

IASOC

ISCHIA ADVANCED SCHOOL OF ORGANIC CHEMISTRY

Ischia (Naples), September 27 - October 2, 2008

(50+10 min)

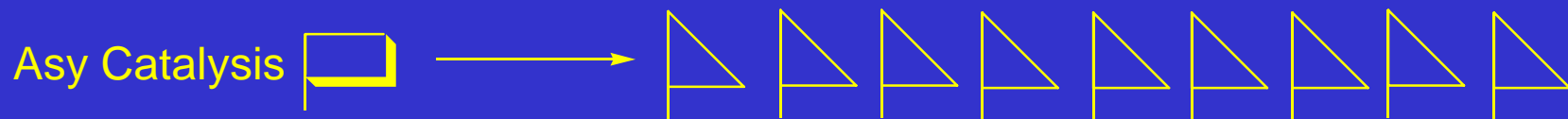




A Self-Supporting Strategy for Chiral Catalyst Immobilization

Kuiling Ding

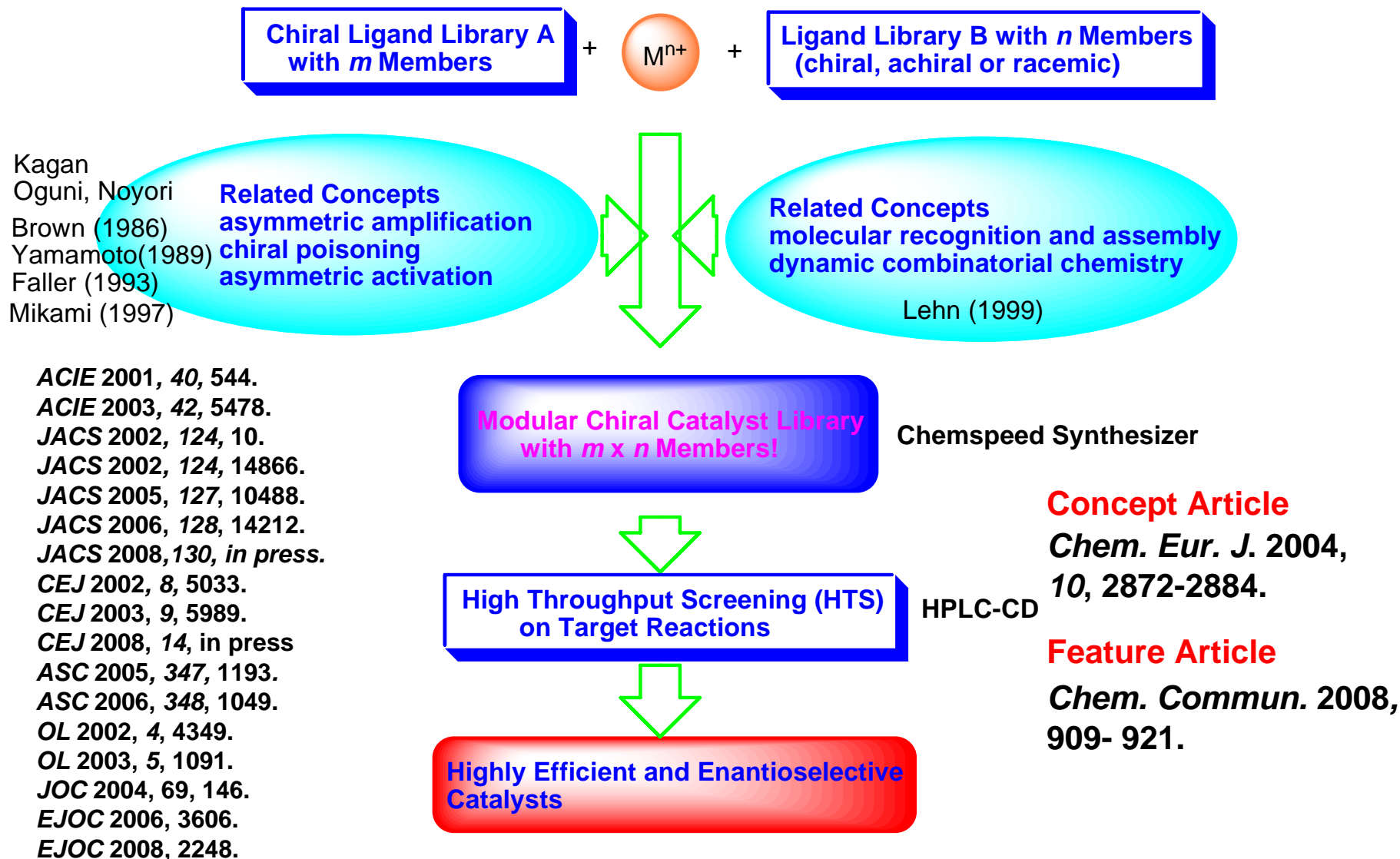
**Shanghai Institute of Organic Chemistry,
Chinese Academy of Sciences, China**

Some Challenges in Asymmetric Catalysis



- **Selectivity:** There is no given catalyst that is universal for all substrates.  **Catalyst diversity**
- **Reactivity and Efficiency :** 1-10 mol% catalyst loading is not practical.  **0.1-0.01% or less.**
- **Process chemistry:** speed up the rate for catalyst discovery in customer synthesis.
- **Catalyst recovery and reuse.**
- **Cost, energy, solvent, safety, and others.....**

Combinatorial Approach to Asymmetric Catalysis



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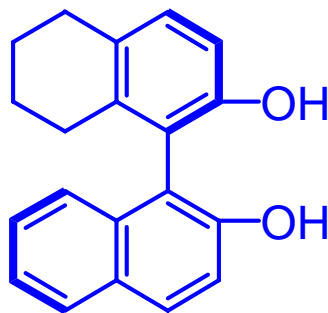
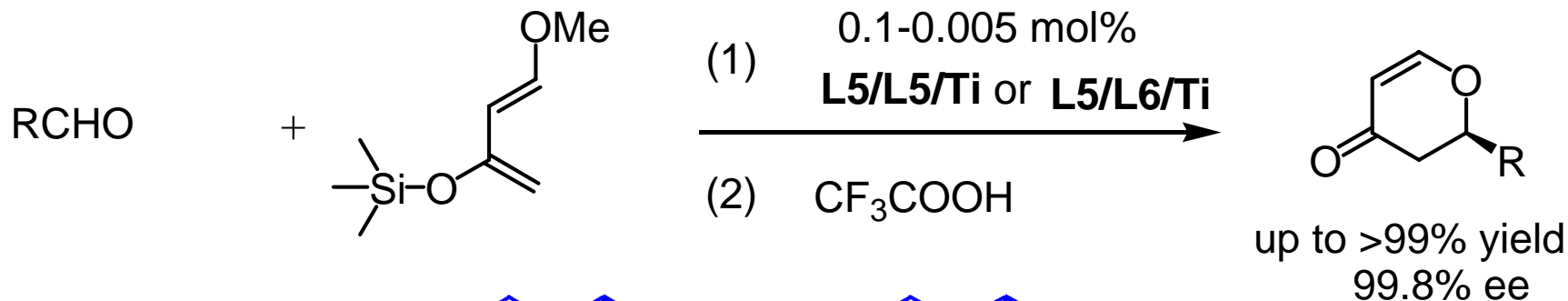
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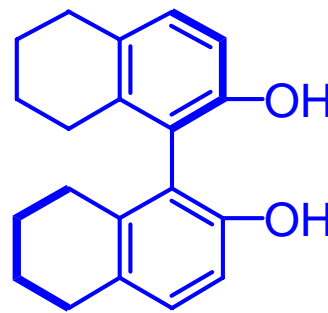
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Exceptionally Efficient Catalysts for Enantioselective Hetero-Diels-Alder Reaction



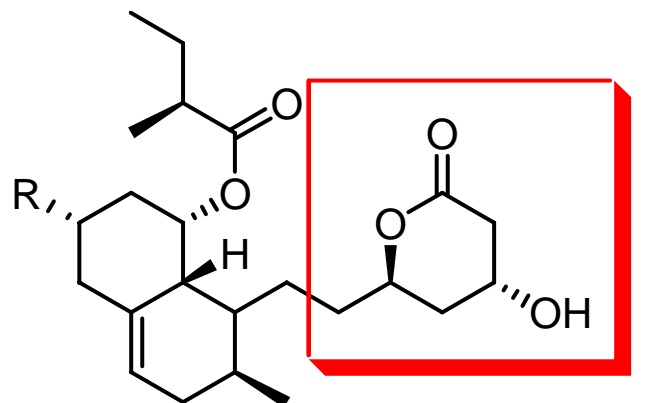
L5



L6

Highly efficient: 0.1% - 0.005% of cat. loading!
Very high yield and enantioselectivity
Room temperature and solvent free

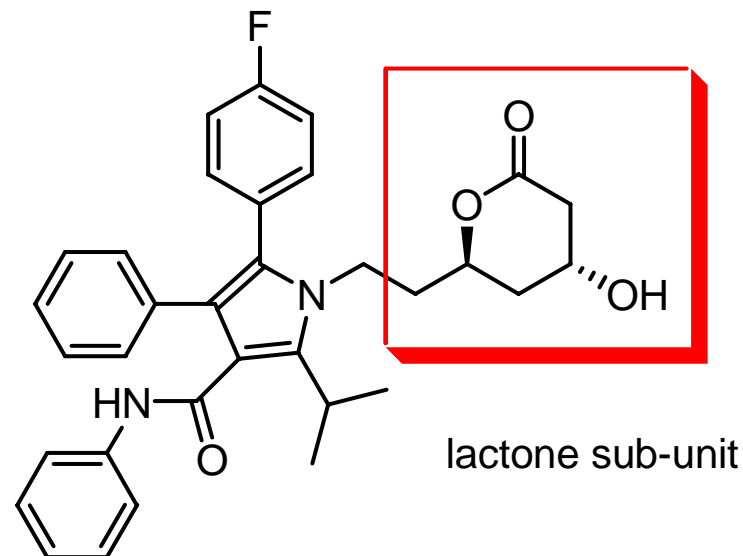
Asymmetric HDA Reaction: A Facile Approach to Lactone Sub-unit of Chiral Drugs



lactone sub-unit

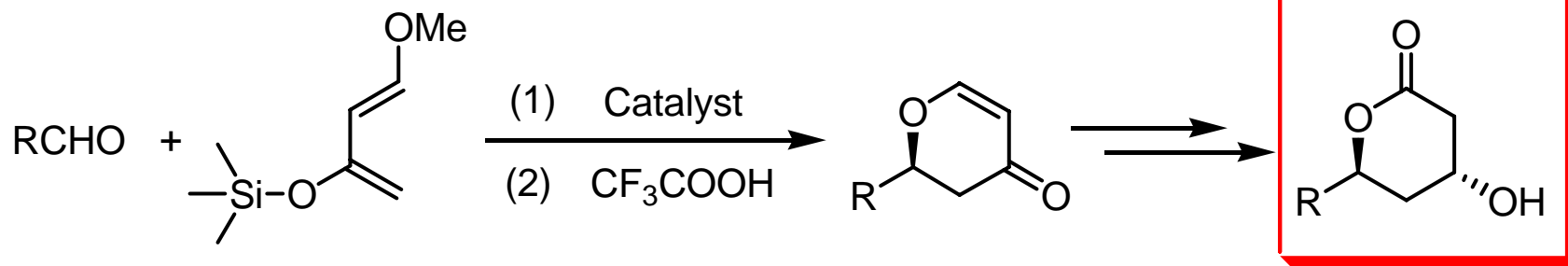
R = H, compactin

R = Me, Mevacor

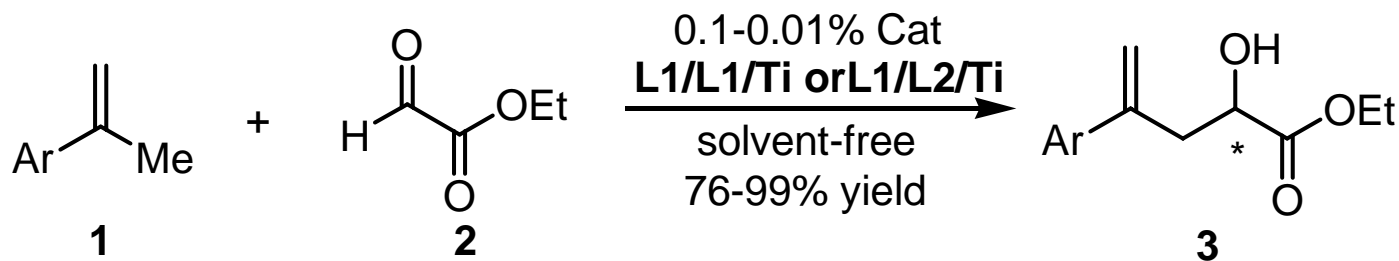


lactone sub-unit

Lipitor, \$13.6 billion (2006)



Exceptionally Efficient Catalysts for Quasi Solvent-Free Enantioselective Carbonyl-Ene Reaction

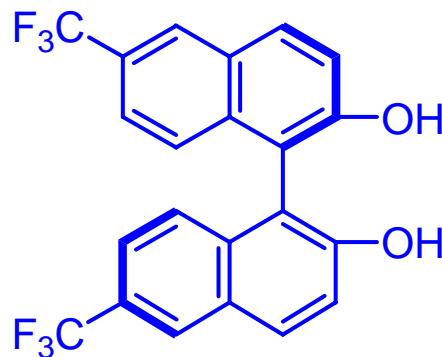


1a: Ar = C₆H₅
1b: Ar = 4-ClC₆H₄
1c: Ar = 4-FC₆H₄
1d: Ar = 4-CH₃C₆H₄

3a: Ar = C₆H₅ (98.2% ee)
3b: Ar = 4-ClC₆H₄ (99.4% ee)
3c: Ar = 4-FC₆H₄ (98.4% ee)
3d: Ar = 4-CH₃C₆H₄ (97.1% ee)

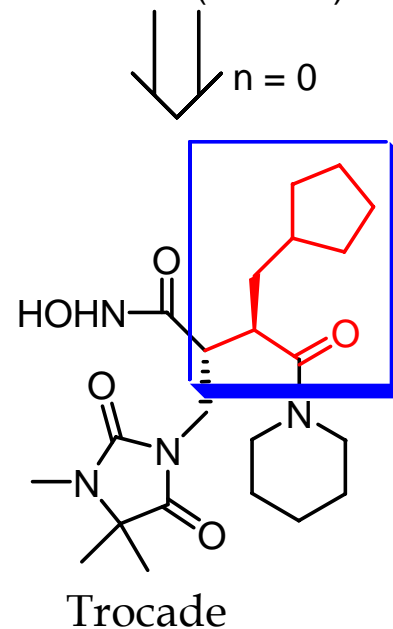
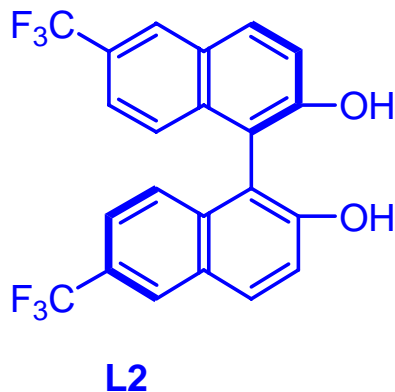
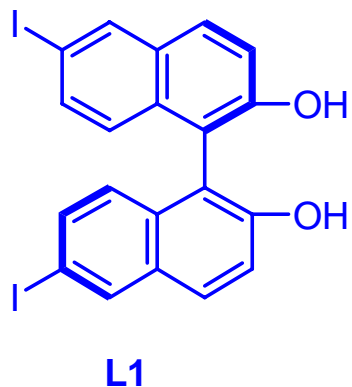
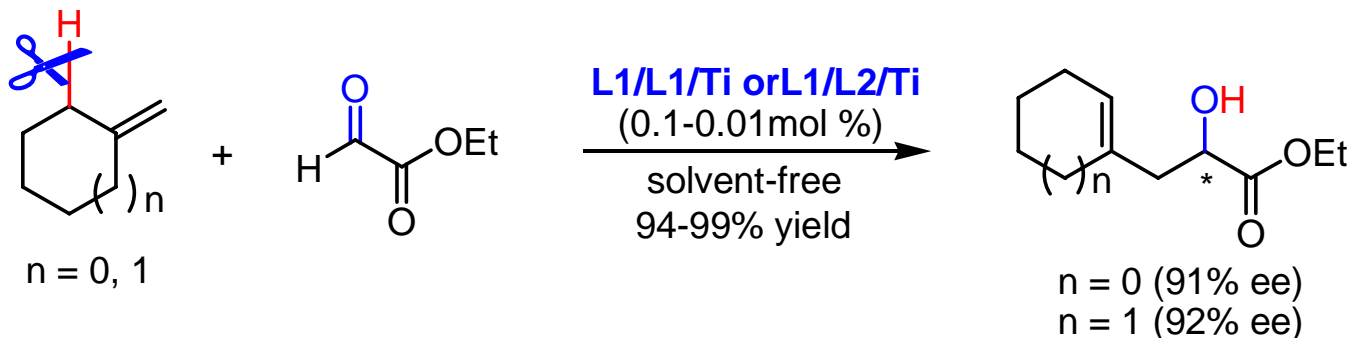


