}^{-1}\text{)}$$

$$\frac{k_{cat}}{K_m} = \text{enzymatic catalytic efficiency (M}^{-1}\text{s}^{-1}\text{)}$$

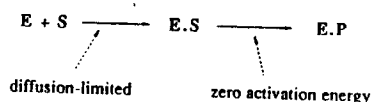


⇒ can experimentally measure V_{max} , K_m

Upper Limit for Catalytic Efficiency

Every E + S encounter gives turnover to E + P

⇒ catalysis limited by rate at which E + S diffuse together



Diffusional upper limit = $10^8 \cdot 10^9 \text{ M}^{-1}\text{s}^{-1}$

$$\Rightarrow \frac{k_{cat}}{K_m} \leq 10^8 \cdot 10^9 \text{ M}^{-1}\text{s}^{-1}$$

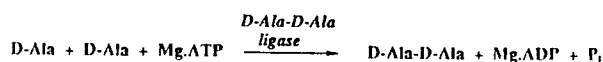
Catalytic Efficiencies of Some Highly Efficient Enzymes

Enzyme	Substrate	k_{cat} (s^{-1})	K_m (M)	k_{cat}/K_m ($\text{M}^{-1}\text{s}^{-1}$)
Acetylcholinesterase	Acetylcholine	1.4×10^4	9×10^{-5}	1.6×10^8
Triose phosphate isomerase	Glyceraldehyde-3-P	4.3×10^3	4.7×10^{-4}	2.4×10^8
β -Lactamase	Benzylpenicillin	2.0×10^3	2×10^{-5}	1×10^8

and some less efficient ones....

Alanine racemase	L-Alanine	14.3	2×10^{-3}	7.2×10^3
D-Ala-D-Ala ligase	D-Alanine	10.7	6×10^{-4}	1.8×10^4
D-Ala-D-Ala adding enzyme	D-Ala-D-Ala	13.1	2.2×10^{-4}	6.0×10^4

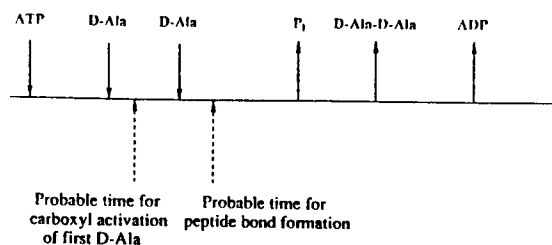
Kinetic Mechanism of D-Ala-D-Ala Ligase



- three substrates/three products.
- ordered vs. random binding and release
- are some products formed and released before all substrates bind?

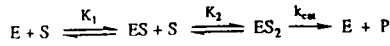
Ordered ter-ter mechanism:

(deduced from steady state kinetics, substrate/product inhibition)



Kinetic Analysis of Two D-Alanine Binding Sites

- D-alanine used as both the **electrophilic** and **nucleophilic** partners in peptide bond formation.
- Two identical molecules of D-alanine bound in distinct binding sites with very different affinities.



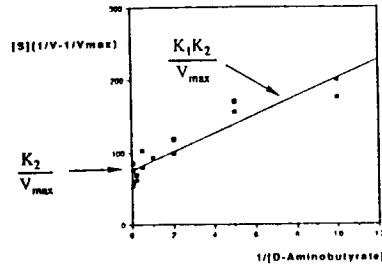
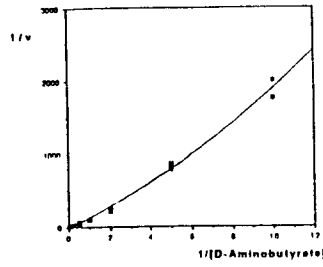
$$v = \frac{v_{max} [S]^2}{K_1 K_2 + K_2 [S] + [S]^2}$$

$$K_1 = 2-6 \mu M$$

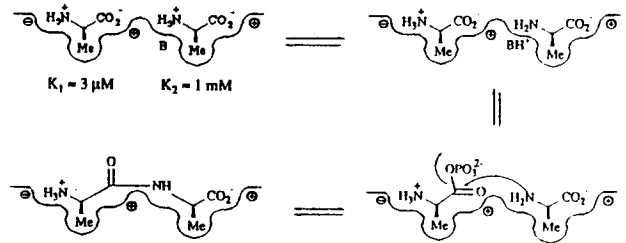
$$K_2 = 540-1200 \mu M (= \text{apparent } K_m)$$

