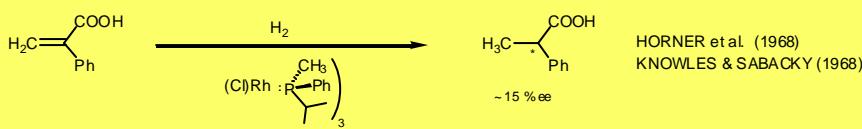


Iridium-Catalyzed Asymmetric Hydrogenation

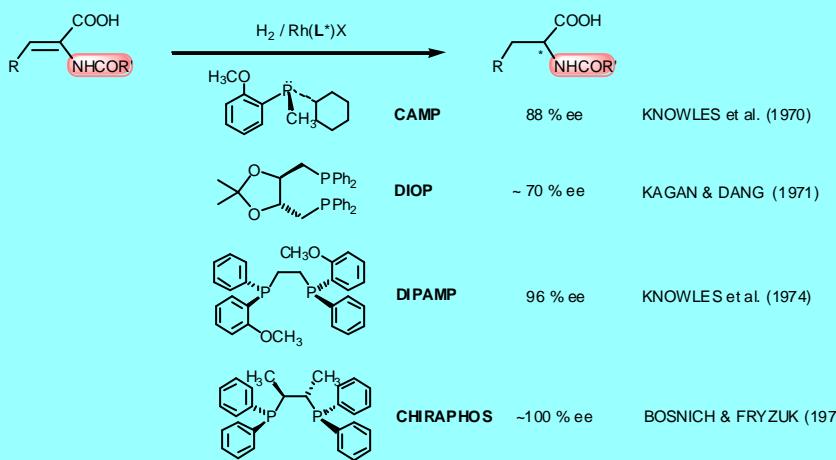
Andreas Pfaltz

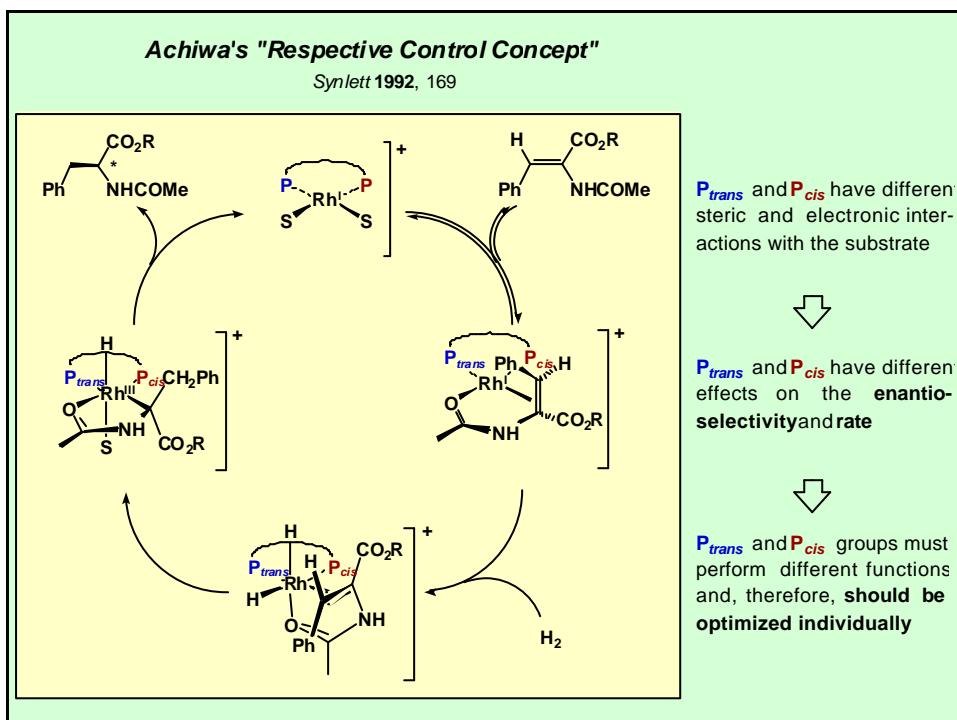
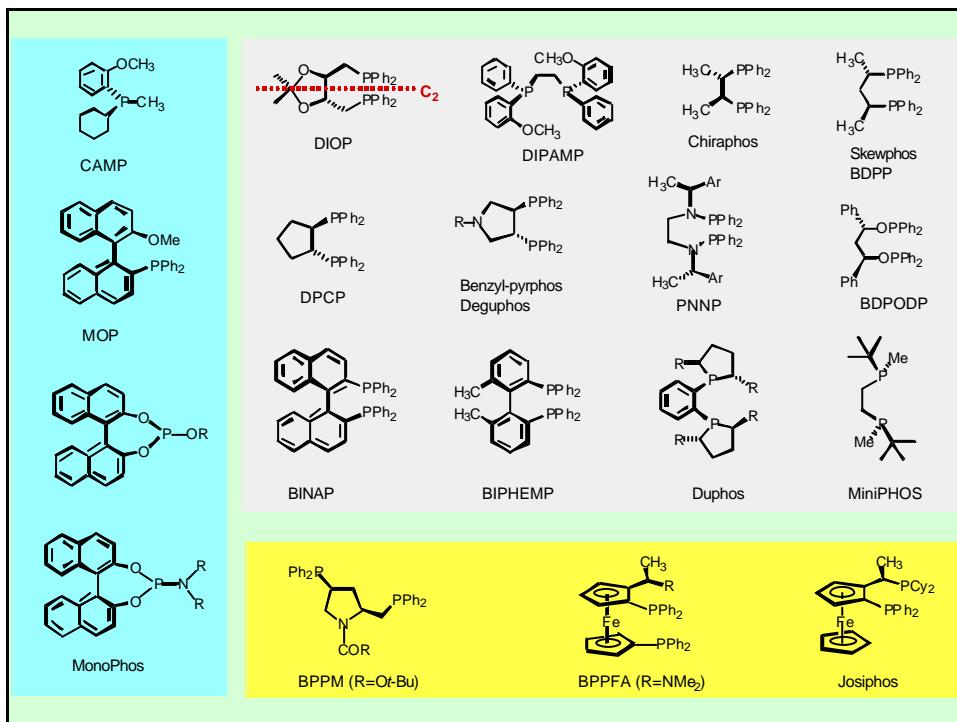
Department of Chemistry , University of Basel
St. Johanns-Ring 19, CH-4056 Basel
Switzerland
E-mail: andreas.pfaltz@unibas.ch