L1



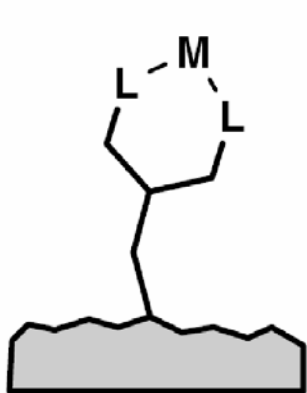
L2

Exceptionally Efficient Catalysts for Quasi Solvent-Free Enantioselective Carbonyl-Ene Reaction

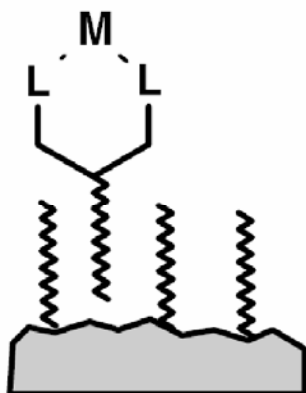


General Strategy for Chiral Catalyst Immobilization

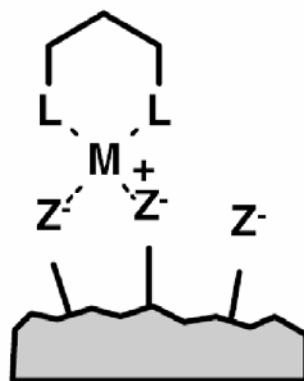
Bonding Patterns



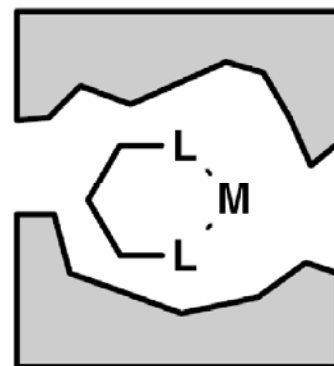
Covalent
bonding



Physisorption



Electrostatic
interactions



Entrapment

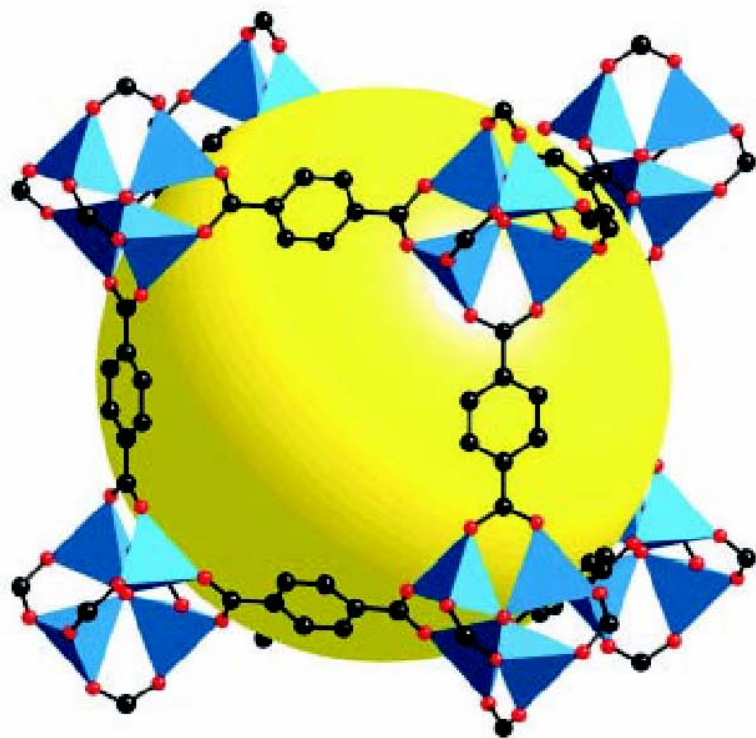
Supports or Media

- ★ Organic Polymers and Dendrimers;
- ★ Inorganic Supports;
- ★ Non-Conventional Media (such as water, fluorinated liquids, ionic liquids, and Sc CO_2 ...)

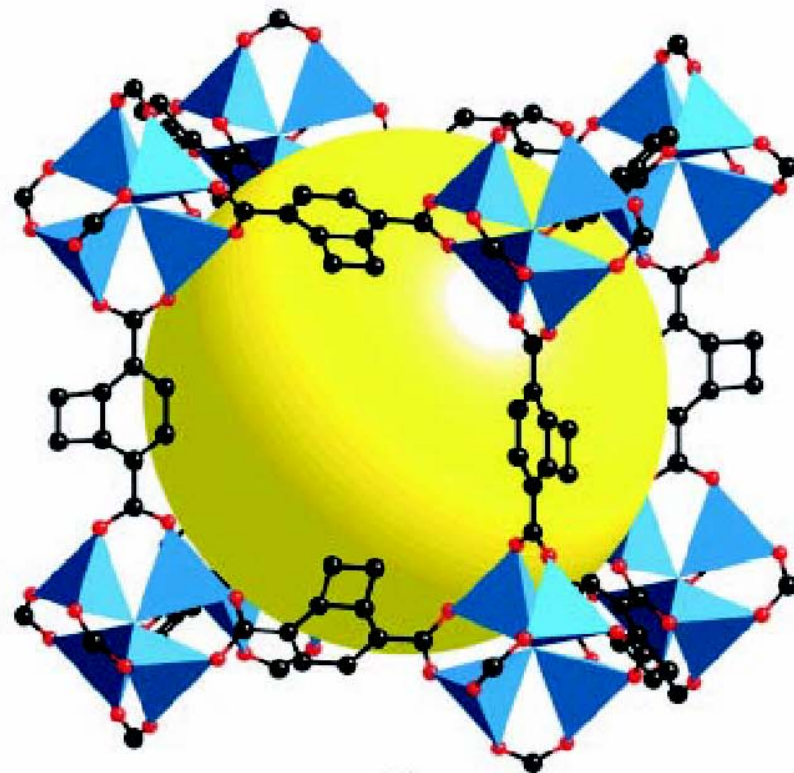
Fan, Q.; Li, Y.-M.; Chan, A. S. C. *Chem. Rev.* 2002, 102, 3385.

D. E. De Vos, I. F. J. Van Keulecom, P. A. Jacobs, Eds. *Chiral Catalyst Immobilization and Recycling*, Wiley-VCH, Weinheim, 2000.

Microporous Metal-Organic Frameworks



A

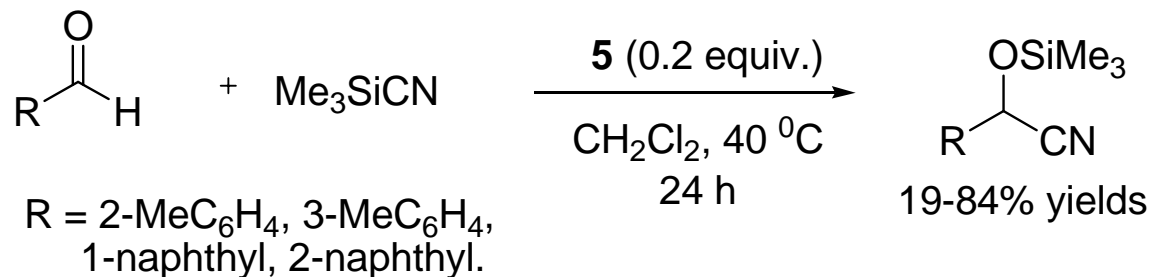
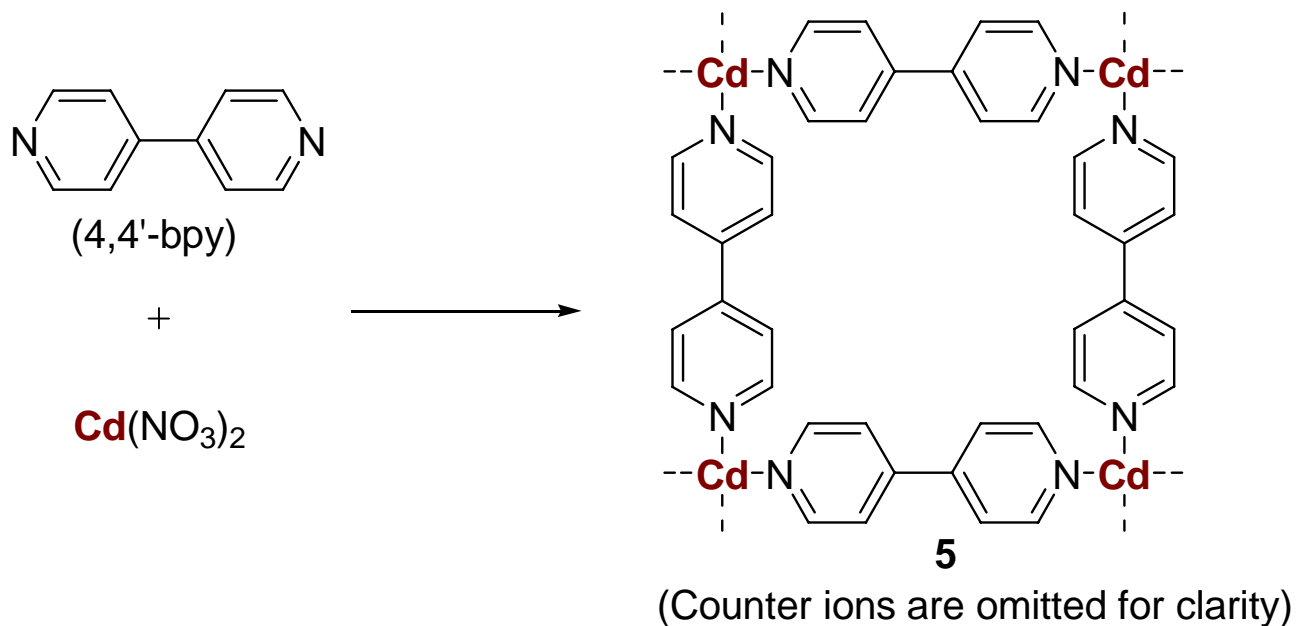


B

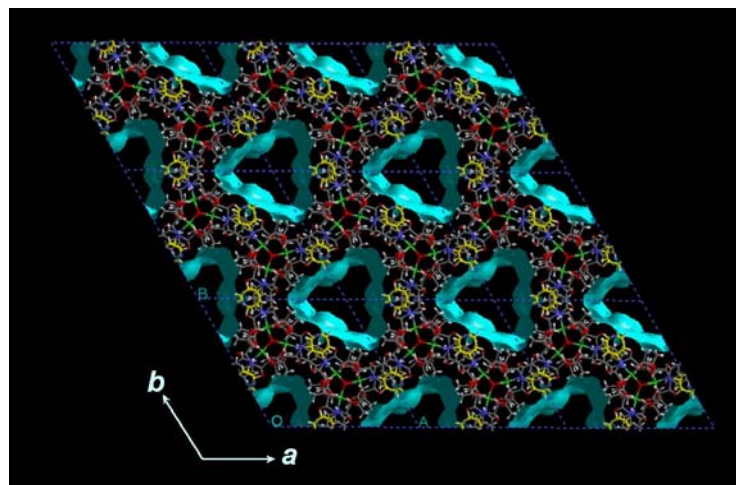
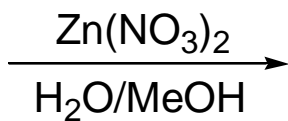
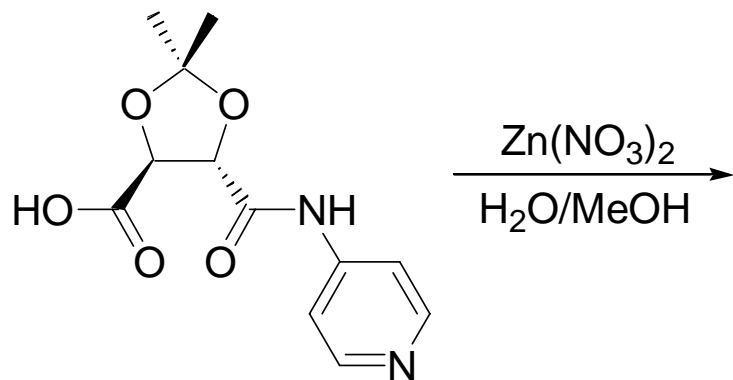
O. M. Yahgi, et al. *Nature*, **2003**, 423, 705;

Science, **2003**, 300, 1127.