⇒ First D-Ala bound 250-500x more tightly than second D-Ala

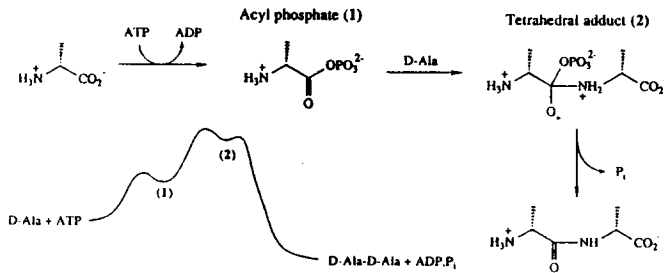
⇒ Specificity leakage at C-terminal D-alanine site? (product D-Ala-D-Xaa)



Proposed Two Subsites for D-Alanine Recognition



Mechanism of D-Ala-D-Ala Ligase



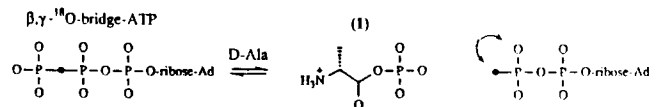
To study reversible formation of (1):

- Positional Isotope Exchange (PIX)
- Molecular Isotope Exchange (MIX)

To study reversible formation of (2):

- make stable analogs of transition state or reaction intermediate; test as inhibitors.
- CPMAS solid state NMR of enzyme-inhibitor complex.

Positional Isotope Exchange (PIX)



Assay by ¹⁸O shift on ³¹P NMR signal of γ-phosphate of ATP.

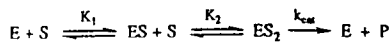
¹⁸O scrambles faster than forward reaction rate:



Kinetic competence vs. **chemical** competence.

Kinetic Analysis of Two D-Alanine Binding Sites

- D-alanine used as both the **electrophilic** and **nucleophilic** partners in peptide bond formation.
- Two identical molecules of D-alanine bound in distinct binding sites with very different affinities.



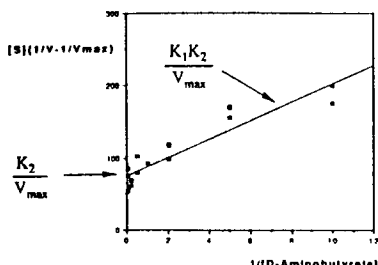
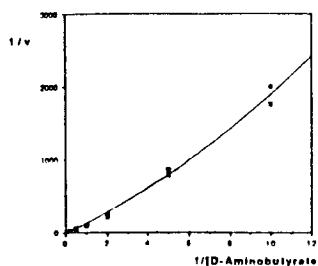
$$v = \frac{v_{max} [S]^2}{K_1 K_2 + K_2 [S] + [S]^2}$$

$$K_1 = 2-6 \mu M$$

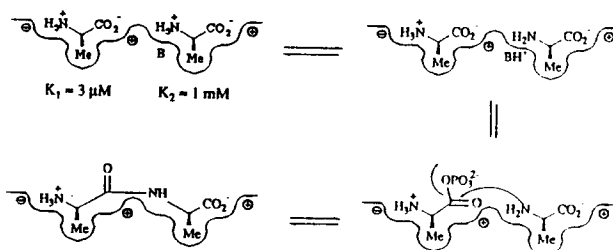
$$K_2 = 540-1200 \mu M (= \text{apparent } K_m)$$

⇒ First D-Ala bound 250-500x more tightly than second D-Ala

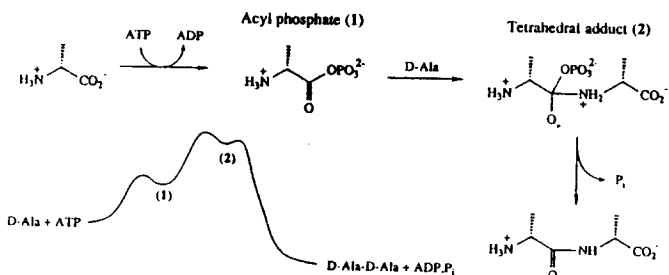
⇒ Specificity leakage at C-terminal D-alanine site? (product D-Ala-D-Xaa)