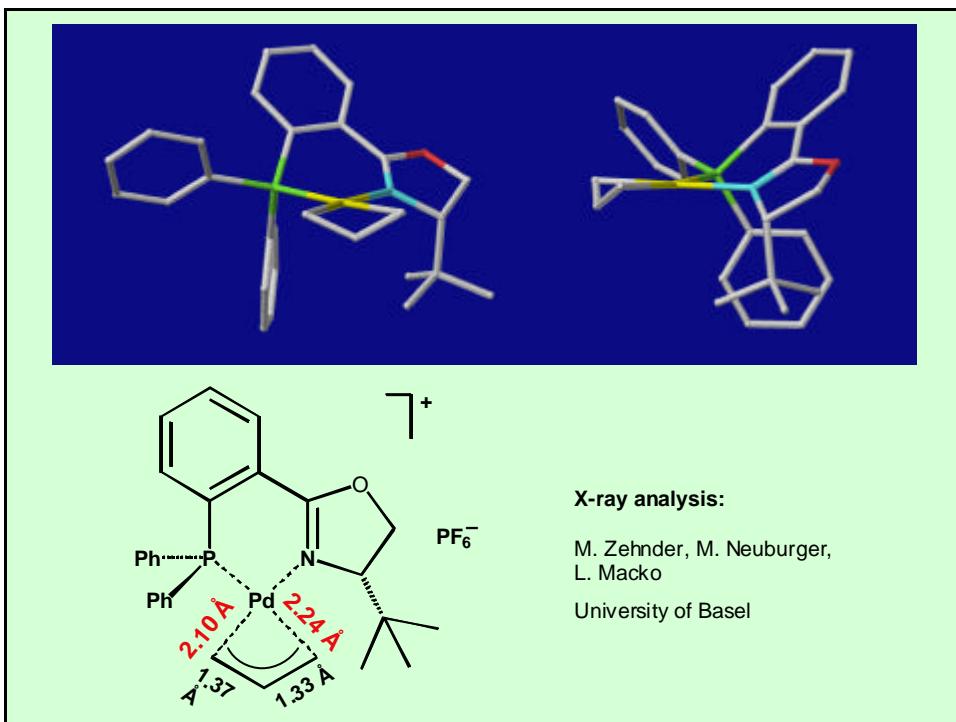
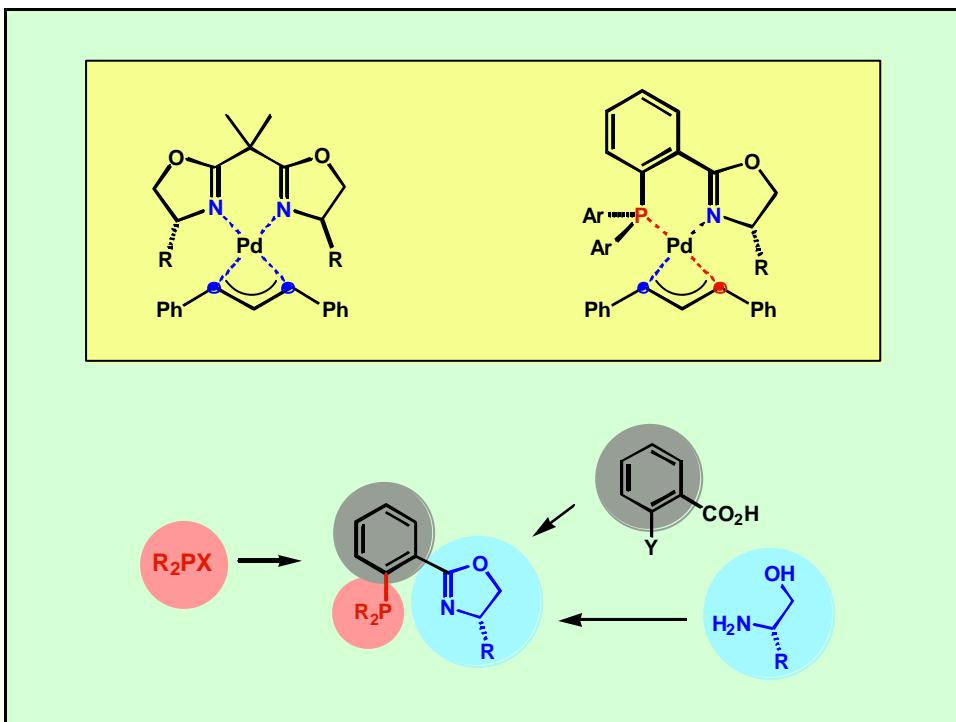
ENANTIOSELECTIVE HYDROGENATION

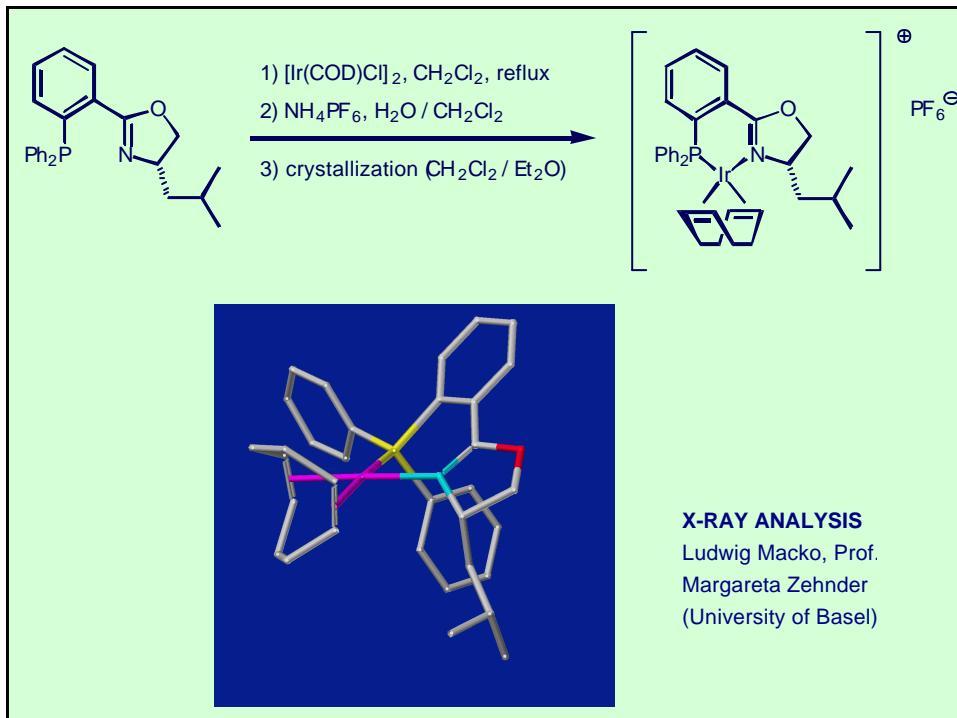
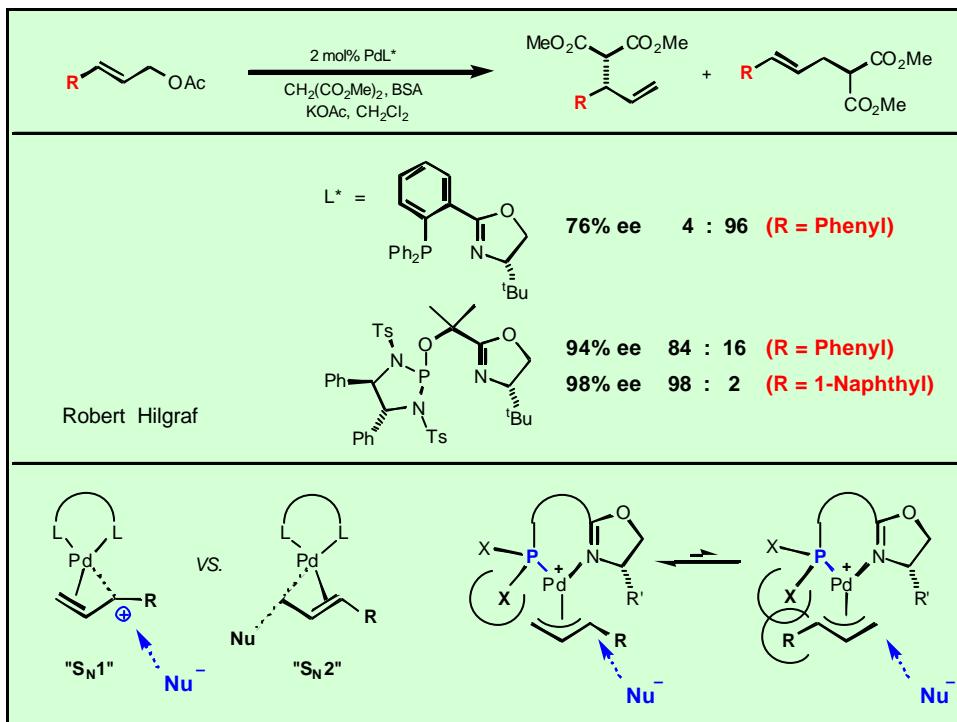