Coordination Network Material as A Zeolite-Like Heterogeneous Catalyst

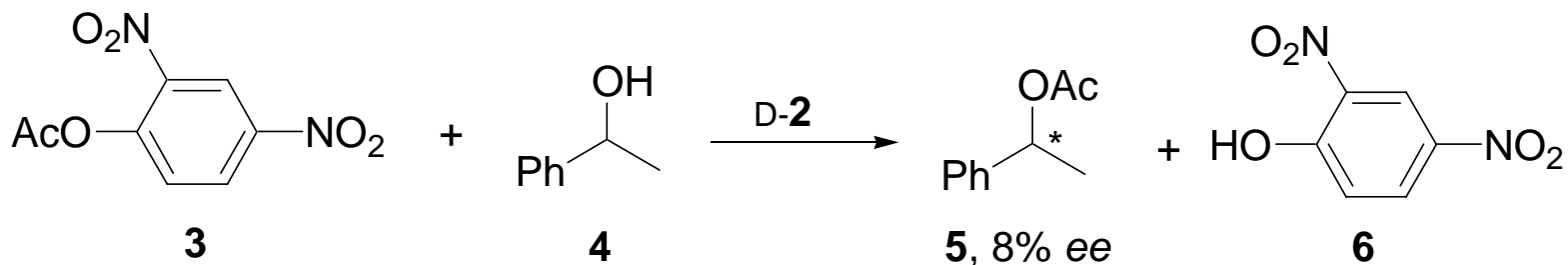


Microporous Homochiral Metal-Organic Frameworks



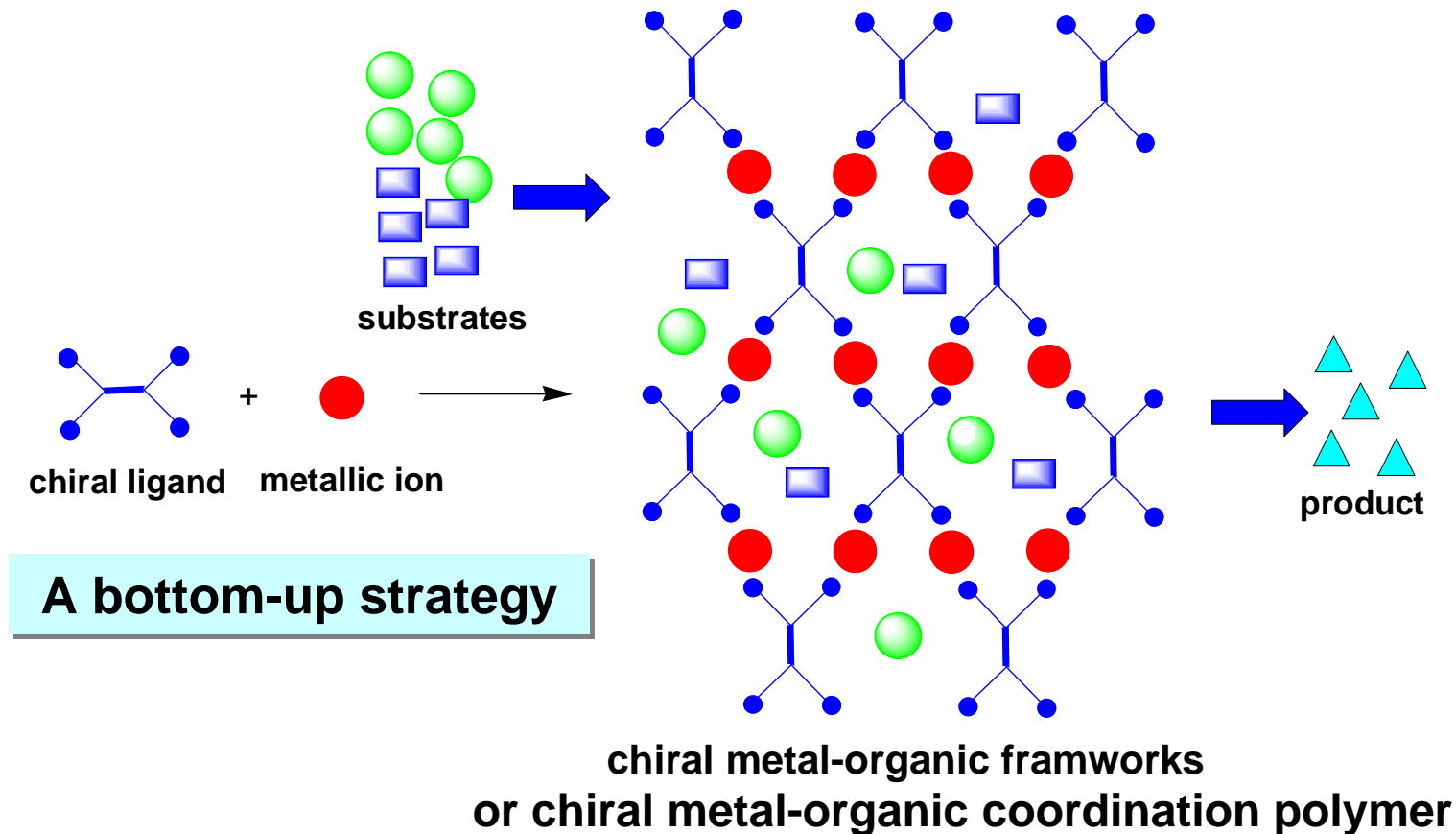
1

2



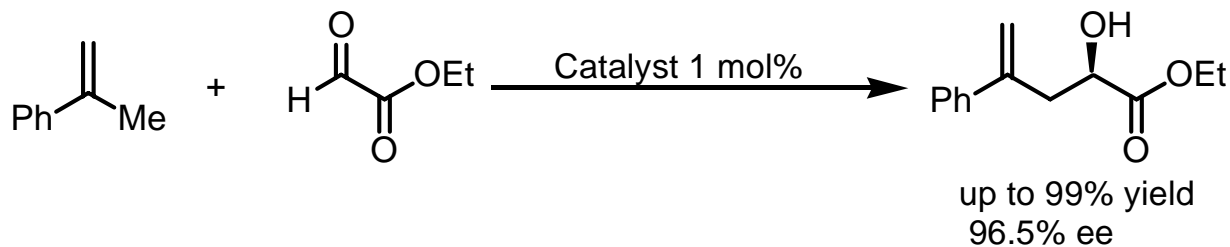
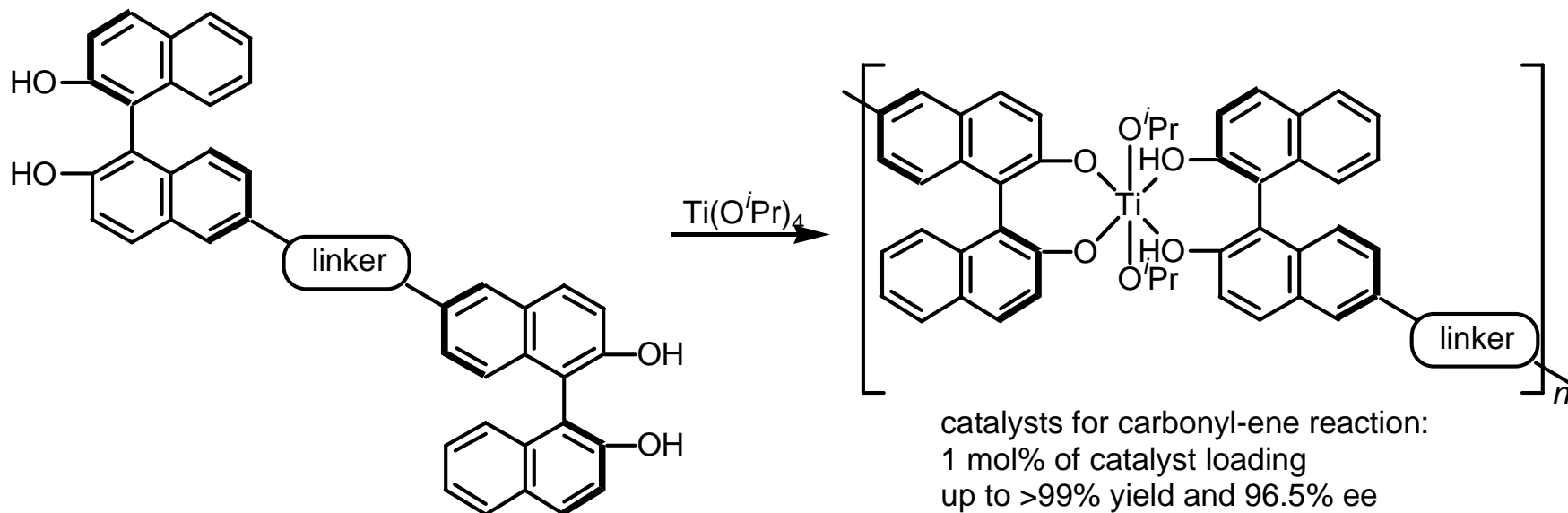
K. Kim et al., *Nature*, **2000**, 404, 982.

Self-Supported Chiral Catalysts for Heterogeneous Enantioselective Catalysis



K. Ding et al. *Chem. Eur. J.* **2006**, *12*, 5188-5197.

Self-Supported BINOL-Ti Catalysts for Carbonyl-Ene Reaction



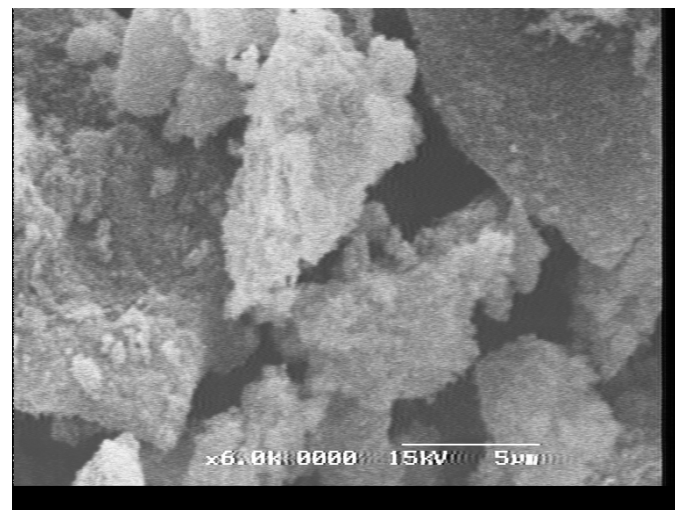
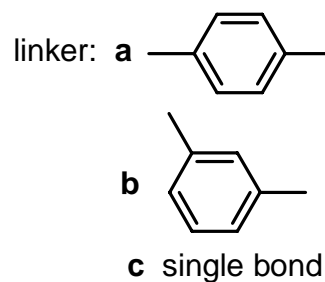
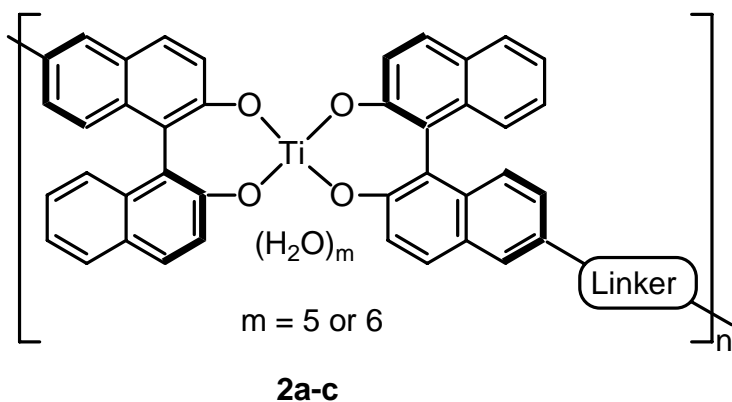
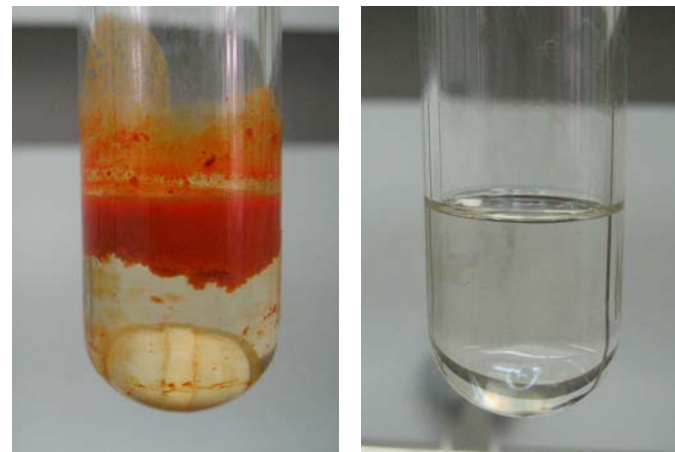
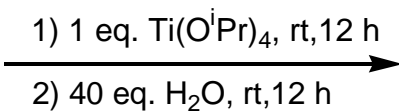
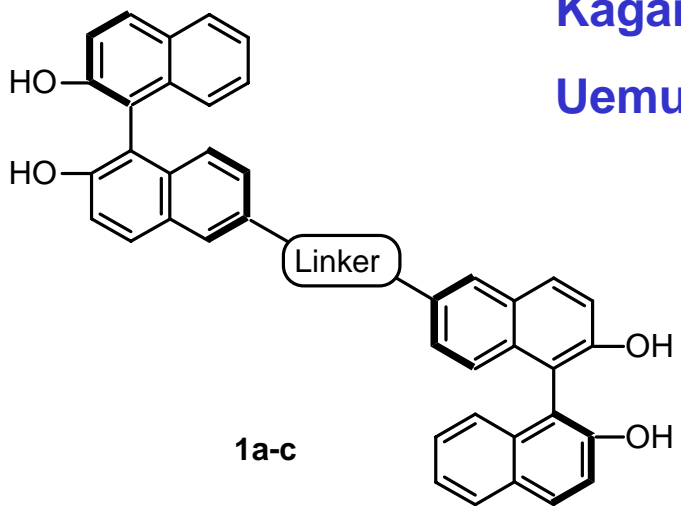
H. Sasai et al., *Angew. Chem. Int. Ed.* **2003**, *42*, 5711-5714.

H. Guo, X. Wang et al, *Tetrahedron Lett.* **2004**, *45*, 2009-2012.

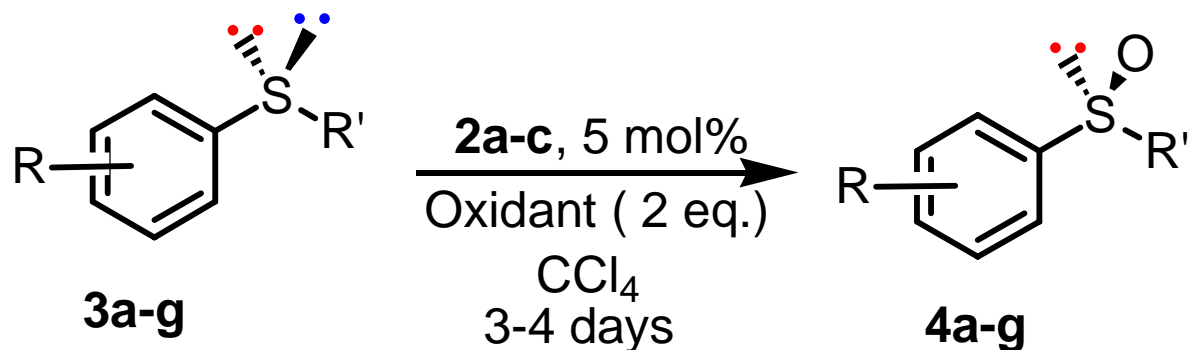
X. Wang, H. Guo et al. *Chem. Eur. J.* **2005**, *11*, 4078-4088.