Proposed Two Subsites for D-Alanine Recognition



Mechanism of D-Ala-D-Ala Ligase



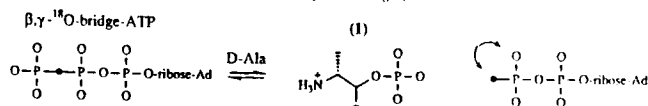
To study reversible formation of (1):

- A. Positional Isotope Exchange (PIX)
- B. Molecular Isotope Exchange (MIX)

To study reversible formation of (2):

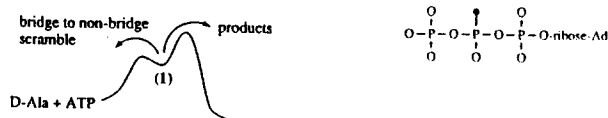
- make stable analogs of transition state or reaction intermediate; test as inhibitors.
- CPMAS solid state NMR of enzyme-inhibitor complex.

Positional Isotope Exchange (PIX)



Assay by ¹⁸O shift on ³¹P NMR signal of γ-phosphate of ATP.

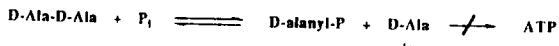
¹⁸O scrambles faster than forward reaction rate:



Kinetic competence vs. chemical competence.

Molecular Isotope Exchange (MIX)

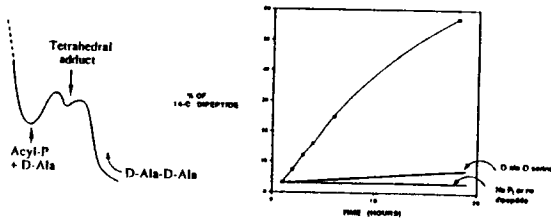
Partial back reaction in absence of ADP (phosphorolysis of the amide bond):



Add a trace of ^{14}C -D-Ala

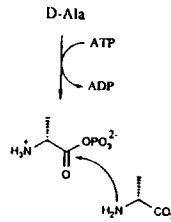
release into solution

Enzyme catalyzes exchange of ^{14}C -D-Ala into D-Ala-D-Ala

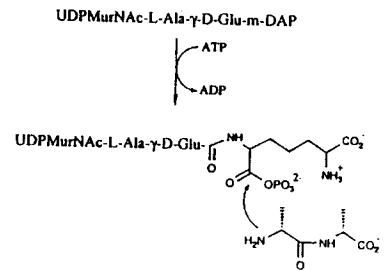


Acyl Phosphoric Anhydrides as Intermediates in ATP-Cleaving Amide Bond Formation

D-Ala-D-Ala ligase

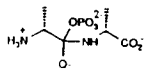


D-Ala-D-Ala adding enzyme

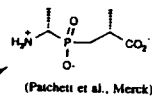


Investigation of Enzymatic Transition State

Proposed transition state

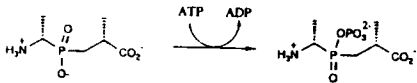


Synthesize stable analogs:



Potent inhibitor that becomes time-dependent in assays containing ATP

Proposed mechanism:

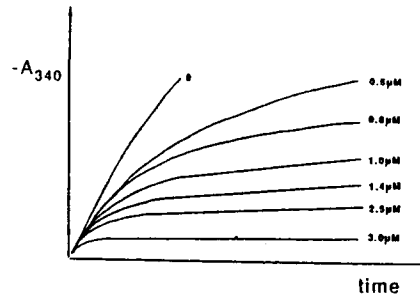


analog to tetrahedral adduct

phosphinate

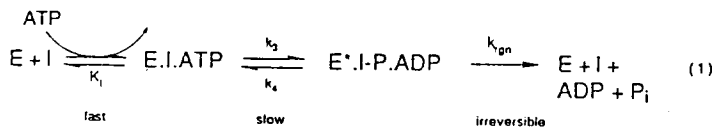
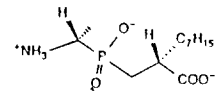
better analog to tetrahedral adduct

phosphinophosphate



D-ALA-D-ALA LIGASE ACTIVITY

Progress curves for time-dependent onset of reduced steady state activity in the presence of:

