

New Catalytic Asymmetric Reactions

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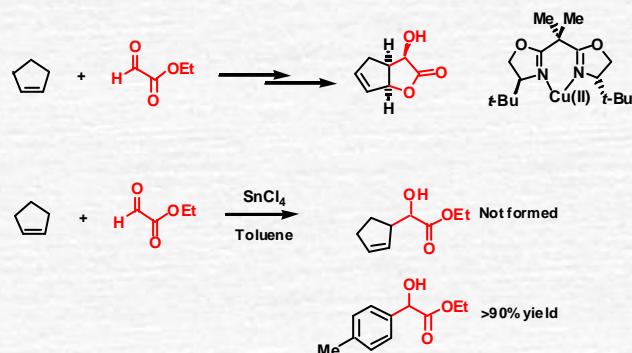


When something goes wrong



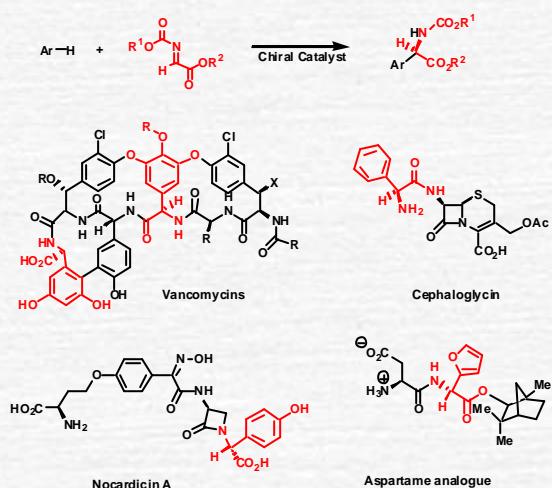
Goya: *Mother showing her deformed child to two women*
Louvre, Paris