Highly Stable and Enantioselective Heterogeneous Titanium Catalysts for Asymmetric Oxidation of Sulfides

Kagan, Modena,
Uemura, Bolm



Highly Stable and Enantioselective Heterogeneous Titanium Catalysts for Asymmetric Oxidation of Sulfides



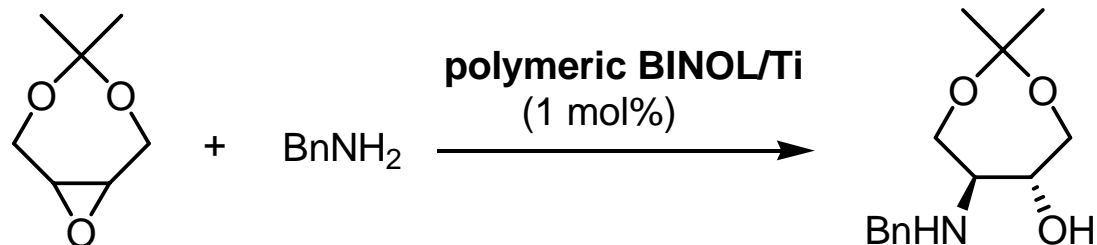
- 3a:** R = H, R' = Me
3b: R = 4-Me, R' = Me
3c: R = 4-Br, R' = Me
3d: R = 4-F, R' = Me
3e: R = 3-Br, R' = Me
3f: R = NO₂, R' = Me
3g: R = H, R' = Et

Reused for 8 times,
Life time > 1 month
98.2->99.9% ee

up to >99.9% ee
35-45% yield

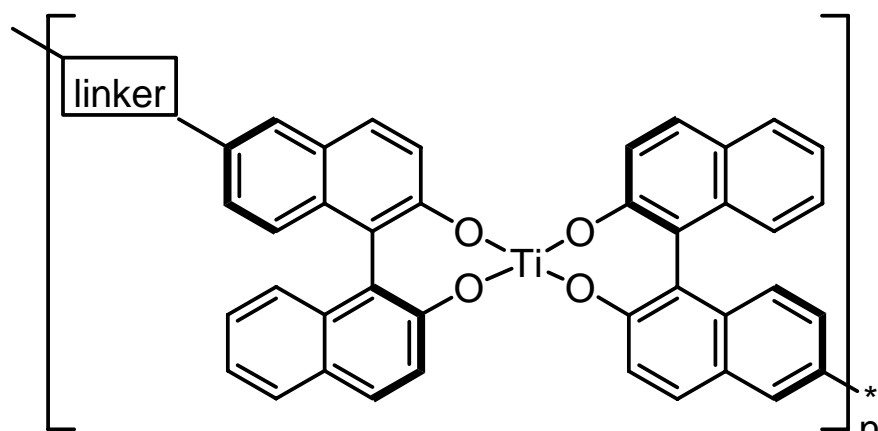
Run	1	2	3	4	5	6	7	8
Y (%)	39	34	42	39	30	38	32	34
Ee (%)	98.2	99.1	99.2	99.2	99.0	99.1	>99.9	99.1

Heterogeneous Titanium Catalysts for Asymmetric Ring Opening of Epoxide



recycled 12 times

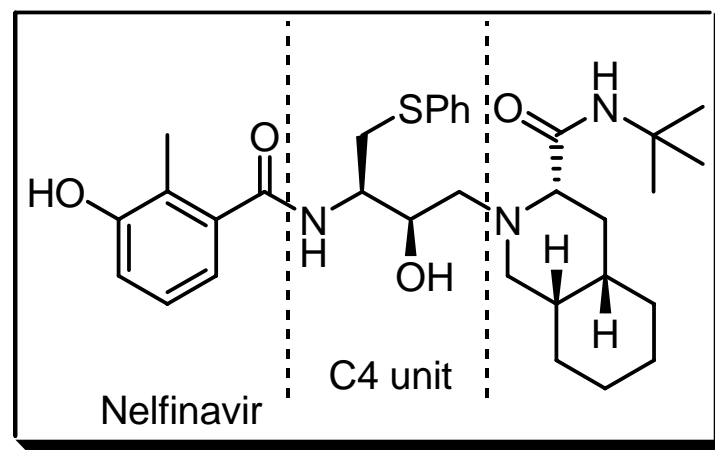
>90% yield
94-97% ee



polymeric BINOL/Ti

linker = 1,4-phenylene

linker = $-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2-$

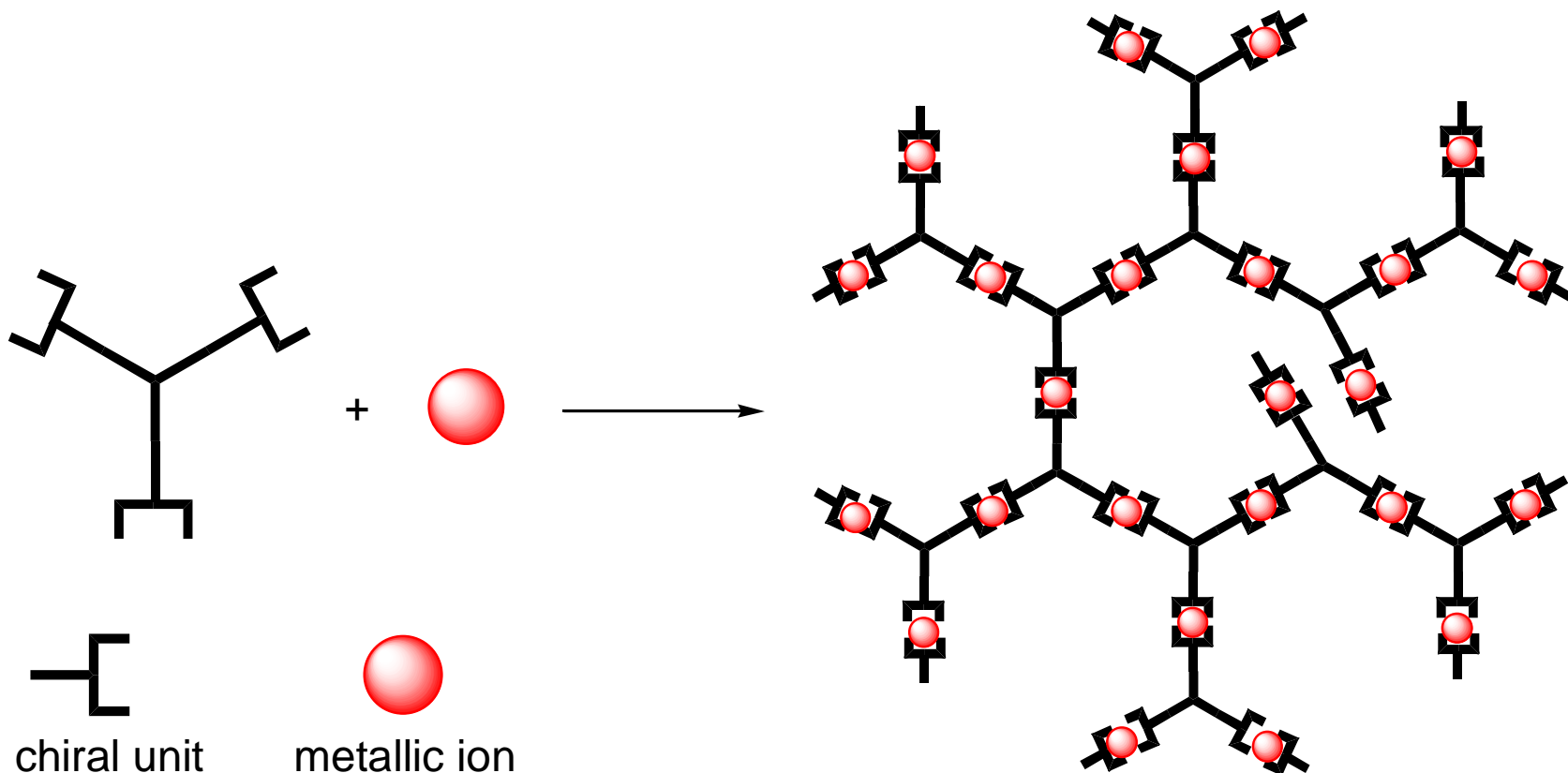
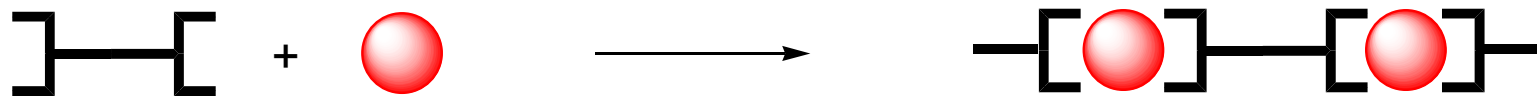


Cf.

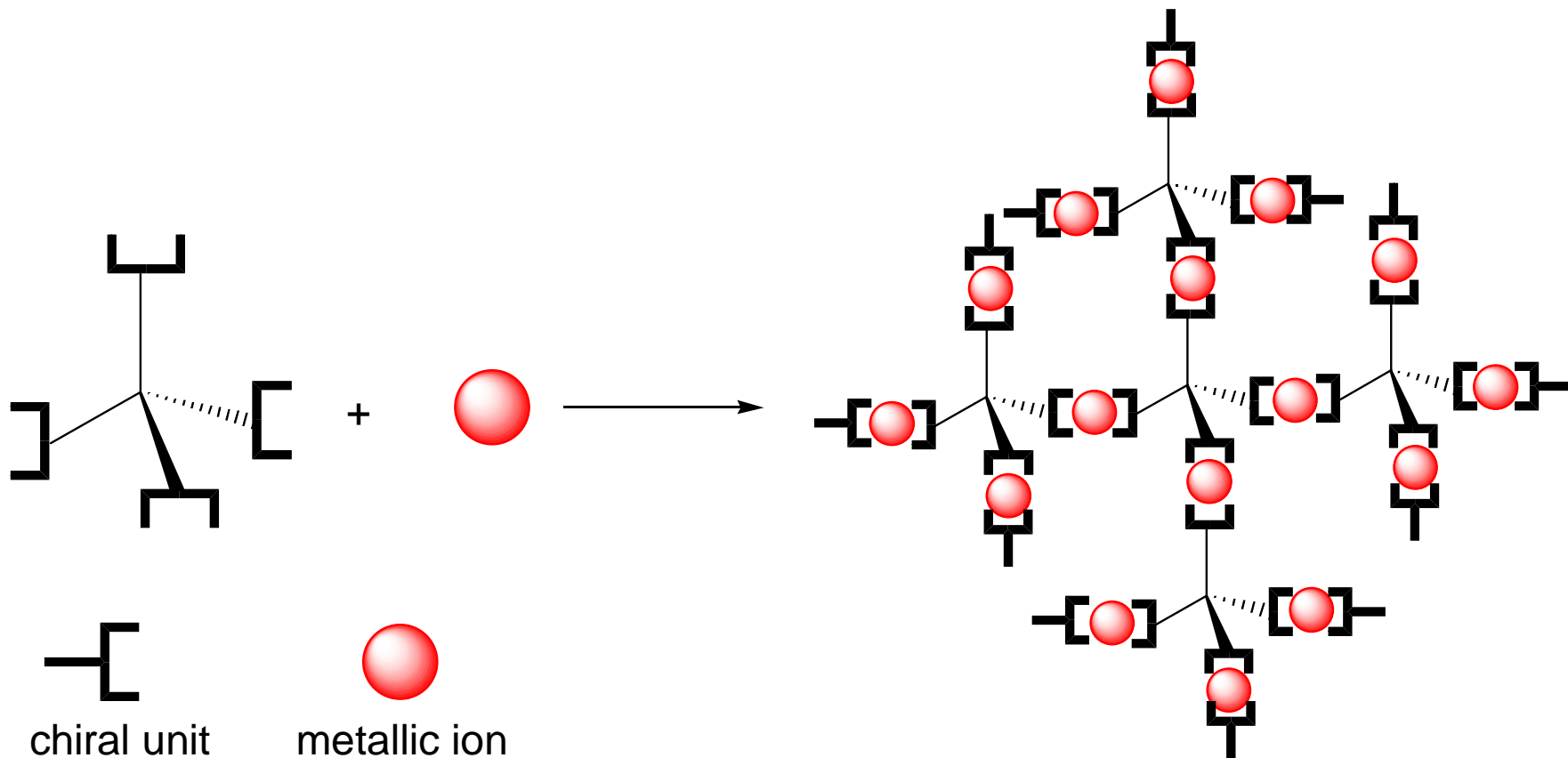
Sagawa, S.; Inaba, T. *J. Org. Chem.* **1999**, *64*, 4962.

H. Bao, Z. Wang & K. Ding et al., *J. Am. Chem. Soc.* **2008**, in the press.

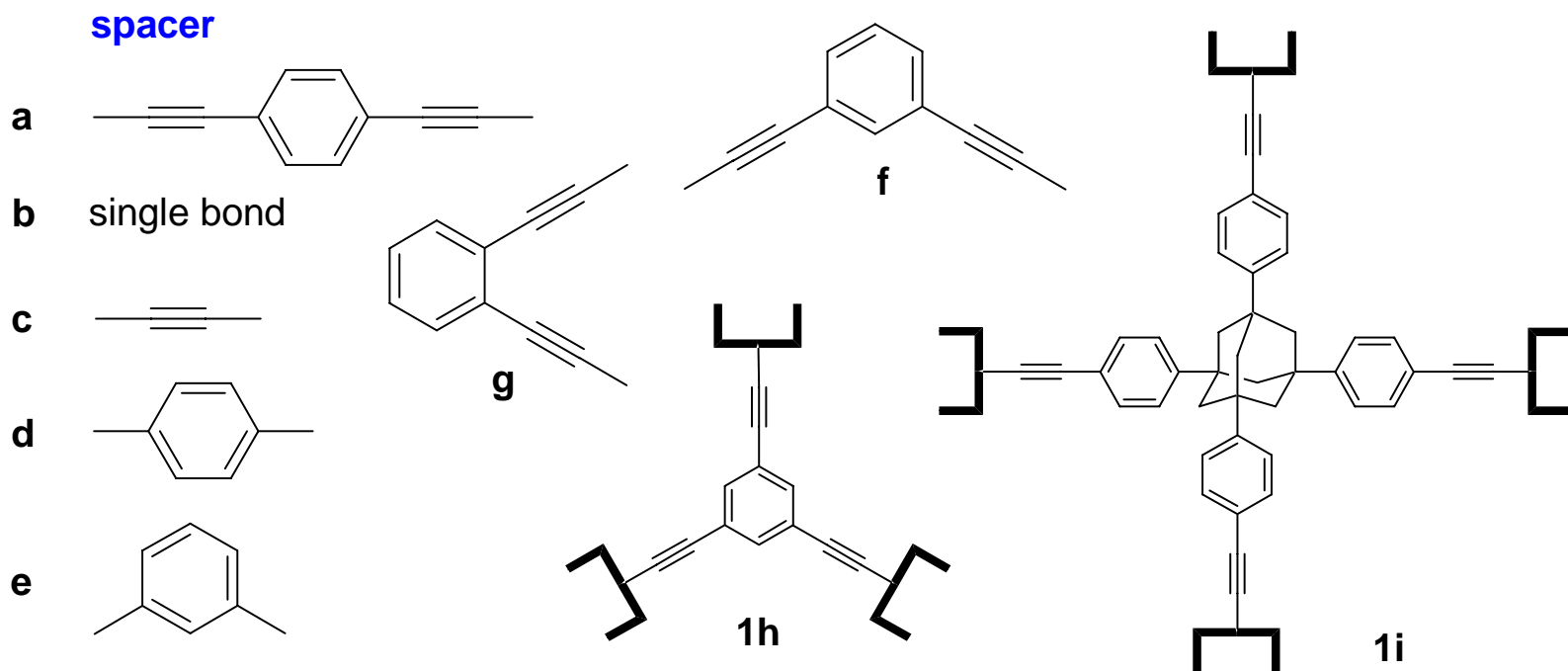
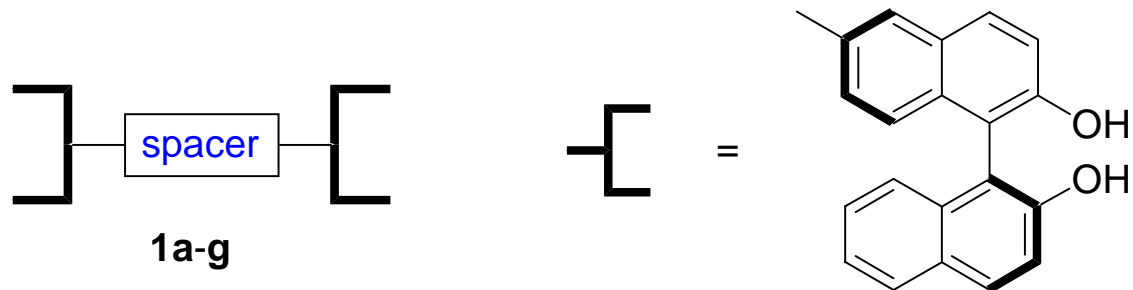
The Impact of Bridging Spacers on the Self-Supported Catalysts