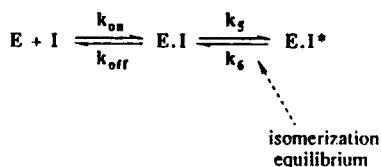


K_d vs. Binding Energy

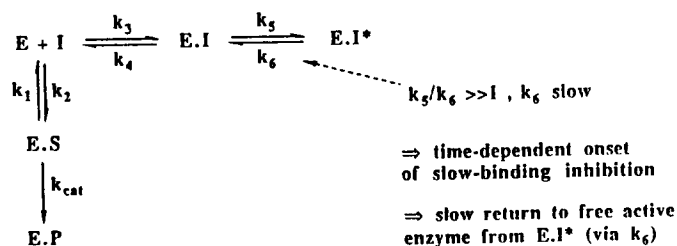
$$\Delta G^\circ = -RT \ln K_{eq}$$

for $K = 10^{-5} \text{ M}$ $\Delta G = -7 \text{ kcal/mol}$
 for $K = 10^{-8} \text{ M}$ $\Delta G = -11.2 \text{ kcal/mol}$
 for $K = 10^{-11} \text{ M}$ $\Delta G = -15.4 \text{ kcal/mol}$

10^{-4} - 10^{-5} M = normal range for substrate K_m
 10^{-7} - 10^{-9} M = range for specific, potent enzyme inhibitors
 10^{-10} - 10^{-12} M = range for "slow binding" inhibitors

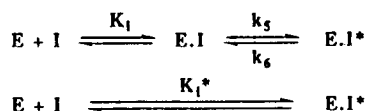


Slow-Binding Inhibition



Kinetic Constants of some Slow-Binding Inhibitors

Enzyme	Inhibitor	K _i (M)	K _i [*] (M)	k ₅ /k ₆	t _{1/2} (regain)
Acetolactate synthase	sulfometuron methyl	6.6×10^{-7}	6.5×10^{-8}	10	30 min
Dihydrofolate reductase	methotrexate	3.6×10^{-9}	1.4×10^{-11}	270	27 min
Pepsin	pepstatin	1.3×10^{-8}	4.5×10^{-11}	290	2.5 hr
Alanine racemase	Ala-P	1×10^{-3}	1×10^{-9}	10^6	20 days
D-Ala-D-Ala ligase	aminoalkyl phosphinate	1.4×10^{-4}	ND	ND	17 days



