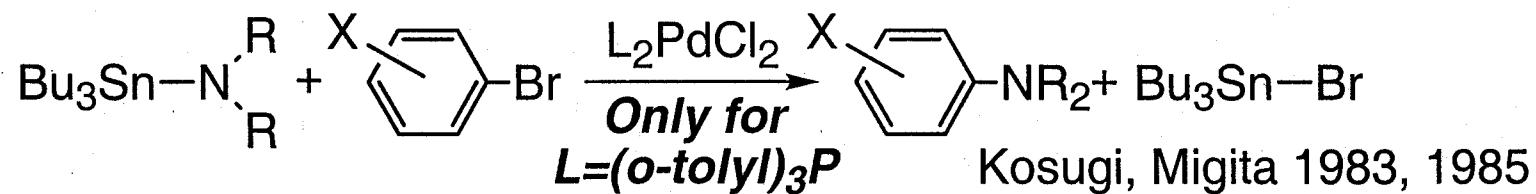


Palladium-Catalyzed Aromatic Amination



Synthesis:

Problems:

- Tin reagents: air sensitive and toxic
- Scope was narrow
- Too much catalyst

Potential:

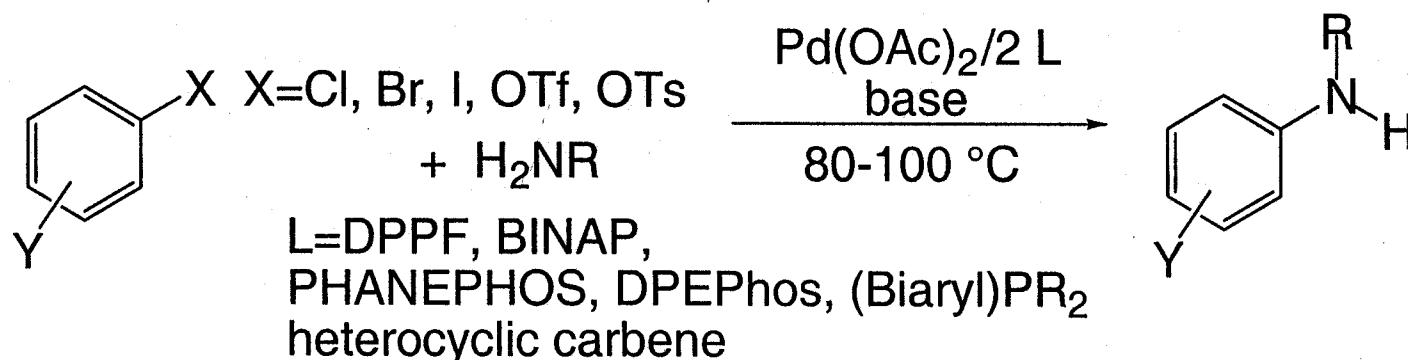
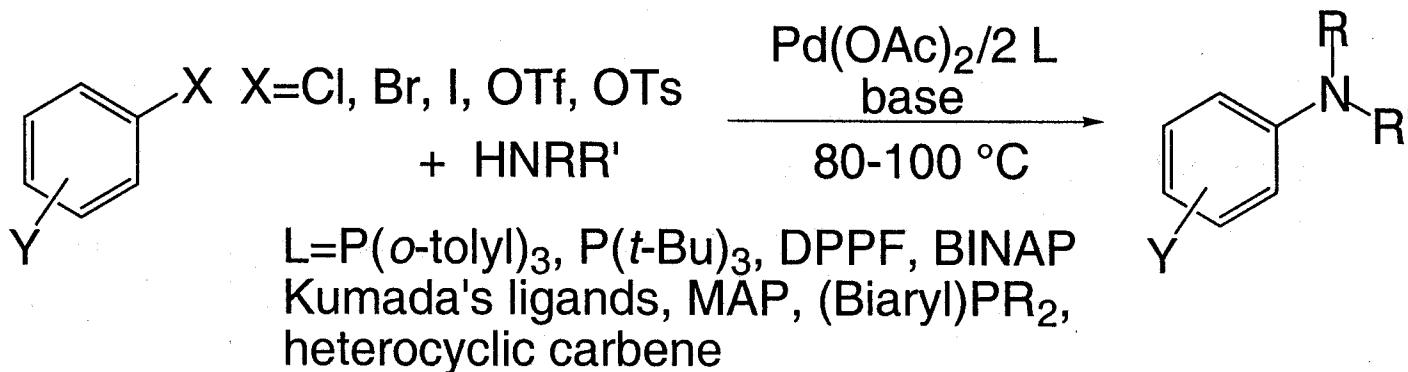
- Pharmaceuticals
- Redox Active Macromolecules
- New Ligands for Metals

Mechanism:

- New chemistry of palladium amides?
- C-X bond formation by organometallic compounds?

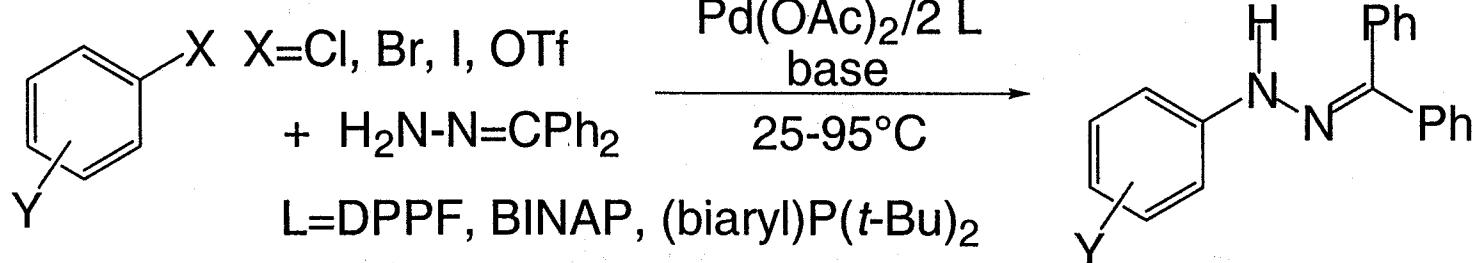
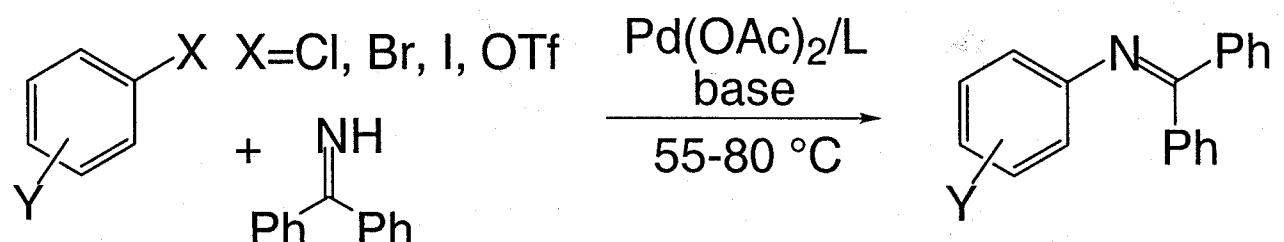
Current Arylation Processes

Review: Hartwig, J.F. *Angew. Chem., Int. Ed. Engl.* 1998, 37, 2046-2067

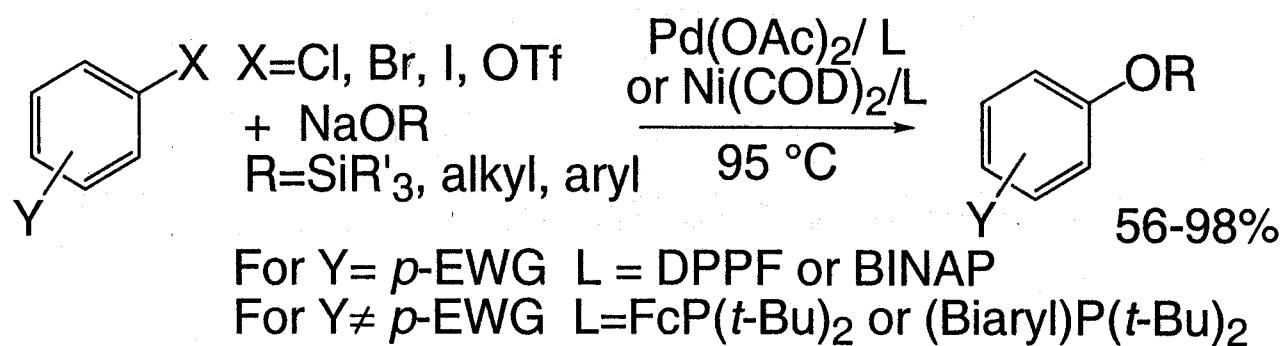
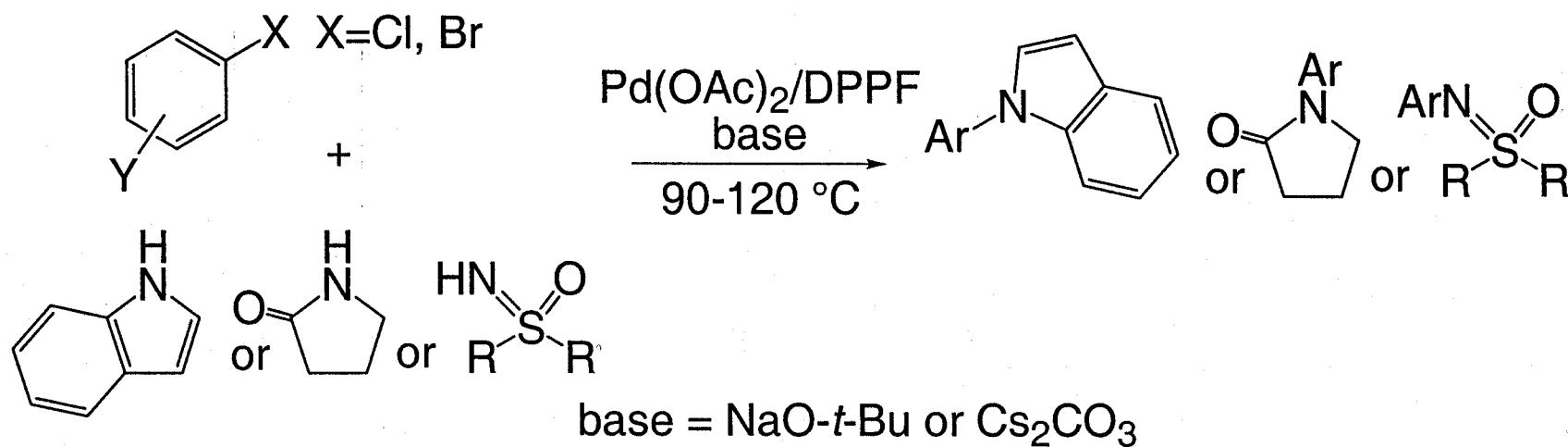