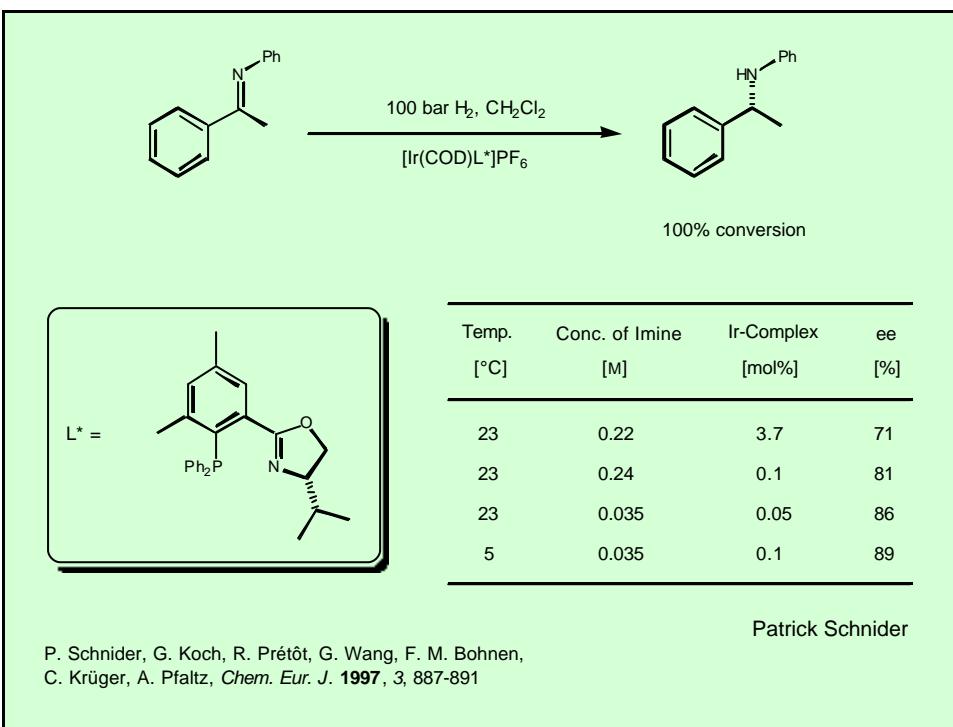
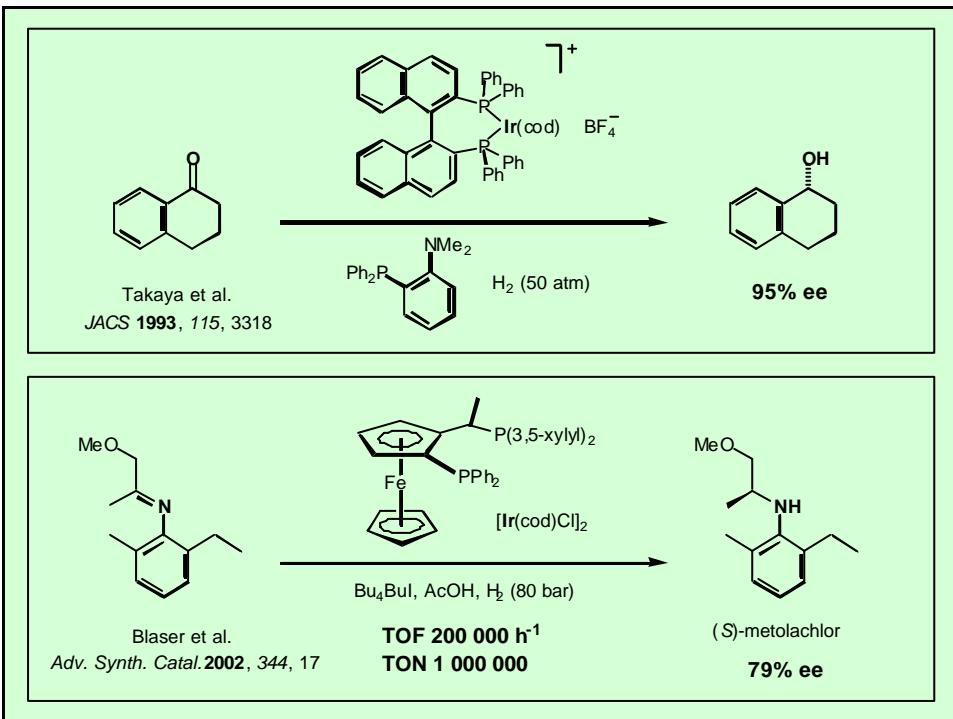
HORNER et al. (1968)
KNOWLES & SABAICKY (1968)





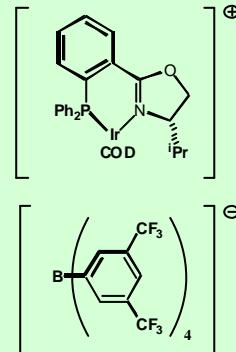
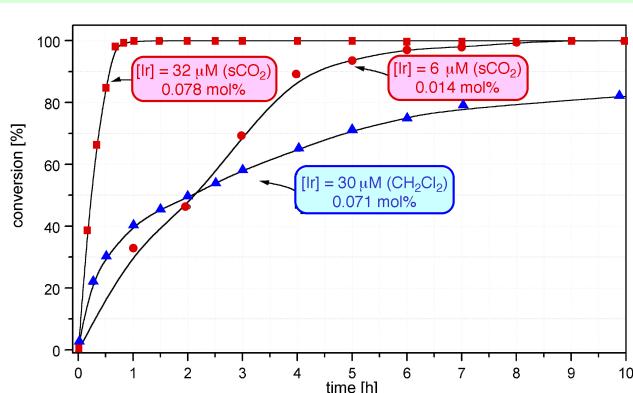
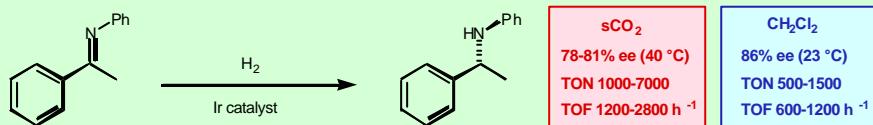