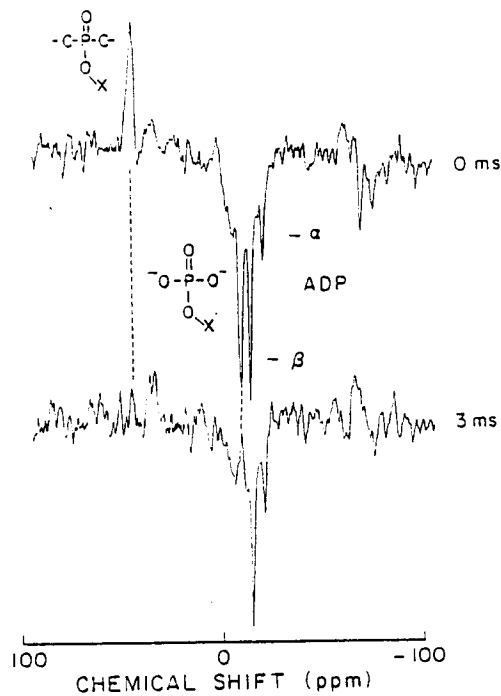
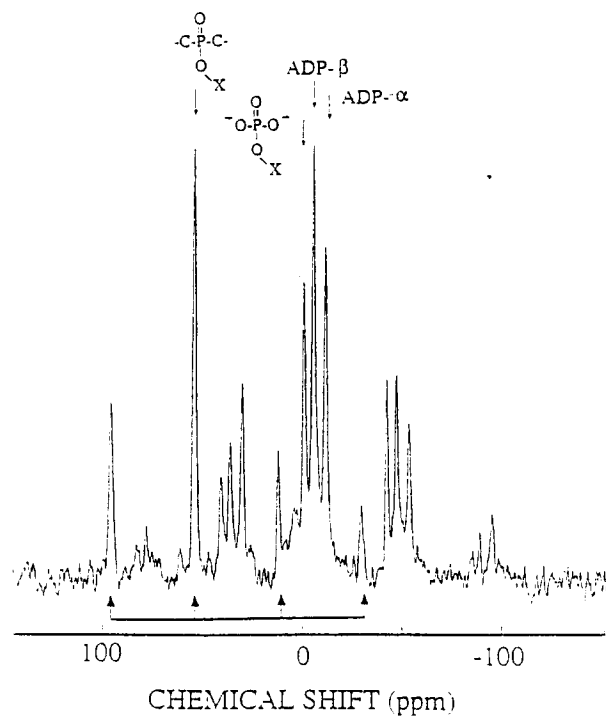
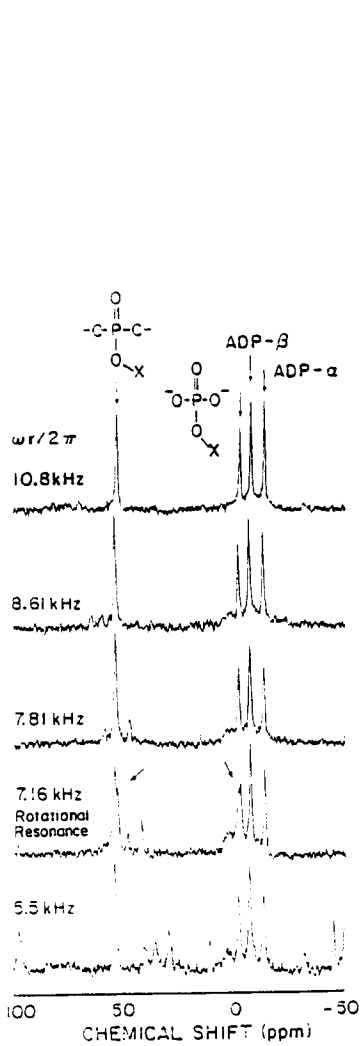
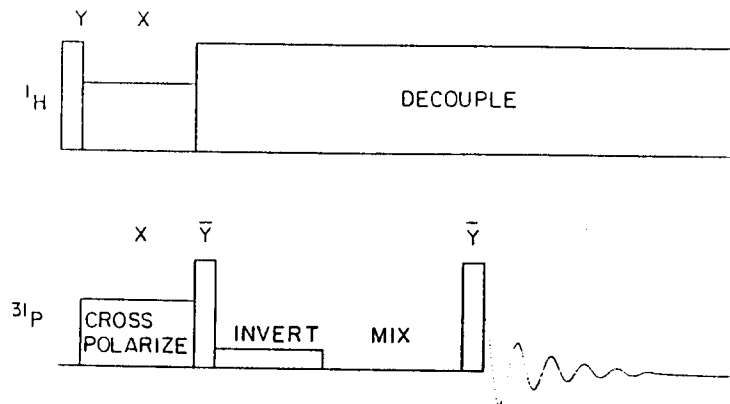
How to detect a labile, enzyme-bound phosphinophosphate?

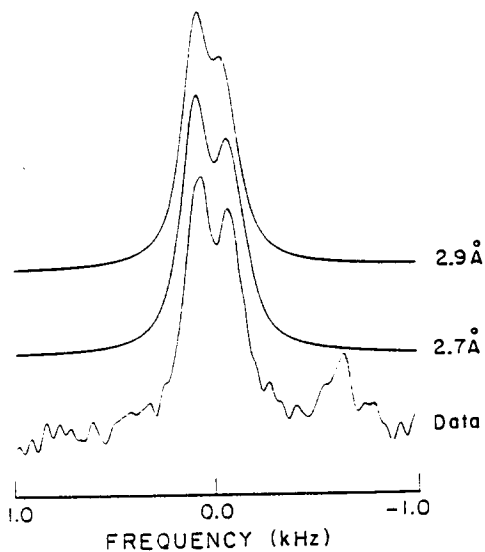
—————> NMR of enzyme-bound species

solution NMR —————> signals too broad to be useful

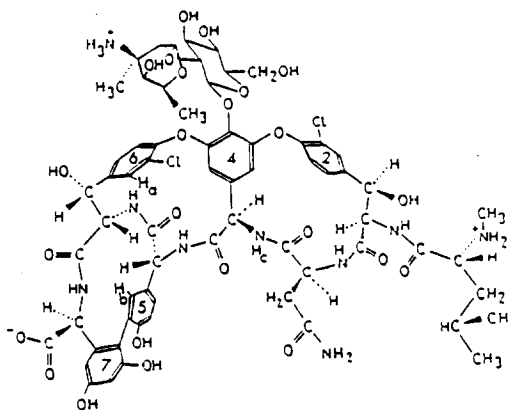
solid state NMR —————> CPMAS ³¹P NMR to get narrow lines
 (Cross Polarization Magic Angle Spinning NMR)

(with A. McDermott, R. Griffin, National Magnet Lab, M.I.T.)