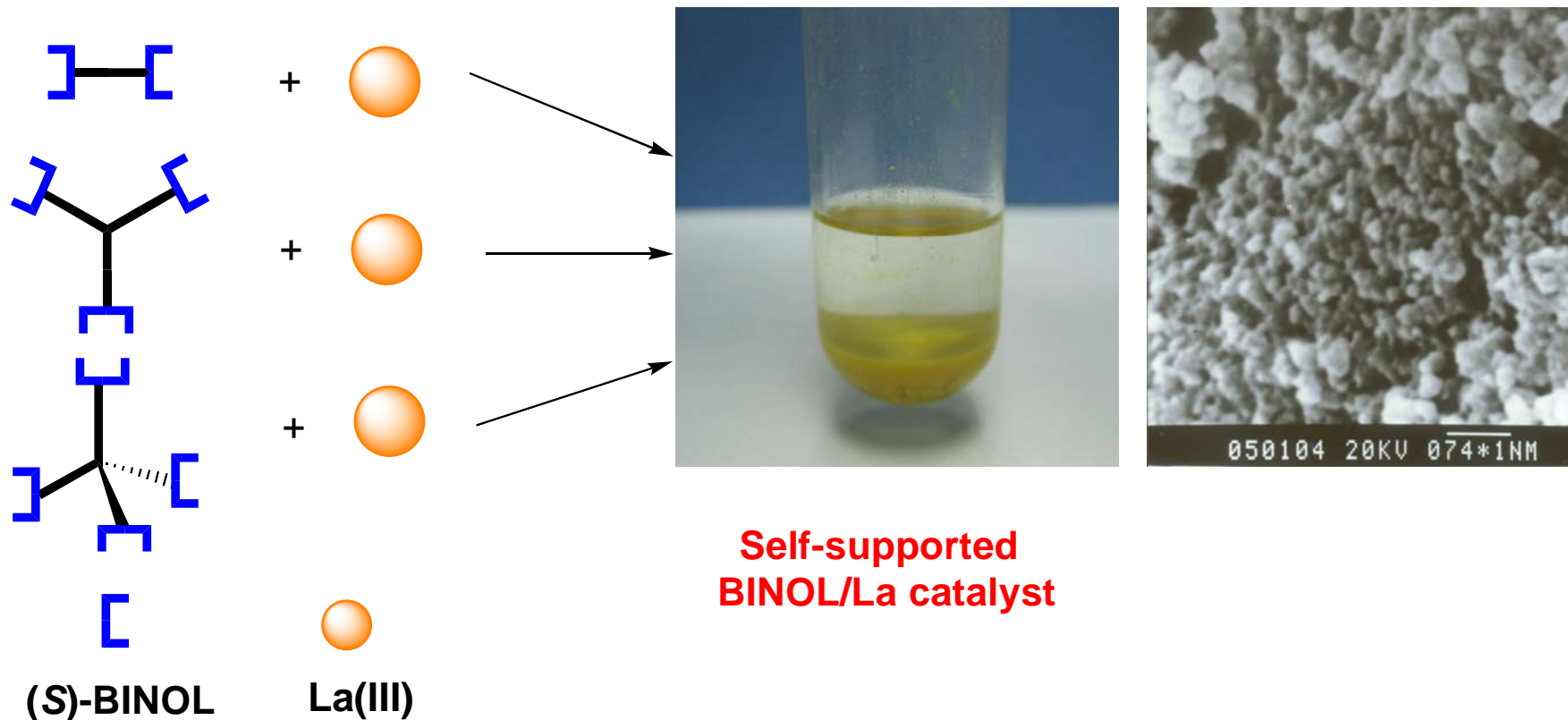
The Impact of Bridging Spacers on the Self-Supported Catalysts



Multitopic Ligands Containing BINOL Units with Various Bridging Spacers



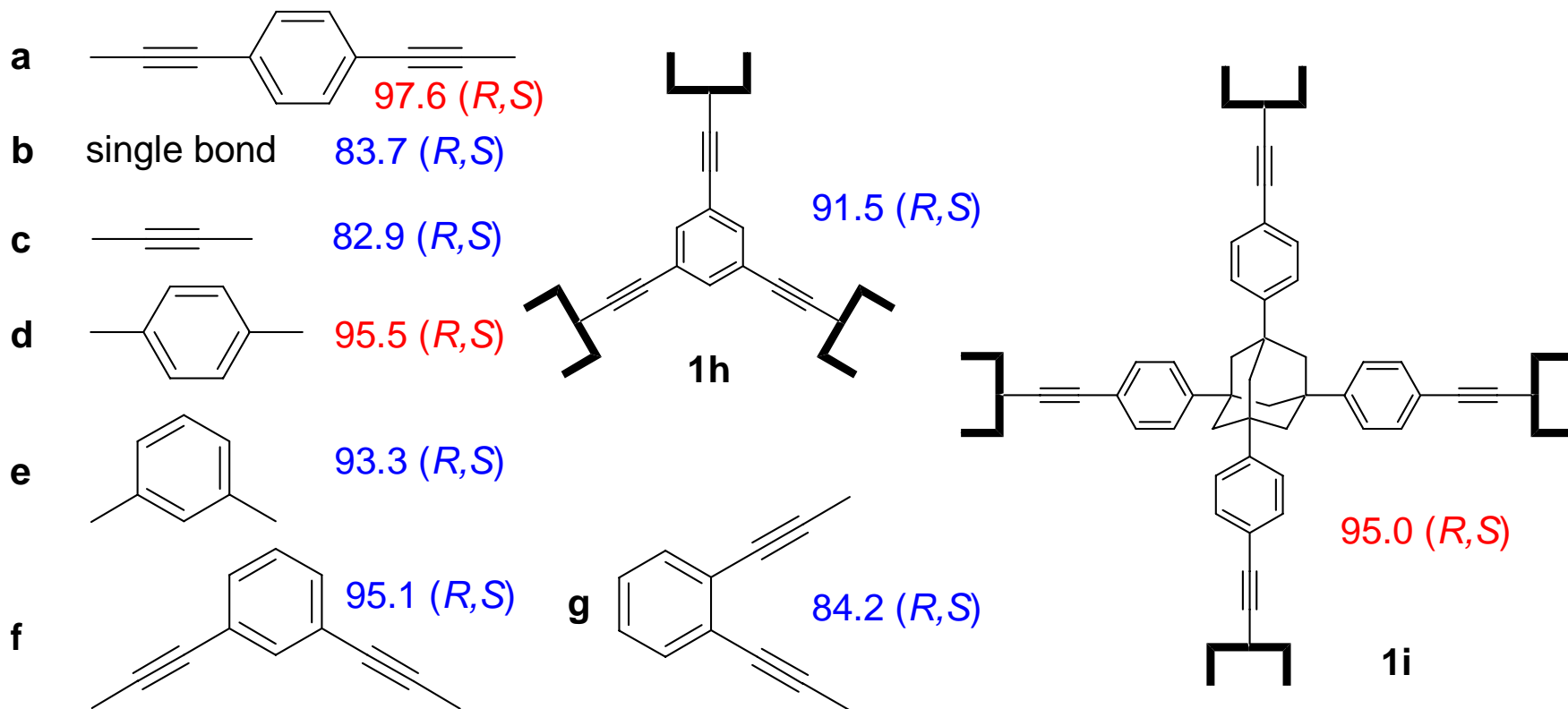
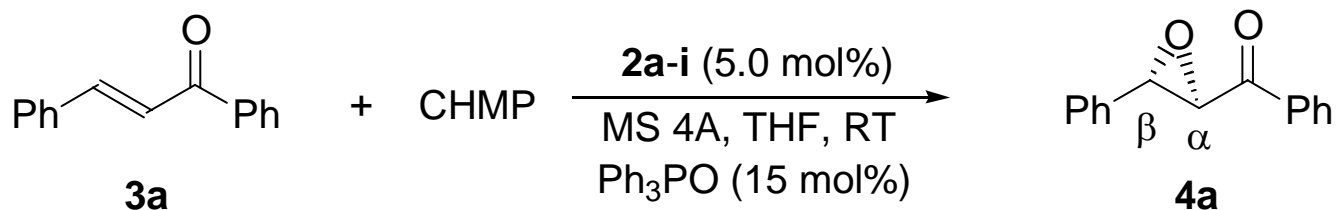
Self-Supported BINOL-La Catalysts with Various Bridging Spacers



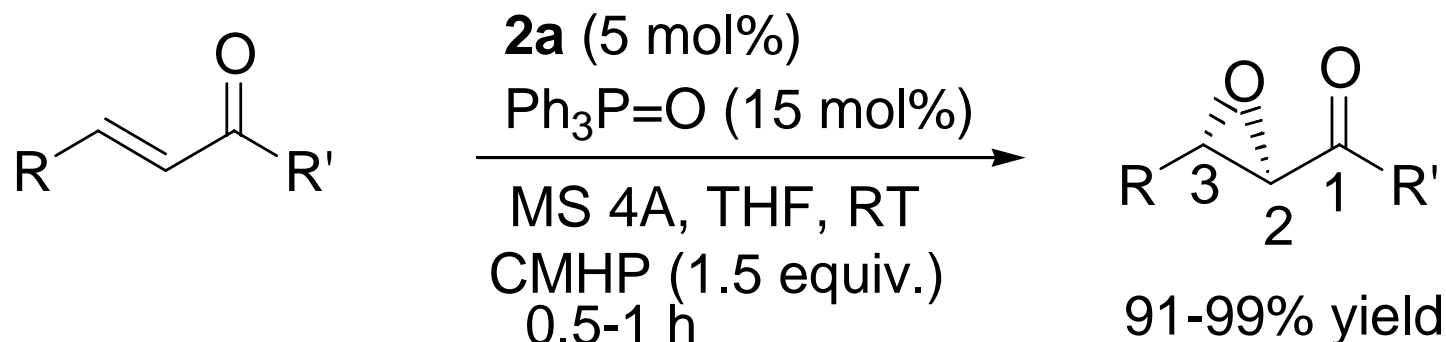
Cf.

M. Bougauchi, S. Watanabe, T. Arai, H. Sasai, M. Shibasaki, *J. Am. Chem. Soc.* **1997**, *119*, 2329.
K. Daikai, M. Kamaura, J. Inanaga, *Tetrahedron Lett.* **1998**, *39*, 7321-7322

Self-Supported BINOL-La Catalysts for Epoxidation of Enones

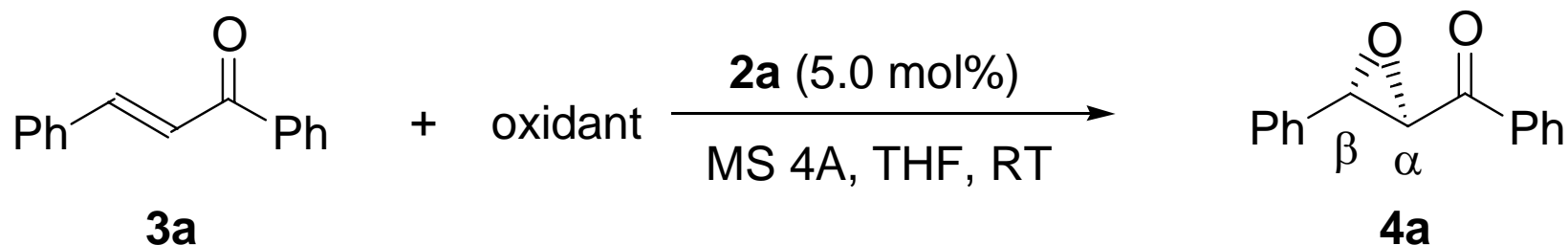


Self-Supported BINOL-La Catalysts for Epoxidation of Enones



4a: R = Ph, R' = Ph	97.6% ee (<i>R,S</i>)
4b: R = 4-F-Ph, R' = Ph	96.2% ee (<i>R,S</i>)
4c: R = 4-Cl-Ph, R' = Ph	96.0% ee (<i>R,S</i>)
4d: R = 4-Br-Ph, R' = Ph	95.6% ee (<i>R,S</i>)
4e: R = 4-NO ₂ -Ph, R' = Ph	95.7% ee (<i>R,S</i>)
4f: R = 4-NC-Ph, R' = Ph	94.3% ee (<i>R,S</i>)
4g: R = Ph, R' = 4-MeO-Ph	95.0% ee (<i>R,S</i>)
4h: R = <i>i</i> -Pr, R' = Ph	84.9% ee (<i>R,S</i>)

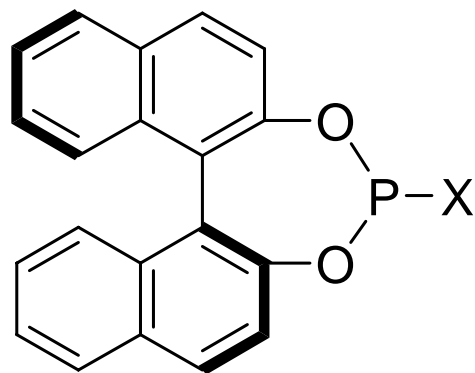
Recovery and Reuse of Self-Supported BINOL-La Catalysts



Run	Ph ₃ PO [mol%]	Time [h]	Yield [%] ^[b]	Ee ^[c] [%] (Config.) ^[d]
1	15	0.5	>99	96.5 (R,S)
2	10	0.5	>99	96.3 (R,S)
3	10	0.5	>99	95.8 (R,S)
4	10	0.5	>99	94.9 (R,S)
5	10	0.5	95	94.5 (R,S)
6	10	1	83	93.2 (R,S)

La leaching:
<0.4 ppm

Modular Monodentate Phosphorous Ligands for Rh(I)-Catalyzed Asymmetric Hydrogenation



1a X = OR

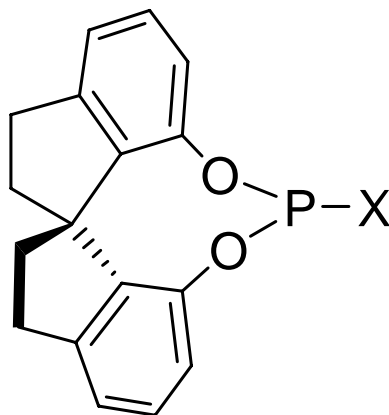
1b X = alkyl or aryl

1c X = NR₂, MonoPhos

M. Reetz (2000)

B. L. Feringa (2000)

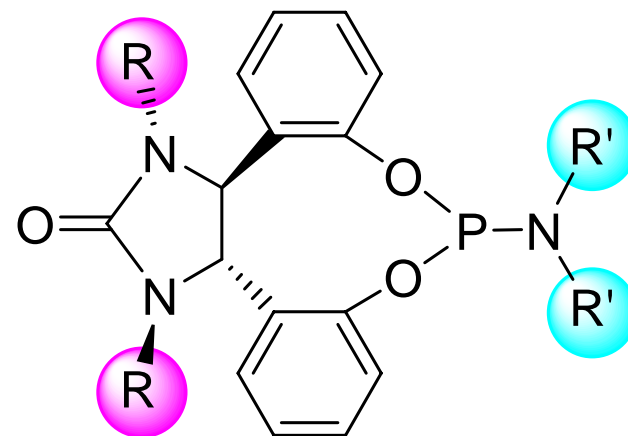
P. Pringle (2000)



2a X = OR

2b X = NR₂, SpiroPhos

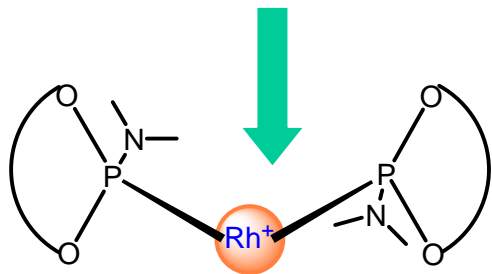
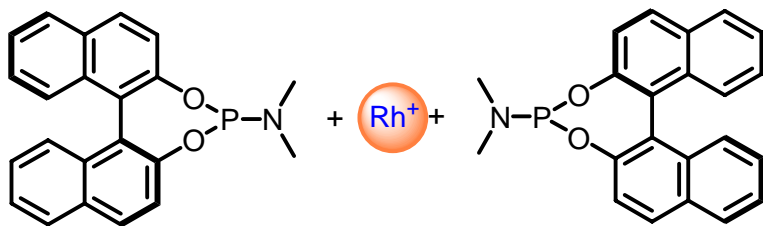
Q. Zhou (2002)