base = NaO-*t*-Bu, Cs₂CO₃
DPPF=1,1'-bis-(diphenylphosphino)ferrocene
BINAP=2,2'-bis-(diphenylphosphino)-1,1'-binaphthyl

Current Arylation Processes



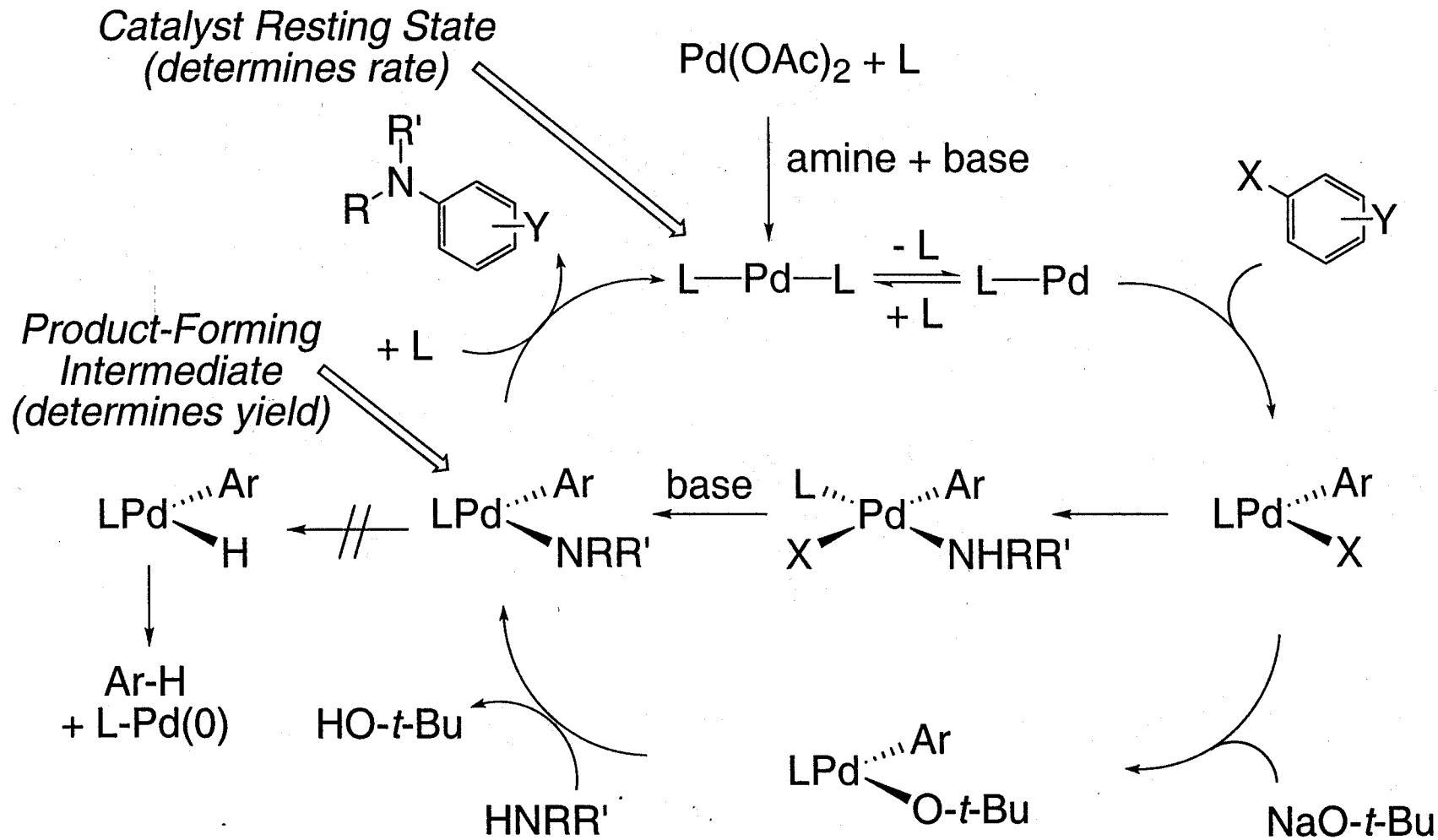
Current Arylation Processes



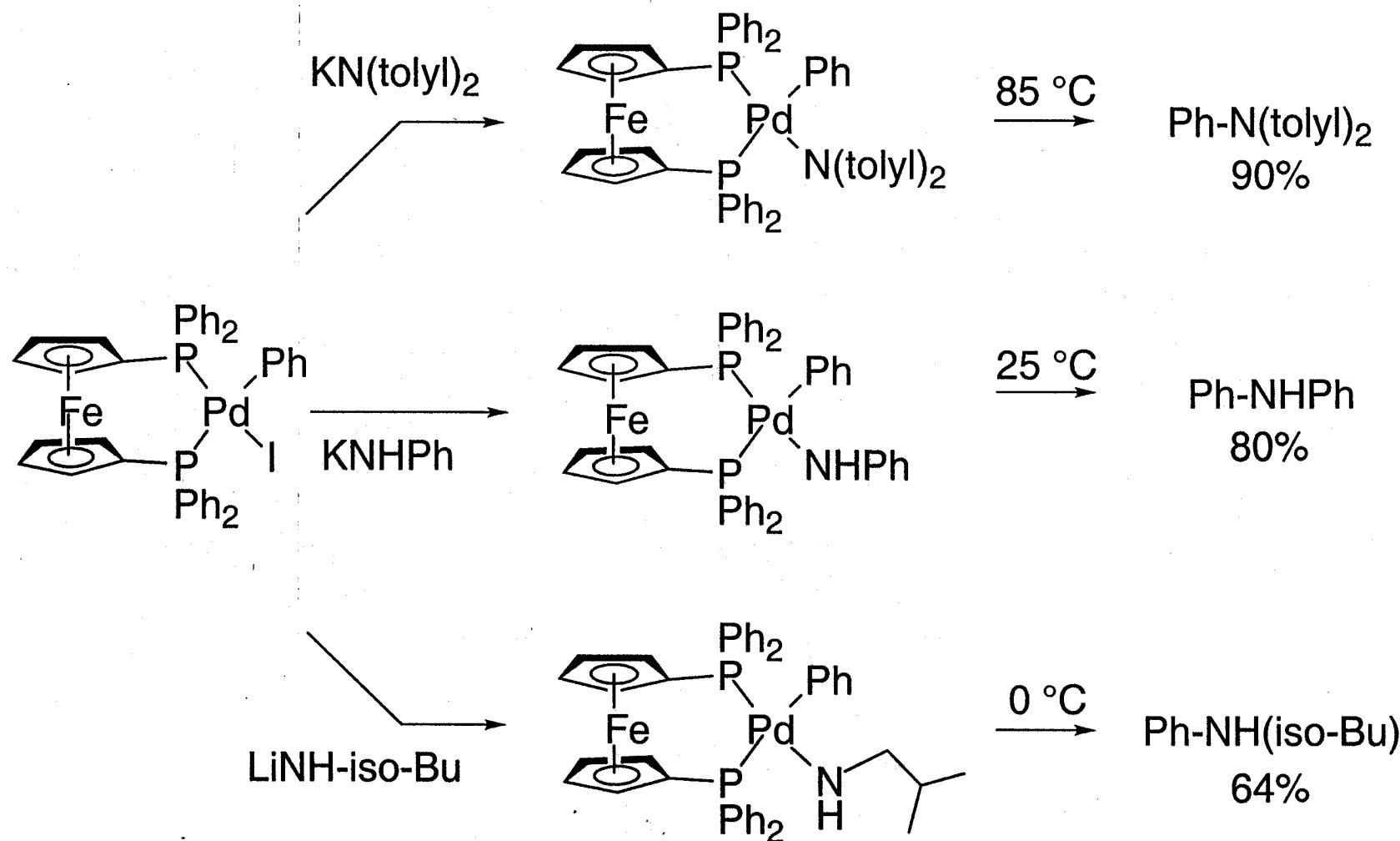
Review: Hartwig, J.F. *Angew. Chem., Int. Ed. Engl.* **1998**, 37, 2046-2067