A failed reaction



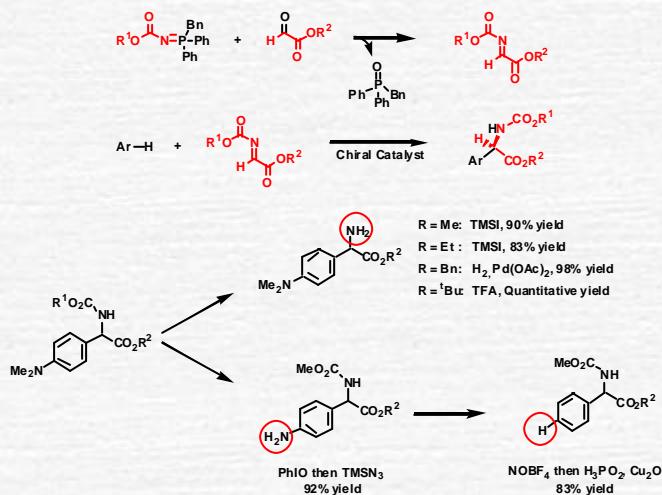
Chem. Commun. 1999, 1869

Catalytic asymmetric aza-Friedel-Crafts reactions



Optically active aromatic α -amino acids

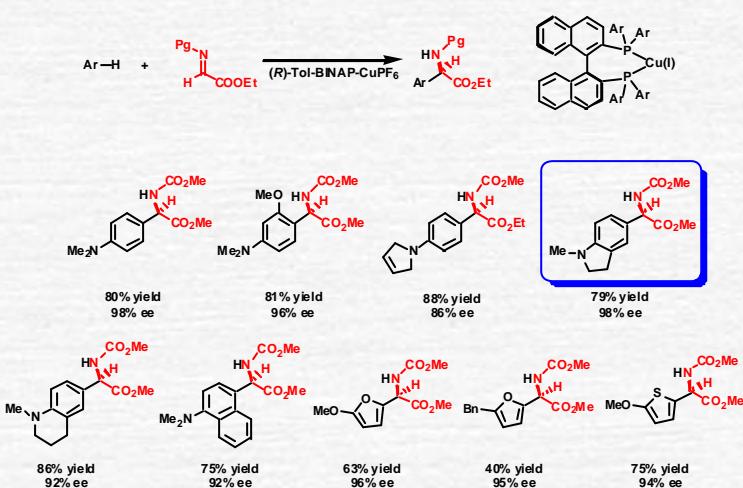
5



Angew. Chem. Int. Ed. 2000, 39, 4114
J. Org. Chem. 2002, 67, 4352

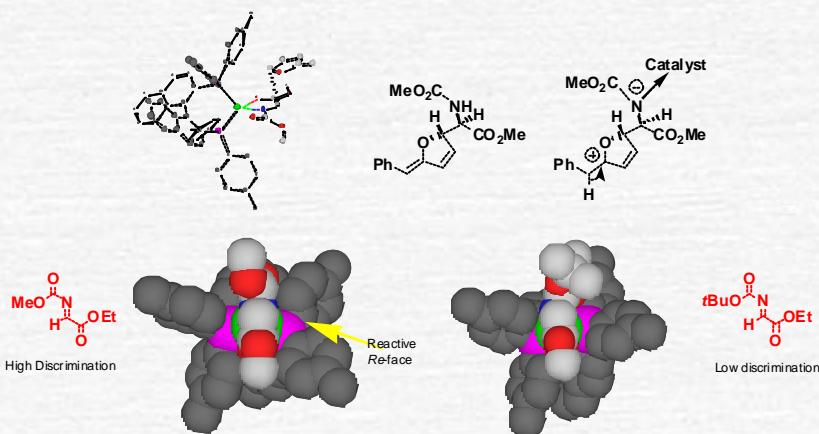
Scope and potential

6



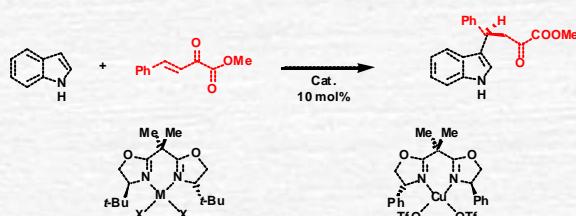
Mechanistic Considerations

7



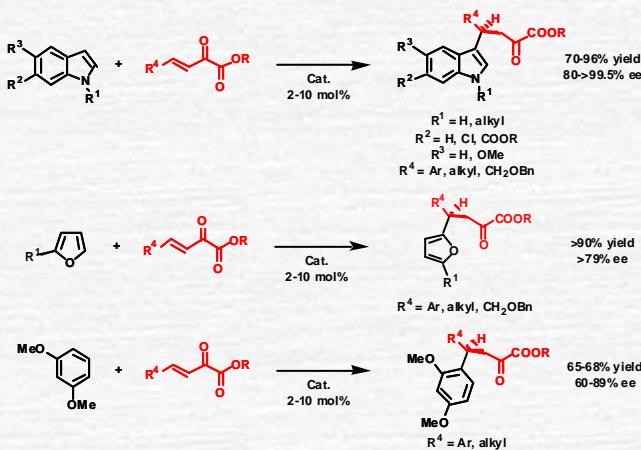
Catalytic asymmetric Friedel-Crafts alkylation reactions

8



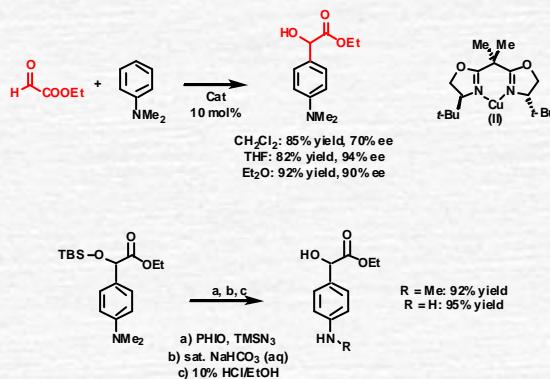
M = Cu(II), X = OTf (CH₂Cl₂): 100% conversion, 88% ee
M = Cu(II), X = OTf (Et₂O): 100% conversion, 97% ee
M = Zn(II), X = OTf (CH₂Cl₂): 100% conversion, 67% ee

100% conversion, 42% ee

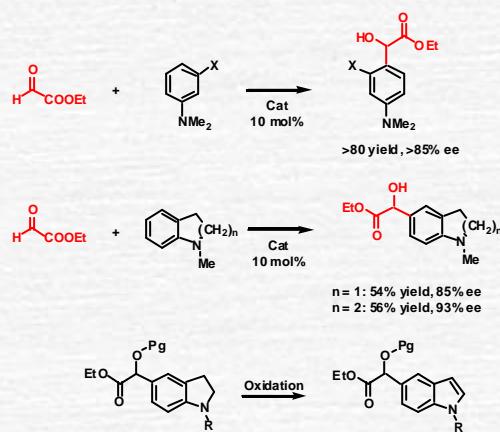


Catalytic enantioselective Friedel-Crafts reactions

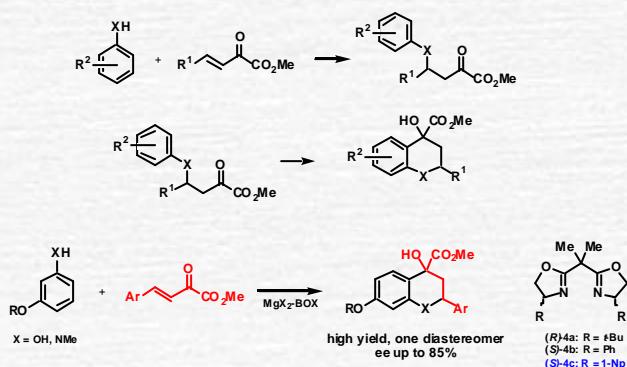
a simple approach to optically active α -aromatic α -hydroxy carboxylic acids



Scope and potential

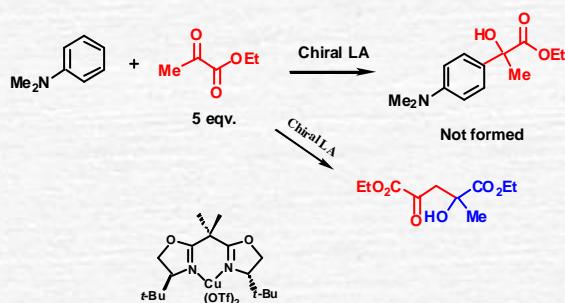


Catalytic asymmetric tandem reactions – formation of optically active dihydro-benzopyrans