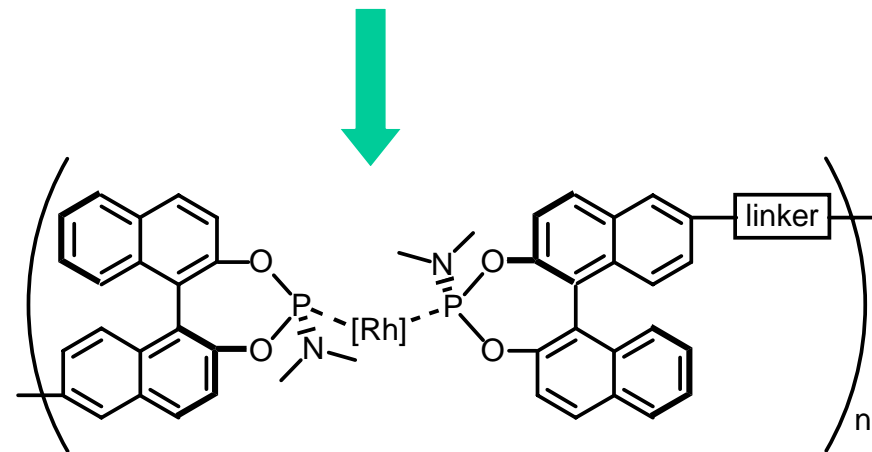
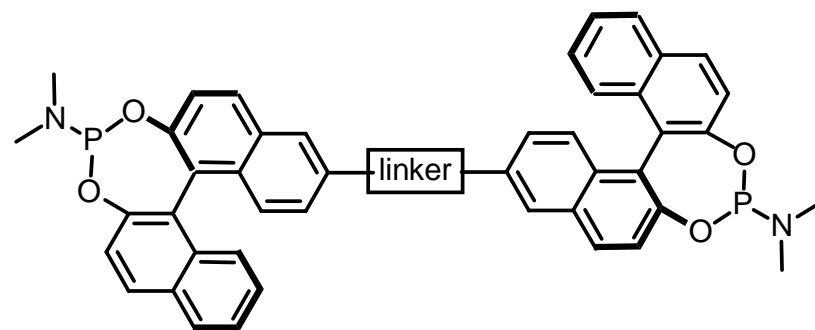
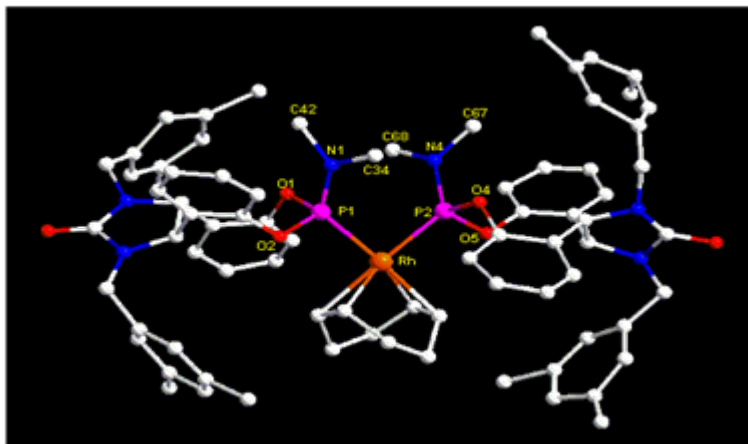
3, DpenPhos

K. Ding (2005)

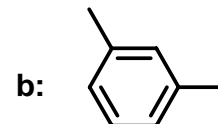
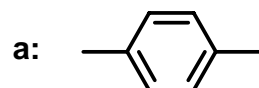
MonoPhos/Rh Catalyst: from Homogeneous to Heterogeneous



Excellent catalyst for asymmetric hydrogenation



Linker =

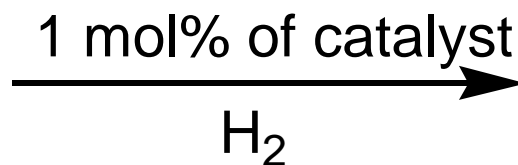
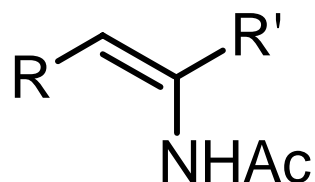


c: single bond

B. L. Feringa, et al, *J. Am. Chem. Soc.* 2000, 122, 11539

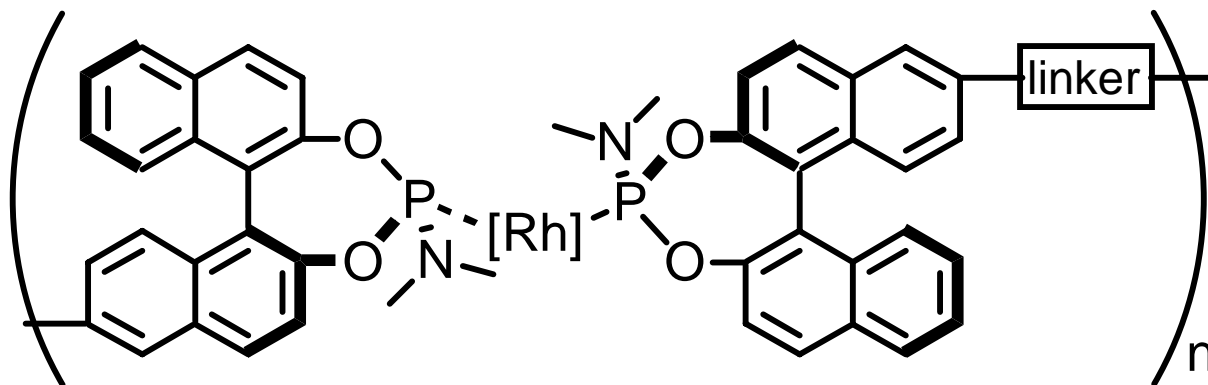
Y. Liu & K. Ding, *J. Am. Chem. Soc.* 2005, 127, 10488-10499

Self-Supported Catalysts for Heterogeneous Enantioselective Hydrogenation

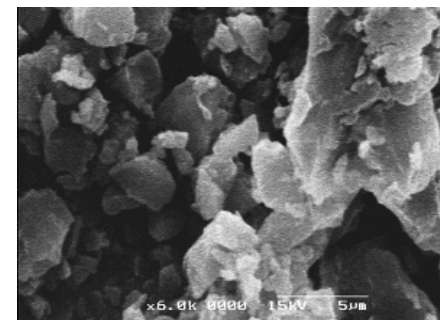


94-97% ee

R = H, Ph, Me; R' = COOMe, H

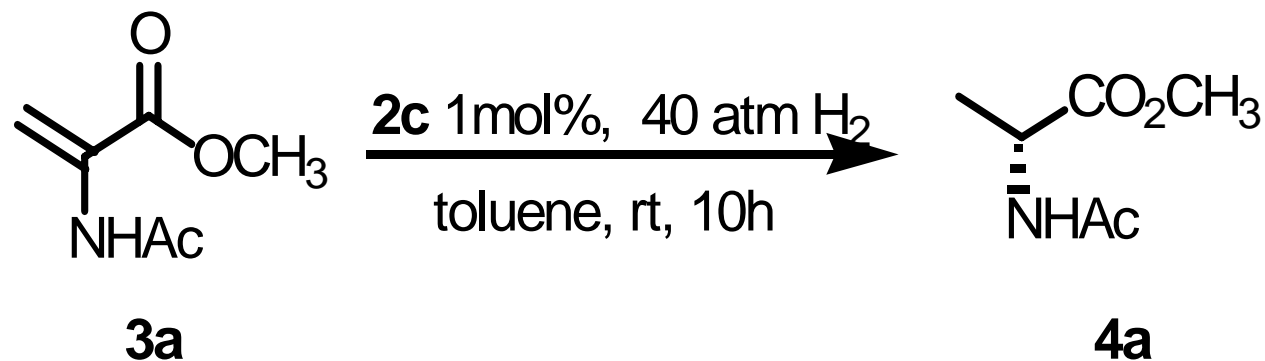


self-supported catalyst



X. Wang *et al.* *J. Am. Chem. Soc.* **2004**, 126, 10524.

Catalyst Recycling

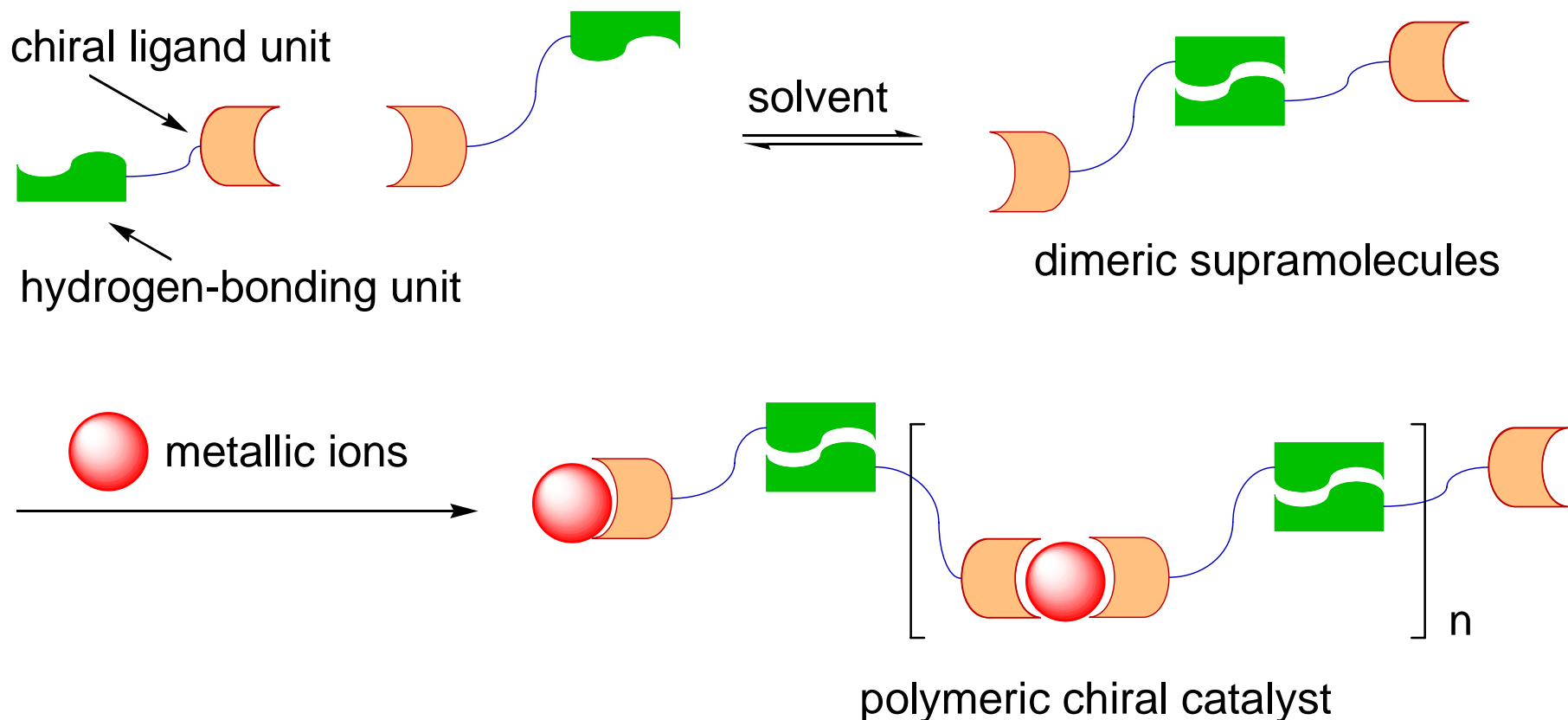


3a

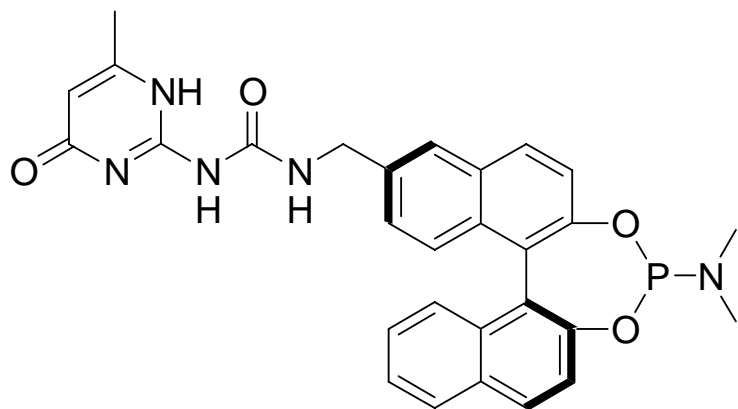
4a

Run	1	2	3	4	5	6	7	8	9	10
Conv.(%) ^b	>99	>99	>99	>99	>99	>99	>99	>99	>99	>99
E.e. (%) ^c	95.0	93.5	90.2	90.9	90.5	90.0	89.5	87.7	87.3	87.3

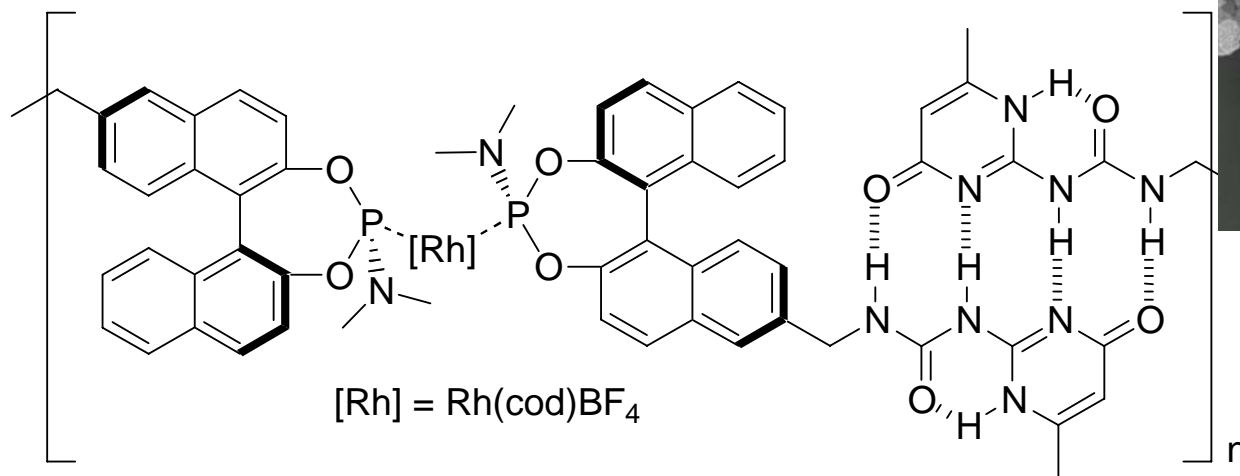
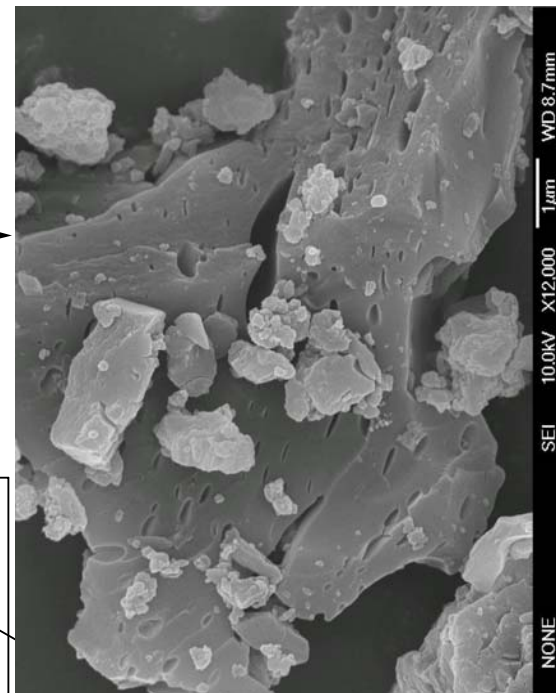
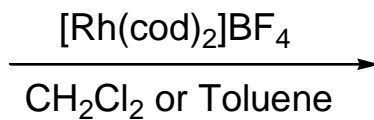
Use of Hydrogen Bonds as the Bridging Linker: From Dimeric Supramolecules to Polymeric Chiral Catalyst