Summary of Amination Mechanism

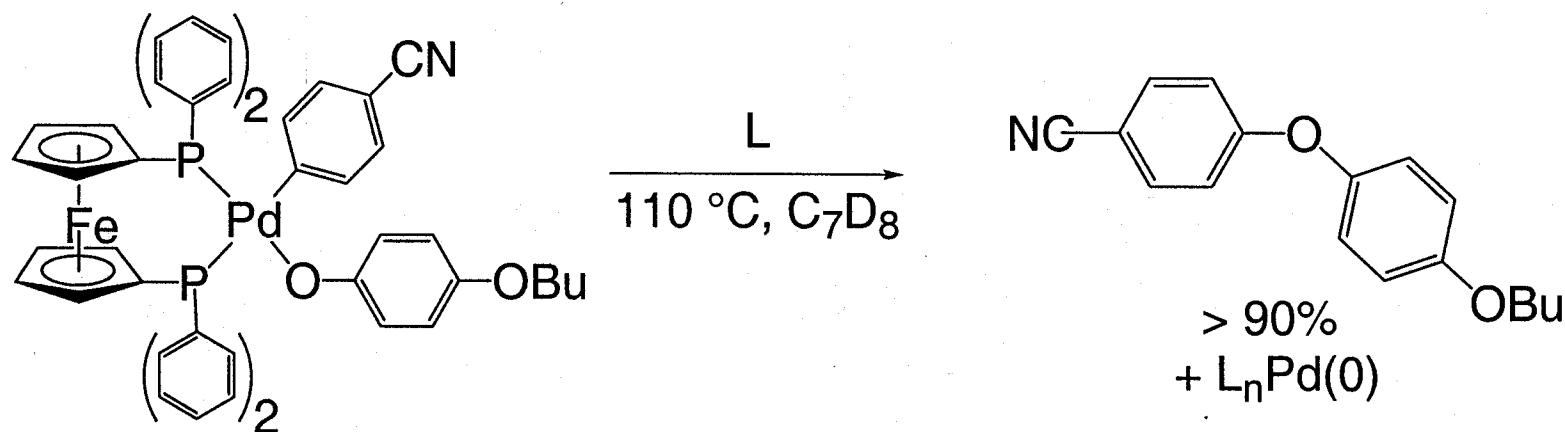
$L = \text{BINAP, DPPF, P}(t\text{-Bu})_3$



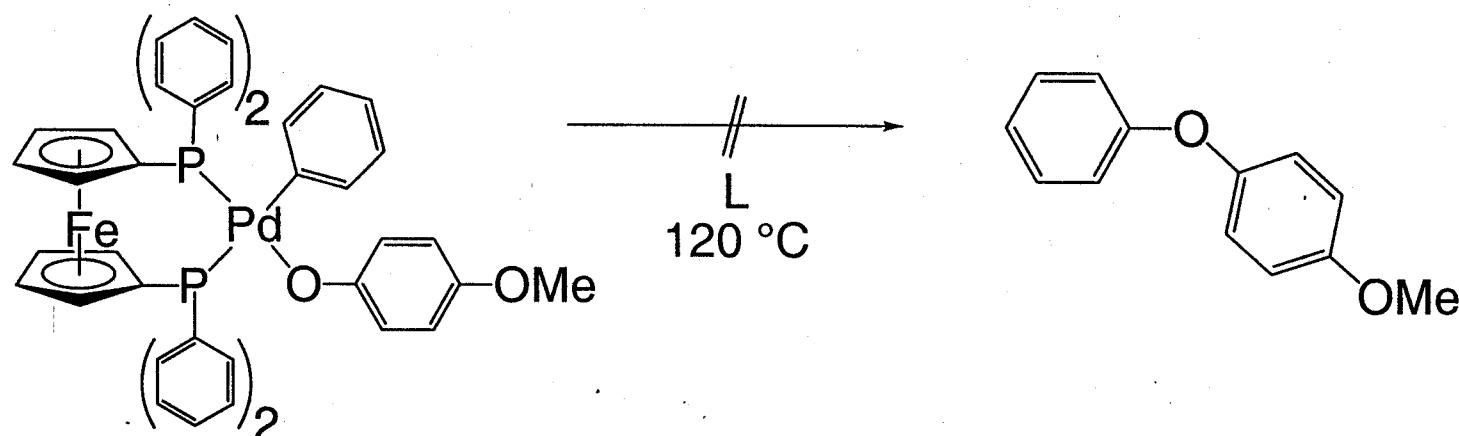
Pd-Amido Complexes with Bis-Phosphines



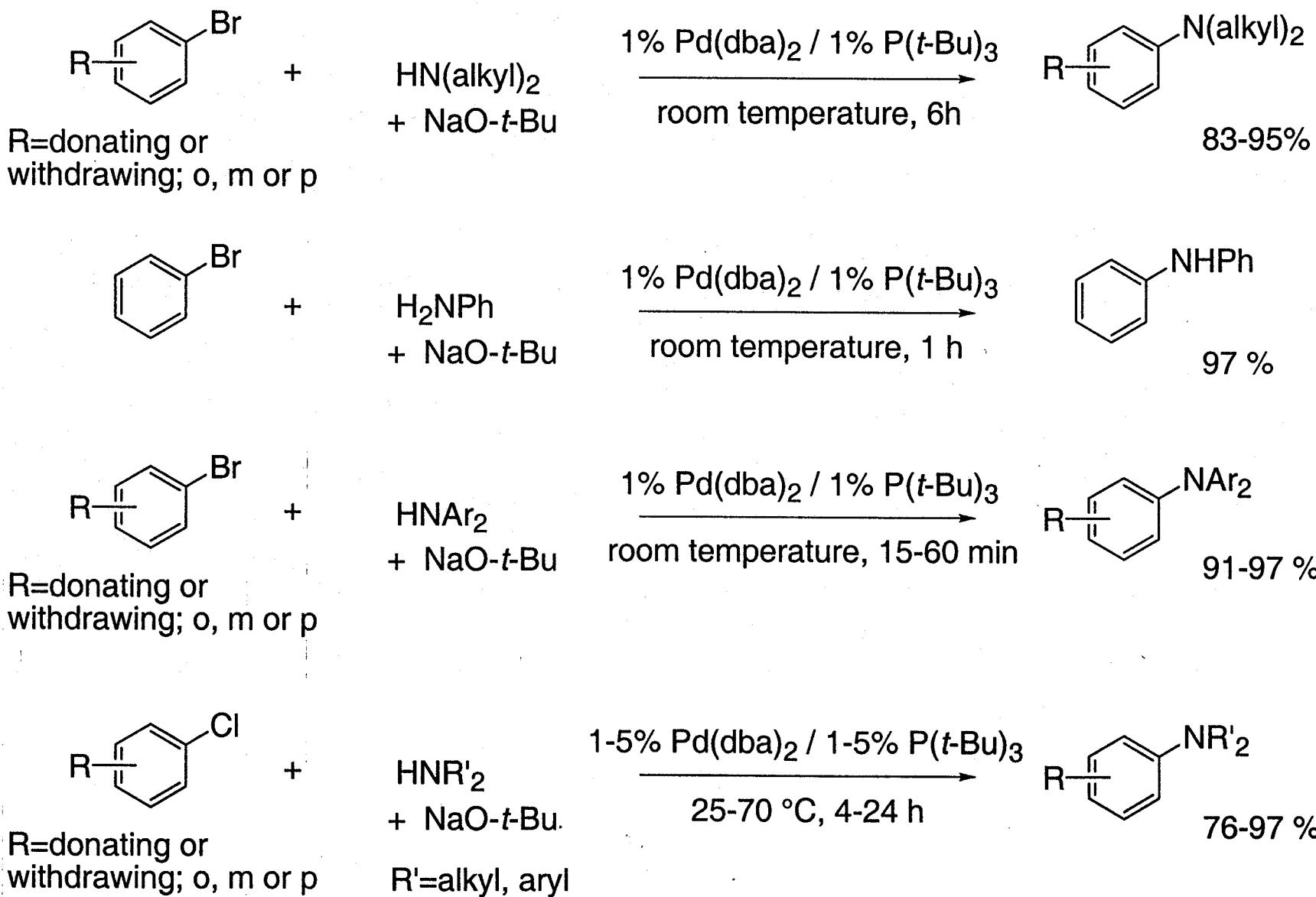
Reductive Elimination of Diaryl Ether...