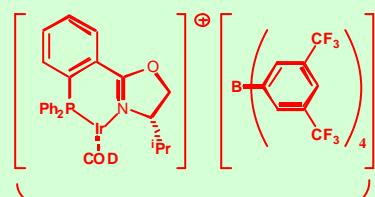
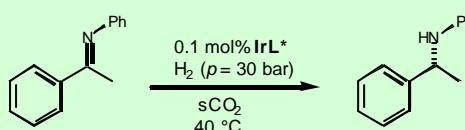
HYDROGENATION IN SUPERCRITICAL CO₂

2

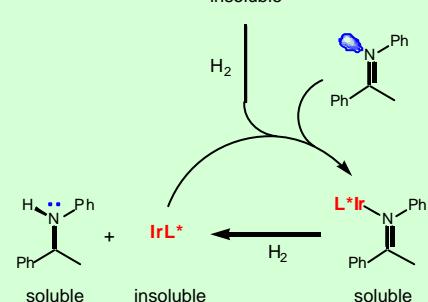


Sabine Kainz, Axel Brinkmann, Walter Leitner (Max-Planck-Institut für Kohlenforschung)

CATALYST RECYCLING IN SUPERCRITICAL CO₂

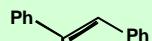
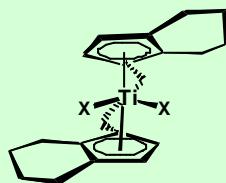


Cycle	Time for full conv.	%ee
1	1 h	79
2	1 h	80
3	1 h	76
4	2 h	75
5	20 h	72
6	40 h	74
7	65 h	70



Unfunctionalized Olefins

Buchwald

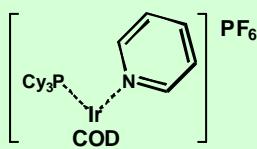


94% yield
99% ee

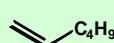
High yield and ee
5-8 mol% catalyst
low TOF ($1\text{-}2 \text{ h}^{-1}$)

Broene & Buchwald, JACS
1993, 115, 12569

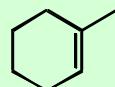
Crabtree



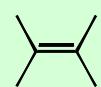
TOF (h^{-1})



6400

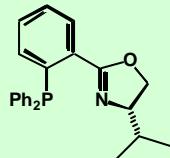
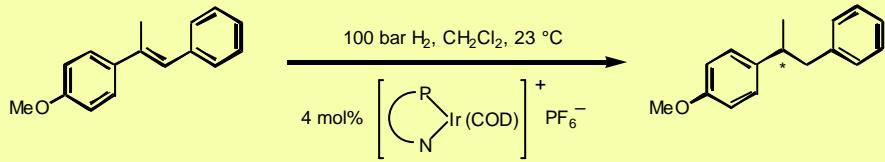


3800

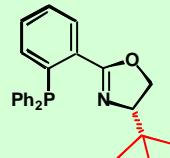


4000

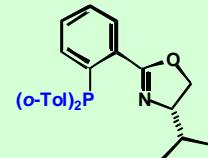
Ligand Variation



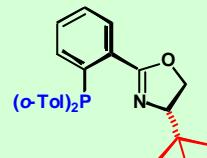
78% conv.
75% ee



98% conv.
90% ee



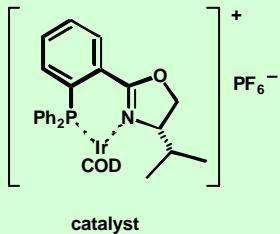
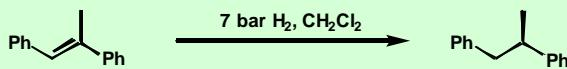
100% conv.
91% ee



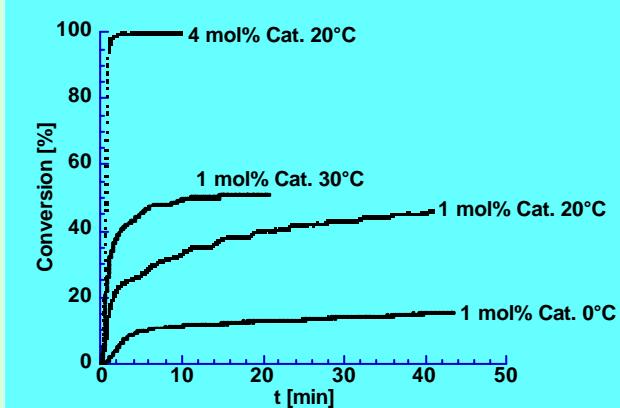
72% conv.
97% ee

PATRICK SCHNIDER, ANDREW LIGHTFOOT

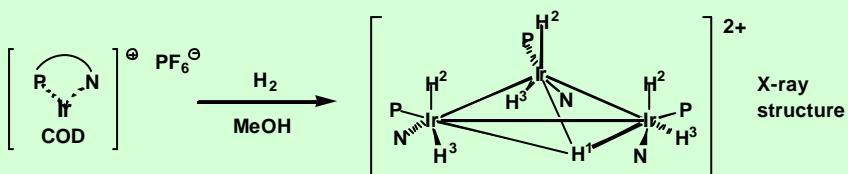
Kinetic Studies



Prof. Donna G. Blackmond
Thorsten Rosner
(MPI für Kohlenforschung)