Self-Supported MonoPhos/Rh(I) Catalyst

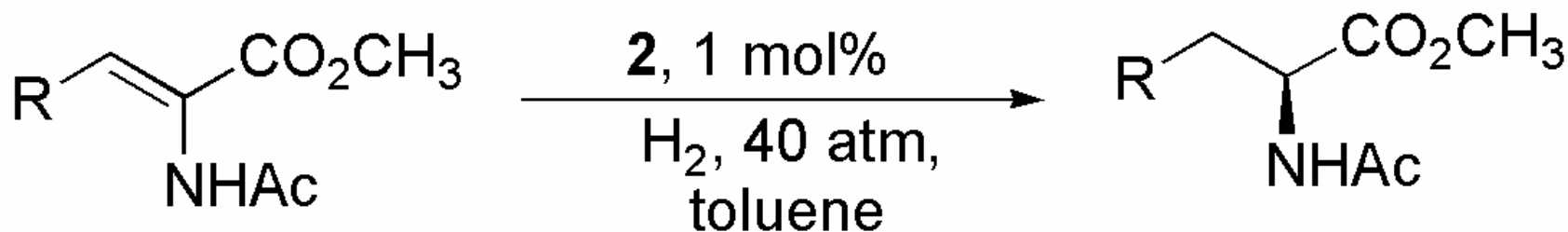


SupraMonoPhos (1a)



2

Self-Supported MonoPhos/Rh(I) Catalyst for Enantioselective Hydrogenation

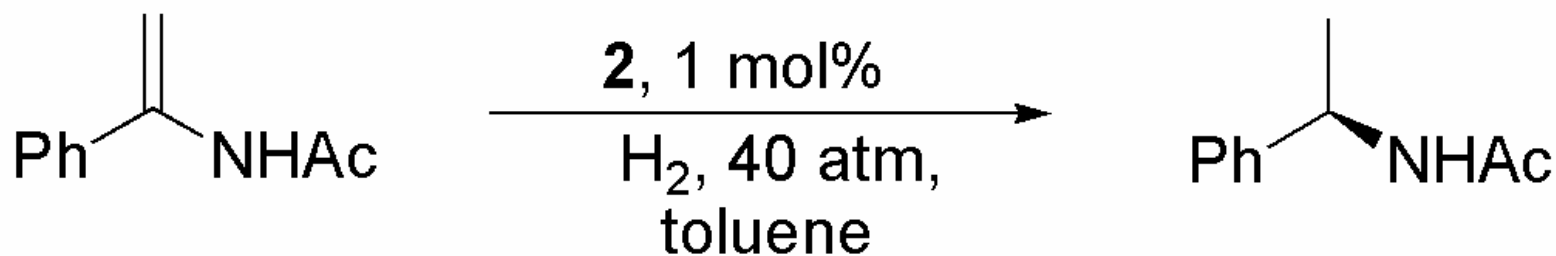


3a-d

4a-d

R = H (a), CH₃ (b), Et (c), *i*Pr (d)

>99% conv.
93-96% ee

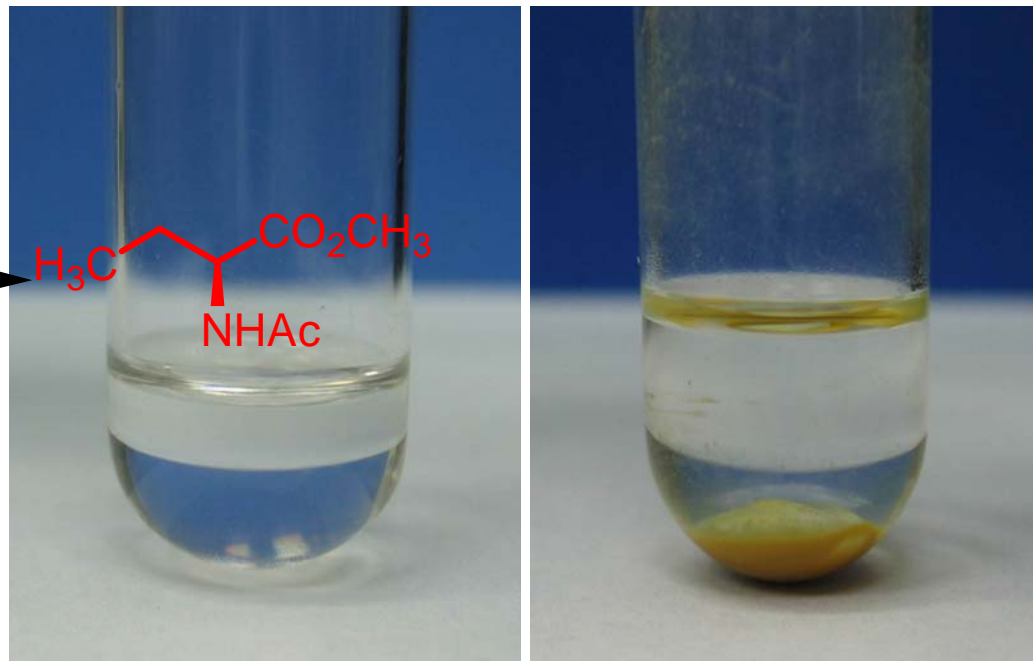
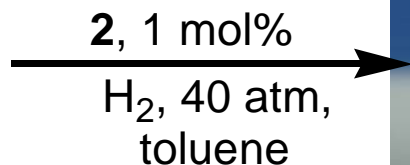
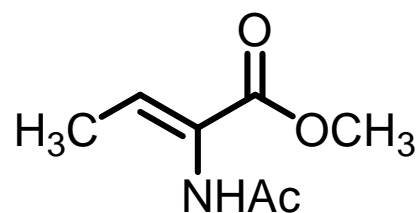


3e

4e

>99% conv.
91% ee

High Enantioselectivity and Facile Recovery

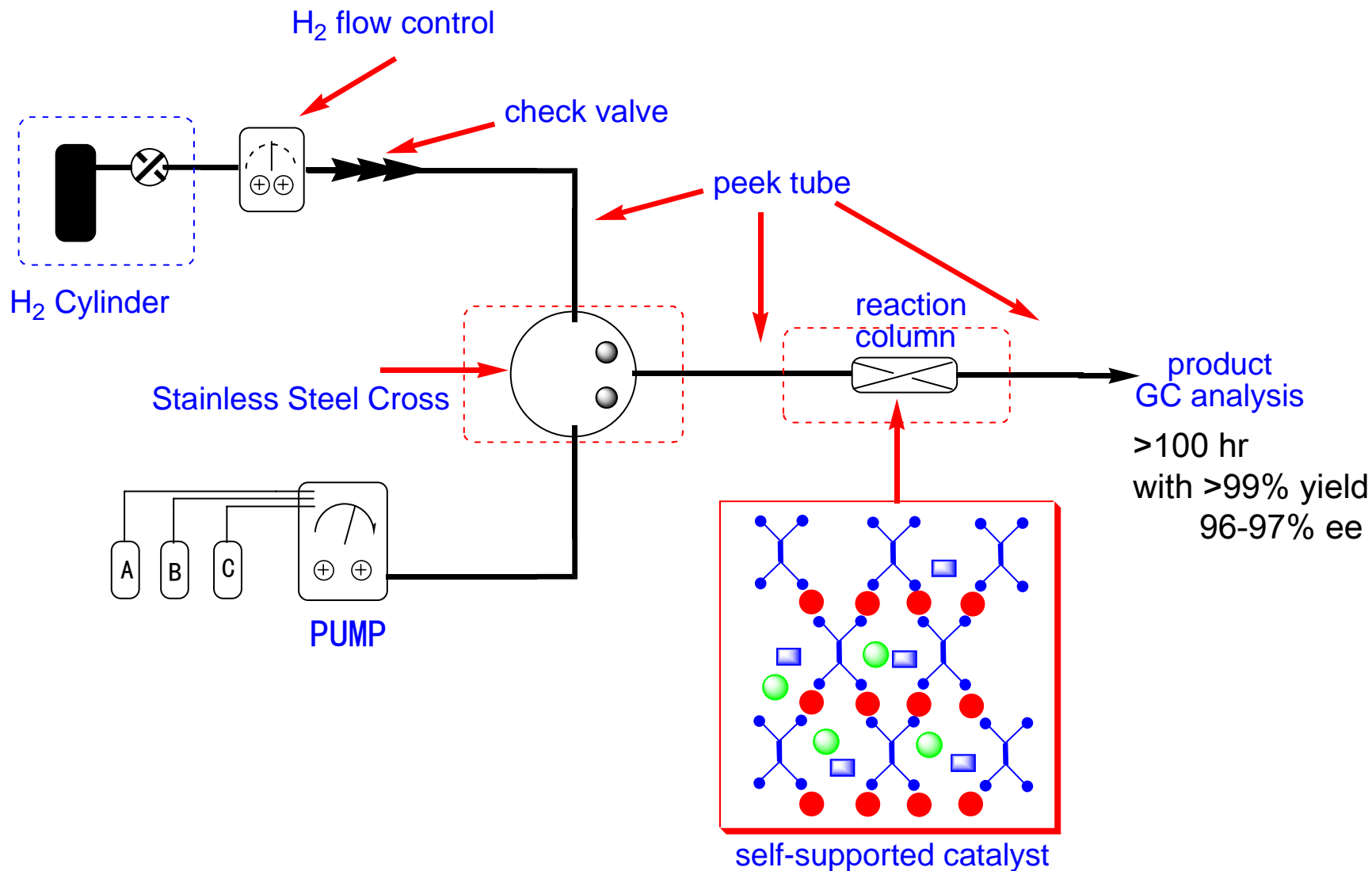


Rh leaching < 1 ppm

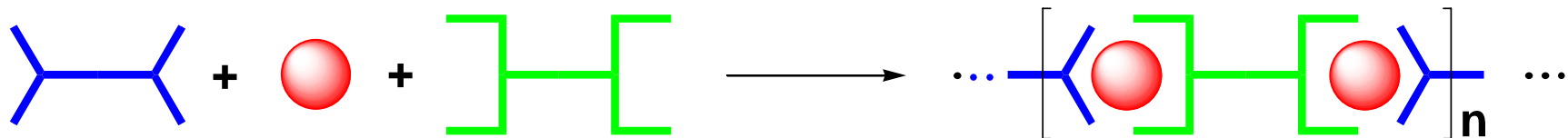
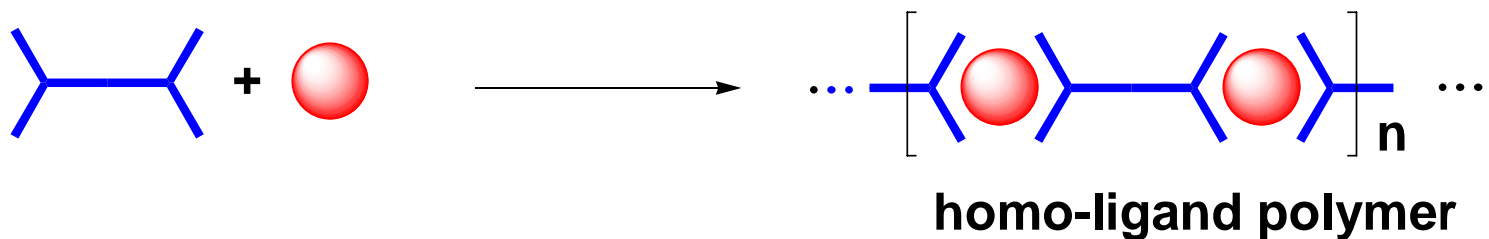
Table 2. Recycling and reuse of the self-supported catalysts 2b in enantioselective hydrogenation of 3b.^a

Run	1	2	3	4	5	6	7	8	9	10	11
conv. [%] ^b	>99	>99	>99	>99	>99	>99	>99	>99	>99	>99	96
Ee [%] ^c	95.7	95.7	95.2	94.7	94.3	94.5	94.6	94.5	92.9	92.0	91.5

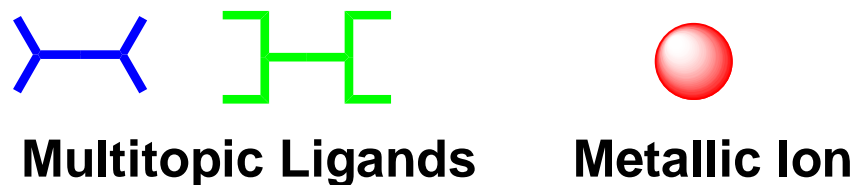
Schematic Representation of Continuous Flow Reaction System for Chiral Catalysis