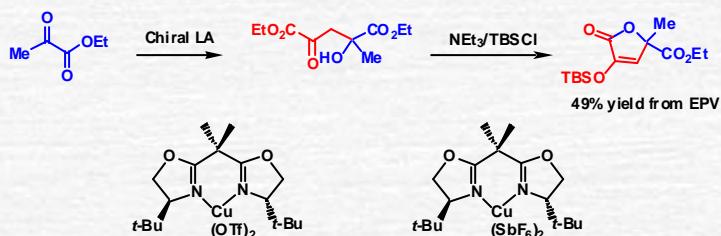


Submitted 2002

Discovery of a direct aldol reaction



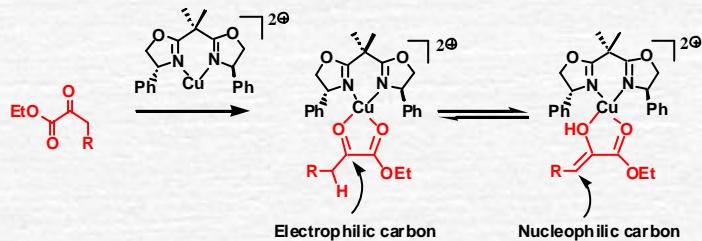
A catalytic asymmetric homo-aldol reaction



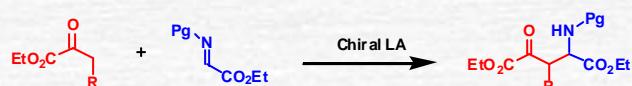
Amine [mol%]	Ee [%]
-	(-) -65
PhNMe ₂ /10	(-) -79
PhNMe ₂ /5	(-) -96
PhNBn ₂ /10	(-) -93
Et(<i>i</i> Pr) ₂ N/10	(-) -67

Amine [mol%]	Solvent	Ee [%]
-	Et ₂ O	(-) -50
PhNMe ₂ /10	CH ₂ Cl ₂	(+) -63
PhNBn ₂ /10	CH ₂ Cl ₂	(+) -75
CyNMe ₂ /10	CH ₂ Cl ₂	(+) -77

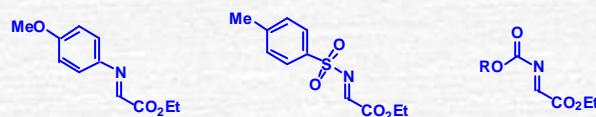
The concept



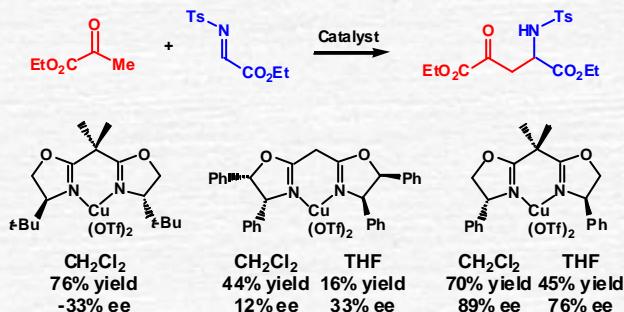
Catalytic asymmetric direct Mannich reactions



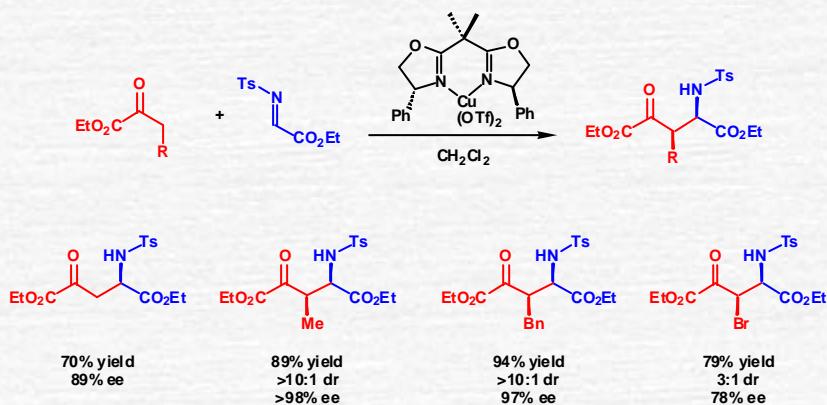
Imine candidates



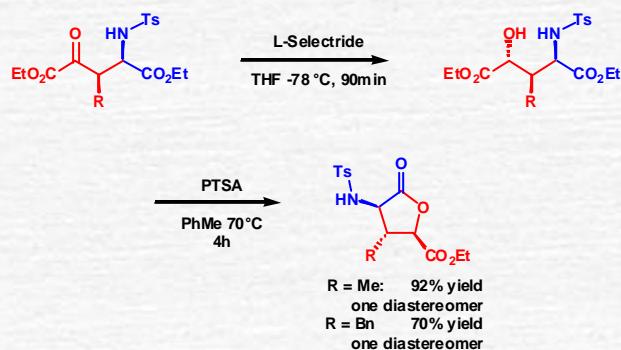
Optimisation of the direct Mannich reaction