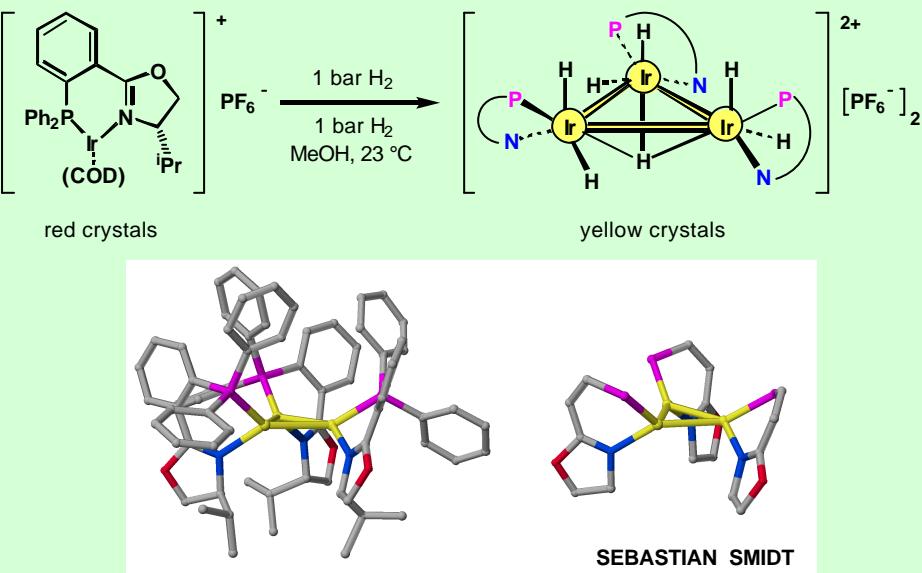
Hydride-Bridged Trimer



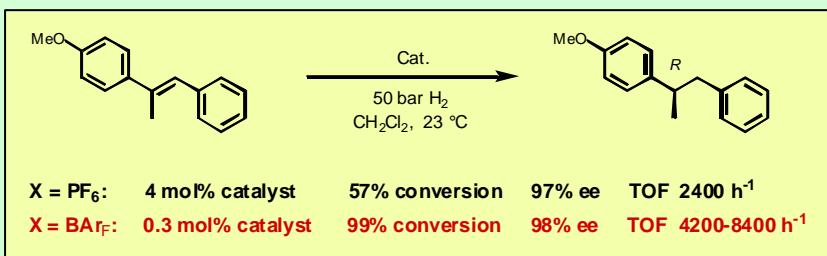
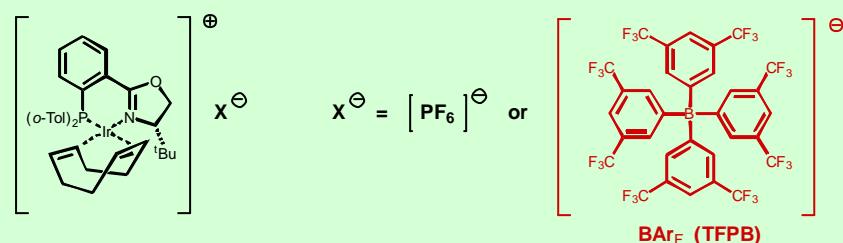
L. Pignolet et al.,
Inorg. Chem.
1988, 325

P N	¹ H-NMR
	- 2.0 (<i>q</i> , <i>J</i> _{PH} , <i>trans</i> = 47 Hz, 1 H; ¹ H) - 18.7 (<i>d</i> , <i>J</i> _{PH} , <i>cis</i> = 12 Hz, 3 H; ² H or ³ H) - 22.5 (<i>d</i> , <i>J</i> _{PH} , <i>cis</i> = 26 Hz, 3 H; ³ H or ² H)
	- 8.0 (<i>q</i> , <i>J</i> _{PH} = 47 Hz, ¹ H) - 16.6 (<i>br. s</i> , 3H) - 20.5 (<i>d</i> , <i>J</i> _{PH} = 21 Hz, 3H)

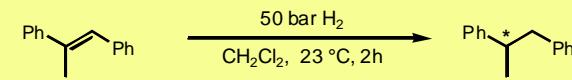
Preparation and X-ray analysis of the hydride-bridged trimer $[\{(\text{Ir}(\text{PHOX})\text{H})_3\}_3\text{H}] [\text{PF}_6]_2$



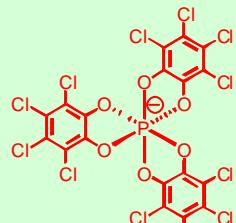
EFFECT OF THE COUNTER ION



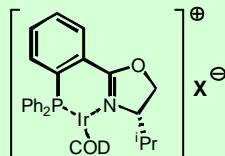
ANDREW LIGHTFOOT



D-TRISPHAT:



Jérôme Lacour
(University of Geneva)

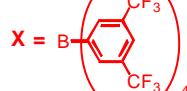
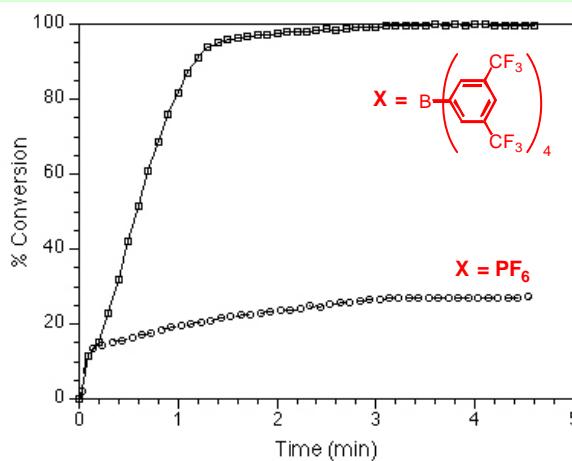
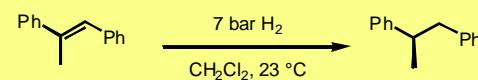


X = BArF:
(1 mol% cat.) 70% ee
100% conv.

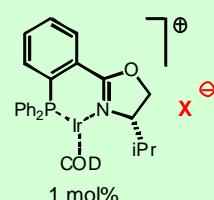
X = D-TRISPHAT:
(4 mol% cat.) 70% ee
100% conv.

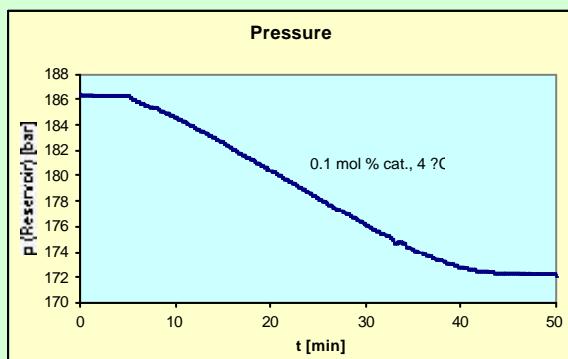
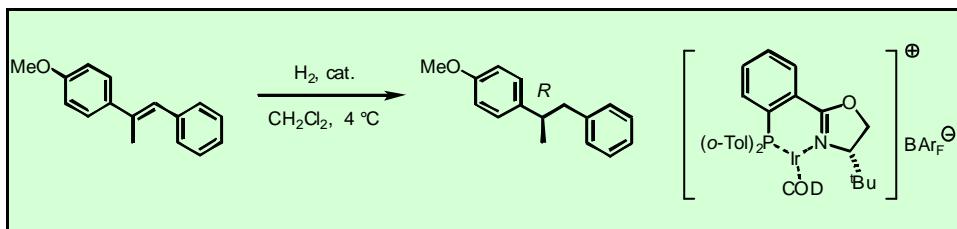
X = D-TRISPHAT:
(4 mol% cat.) 0% ee
100% conv.

X = D-TRISPHAT:
(1 mol% cat.) 0% ee
70% conv.