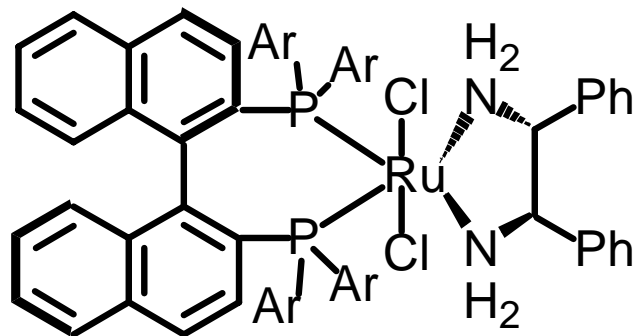
Schematic Representation of the Strategy



hetero-ligand polymer

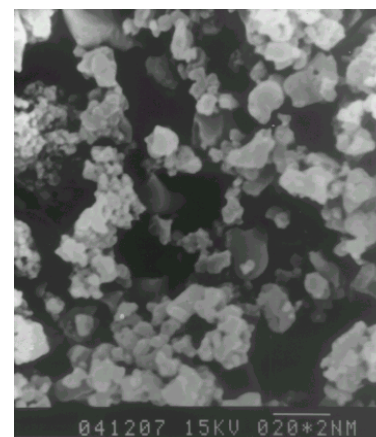
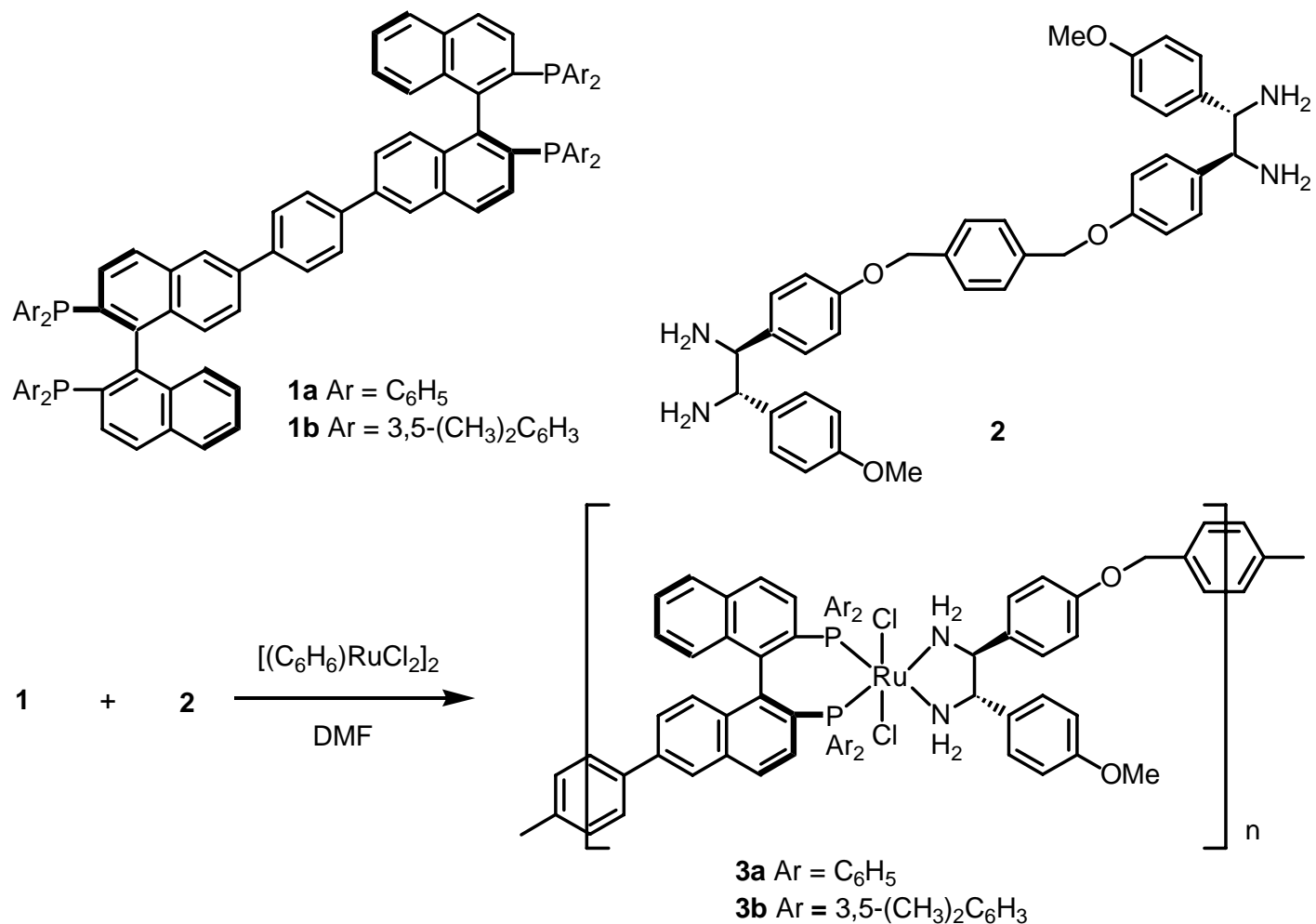


Programmed Assembly



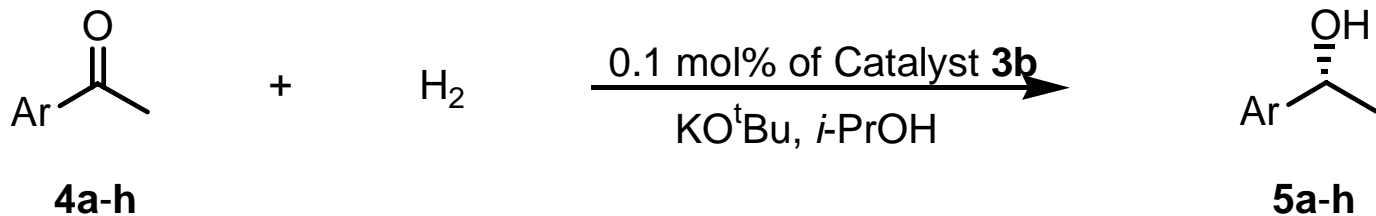
Cf. R. Noyori, et al.
J. Am. Chem. Soc. 1995, 117, 2675.

Self-Supported Noyori-Type Catalyst



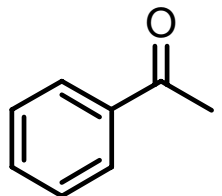
Liang, Y. et al. *J. Am. Chem. Soc.* **2005**, 127, 7694-7695.

Self-Supported Noyori-Type Catalyst for Asymmetric Hydrogenation of Ketones

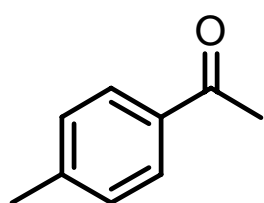


4a-h

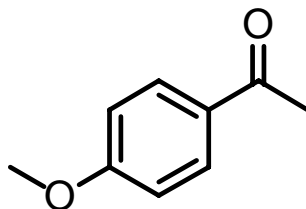
5a-h



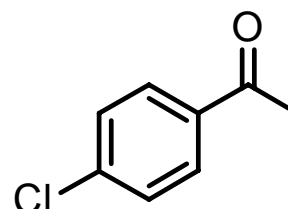
97.4% ee



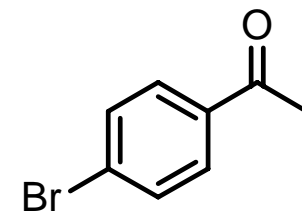
97.5% ee



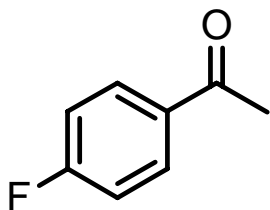
96.2% ee



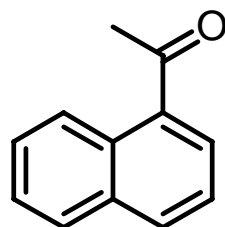
96.9% ee



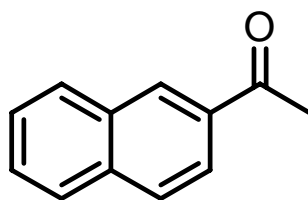
97.2% ee



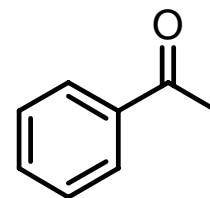
96.2% ee



98.1% ee

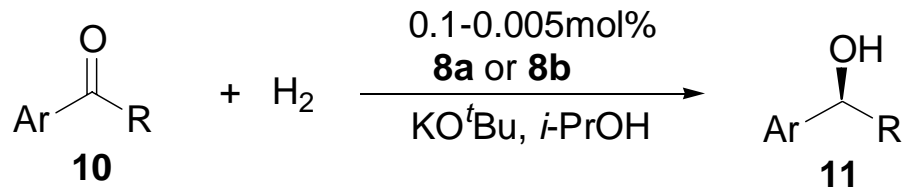
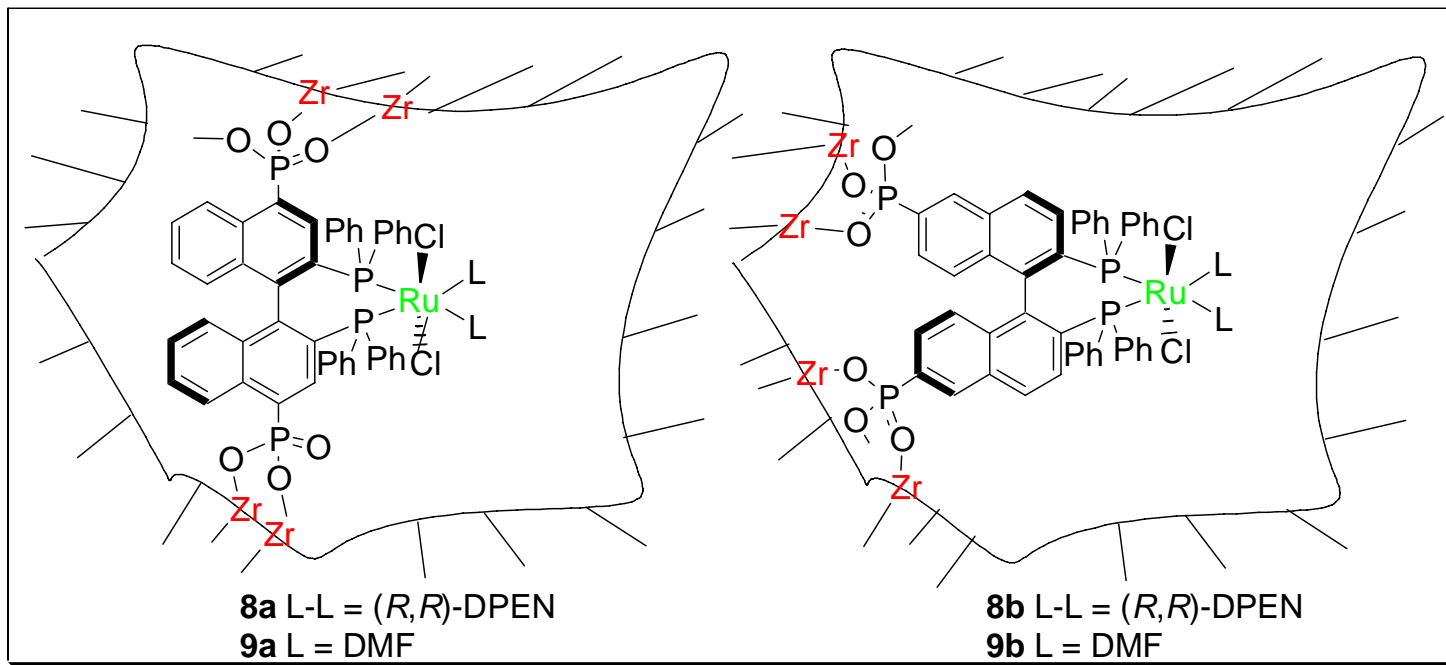


94.5% ee

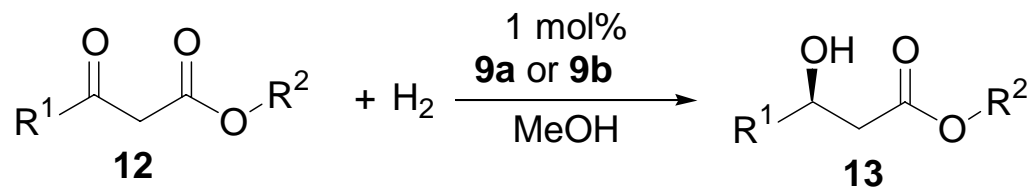


94.2% ee (0.01 mol % of cat)

Run	1	2	3	4	5	6	7
Y (%)	>99	>99	>99	>99	>99	>99	97
Ee (%)	97.4	97.6	97.3	96.5	95.6	96.1	95.4



up to >99% yield; 99% ee, 8 recycles



up to >99% yield, 95% ee, 5 recycles

Lin, W. et al. *J. Am. Chem. Soc.* 2003, 125, 11490;

Lin, W. et al. *Angew. Chem. Int. Ed.* 2003, 42, 6000.

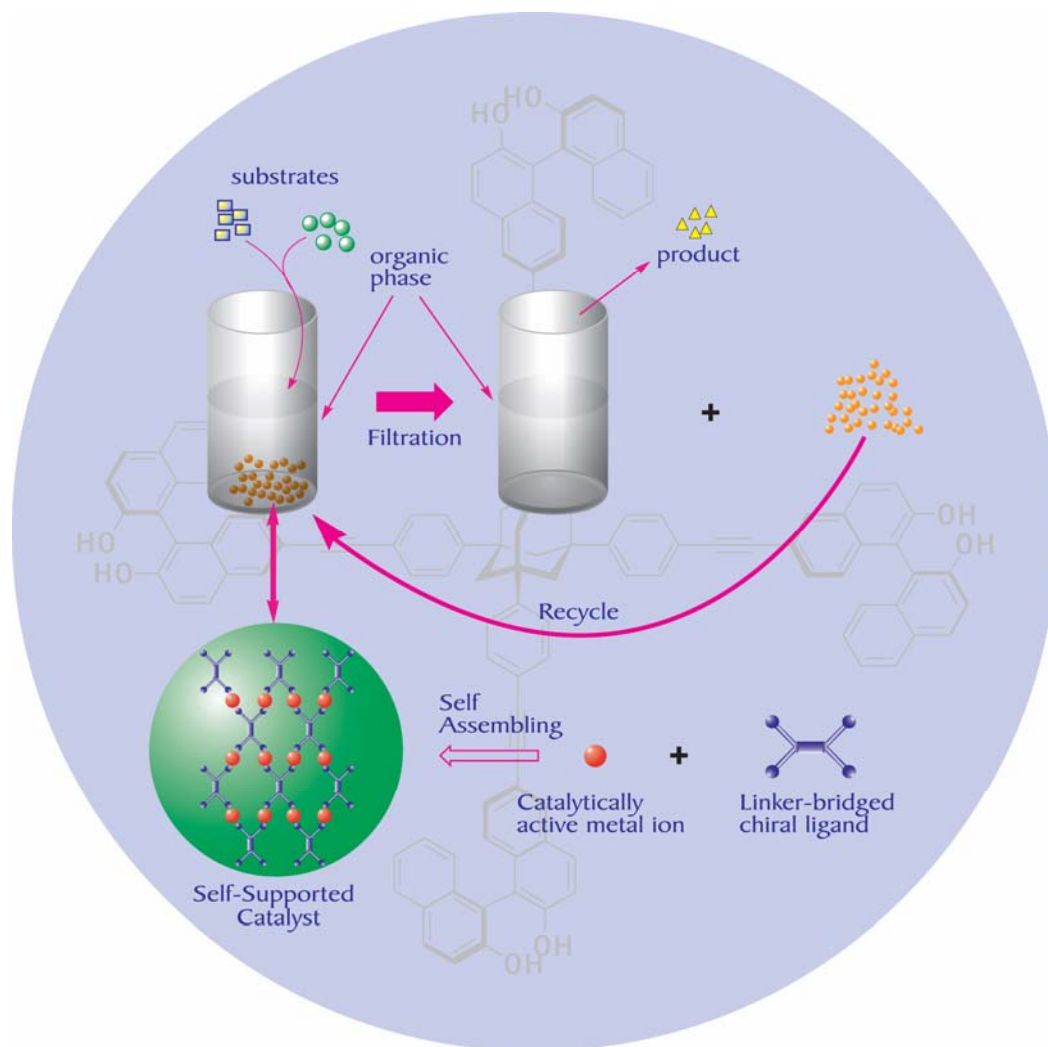
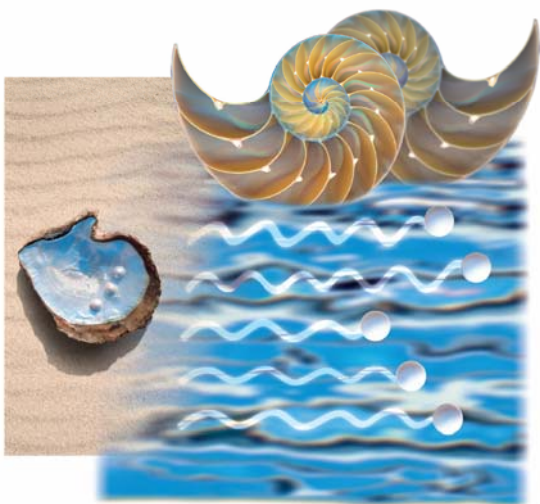
Summary and Outlook

- Carbonyl-ene reaction
- Epoxidation
- Sulfoxidation
- Ring Opening of Epoxide
- Hydrogenation

Edited by Kuiling Ding
and Yasuhiro Uozumi

WILEY-VCH

Handbook of Asymmetric Heterogeneous Catalysis



A Concept Article, see:
K. Ding, et al. *Chem. Eur. J.* **2006**, *12*, 5188-5197.

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