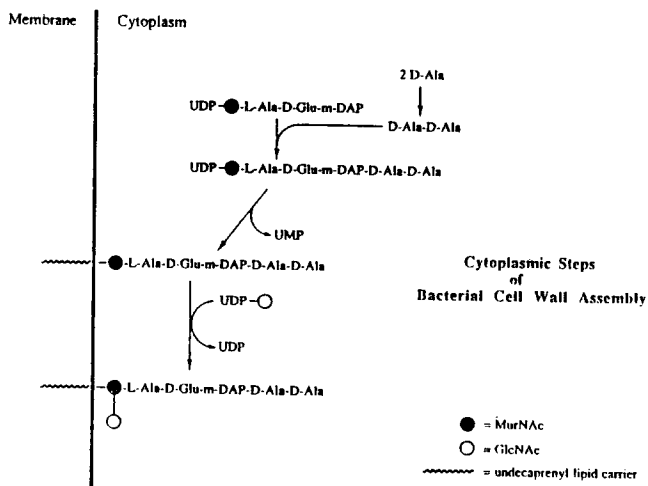




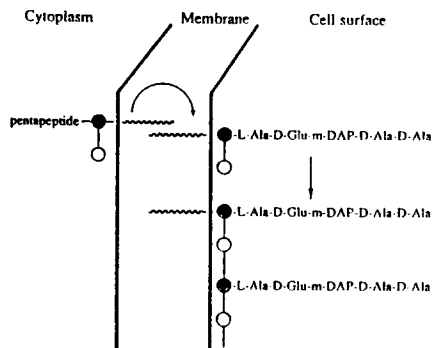
Vancomycin Group of Glycopeptide Antibiotics



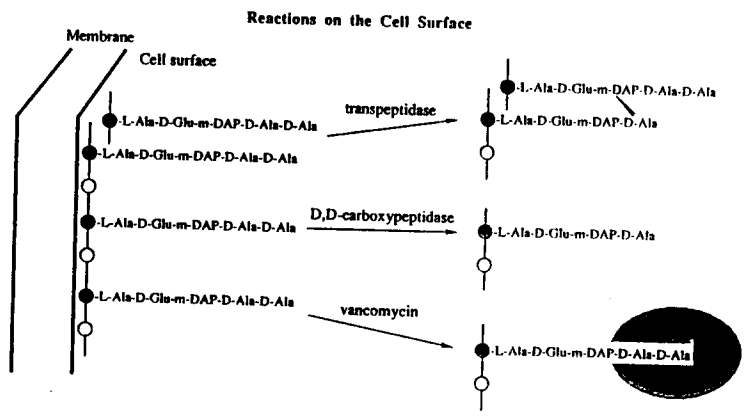
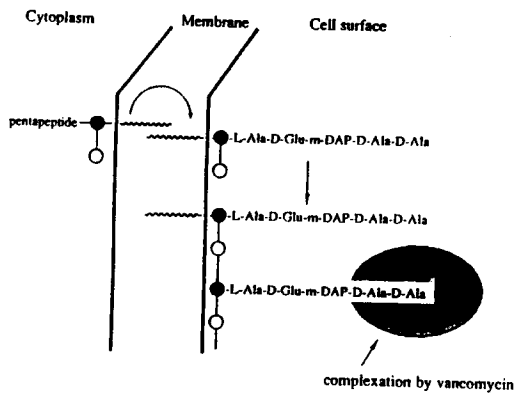
- Examples:** Vancomycin, ristocetin, teicoplanin
- Target:** Surface peptidyl chains terminating in D-Ala-D-Ala
- Mechanism:** Tight complexation of vancomycin to peptidyl-D-Ala-D-Ala termini, preventing both translocation and cross-linking
- Clinical use:** Life-threatening infections in hospital settings (e.g. multiple drug-resistant *Streptococci*, *Enterococci*)