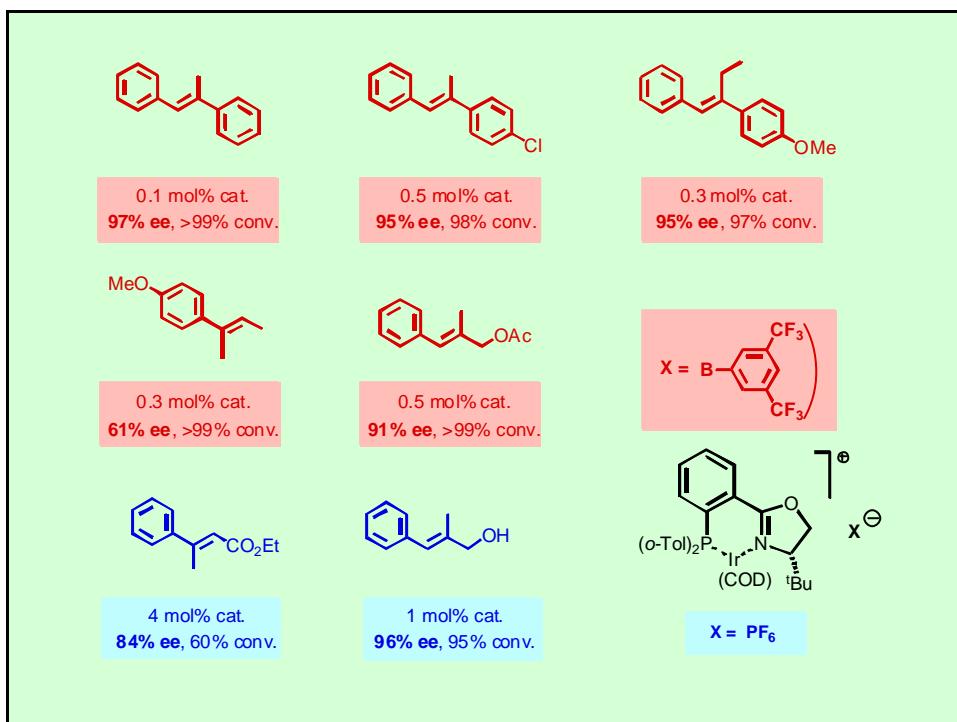
X = PF6-

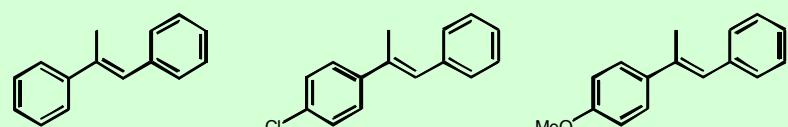




- ◆ 4 °C: up to 5000 TON
- ◆ 1st order in [H₂]
- 5-30 bar (97-98% ee)
- ◆ 0 order in [olefin]
(at high conc. ~0.2)
- ◆ 1st order in [cat] at low conc.
0.3 at high conc.
- ◆ 23 °C: rate diffusion-limited

Nicole Zimmermann, Dr. Martin Studer (Solvionic)





1 97 % ee

2 99 % ee

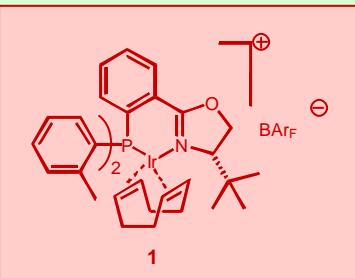
95 % ee

99 % ee

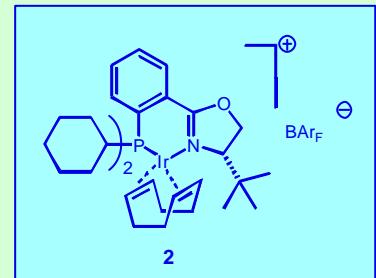
MeO

98 % ee

99 % ee



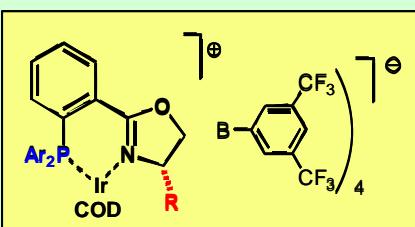
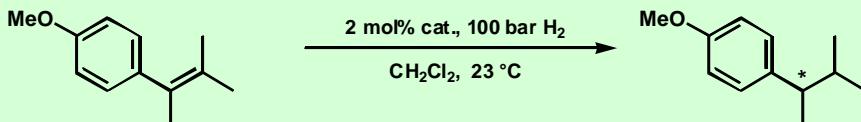
1



2

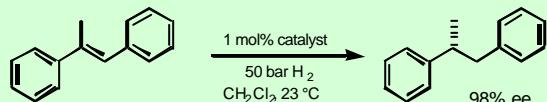
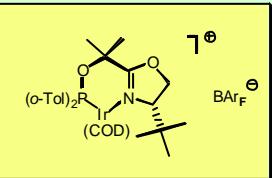
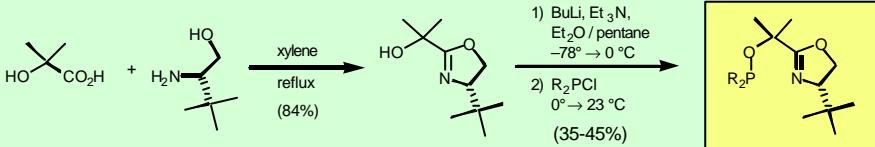
NICOLE ZIMMERMANN

TETRA SUBSTITUTED OLEFINS

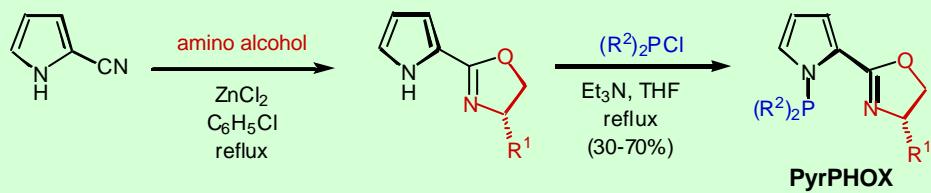
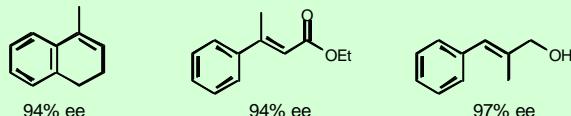


Ar	R	Conv.	ee
o-Tol	t Bu	37%	3% ee
o-Tol	i Pr	97%	8% ee
Ph	t Bu	23%	31% ee
Ph	i Pr	>99%	58% ee
Ph	CH ₂ t Bu	>99%	81% ee