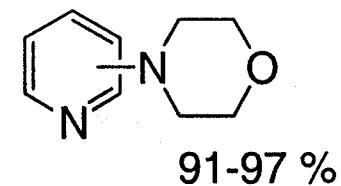
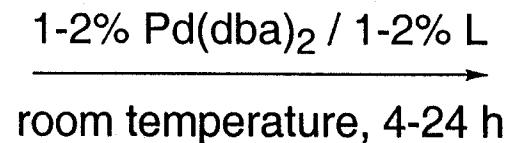
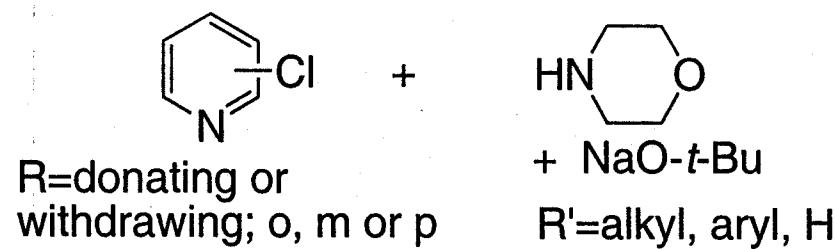
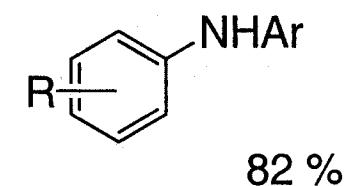
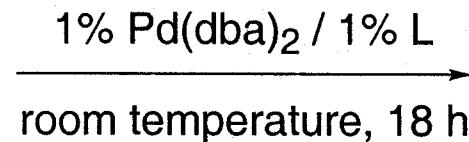
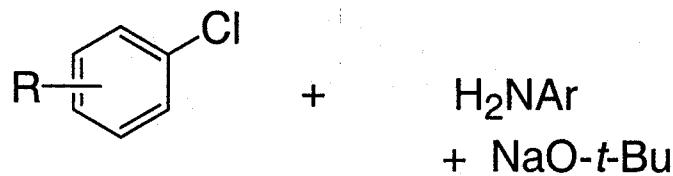
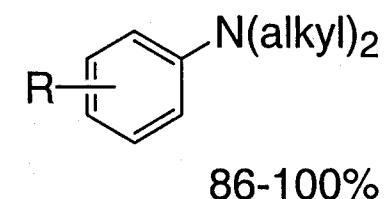
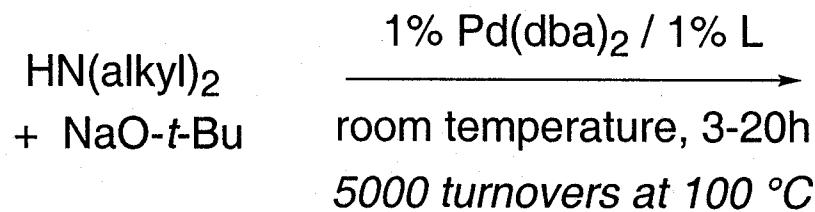
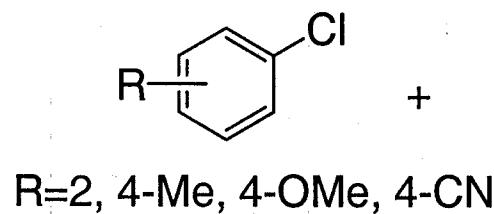
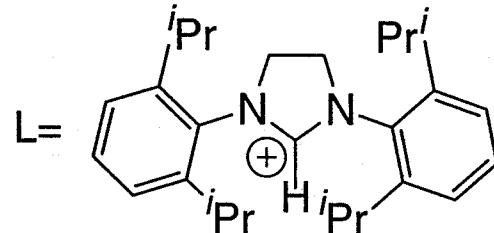
But No Elimination Without a Withdrawing Group



Mild Amination with $P(t\text{-}Bu)_3$

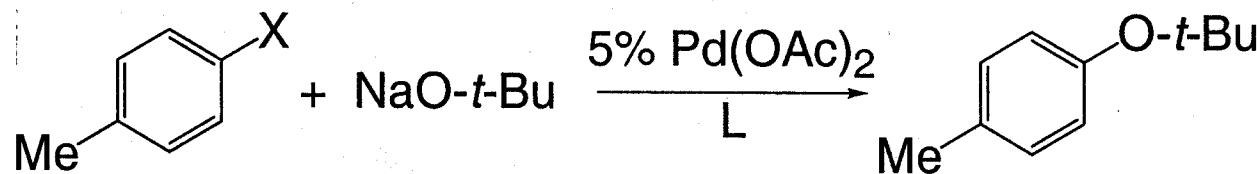


Mild Amination with Saturated Carbene Ligands

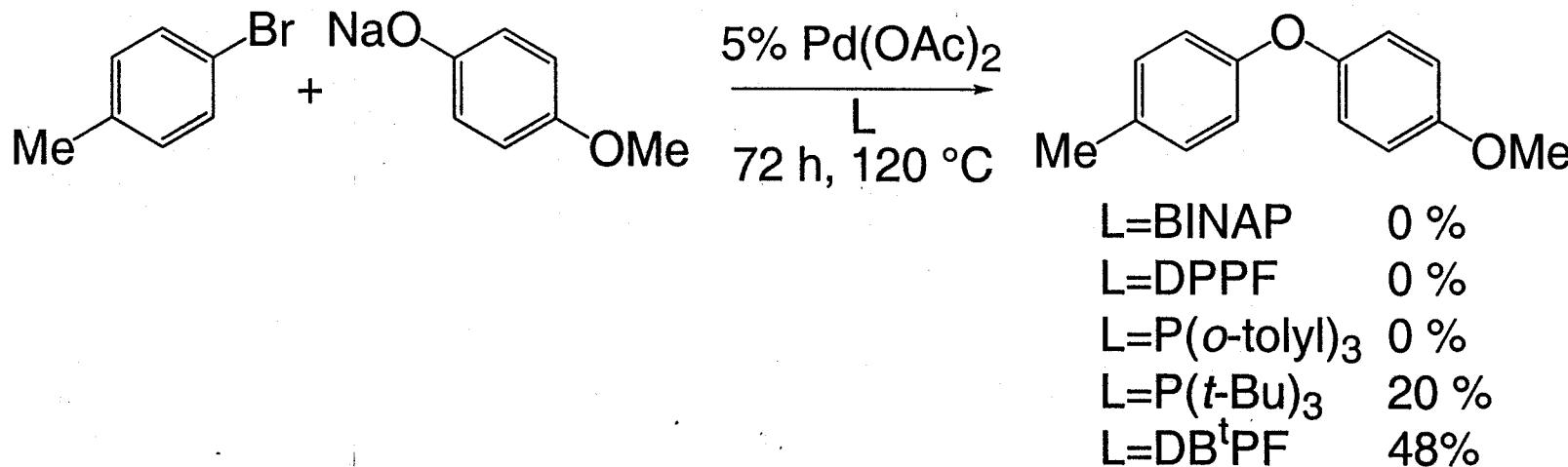


R'=alkyl, aryl, H

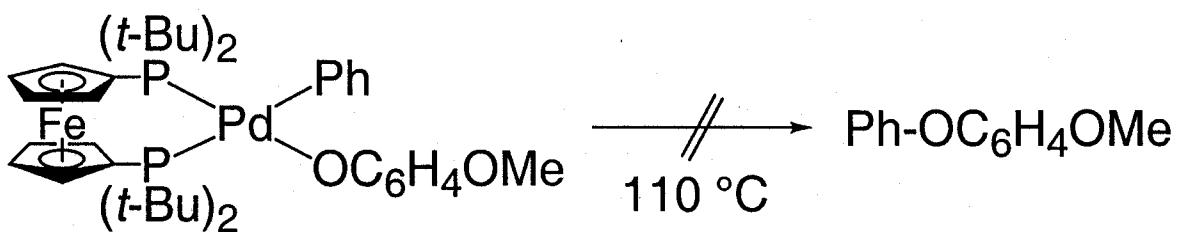
Hindered Alkylphosphines Improve Ether Synthesis



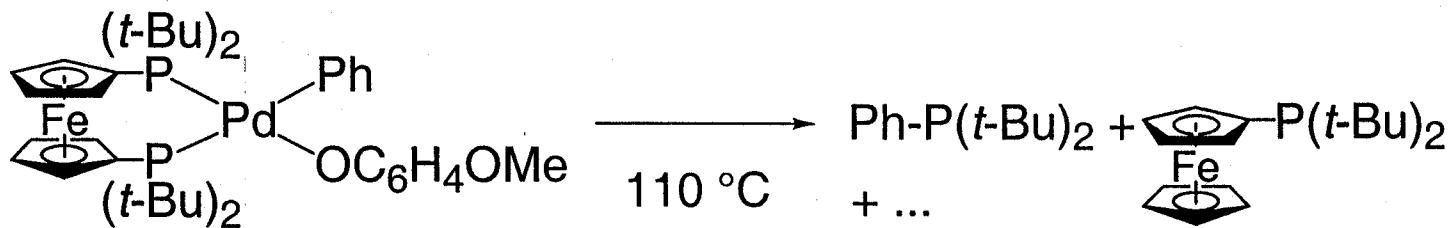
$\text{DB}^t\text{PF} =$		$L = \text{BINAP}$	$X = \text{Br}$	53 %
		$L = \text{DPPF}$	$X = \text{Br}$	38 %
		$L = \text{P}(o\text{-tolyl})_3$	$X = \text{Br}$	0 %
		$L = \text{P}(t\text{-Bu})_3$	$X = \text{Cl}$	100 %
		$L = \text{DB}^t\text{PF}$	$X = \text{Cl}$	89 %