Translocation of the Cell Wall Precursor Across Cell Membrane



Translocation of the Cell Wall Precursor Across Cell Membrane



Emerging Resistance to Vancomycin

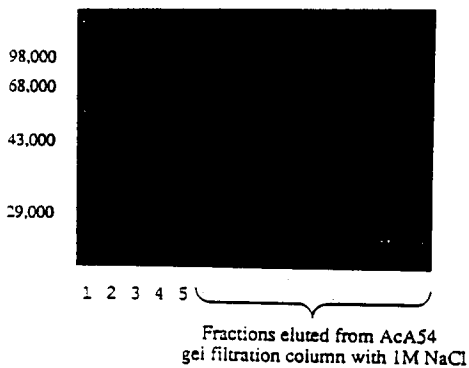
- 1986 Glycopeptide-resistant strains of *Enterococci* first isolated in France & U.K.
- 1988 Vancomycin resistance found to be plasmid-encoded. Origin of plasmids unknown.
- 1990 *VanA* gene subcloned from resistance plasmid, found to correlate with vancomycin resistance. *VanA* gene sequenced, revealed substantial homology to *ddl* genes.

Is VanA a D-Ala-D-Ala ligase?

% Amino Acid Sequence Homology Between VanA and D-Ala-D-Ala Ligases

	VanA	<i>S.typh.</i> DdlA	<i>E. coli</i> DdlB
Predicted MW	37,400	39,271	32,840
VanA	-	36%	28%
<i>S. typh.</i> DdlA	36%	-	33%
<i>E. coli</i> DdlB	28%	33%	-

Purification of VanA from JM105/pAT214
(with M. Arthur, P. Courvalin, Pasteur Inst., Paris)



- Lane 1: MW standards
- Lane 2: Crude extract
- Lane 3: Membrane fraction
- Lane 4: 25%-50% Ammonium sulphate fraction
- Lane 5: 50% Ammonium sulphate supernatant

Kinetic Parameters of VanA

		<i>E. coli</i> DdIA	<i>E. coli</i> DdIB	VanA
D-alanine	k_{cat}	444 min ⁻¹	1018 min ⁻¹	228 min ⁻¹
	K_1	5.7 μM	3.3 μM	3.4 mM
	K_2	0.55 mM	1.2 mM	38 mM
	k_{cat}/K_m	1.3×10^4 M ⁻¹ s ⁻¹	1.4×10^4 M ⁻¹ s ⁻¹	1.0×10^2 M ⁻¹ s ⁻¹
D-cycloserine	K_i	8.9 μM (C)	27 μM (C)	730 μM (C)
Aminoalkylphosphinate	K_i	130 μM (C)	28 μM (C)	430 μM (C)
	k_{inact}	2.4 min ⁻¹	2.3 min ⁻¹	4.9 min ⁻¹

Comparison of VanA and D-Ala-D-Ala Ligases

Similarities

Similar size (37,000 vs. 32,000/39,000)
 28-36% amino acid sequence homology
 VanA possesses D-Ala-D-Ala ligase activity
 VanA susceptible to inhibition by D-cycloserine and transition state analog for D-Ala-D-Ala ligase (of clinical use?)

Differences

VanA much more hydrophobic protein (stabilised by high salt conditions)
 VanA probably membrane-associated, whereas D-Ala-D-Ala ligases are cytoplasmic
 VanA has 1% catalytic efficiency of DdIA/B
 K_m , K_i values for VanA 10-1000 fold higher than for D-Ala-D-Ala ligases

Dilemma

How might VanA as a "low efficiency", membrane associated D-Ala-D-Ala ligase impart vancomycin resistance?

Acknowledgements

D-Ala-D-Ala Ligase

L. Zawadzke
 K. Duncan
 T. Bugg

A. Patchett (Merck)
 A. McDermott, R. Griffin (MIT)
 L. Mullins, F. Rauschel (Texas)

D-Ala-D-Ala Adding Enzyme

K. Duncan
 T. Bugg

J. van Heijenoort (Paris)

VanA

T. Bugg

M. Arthur, P. Courvalin
 (Pasteur Inst., Paris)

National Science Foundation