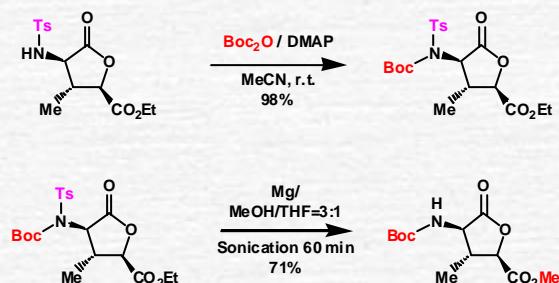
Scope of the direct Mannich reaction



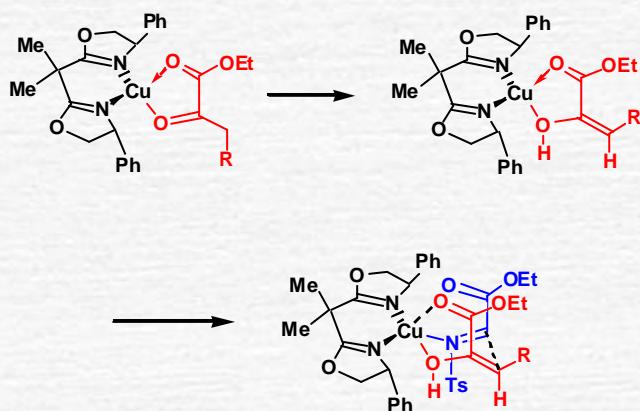
Reduction of the Mannich product



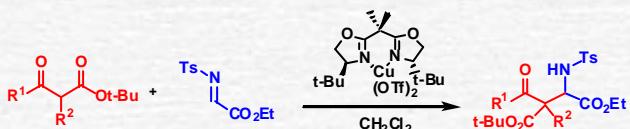
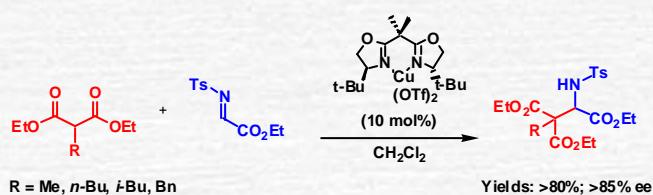
Detosylation



Proposed Mechanism

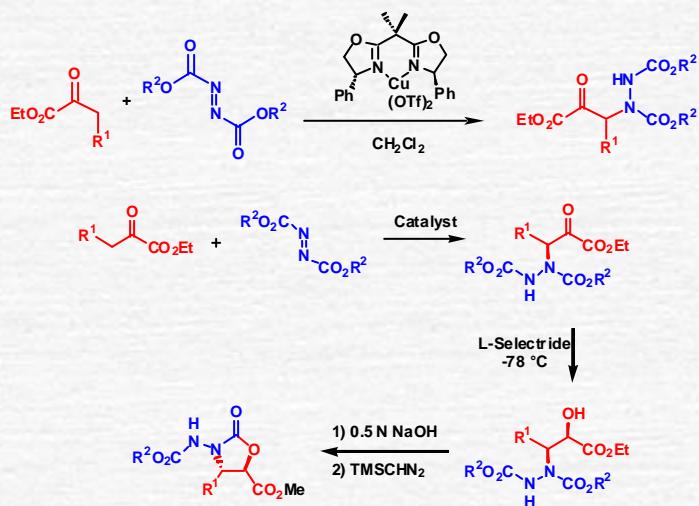


Malonic esters and β -keto esters as nucleophiles in direct Mannich reactions



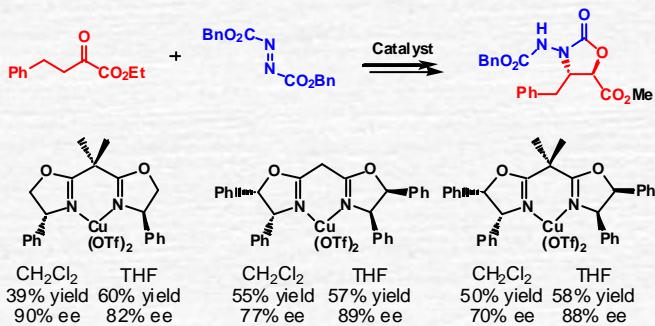
$\text{R}^1 = \text{R}^2 = \text{Me}$: 87%, 93:7 dr, 88% ee
 $\text{R}^1 = \text{Et}, \text{R}^2 = \text{Me}$: 80%, 98:2 dr, 92% ee
 $\text{R}^1 = \text{i-Pr}, \text{R}^2 = \text{Me}$: 55%, 84:16 dr, 91% ee
 $\text{R}^1 = \text{Bn}, \text{R}^2 = \text{Me}$: 83%, 96:4 dr, 93% ee
 $\text{R}^1 = \text{Me}, \text{R}^2 = \text{allyl}$: 69%, 92:8 dr, 87% ee
 $\text{R}^1 = \text{R}^2 = (\text{CH}_2)_4$: 89%, 99:1 dr, 86% ee