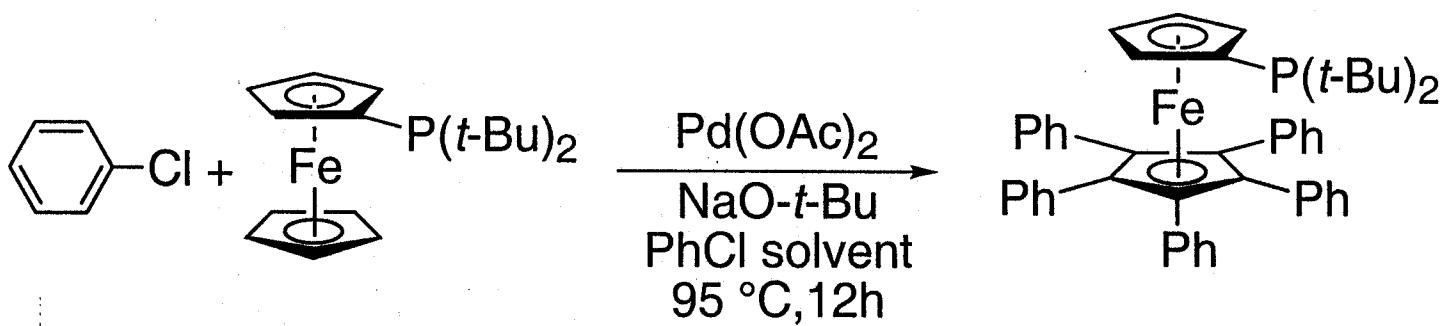
Can DB^tPF induce Ether Formation without EWG?



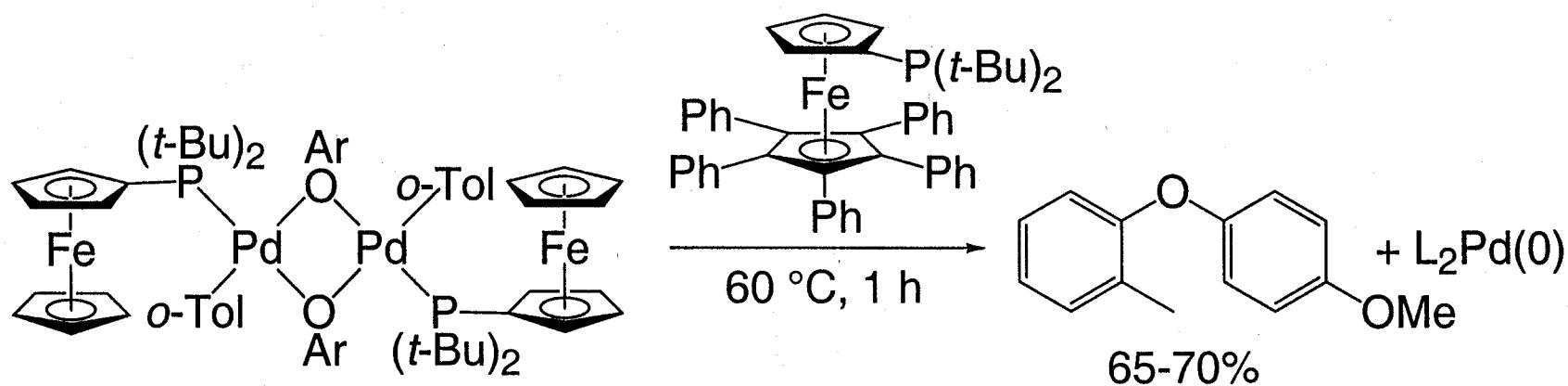
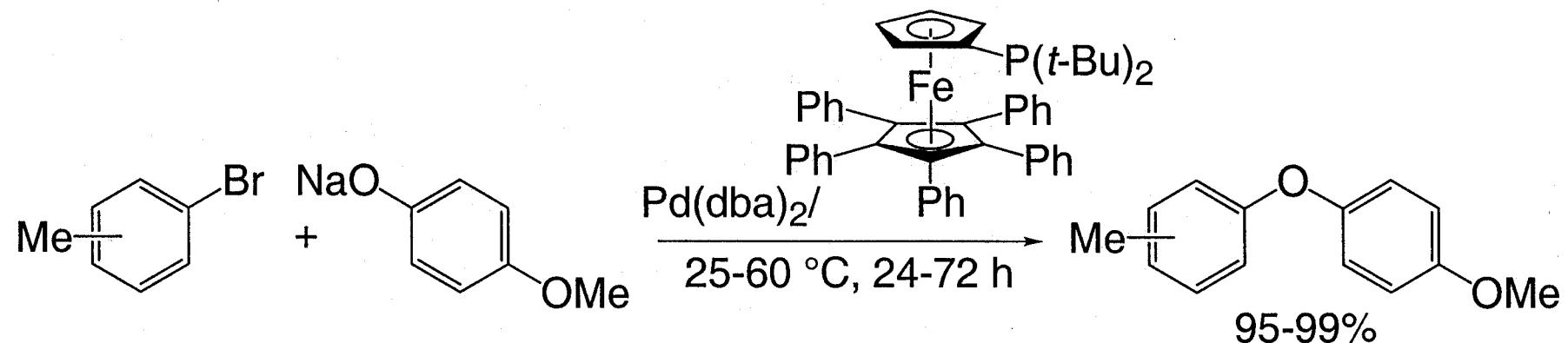
A Closer Look:



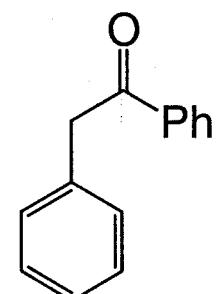
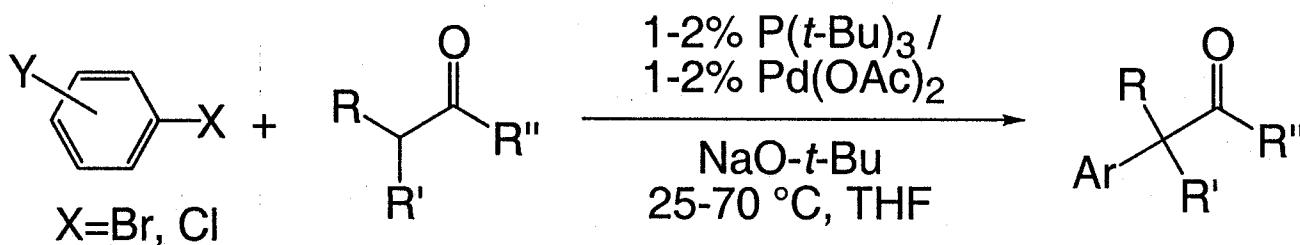
The Catalyst Changes



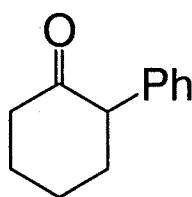
Very Mild Synthesis of Diaryl Ethers



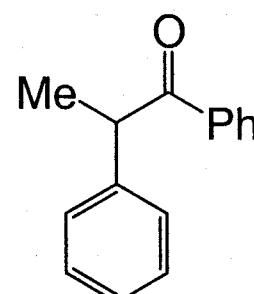
Mild Ketone Arylations



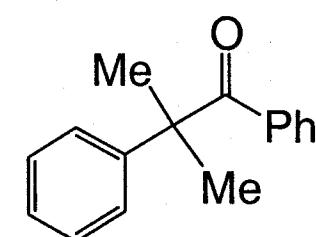
X=Br: 96%



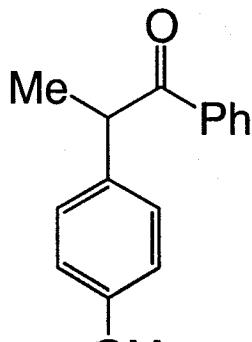
X=Br: 73 %



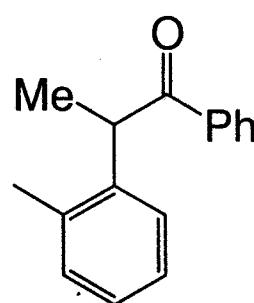
0.5 mol% 97%
 0.005 mol% <24h, 97%
 X=Cl 2mol% 90%



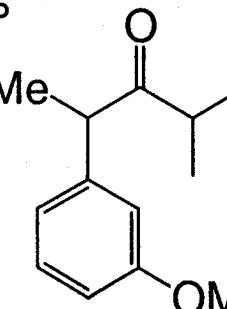
X=Br 92 %
 X=Cl 82% (3-OMe)