ANDREW LIGHTFOOT



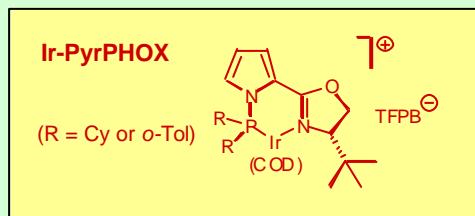
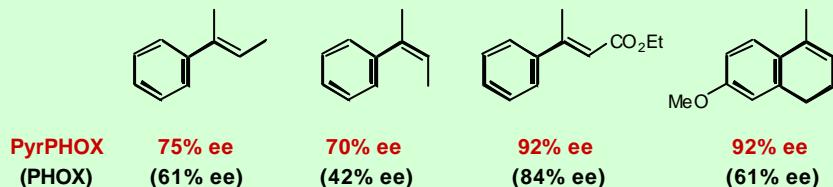
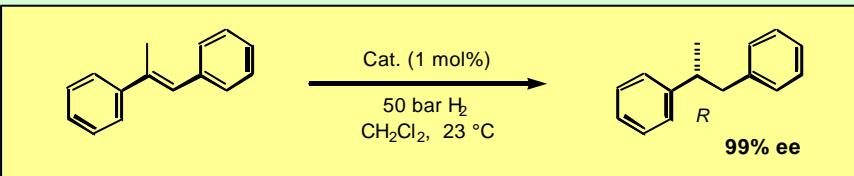
Sebastian Smidt



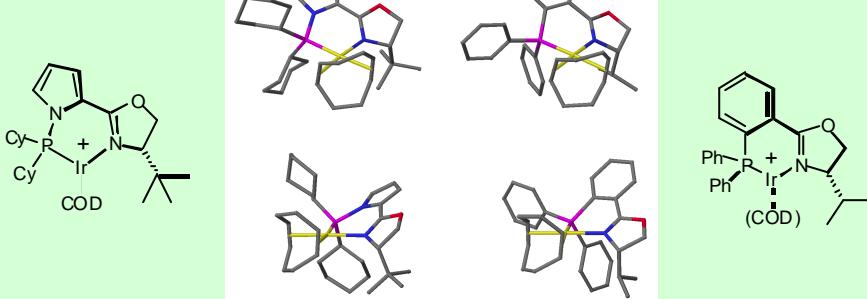
(H. Brunner, B. Haßler, *Z. Naturforsch.* **1998**, *53b*, 476)

- $R^1 = i\text{-Pr}; R^2 = \text{Ph}$
- $R^1 = t\text{-Bu}; R^2 = \text{Ph}$
- $R^1 = t\text{-Bu}; R^2 = o\text{-Tolyl}$
- $R^1 = t\text{-Bu}; R^2 = \text{Cyclohexyl}$

Prof. Pier Giorgio Cozzi
(University of Bologna)

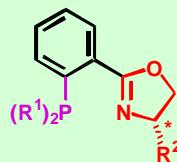


P. G. Cozzi
Nicole Zimmerman

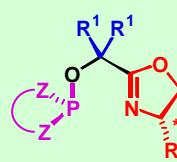


X-ray: Dr. Silvia Schaffner, Prof. Margaret Neuburger-Zehnder (University of Basel)

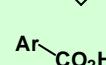
FURTHER VARIATION OF THE PHOSPHINOXAZOLINE STRUCTURE



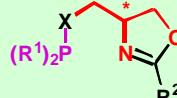
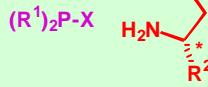
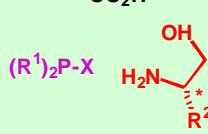
PHOX



Z = aryl, alkyl, O, NR

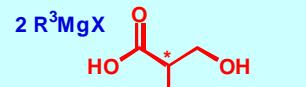
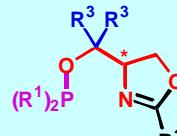


↓

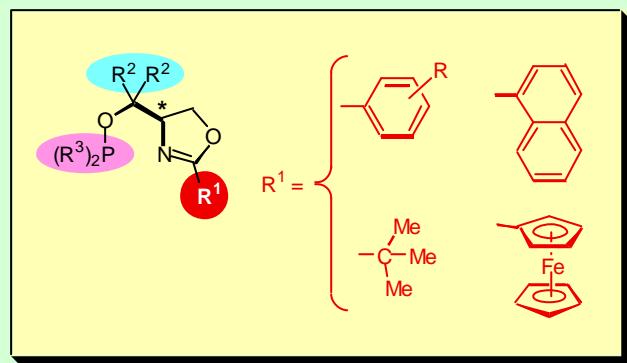
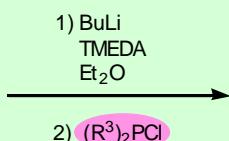
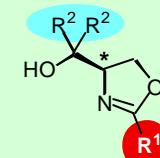
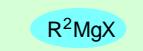
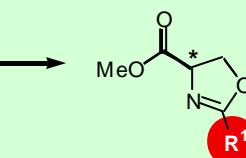
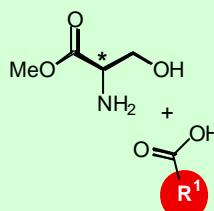


X = CH₂: Burgess,
JOC 2001, 66, 206

X = O: Richards,
TL 2001, 42, 5553

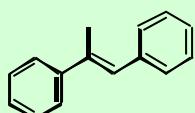


JÖRG BLANKENSTEIN

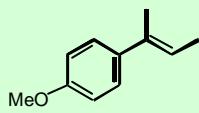


JÖRG BLANKENSTEIN

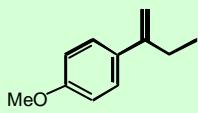
Iridium-Catalyzed Hydrogenation



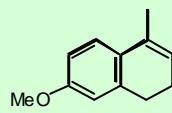
97% ee



96% ee



80-90% ee



82% ee

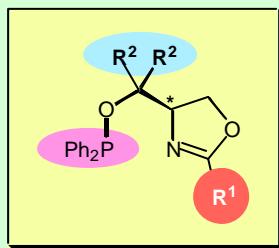
$R^1 = \text{Ph, ferrocenyl}$
 $R^2 = \text{isopropyl}$

$R^1 = 3,5-(\text{Bu})_2\text{Ph}$
 $R^2 = \text{benzyl}$

$R^1 = \text{ferrocenyl}$
 $R^2 = \text{isobutyl}$

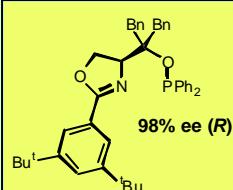
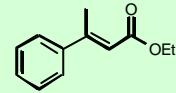
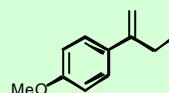
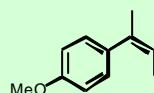
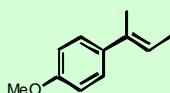
$R^1 = \text{ferrocenyl}$
 $R^2 = \text{benzyl}$

JÖRG BLANKENSTEIN



Conditions:

0.1-0.3 mol% $[\text{Ir}(\text{L}^*)(\text{COD})]\text{BArF}$
50 bar H_2 , toluene or CH_2Cl_2 , RT

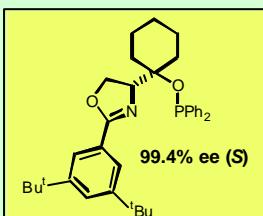


98% ee (*R*)