Catalytic asymmetric direct α -amination

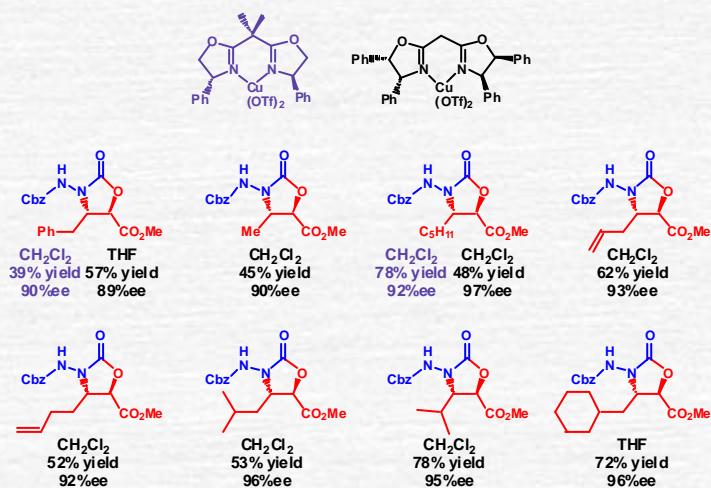


J. Am. Chem. Soc. **2002**, 124, 2420

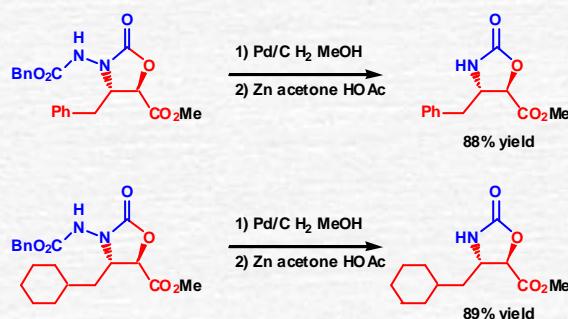
Optimisation of the direct α -amination reaction



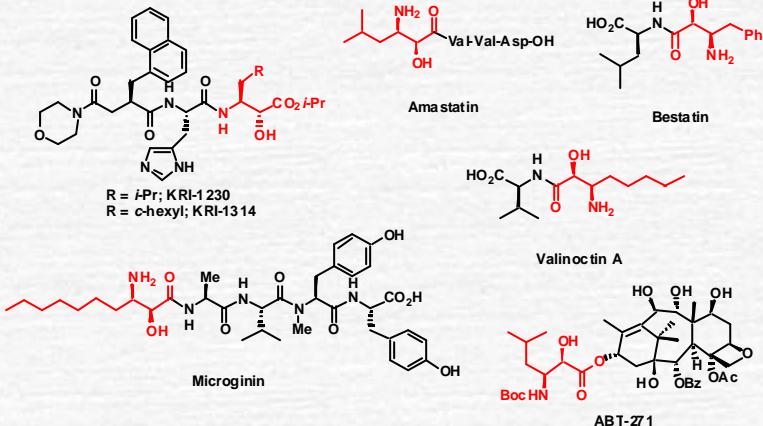
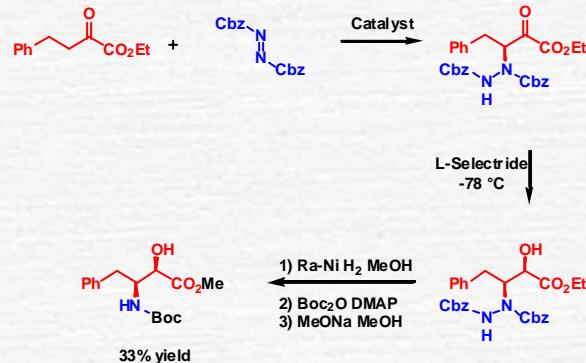
Scope of the direct α -amination reaction



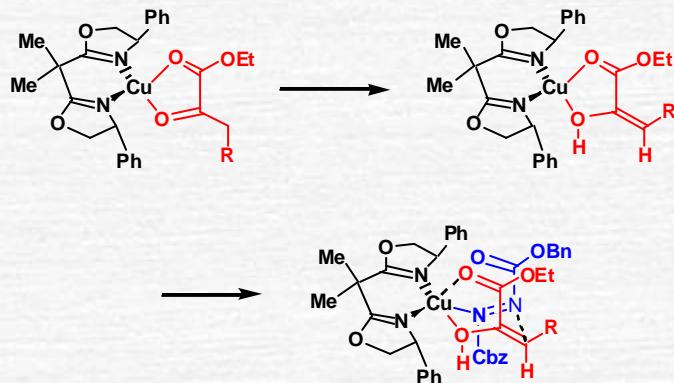
Synthesis of oxazolidinones



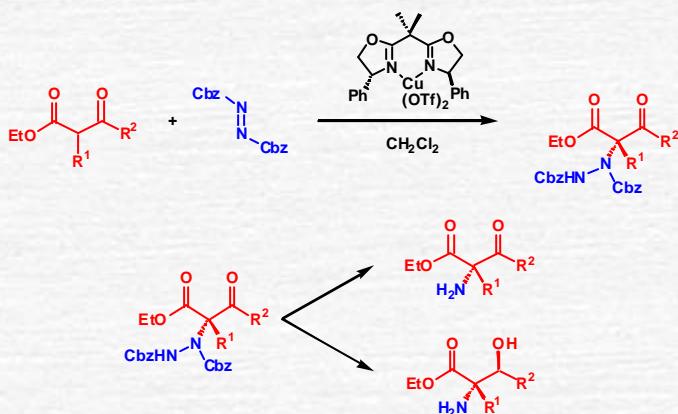
Synthesis of *syn*- α -hydroxy- β -amino acids



Proposed mechanism

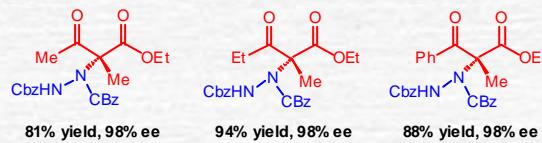


Catalytic asymmetric direct α -amination of β -keto esters – optically active β -keto- and β -hydroxy- α -amino acids