X=Cl: 90%

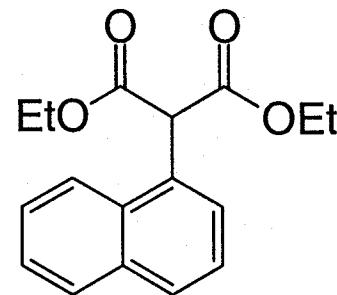
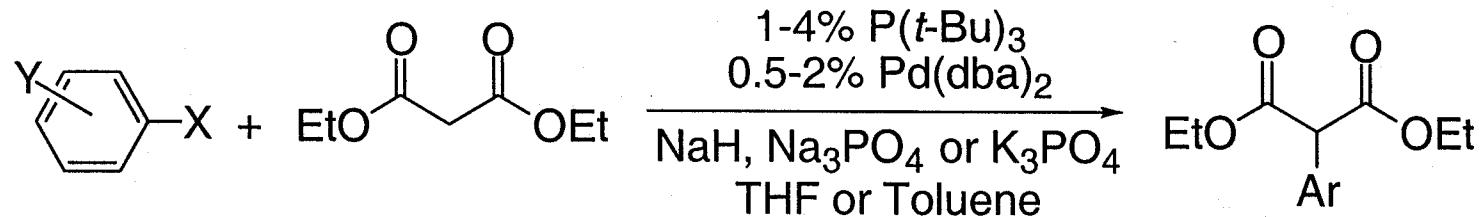


X=Cl 80%

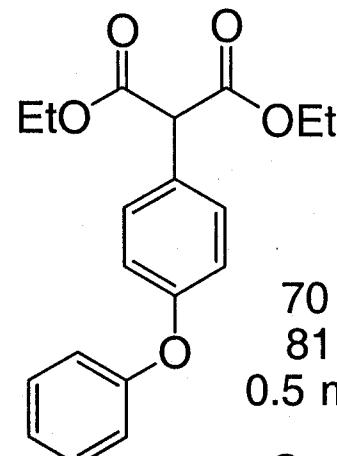


X=Br 83 %
 (74:26 ratio)
 X=Cl 69 %
 (76:24 ratio)

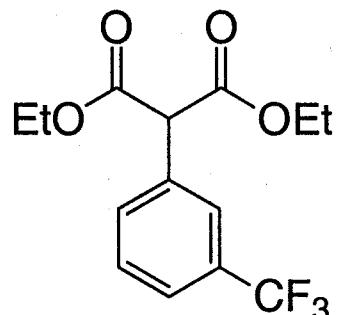
Arylation of Malonates



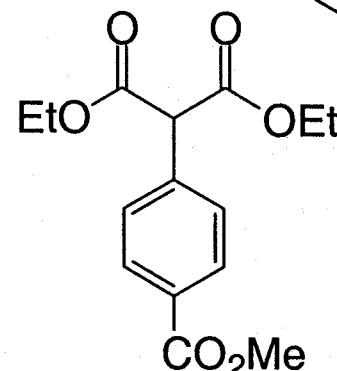
70 °C
84 %



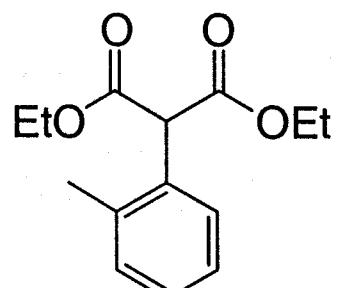
70 °C
81 %
0.5 mol%



70 °C
92 %

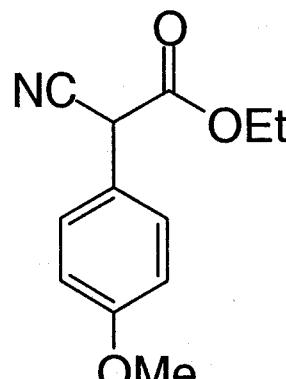
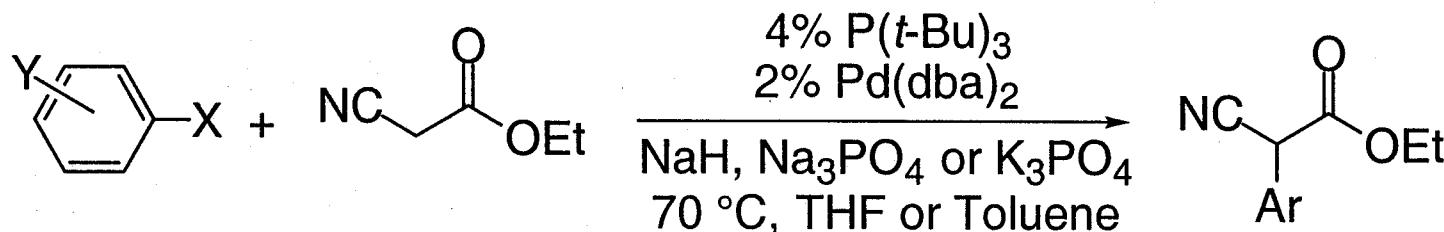


50 °C
86 %

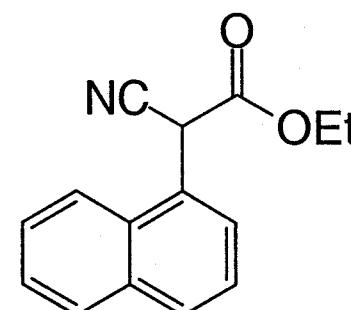


70 °C
82 %

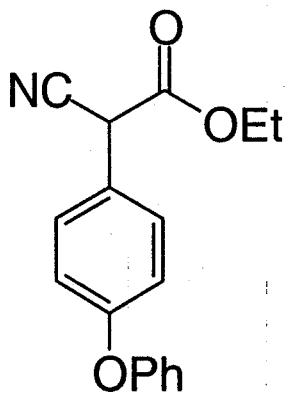
Arylation of Cyanoesters



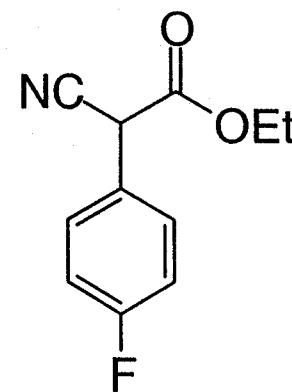
80 %



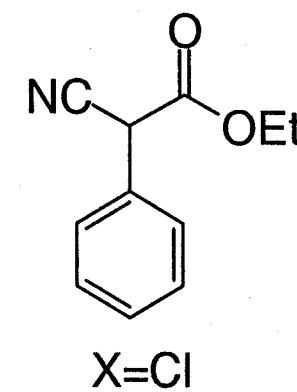
83 %



90%



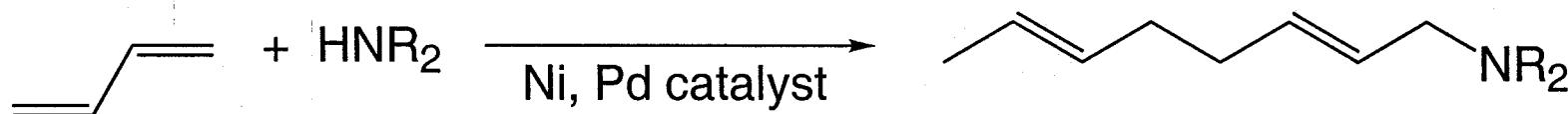
93%



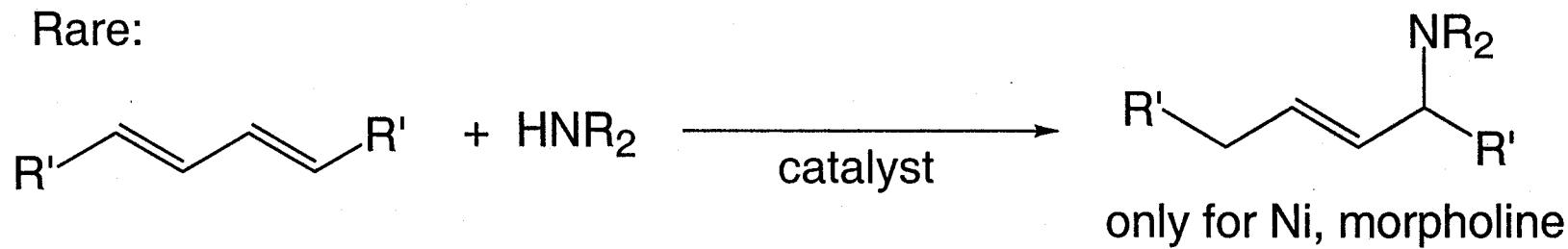
X=Cl
100 °C
91 %

Catalytic Hydroamination of Dienes

Well known:



Rare:



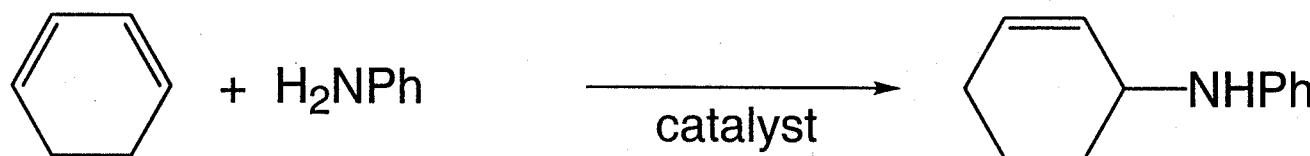
Objectives:

- Substituted dienes
- Arylamines amines

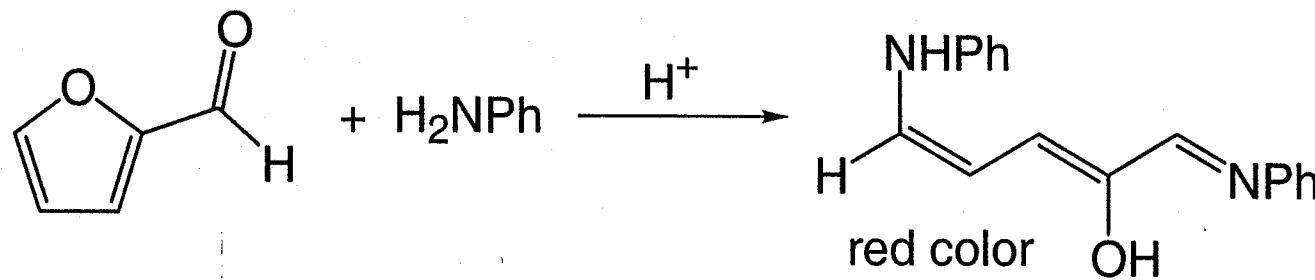
Questions:

- Is nickel necessary?
- What ligand properties are optimal?