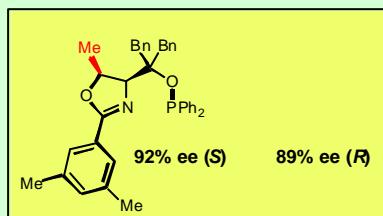
74% ee (*S*)

89% ee (*R*)

55% ee (*R*)

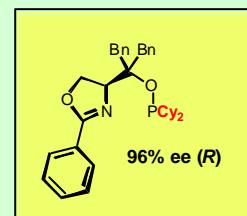


99.4% ee (*S*)



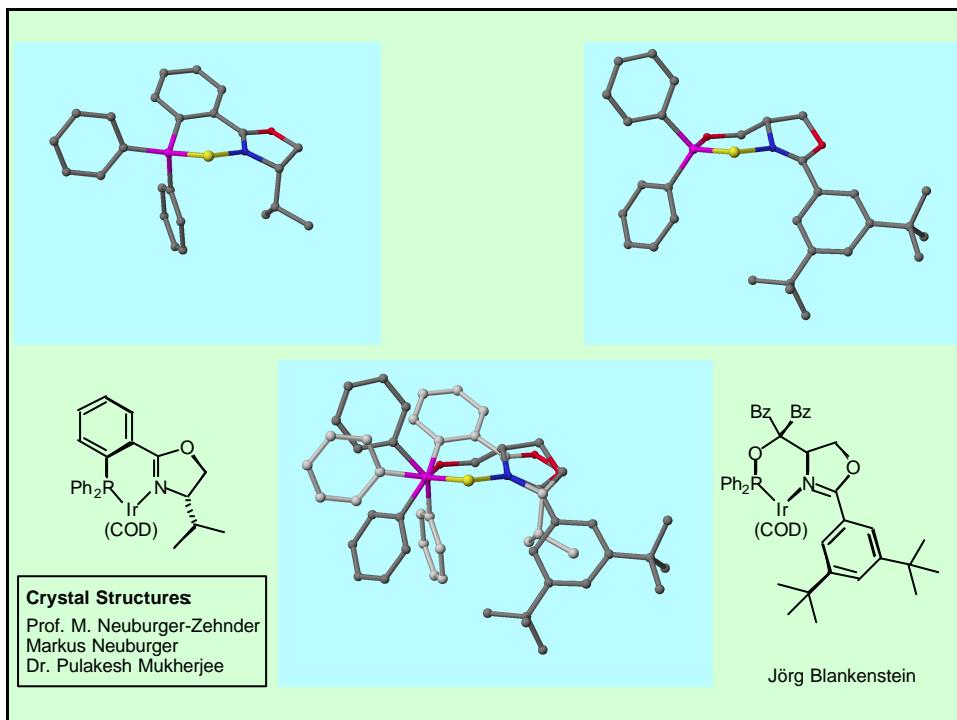
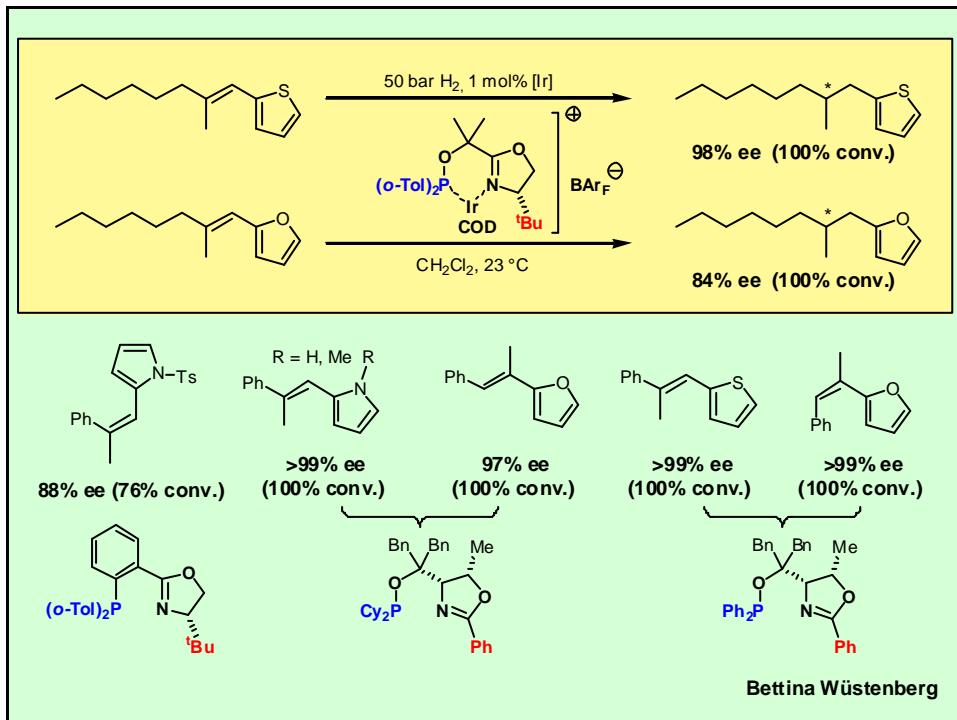
92% ee (*S*)

89% ee (*R*)

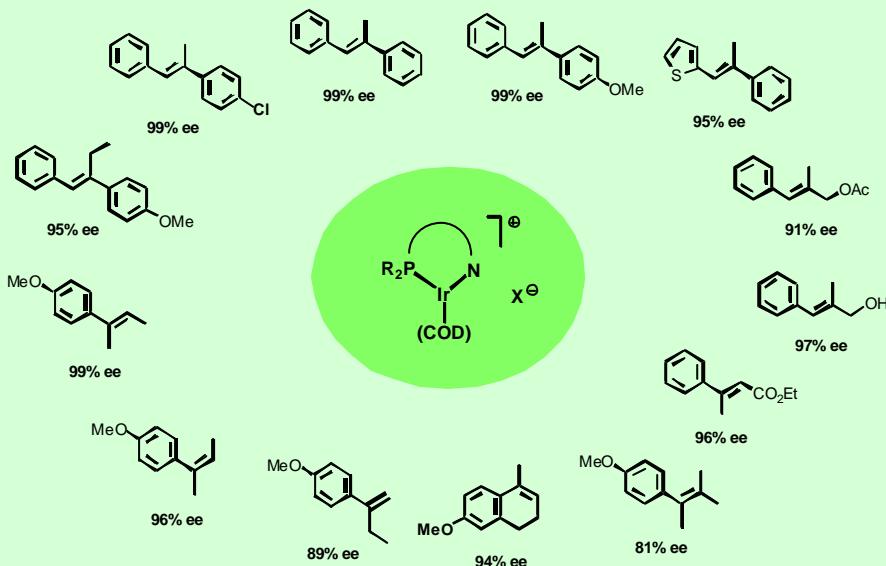


96% ee (*R*)

Frederik Menges, Jörg Blankenstein, Steve McIntyre



IRIDIUM-CATALYZED HYDROGENATION OF C=C BONDS



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