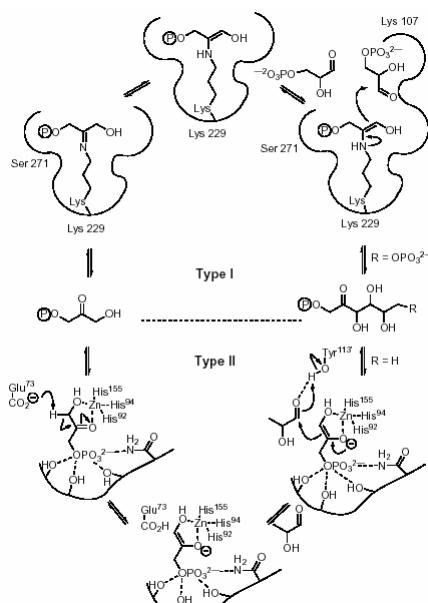
Submitted for publication

Scope

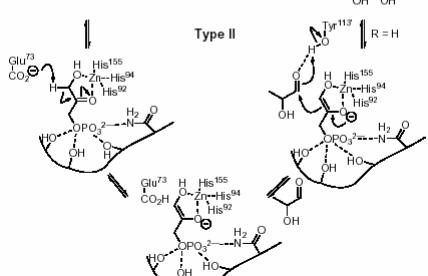


n = 1: 99% yield, 99% ee
n = 2: 94% yield, 98.5% ee
n = 2: 92% yield, 98% ee

Organic Catalytic Aldol Reaction: Type I Aldolase

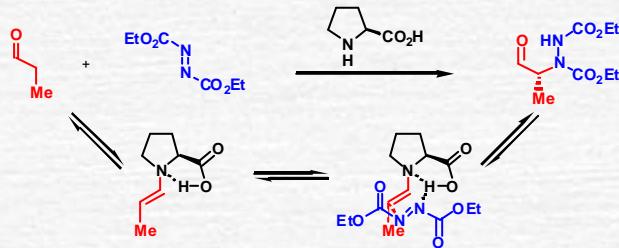


Metal Catalysed Aldol Reaction: Type II Aldolase



L-Proline catalysed asymmetric α -amination of aldehydes

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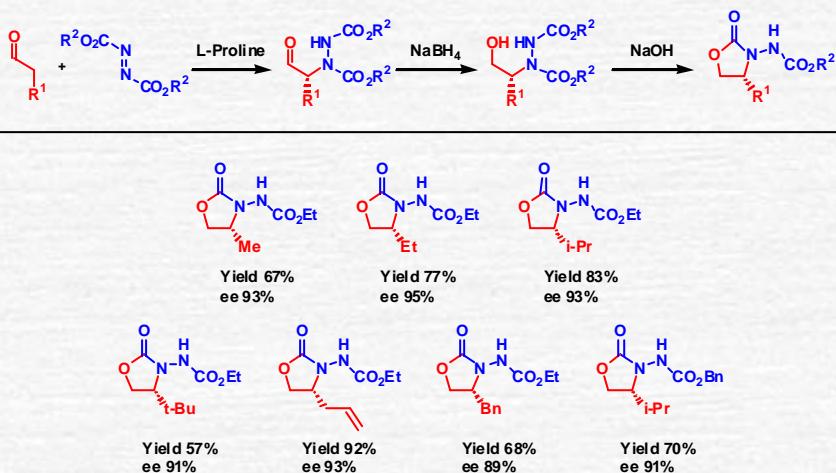


Solvent	L-Proline (mol%)	Reaction time [min]	Yield [%]	ee [%]
CH ₂ Cl ₂	50	45	93	92
CH ₂ Cl ₂	5	105	87	91
CH ₂ Cl ₂	2	300	92	84
MeCN	50	30	70	91
EtOAc	50	300	77	81
PhMe	50	450	81	86

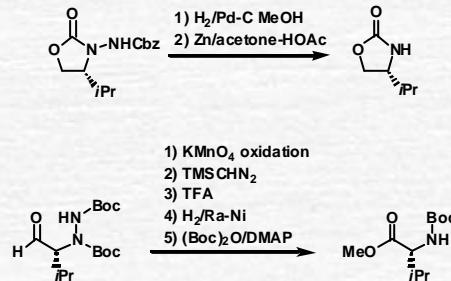
Angew. Chem. Int. Ed. 2002, 41, 1790 (rec. 01.03.02)
List, J. Am. Chem. Soc. 2002, 124, 5656 (rec. 07.03.02)