Spot Test: Catalytic Diene Hydroamination

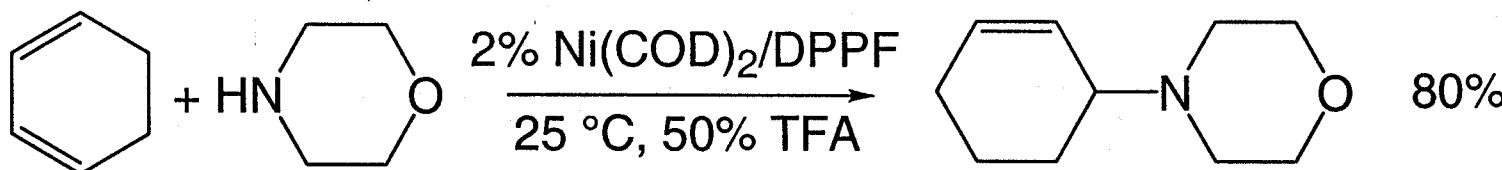
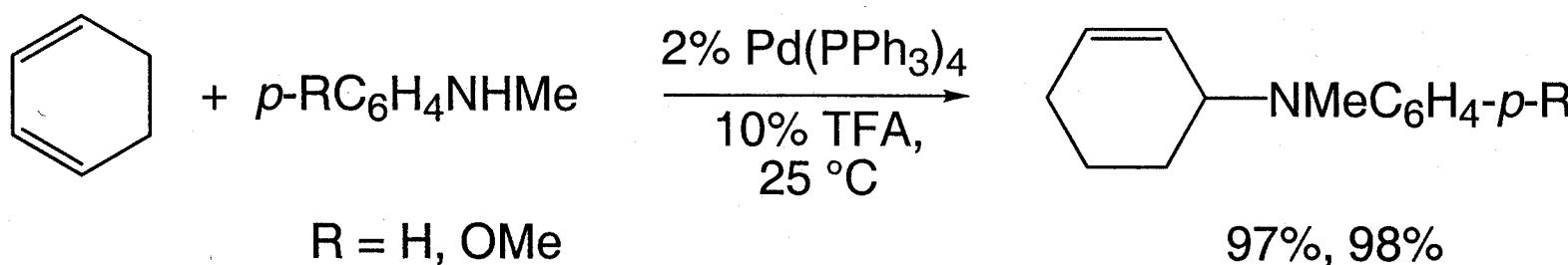
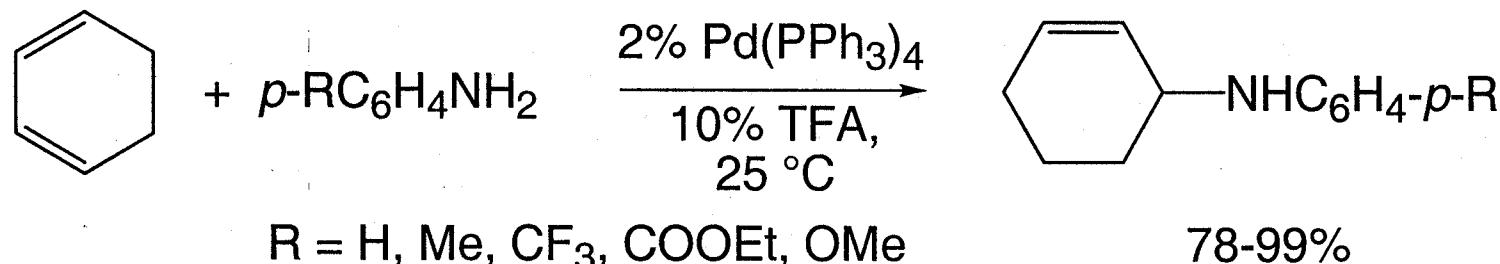


Analysis for Aniline: Furfural, HOAc

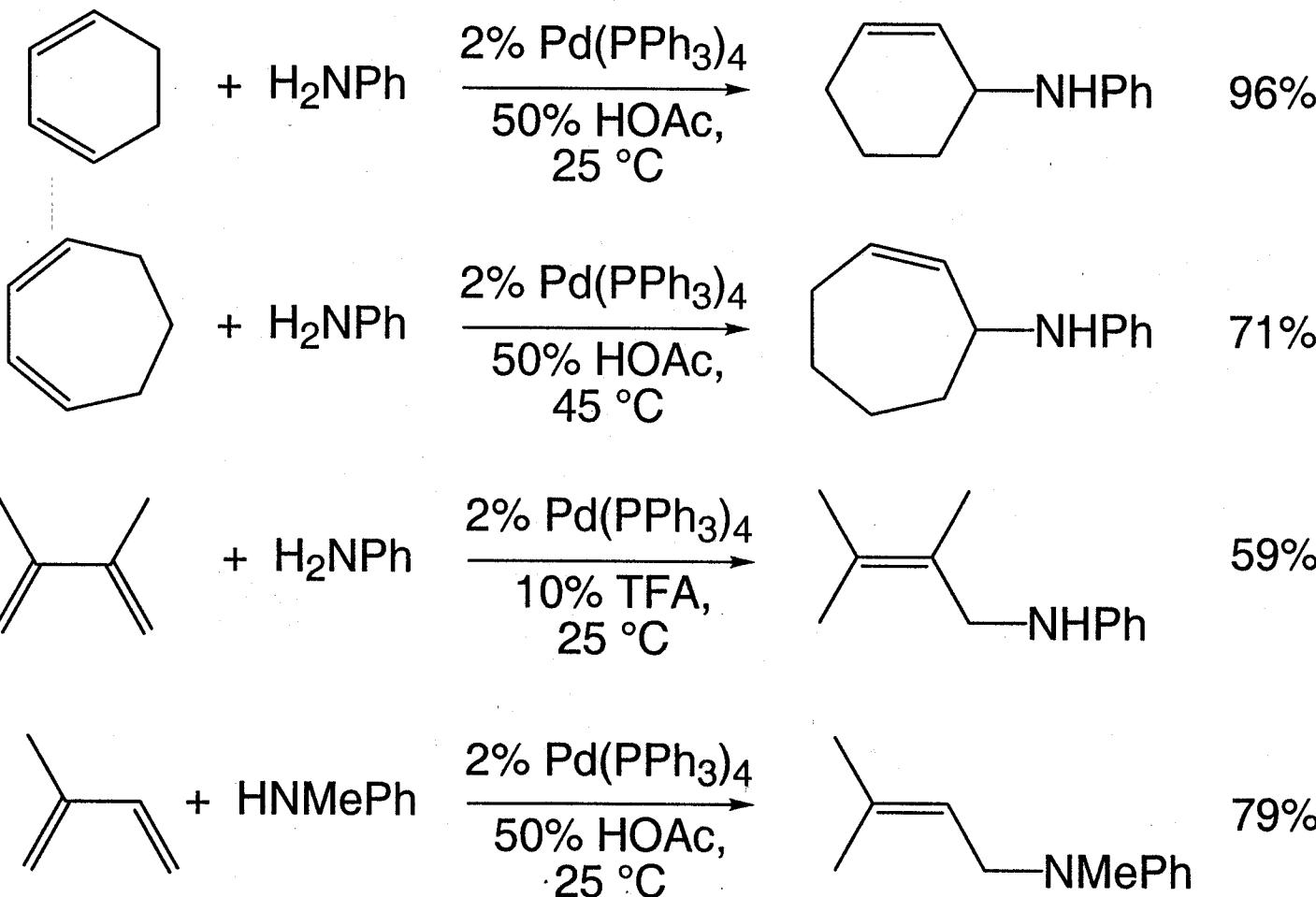


Unreacted amine: Red
Complete reaction: clear or the color of the catalyst

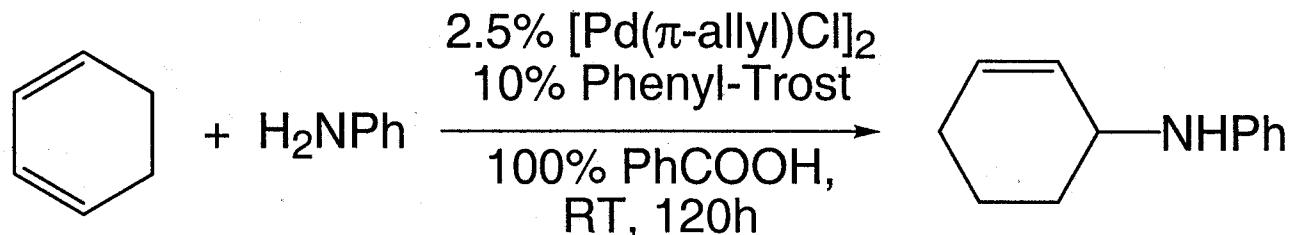
Preliminary Scope of Diene Hydroamination