Scope of the L-proline catalysed asymmetric α -amination of aldehydes

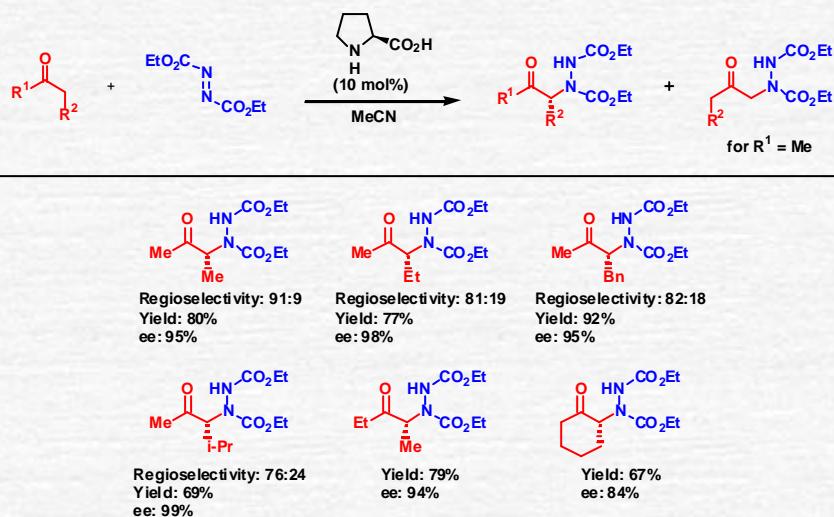
34



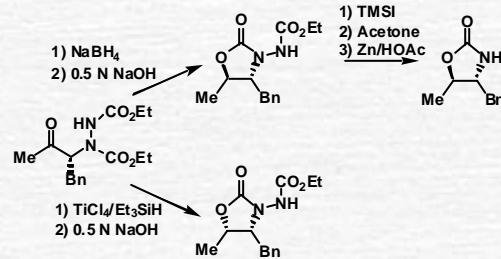
Product modification / Determination of absolute configuration



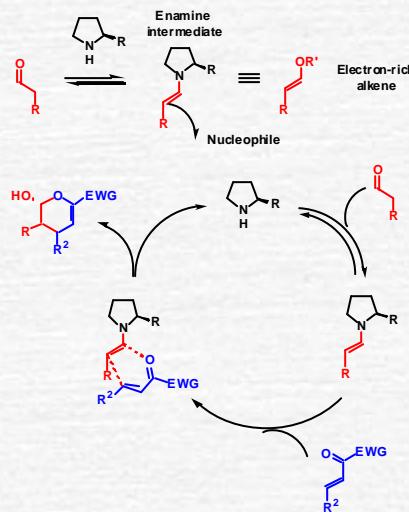
L-proline catalysed asymmetric α -amination of ketones



Stereoselective formation of optically active oxazolidinones

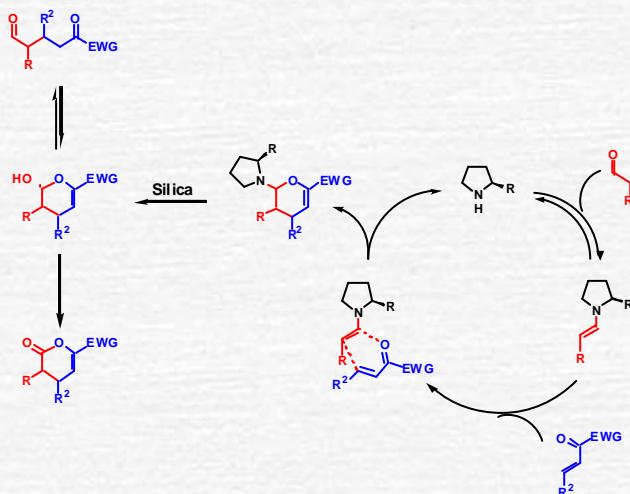


The first organo-catalytic inverse-electron demand hetero-Diels-Alder reaction – direct evidence for enamine intermediate

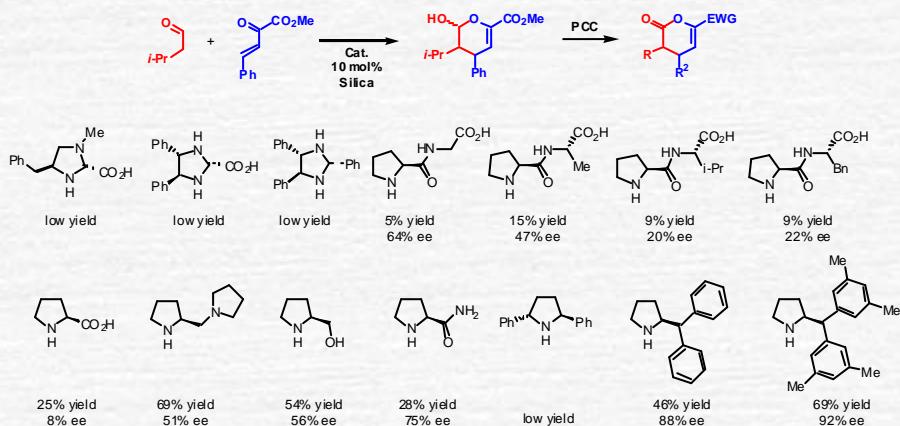


Submitted for publication

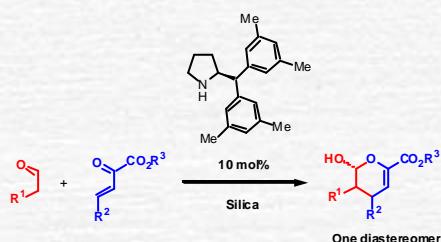
Further into the mechanism



Screening of catalysts



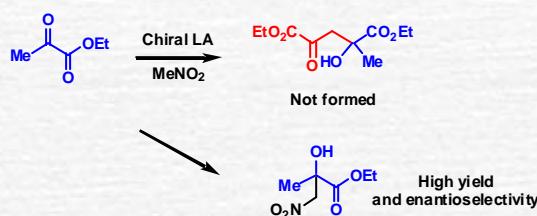
Potential of the reaction



R^1	R^2	R^3	Yield [%]	ee [%]
Et	Ph	Me	50	88
i-Pr	Ph	Me	93	89
Bn	Ph	Me	70	81
Et	4-ClPh	Me	79	85
i-Pr	4-ClPh	Me	70	90
i-Pr	Me	Et	75	94