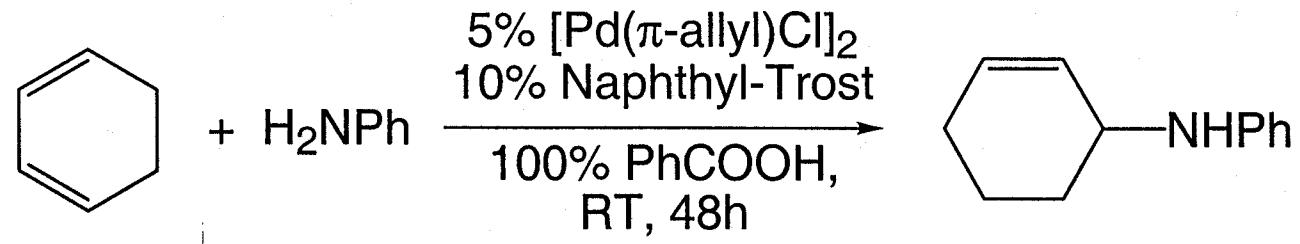
Preliminary Scope of Diene Hydroamination



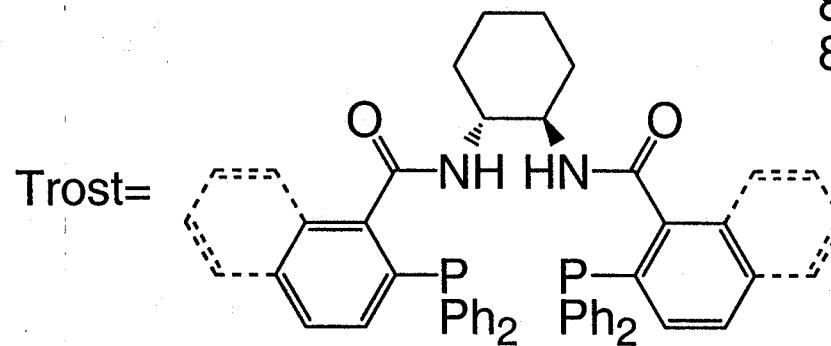
Toward An Enantioselective Version



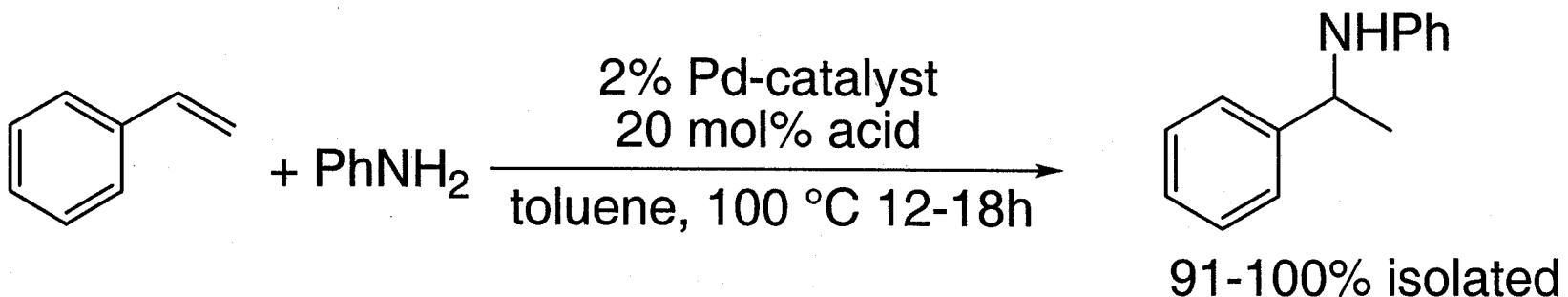
98% conversion
55% ee



84% conversion
88% ee



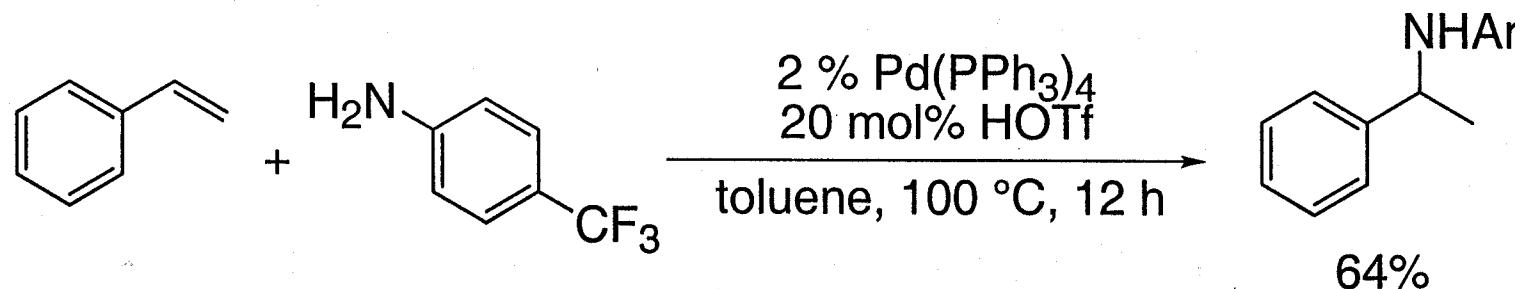
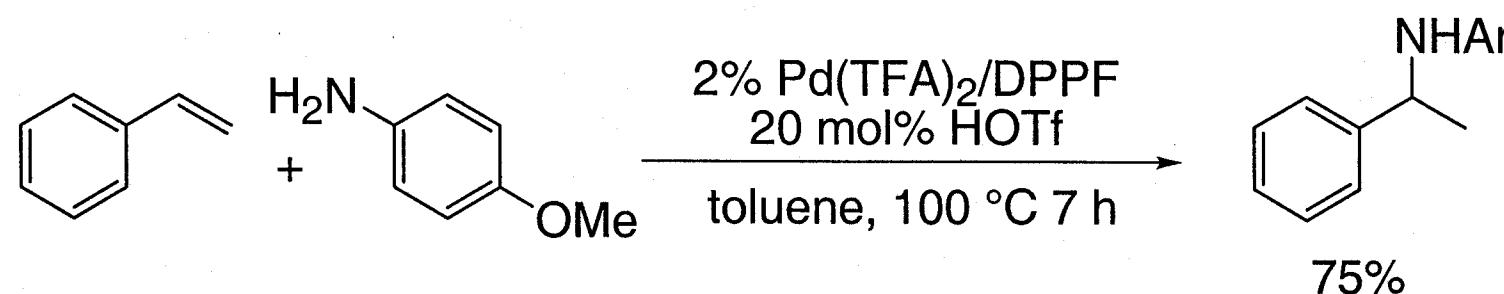
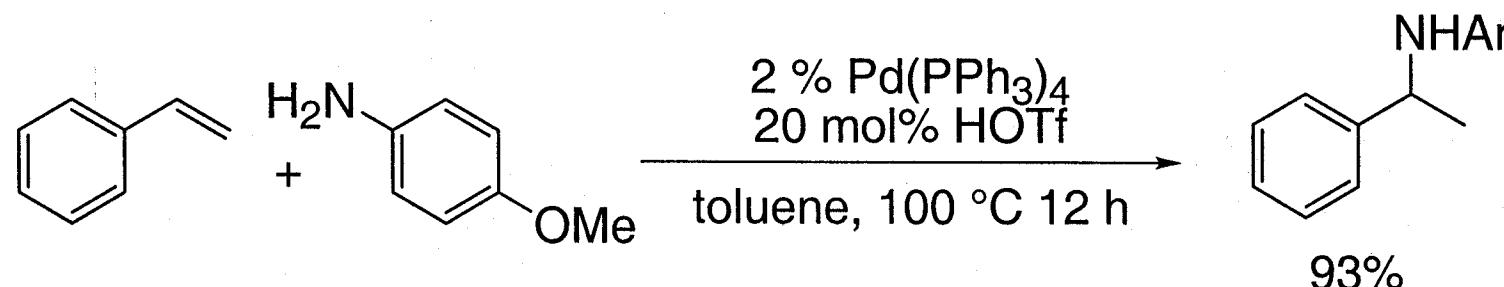
Vinylarene Hydroamination



Role of the Acid:

Palladium / Ligand	Acid	Yield
2% [Pd(PPh ₃) ₄]		0%
2% [Pd(PPh ₃) ₄]	20% TFA	63%
2% [Pd(PPh ₃) ₄]	20% TfOH	91%
2% Pd(TFA) ₂ / 8% PPh ₃		30%
2% Pd(TFA) ₂ / 8% PPh ₃	20% TFA	68%
2% [Pd(TFA) ₂] / 3% DPPF		71%
2% [Pd(TFA) ₂] / 3% DPPF	20% TfOH	100%
2% [(DPPF)Pd(OTf) ₂]		100%

Vinylarene Hydroamination



Asymmetric Vinylarene Hydroamination