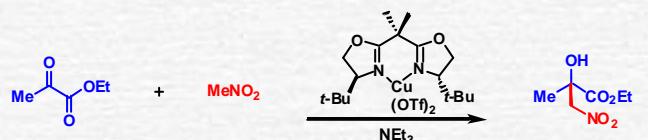
Catalytic asymmetric Henry reactions

a simple approach to optically active **b**-nitro and **b**-amino **a**-hydroxy carboxylic acids



Catalytic Enantioselective Henry Reaction of 2-Keto Esters

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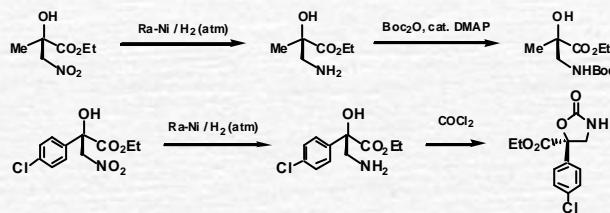


Yield: 95% ee: 92%	Yield: 73% ee: 87%	Yield: 47% ee: 77%	Yield: 91% ee: 93%
Yield: 97% ee: 94%	Yield: 92% ee: 94%	Yield: 90% ee: 94%	Yield: 99% ee: 92%
Yield: 81% ee: 86%	Yield: 91% ee: 88%	Yield: 99% ee: 93%	Yield: 68% ee: 57%

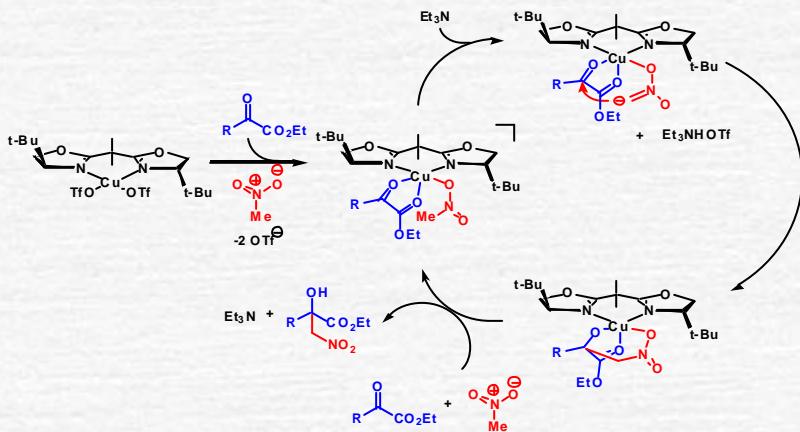
Chem. Commun. 2001, 2222
J. Org. Chem. 2002, 67, 4875

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Reduction of nitro functionality

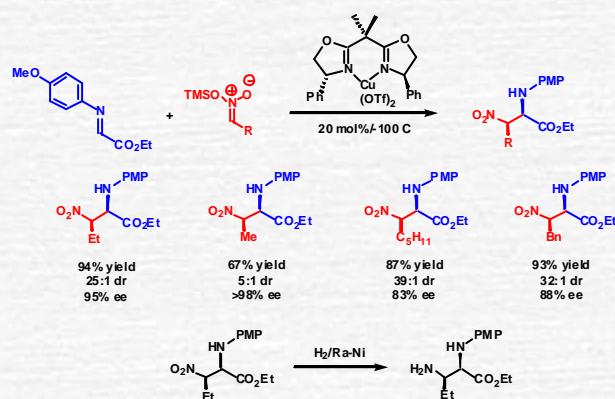


Proposed mechanism

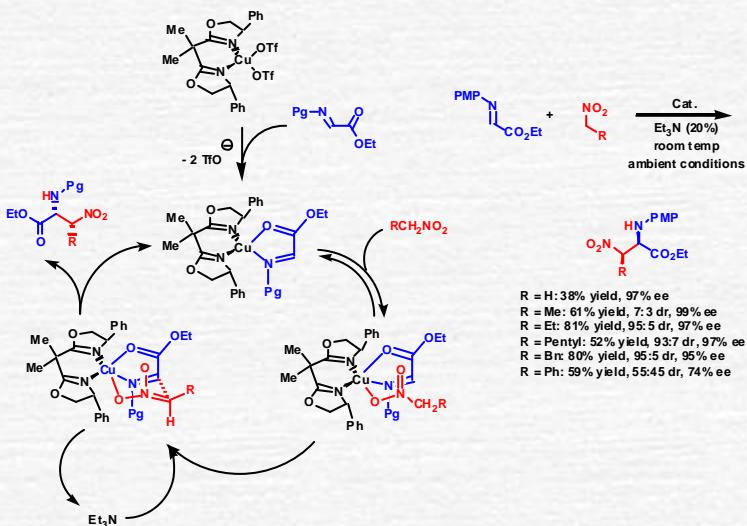


The catalytic enantioselective aza-Henry reaction

a simple approach to optically active *b*-nitro and *b*-amino *a*-amino carboxylic acids



The catalytic asymmetric nitro-Mannich reaction



Angew. Chem. Int. Ed. 2001, 40, 2992

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